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High pressure synthesis and properties of $Bi_{0.5}Pb_{0.5}CrO_3$: A novel Cr^{4+}/Cr^{3+} perovskite



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ABSTRACT

We have synthesized a new Bi_{0.5}Pb_{0.5}CrO₃ perovskite phase by means of a high pressure reaction at 70 kbar and 1000 °C. The distorted orthorhombic perovskite structure can be indexed in the space group *Pnma* with lattice parameters *a*=5.4768 (1) Å, *b*=7.7450 (2) Å, and *c*=5.4574 (1) Å at room temperature, but undergoes a structural phase transition and enters into a *P*₂₁/*m* monoclinic distorted perovskite phase below 150 K with *a*=5.4173 (2), *b*=7.7286 (4) and *c*=5.4930 (3). The structural transition is coincident with the onset of magnetic interactions. At lower temperatures a weak ferromagnetic structure is evident related to antiferromagnetic Cr-spin canting and a spin-glass transition is observed at \approx 40 K. The semiconducting-type electrical resistivity is relatively low, associated with Cr³⁺/Cr⁴⁺ electron hopping, and shows considerable magneto-resistance (up to 15%). Due to the low resistivity the dielectric permittivity $\varepsilon_{\rm r}$ could be determined only below *T* < 80 K to be \approx 300 and did not show any strong temperature-dependence. Ferroelectricity was not detected in the *T*-range investigated and no magnetocapacitance effects were observed.

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1. Introduction

Among mixed oxide phases, perovskites represent one of the most important and well-studied structures. Perovskites or perovskite related structures can accommodate a vast range of different cations, which can lead to distinct properties such as superconductivity (e.g. in $Ba_{1-x}K_xBiO_3$), colossal magneto-resistance ($La_{1-x}Ca_xMnO_3$), itinerant electron ferromagnetism (SrRuO₃), ferroelectricity (BaTiO₃), piezoelectricity (PbZr_{1-x}Ti_xO₃), ionic conductivity ($La_{0.67-x}Li_{3x}TiO_3$, BaCeO_{3-x}) [1,2], as well as multiferroic behavior (TbMnO₃) which has attracted particularly strong interest in recent years [3,4]. One among the different strategies for the chemical search for new candidate multiferroic materials is the accommodation of different cations on the *A*- and *B*-site of the perovskite structure. The *B*-site cation would possess partially filled *d* or *f* electron levels as a source

* Corresponding author. E-mail address: emoran@quim.ucm.es (E. Morán). for magnetism, whereas the A-site cation would exhibit lone-pair electrons to induce non-centrosymmetry into the crystal structure and possibly ferroelectric polarization. The parent compounds for the title material, BiCrO₃ and PbCrO₃, are both high pressure/high temperature (HP/HT) phases and require 40 kbar and 55 kbar respectively, and temperatures of $800 < ^{\circ}C < 1000$ to be synthesized [5–10].

The HP synthesis of BiCrO₃ at 35–55 kbar has been largely studied in the past. The compound undergoes a structural phase transition at \approx 410–420 K from a high temperature monoclinic to a low temperature non-centrosymmetric triclinic perovskite phase, and the Cr³⁺ cations order antiferromagnetically below \approx 120 K [7]. Across the structural transition the dielectric permittivity exhibits an anomaly, which signals ferroelectricity and a concomitant ferroelectric polarization. Therefore, BiCrO₃ may be regarded a multiferroic system due to the coexistence of ferroelectric and magnetic ordering below \approx 120 K.

 $PbCrO_3$ has been synthesized and characterized for the first time by Roth and De Vries [6]. They reported a cubic perovskite

symmetry (*a*=4.00 Å), and an antiferromagnetic structure with $T_N \approx 240$ K. More recently, a deeper investigation revealed the coexistence of an occupational and displacive modulation of Pb²⁺ cations [9]. Further magnetic measurements validated the AFM behavior and revealed a novel temperature-driven and gradual spin reorientation between 185 and 62 K [9,10], which was believed to be correlated to the occupational and displacive Pb²⁺ modulations. Within the context of the solid solution Bi_{1-x}Pb_xCrO₃ that we have recently investigated [11], we here describe the synthesis and characterization of the double perovskite Bi_{0.5}Pb_{0.5}CrO₃. Our work is motivated by the above-mentioned strategy of combining two different cations on the perovskite *A*-site with both having lone-pair electrons and a magnetic cation in the *B*-site leading to an $A_{0.5}A'_{0.5}BO_3$ perovskite.

2. Experimental section

Bi_{0.5}Pb_{0.5}CrO₃ was prepared under HP/HT conditions using a Belt-type press located in the "Laboratorio Complutense de Altas Presiones". Stoichiometric amounts of α -PbO, Bi₂O₃, Cr₂O₃ and CrO₂ were intimately mixed and ground in an agate mortar and placed in a Pt capsule for HP/HT, the synthesis conditions being explored at temperatures in the 900-1100 °C range, at pressures between 40-80 kbar and reaction times between 3 and 30 min. The resulting powders were examined by powder X-ray diffraction (XRD) using a Panalytical model X'Pert PRO MPD Alfa1 diffractometer with monochromatic CuK_{α} (λ = 1.5405 (6) Å) radiation obtained with a germanium primary monochromator. Data were collected in the range of $2\theta = 5 - 120^{\circ}$, with a step size of 0.017° (2 θ) over a total exposure time period of 12 h. The powder diffraction patterns were indexed by TREOR and DICVOL software [12,13], and fitted to a structural model using the Rietveld method by means of the FullProf-Suite program [14,15].

Low temperature thermodiffraction measurements have been performed in an X'Pert PRO-TT diffractometer using Cu radiation ($K_{\alpha 1}$ =1.5405 (6) Å and $K_{\alpha 2}$ =1.5443 (9) Å). The temperature was varied between 14 and 298 K in a Phenix Oxford Cryosystems low temperature chamber.

Samples for high resolution transmission electron microscopy (HRTEM) and convergent beam electron diffraction (CBED) were prepared by grinding the powder, dispersing it in *n*-butyl alcohol and collecting few drops of the suspension on a holey carbon-coated copper grid. Selected area electron diffraction (SAED) was performed using a JEOL JEM FX2000 microscope, equipped with a double tilt $\pm 45^{\circ}$ holder, working at 200 kV. HRTEM was carried out on a JEOL JEM 3000FEG microscope, with C_s =0.6 nm, operating at 300 kV. Both microscopes were equipped with an analyzer for Energy Dispersive Analysis of X-rays (EDAX) to verify the stoichiometry of Bi_{0.5}Pb_{0.5}CrO₃.

HRTEM images have also been simulated using Mac Tempas software, whereby the focus was varied in the model and optimized between -600 Å and -100 Å in 6 steps of 100 Å, and the sample thickness between 11 and 93.1 Å in 4 steps [16,17].

A SQUID Quantum Design XLMPM was used to perform measurements of magnetic susceptibility (χ) vs temperature (*T*) under ZFC and FC conditions (1 and 5 kOe) between 2 and 300 K, magnetization *M* vs applied field (*H*) at 5 K between *H*=0–5 T, and magnetic AC susceptibility (χ) vs *T* at 33, 330, 660 and 1000 Hz between 20 and 60 K.

A QuadTech impedance analyzer was used for dielectric spectroscopy with an applied ac amplitude of 20 mV operating between 20 Hz and 1 MHz. Data were collected in the format of real and imaginary part of the impedance (Z''-Z'). To facilitate frequency (f), Tand H dependent dielectric measurements a special sample holder was used (Janis Research Ltd., USA), which can fit into a Quantum Design PPMS measurement system providing variable *H* (up to 140 kOe) and variable *T* (10–320 K). The custom built probe was necessary to minimize the internal probe capacitance ($\approx 0.2 \text{ pF}$) and maximize the internal probe resistance ($\approx 10 \text{ G}\Omega$), which is both detrimental for reliable alternating current (*ac*) dielectric measurements. The sample tray at the bottom of the probe was equipped with spring loaded drop-down pins to ensure optimal contact between the pins and sample electrodes. As an electrode material Au was sputter-deposited onto two opposing faces of the Bi_{0.5}Pb_{0.5}CrO₃ bulk samples and both such electrodes were covered with Ag paint.

Equivalent circuit fitting of the dielectric data collected at variable f, T and H was performed by employing commercial Z-View[®] fitting software, where each equivalent circuit model was fitted to both the real and imaginary parts of the f-dependent impedance data by minimizing the statistically weighted linear least-squares.

3. Results and discussion

3.1. Optimization of the synthesis conditions

For the synthesis of $Bi_{0.5}Pb_{0.5}CrO_3$ the reaction temperature, pressure and time needed to be optimized by trial and error. For any of these 3 parameters, too low a value leads to an incomplete reaction of the precursors, whereas too high values lead to decomposition. The optimum conditions were found to be 70 kbar and 1000 °C with a reaction time of 15 min.

3.2. Structural characterization

3.2.1. Room temperature phase

The room temperature Rietveld refinement provided good agreement between model and data (Fig. 1). No indications for a Bi^{3+}/Pb^{2+} *A*-site cation order were found by XRD. This is not surprising though, because Pb^{2+} and Bi^{3+} are isoelectronic ions and X-ray diffraction cannot distinguish between them. The $Bi_{0.5}Pb_{0.5}CrO_3$ phase was indexed as an orthorhombic perovskite structure, because we observe an unusual broadening of the 220 peak in the XRD pattern at around 40° . We interpret this as a weak orthorhombic distortion in $Bi_{0.5}Pb_{0.5}CrO_3$ perovskite, which cannot fully split the 220 reflection. Rietveld refinements using tetragonal settings give significantly higher fitting errors.

Refinements were carried out with space group (S. G.) *Pnma*, with lattice parameters a=5.4768 (1) Å, b=7.7450 (2) Å and c=5.4574 (1) Å. This can be regarded as the diagonal cell $\sqrt{2a_c} \times 2a_c \times \sqrt{2a_c}$ in a pseudo-cubic perovskite setting with $a_c \approx 3.8$ Å [18,19]. The ED micrographs of the room temperature phase shown below suggest *Pnma* orthorhombic symmetry due



Fig. 1. Experimental and calculated XRD patterns and their difference for perovskite $Bi_{0.5}Pb_{0.5}CrO_3$ at 298 K. Tic marks show positions of expected peaks. The bottom curve shows the difference between observed and calculated patterns. Secondary phases are $PbCrO_3$ and δ -Bi₂O₃.

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