



Bioisostere of valtrate, anti-HIV principle by inhibition for nuclear export of Rev

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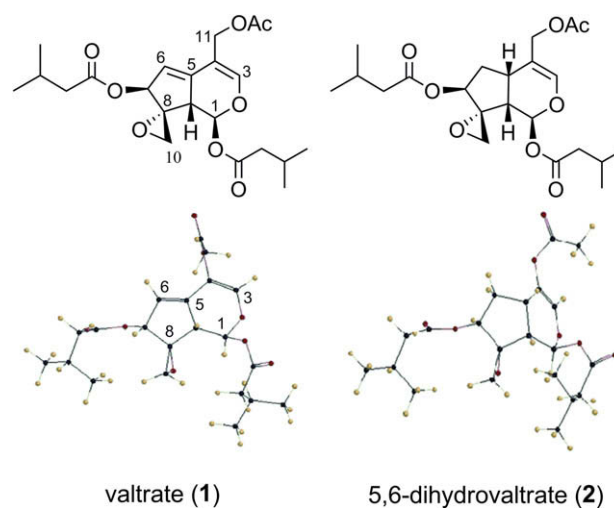
ABSTRACT

Rational design by the MO calculation disclosed 5,6-dihydrovaltrate (**2**) as the bioisostere of valtrate (**1**), the Rev-export inhibitor with anti-HIV activity. The synthesis of **2** was accomplished by ingenious use of asymmetric Diels–Alder reaction and stereoselective epoxidation associated with the adjacent hydroxyl group. Because of similar biological potency to **1**, the analog **2** should be recognized as a promising scaffold for new anti-HIV agents with an unprecedented mechanism of action, inhibition for nuclear export of Rev protein, in the conventional remedy.

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The acquired immunodeficiency syndrome (AIDS) is a life-threatening disease caused by HIV-1 and the HIV pandemic remains one of the most serious threats to worldwide public health.¹ Rev protein of HIV-1 was clarified to play an essential role in viral replication by constructing the structural proteins after export of the viral mRNA from the nucleus to the cytoplasm with the aid of the cargo protein CRM1.^{2,3} Inhibition for nuclear export of Rev was, therefore, recognized as one of the attractive targets for new anti-HIV drugs with an unprecedented mechanism of action. In practice, leptomycin B, the first nuclear export inhibitor of Rev, was reported to show anti-HIV activity.^{4,5} On the other hand, we found out valtrate (**1**) as the inhibitor for nuclear export of Rev from the medicinal plant *Valeriana Radix* (root of *Valeriana fauriei*) and revealed **1** to inhibit proliferation of HIV-1. Furthermore, valtrate (**1**) was shown to be linked to the Cys-529 residue in CRM1 by the covalent bond between the epoxy moiety in **1** and the thiol group in Cys-529, this indicating that the epoxy portion is the conclusive function to exert the bioactivity of **1** (Fig. 1).⁶ Despite the potential biological activity as well as the elucidated mechanism of action, utilization of **1** as a scaffold toward anti-HIV agents seemed infeasible because of both scarce supply from natural resource ($6.3 \times 10^{-3}\%$ from crude drug) and strict limita-

tion of derivatization of **1**; removal of the isovaleryl group at C-1 faceably brought about decomposition by way of dial with involving elimination of the acetoxy group. In this context, search for the bioisosteres of **1** was intensively considered to be important for development of new anti-HIV agents. This manuscript deals with the



valtrate (**1**)

5,6-dihydrovaltrate (**2**)

Figure 1. Stable conformations of valtrate (**1**) and 5,6-dihydrovaltrate (**2**) by MO calculation.

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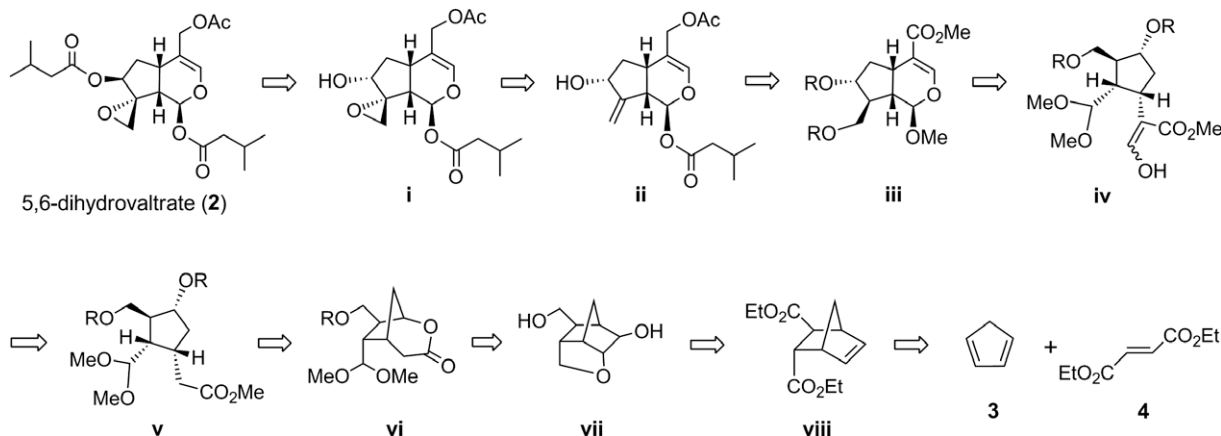


Figure 2. Retrosynthetic analysis of 5,6-dihydrovaltrate (**2**).

synthesis and biological property of 5,6-dihydrovaltrate (**2**), the bioisostere of valtrate (**1**), which will be anticipated to possess fairly similar steric environment around the epoxy pharmacophore to **1** through rational design by the molecular orbital calculation.

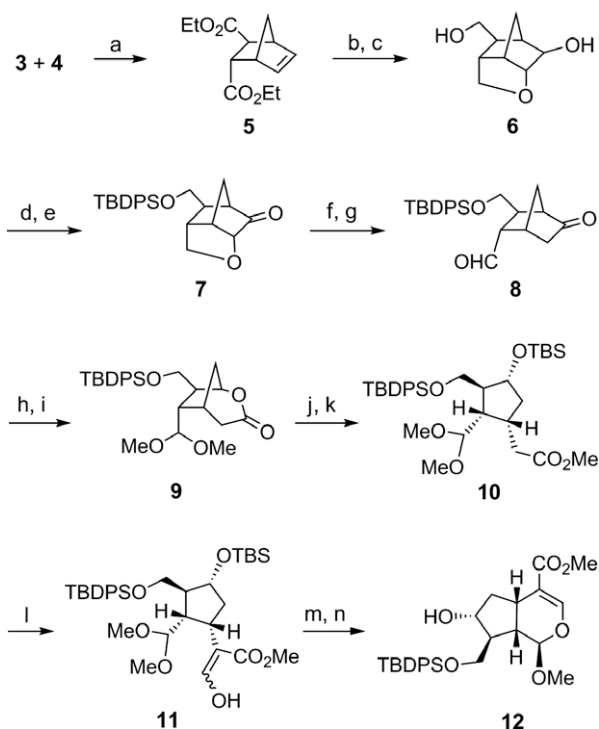
In the first instance, the most stable conformations of **1** and **2** were calculated by a semi-empirical molecular orbital method PM3. As a result of the MO calculation, no serious difference in steric environment around the C-8 epoxy moiety appeared between **1** and **2**, which intensively presented that 5,6-dihydroanalogue (**2**) exhibited nearly similar inhibitory activity for nuclear export of Rev as compared with valtrate (**1**) (Fig. 1).

Since 5,6-dihydroanalogue (**2**) was presumed to be synthesized from the highly optical pure cyclopentane **v** by way of the bicyclic

methyl acetal **iii** bearing the fully oxygen-functionalized iridoid skeleton, we conducted retrosynthetic analysis of **2** as illustrated in Figure 2. Namely, the labile 1,1-disubstituted epoxy moiety would be constructed by taking advantage of the α -oriented hydroxyl group at C-7 in the late stage. The isovaleryloxy group would be introduced by Mitsunobu inversion for the hydroxyl group at C-7 in the final stage. *Exo*-olefin **ii** as the precursor of epoxide **i** would be afforded from methyl acetal **iii**. Moreover, the acetal **iii** was planned to be yielded by intramolecular transacetalization of enol **iv**, which would be obtained by condensation between the optical active cyclopentane **v** and ethyl formate. The cyclopentane **v** would be prepared by methanolysis of lactone **vi**, which would be available from cyclic ether **vii** by oxidation followed by Baeyer–Villiger rearrangement. The cyclic ether **vii** was expected to be attained by stereoselective epoxidation and intramolecular etherization from the asymmetric Diels–Alder adduct **viii** between cyclopentadiene (**3**) and diethyl fumarate (**4**) in high optical purity.

The synthesis of the methyl acetal **12** with the optical active iridoid skeleton was executed as shown in Scheme 1. Asymmetric Diels–Alder reaction using Corey's chiral ligand⁷ between cyclopentadiene (**3**) and diethyl fumarate (**4**) afforded the adduct **5** quantitatively with 95% ee. After reduction of the two ester functions in **5**, treatment of the resulting diol with mCPBA promoted not only epoxidation but also intramolecular etherization to give tricyclic ether **6**. The primary hydroxyl group in **6** was selectively protected as TBDPS ether and subsequent Dess–Martin oxidation⁸ provided cyclic ketone **7**. Reductive cleavage of the cyclic ether in **7** with Al–Hg followed by oxidation of the resultant primary hydroxyl function afforded aldehyde **8**. The formyl group in **8** was subjected to selective acetalization by treatment with methoxytrimethylsilane (MeOTMS) in the presence of TMSOTf, then Baeyer–Villiger rearrangement by mCPBA gave seven-membered lactone **9**. Successive methanolysis of **9** with NaOMe and protection of the resulting hydroxyl group with the TBS residue provided cyclopentane **10**, which was submitted to installation of the formyl group mediated with ethyl formate and lithium diisopropyl amide to provide enol **11**. BF₃ catalyzed transacetalization of **11** smoothly proceeded concomitant with deprotection of the TBS function. In the last step, the partly obtained hemiacetal was entirely converted to methyl acetal **12** possessing the desired iridoid skeleton by treatment with CH(OMe)₃ in the presence of *p*-TsOH.

Subsequently, the synthesis of 5,6-dihydroanalogue (**2**) was accomplished from **12** by ingenious use of stereoselective epoxidation associated by the adjacent hydroxyl group as displayed in Scheme 2. The secondary hydroxyl group in **12** was protected as MOM ether, then the TBDPS group was removed by treatment of



Scheme 1. Reagents and conditions: (a) ligand, CH₂Cl₂, –35 °C, quant.; (b) LiAlH₄, THF, 0 °C, quant.; (c) mCPBA, CH₂Cl₂, 90%; (d) TBDPSCI, DBU, CH₂Cl₂, 95%; (e) Dess–Martin periodinane, CH₂Cl₂, 88%; (f) Al–Hg, THF, EtOH, 92%; (g) Dess–Martin periodinane, CH₂Cl₂, quant.; (h) MeOTMS, TMSOTf, CH₂Cl₂, 80%; (i) mCPBA, KH₂PO₄, CH₂Cl₂, 81%; (j) NaOMe, MeOH, 90%; (k) TBSOTf, 2,6-lutidine, CH₂Cl₂, 80%; (l) LDA, HCOEt, THF, –78 °C; (m) BF₃·OEt₂, CH₂Cl₂, 0 °C; (n) *p*-TsOH, CH(OMe)₃, MeOH, three steps 54%.

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