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# Study on H-bond patterns in phosphoric triamides having a P(O)NHC(O) skeleton, a *gauche* orientation of P(O) vs C(O) in new compounds

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#### 1. Introduction

Hydrogen bonds (HBs) are the most important directional intermolecular interactions which may affect the preference of a molecular conformation and arrangement [1]. Many efforts have been performed to classify HBs based on their strengths [2,3] and morphology [4-6]. Such considerations help to predict crystal structures based on molecular structures. However, this is not yet completely possible, but it leads to classify HB patterns in analogous compounds and to obtain empirical "rules" which are most useful for predicting hydrogen-bond patterns of systems with a limited number of functional groups [7]. In this area, HB patterns have been discussed for nitroanilines [7], diarylureas [7], amides [8], imides [8], diamides [8] and different co-crystals [8]. Now, we wish to organize the existing data related to a biologically important functional group, P(O)NHC(O), belonging to N-carbonyl phosphoric triamides (CPAs, RC(O)NHP(O)[NR<sup>1</sup>R<sup>2</sup>]<sub>2</sub>) [9]. Some derivatives have been investigated structurally [10-64], the cif files of which have been used for studying and classifying hydrogen bond patterns in CPAs and evaluating the behavior of different parts of the molecule in relation to the hydrogen bonds. This paper attempts to study the collective behavior of HBs in the crystal packing of CPAs. Among the 91 crystal structures found in the Cambridge Structural Database (CSD version 5.31 (November 2009)) and recently published papers (Table 1), there are only 10 structures with a non-anti orientation of C=O versus P=O. Here, we report 4 new structures with a *gauche* orientation and classify

#### ABSTRACT

The crystal and packing structures of new phosphoric triamides, in a rare *gauche* orientation of P(O) *versus* C(O), with the formula (CCl<sub>2</sub>HC(O)NH)X<sub>2</sub>P(O), X = NC<sub>4</sub>H<sub>8</sub> (1), N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub> (2), N(CH<sub>3</sub>)(C<sub>6</sub>H<sub>11</sub>) (3) and (CCl<sub>2</sub>HC(O)NH)(Y)P(O), Y = NHCH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>NH (4) have been investigated. This article also reviews 91 similar structures deposited in the CSD aiming to classify hydrogen bond patterns in this category of phosphorus compounds. The present X-ray structural analysis shows that the H-bond pattern in the studied structures strongly depends on the conformation in the P(O)NHC(O) skeleton and the kind of amide linked to the P atom. The spectroscopic features (<sup>31</sup>P{<sup>1</sup>H}, <sup>1</sup>H and <sup>13</sup>C NMR, IR) of the new compounds have been investigated.

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the H-bond patterns affected by the conformation of the P(O)NH-C(O) skeleton and the kind of amide substituents linked to the P atom.

#### 2. Experimental

### 2.1. Spectroscopic measurements

<sup>1</sup>H, <sup>13</sup>C and <sup>31</sup>P{<sup>1</sup>H} NMR spectra were recorded on a Bruker Avance DRS 500 spectrometer. <sup>1</sup>H and <sup>13</sup>C chemical shifts were determined relative to TMS and <sup>31</sup>P chemical shifts relative to 85% H<sub>3</sub>PO<sub>4</sub> as the external standards. The field strengths for the acquisition of <sup>1</sup>H, <sup>13</sup>C and <sup>31</sup>P spectra were 500.13, 125.77 and 202.46 MHz, respectively. Infrared (IR) spectra were recorded on a Buck 500 scientific spectrometer using KBr discs.

#### 2.2. X-ray measurements

Single colorless crystals of compounds **1–4** were selected and glued on glass fibers. Diffraction data were collected on an Oxford Diffraction KM4 four-circle goniometer equipped with a Sapphire CCD detector. The crystals to detector distance was 45.0 mm and graphite monochromated Mo K $\alpha$  ( $\lambda$  = 0.71073 Å) X-radiation was employed in all measurements. The frame widths of 1° in  $\omega$ , with exp. time between 8 and 60 s were used to acquire each frame for **1**, **2**, **3** and **4**. More than one hemisphere of three-dimensional data was collected in all measurements. The data were reduced using the Oxford Diffraction program CRYSALISPRO [65]. A semi-empirical absorption-correction based upon the intensities of equivalent reflections was applied, and the data were corrected for Lorentz,



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## Table 1

<sup>a</sup>H-bond pattern for RC(O)NHP(O)[X]<sub>2</sub> compounds (X = Cl; XH = primary or secondary amine, alcohol;  $[XH]_2$  = diamine); the point groups of H-bonded dimers are given in parenthesis as C<sub>i</sub> for centrosymmetric and C<sub>1</sub> for non-centrosymmetric.

No	R	X/[X] <sub>2</sub> for diamine	O-P-N-C	Orientation	Packing	Refcode	Ref
1	4-F-C <sub>6</sub> H <sub>4</sub>	, ∧NH	-177.6(3)	anti	1-D ladder	SAYJIM	[38]
2	(Cl)		-62 3(3)	gauche	2-D		[10]
2			-02.5(5)	guuene	2.0		[10]
3	CcHr	× ×	173 85(19)	anti	1-D chain	токхон	[29]
-	-05	NH NH	()				()
4	4-F-C <sub>6</sub> H <sub>4</sub>	b	-179.04(11)	anti	1-D chain	KEBPOX	[39]
5	CNCH <sub>2</sub> CCl <sub>2</sub>		164.06(17) 177.7(3)	anti	I-D ladder 1-D chain	PACNIR	[28]
0	ceij		177.7(3)	unti		Therun	[11]
7	$3-NO_2-C_6H_4$	b	177.6(1)	anti	Hexamer		[50]
	2 0 1		179.0(1)				
		b	-180.0(1)				
8	$C_6H_5$	D	-174.8(2)	anti	Two symmetrically independent 1-D chains	PACNEN	[11]
9	4-F-CcH₄	b	-174.16(11)	anti	1-D chain	KEBPIR	[39]
10	NC <sub>5</sub> H <sub>4</sub>	b	c	anti	c		[64]
11	NC <sub>5</sub> H <sub>4</sub>	b	с	anti	c		[64]
12	$4-NO_2-C_6H_4$	NH	170.69(13)	anti	1-D chain		[46]
13	CCl <sub>3</sub>	NH	-168.55(19)	anti	1-D chain	YIVPAV	[12]
		$\frown$					
14	4-Cl-C <sub>6</sub> H <sub>4</sub>	b	173.92(19)	anti	1-D chain	NOBHUI	[18]
15	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub>		166.55(12)	anti	1-D chain	YIVPEZ	[12]
10	CNCH <sub>2</sub>	NH	-105.65(10)	unn	I-D laddel	LUPCAV	[20]
		0					
		Ч <u> </u>					
17	CHCl <sub>2</sub>	NH NH	61.32(2)	gauche	1-D chain		d
		н.с Сн.					
18	4-F-C <sub>6</sub> H₄	b	-57.2(2)	gauche	2-D		[40]
19	CCl <sub>3</sub>	HN-	-169.9(3)	anti	Two symmetrically independent 1-D chains		[13]
		CH <sup>2</sup>	-161.9(3)				
		- 2	161.1(3)				
20	4-F-CcH	b	168.8(3)	anti	1-D chain	TOKXEX	[29]
20	$C_6H_5$	∧ ∧ .Ft	178.5(4)	anti	1-D chain	GOKMEZ	[30]
		Ét					
22	CCl <sub>3</sub>	ŅН	174.8(6)	anti	<sup>e</sup> Dimer (C <sub>i</sub> )	POKZAQ	[14]
23	CCl <sub>2</sub>	A N	174.9(4)	anti	Dimer (C <sub>1</sub> )		[15]
	;	CH3	175.3(4)		(-1)		()
24	CCl <sub>3</sub>	Ň	162.01(16)	anti	Dimer (C <sub>i</sub> )	RAMBEN	[16]
25	CUCI	UH <sub>3</sub>	170 10(17)		Dimor (C)		[27]
25 26	CHCl <sub>2</sub>	b	-1/8.16(1/) -165.2(2)	anti	Dimer $(C_i)$	RAIZUA	[27]
20	4-F-C <sub>6</sub> H <sub>4</sub>	b	172.10(11)	anti	Dimer (C <sub>i</sub> )	KEBPUD	[39]
28	$4-Cl-C_6H_4$	b	-171.90(14)	anti	Dimer (C <sub>i</sub> )		[27]
29	$4-Br-C_6H_4$	b	-171.6(3)	anti	Dimer (C <sub>i</sub> )	LIVWAP	[53]
30	CF <sub>3</sub>	b	-176.2(2)	anti	Dimer (C <sub>i</sub> )	WALRAD	[22]
31 32	∠-r-∪ <sub>6</sub> п <sub>4</sub> (Сl₂		154.02(9) 171.7(3)	anti	Dimer $(C_i)$ Two symmetrically independent dimers $(C_i)$	XIQZAZ VEHTAF	[42] [17]
32		) N	-151.1(4)		····· symmetrically independent amers (C1)	* LITTAL	(**)
33	CHCl <sub>2</sub>	b	56 2(3)	gauche	1-D chain		d
			33.2(3)	Sumeric			

(continued on next page)

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