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Thermolysis of $Ru_3(CO)_{10}(\mu$ -dppm) and $Ru_3(CO)_9(PRPh_2)(\mu$ -dppm) (R = Ph, H) in presence of O₂: Synthesis and structure of triruthenium clusters containing a capping-oxo ligand



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ABSTRACT

Thermolysis of Ru₃(CO)₁₀(μ -dppm) [dppm = bis(diphenylphosphino)methane] in boiling xylene in presence of molecular oxygen (O₂) affords the trinuclear oxo-capped cluster Ru₃(CO)₇(μ_3 -CO)(μ_3 -O)(μ -dppm) (**1**). A similar reaction between Ru₃(CO)₉(PPh₃)(μ -dppm) and O₂ in refluxing benzene also furnishes the oxo-capped Ru₃(CO)₆(PPh₃)(μ_3 -CO)(μ_3 -O)(μ -dppm) (**2**). Crystal structures of **1** and **2** reveal that both contain a capping oxygen at one face and a triply-bridging carbonyl on the opposite face of the triruthenium plane. In contrast, Ru₃(CO)₉(PHPh₂)(μ -dppm) does not react with O₂ under similar conditions, but undergoes thermal transformations to give Ru₃(CO)₇(μ -H)(μ -PPh₂){ μ_3 -PhPCH₂P(C₆H₄)Ph} (**3**) and Ru₃(CO)₆(μ -CO)(μ -PPh₂)₂(μ_3 -CH₂PPh)] (**4**) via C–P, P–H and C–H bond activation. All the new clusters have been characterized by a combination of spectroscopic data and single crystal X-ray diffraction analysis.

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1. Introduction

It has been known for some years that zero-valent triruthenium clusters react with elemental sulfur and its heavier congeners to afford chalcogenide-capped clusters [1-4]. In 1979, Johnson and co-workers synthesized Ru₃(CO)₉(μ ₃-X)₂ and Ru₃(CO)₉(μ -H)₂(μ ₃-X) from the reaction of Ru₃(CO)₁₂ with X_n (X = S, Se, Te) in presence of CO [3]. In contrast, Ru₃(CO)₁₂ does not react with molecular oxygen (O₂) to afford oxo-capped cluster since the triruthenium core is not electron-rich enough to undergo oxidation. One way to increase the electron-richness of the trinuclear core is the substitution of carbonyl(s) by more electron-donating phosphines and the phosphine-substituted derivatives of Ru₃(CO)₁₂ such as Ru₃(CO)₉{P(C₄H₃S)₃}(μ -dppm), Ru₃(CO)₈(μ -dppm)₂ and Ru₃(CO)₆(μ -dppm)₃ have been found to react with O₂ which lead to the formation of oxo-capped clusters (Chart 1) [4–6].

A literature review in this field shows that at least three carbonyls have to be replaced by phosphines from $Ru_3(CO)_{12}$ in order to make it reactive towards O_2 that leads to the formation of oxocapped cluster [4–6]. In the present work, we have investigated this area further and found that bis(phosphine) substituted cluster $Ru_3(CO)_{10}(\mu$ -dppm) also reacts with O_2 to afford oxo-capped cluster, but only at more drastic conditions. The possibility of side reactions when the incorporated phosphine (e.g., diphenylphosphine) is vulnerable to thermal transformation on cluster surface at moderate temperatures have also been examined.

2. Results and discussion

2.1. Reactions of $Ru_3(CO)_{10}(\mu$ -dppm) and $Ru_3(CO)_9(PPh_3)(\mu$ -dppm) with O_2 – Synthesis of oxo-capped clusters $Ru_3(CO)_7(\mu_3$ -CO)(μ_3 -O) (μ -dppm) (1) and $Ru_3(CO)_6(PPh_3)(\mu_3$ -CO)(μ_3 -O)(μ -dppm) (2)

High temperature reaction between $\text{Ru}_3(\text{CO})_{10}(\mu\text{-dppm})$ and O_2 in boiling xylene led to the formation of oxo-capped cluster $\text{Ru}_3(\text{CO})_7(\mu_3\text{-CO})(\mu_3\text{-O})(\mu\text{-dppm})$ (**1**) in 13% yield (Scheme 1). In contrast, $\text{Ru}_3(\text{CO})_9(\text{PPh}_3)(\mu\text{-dppm})$ afforded the oxo-capped cluster $\text{Ru}_3(\text{CO})_6(\text{PPh}_3)(\mu_3\text{-CO})(\mu\text{-dppm})$ (**2**) in 18% yield from a similar reaction with O_2 in boiling benzene (Scheme 1). In these reactions face-capping with the oxo-ligand would lead to some level of oxidation of the ruthenium atoms in the resultant clusters. Since the triruthenium core in $\text{Ru}_3(\text{CO})_9(\text{PPh}_3)(\mu\text{-dppm})$ is more basic than that in $\text{Ru}_3(\text{CO})_{10}(\mu\text{-dppm})$, oxo-capping of the former is expected to be easier than the later. This is evident by the relatively low temperature required in the reaction of O_2 with



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Chart 1. Examples of oxo-capped triruthenium carbonyl cluster.



Scheme 1. Thermolysis of Ru₃(CO)₁₀(µ-dppm) and Ru₃(CO)₉(PPh₃)(µ-dppm) in presence of O₂.

 $Ru_3(CO)_9(PPh_3)(\mu$ -dppm) compared to that with $Ru_3(CO)_{10}$ (μ -dppm). Both the new oxo-capped clusters have been adequately characterized in solution by IR, ¹H NMR and ³¹P{¹H} NMR spectroscopic data and in the solid-state by single crystal X-ray diffraction analysis.

The solid-state molecular structures of **1** and **2** are depicted in Figs. 1 and 2, respectively, with the captions containing selected bond distances and angles. Both clusters contain a triruthenium core capped with a triply bridging oxo-ligand on one face and a triply bridging carbonyl ligand on the opposite face. They are structurally very similar to the related clusters $Ru_3(CO)_6$ {P(C₄H₃S)₃(μ_3 -CO)(μ_3 -O)(μ -dppm) (**A**) [4] and $Ru_3(CO)_5(\mu_3$ -CO)(μ_3 -O)(μ -dppm) and $Ru_3(CO)_8(\mu$ -dpam)₂ with molecular oxygen respectively. Ruthenium–ruthenium distances in **1**



Fig. 1. The solid-state molecular structure of $Ru_3(CO)_7(\mu_3-CO)(\mu_3-O)(\mu-dppm)$ (1) showing 50% probability thermal ellipsoid. Hydrogen atoms were omitted for clarity. Selected bond lengths (Å) and angles (°): Ru(1)-Ru(2) 2.7302(7), Ru(1)-Ru(3) 2.7238(7), Ru(2)-Ru(3) 2.7084(7), Ru(1)-P(1) 2.3358(16), Ru(2)-P(2) 2.3310(17), Ru(1)-O(10) 2.059(4), Ru(2)-O(10) 2.055(4), Ru(3)-O(10) 2.047(4), Ru(1)-C(6) 2.158(6), Ru(2)-C(6) 2.131(6), Ru(3)-C(6) 2.228(7), Ru(3)-Ru(1)-Ru(2) 59.548(19), Ru(3)-Ru(2)-Ru(1) 60.107(18), Ru(2)-Ru(3)-Ru(1) 60.345(18), Ru(3)-O(10)-Ru(1) 63.17(16), Ru(2)-C(6)-Ru(1) 79.1(2), Ru(2)-C(6)-Ru(3) 76.8(2), Ru(1)-C(6)-Ru(3) 76.8(2), P(2)-C(1)-P(1) 109.2(3).



Fig. 2. The solid-state molecular structure of $Ru_3(CO)_6(PPh_3)(\mu_3-CO)(\mu_3-O)(\mu_4ppm)$ (**2**) showing 50% probability thermal ellipsoid. Hydrogen atoms were omitted for clarity. Selected bond lengths (Å) and angles (°): Ru(1)-Ru(2) 2.7218(5), Ru(1)-Ru(3) 2.7501(5), Ru(2)-Ru(3) 2.7415(5), Ru(1)-P(1) 2.3229(10), Ru(2)-P(3) 2.3595(10), Ru(3)-P(2) 2.3607(10), Ru(1)-O(2) 2.037(2), Ru(2)-O(2) 2.049(2), Ru(3)-O(2) 2.066(2), Ru(3)-C(1) 2.132(4) Ru(1)-C(1) 2.170(4), Ru(2)-C(1) 2.178(4); Ru(2)-Ru(1)-Ru(3) 60.132(13), Ru(1)-Ru(2)-Ru(3) 60.447(12), Ru(2)-Ru(3) -Ru(1) 59.421(13), Ru(1)-O(2)-Ru(2) 83.53(8), Ru(1)-O(2)-Ru(3) 84.18(9), Ru(2)-O(2)-Ru(3) 83.55(9), Ru(3)-C(1)-Ru(1) 79.46(13), Ru(3)-C(1)-Ru(2) 78.98(13), Ru(1)-C(1)-Ru(2) 77.50(12).

[Ru-Ru 2.7084(7)-2.7302(7)Å] and 2 [Ru-Ru 2.7218(5)-2.7501(5)Å] are comparable to those found in related oxo-capped clusters **A** [Ru-Ru 2.7069(5)-2.7462(5)Å] [4] and **B** [Ru-Ru 2.670(2)–2.750(2) Å] [5], but are significantly shorter than those observed in the parent clusters Ru₃(CO)₈(µ-dppm) [Ru-Ru 2.834(1)–2.860(1)Å] [7] and $Ru_3(CO)_9(PPh_3)(\mu-dppm)$ [Ru–Ru 2.8523(3)–2.8938(3) Å] [8] which indicate some level of oxidation of the ruthenium atoms in both 1 and 2. The ruthenium-oxygen bond distances in both of them [Ru-O 2.047(4)-2.059(4) Å (in 1) and 2.037(2)-2.066(2) Å (in 2)] are also comparable to those found in related A and B [Ru-O 2.051(3)-2.057(3) Å (in A) and 2.06(1)-2.11(1) Å (in 2)] [4,5]. In addition, all three Ru–O bond lengths are almost similar within the experimental error in both 1 and 2. This suggests that they may have delocalized electronic states with an averaged ruthenium oxidation number. In order to check this assumption, we compare the ruthenium-oxygen bond distances

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