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Zwitterionic uracil derivatives as potent GnRH receptor antagonists with improved pharmaceutical properties

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The potential of non-peptide Gonadotropin-releasing hormone (GnRH) receptor antagonists serving as novel therapeutics for hormone-dependent disease states such as prostate cancer, endometriosis, and benign prostate hyperplasia has led to discovery of a wide range of small molecule antagonists.^{1,2} We have reported that uracil-based analogs are potent GnRH receptor antagonists based on in vitro and in vivo characterizations.³ However, CYP3A4 inhibition was a common issue for early uracil analogs that contained a basic amine such as 1 in Figure 1. Since inhibition of CYP3A4 enzyme is well known to potentially induce drug-drug interactions, elimination of such undesired property from this class of molecules was clearly necessary. Recently, we have shown⁴ that addition of an acid can drastically reduce the possibility of this class of molecules to inhibit CYP3A4 enzyme activity regardless of the location of the acid. However, the GnRH receptor-binding affinity was heavily dependent on the exact location of this acidic functional group. For example, the acid linked to the phenyl group at the right hand side of the molecule (2b) diminishes the GnRH receptor-binding affinity compared to its ester precursor (2a), yet the acid attached to the amine group through a propylene chain (**3b**) is highly potent GnRH receptor binder. As a matter of fact, such combination of the

ABSTRACT

A novel series of potent zwitterionic uracil GnRH antagonists were discovered that showed reduced liability for CYP3A4 enzyme inhibition.

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amino and acid functionalities offered similar potency to its amine precursor **1**, yet without the CYP3A4 liability. Interestingly, such zwitterionic molecules show good oral bioavailability in cynomolgus monkeys, albeit relatively poor exposure in rats.⁴ To further expand the SAR on zwitterionic uracils, we report here a novel way of linking an acid group to the 5-phenyl uracils, which generated a series of novel and potent zwitterionic molecules without inhibition of CYP3A4 enzyme.

The initial syntheses of such molecules are outlined in Scheme 1. The starting compounds $(4a-d)^{3,4}$ were first treated with BBr₃ to remove the methyl group; subsequently, the amino group was protected using Boc₂O to give compounds **5a-d**. Alkylation with $Br(CH_2)_n CO_2 Et$, followed by hydrolysis of the ethyl ester and removal of the Boc protecting group, yielded the desired zwitterionic compounds 6a, 6b, 7a-c, 8a-c, 8e, and 9a. Compound 8d was prepared alternatively according to Scheme 2 where 5c was first alkylated with 3-bromopropanol, followed by the oxidation of the hydroxyl group to the corresponding carboxylic acid functionality and then removal of the Boc-protecting group. These compounds were assayed against the human GnRH receptor binding, IP₃ function, and CYP3A4 inhibition.⁶ The results are summarized in Table 1. Our previous SAR has indicated that polar group cannot be tolerant around the 3-methoxylphenyl region at 5-uracil, thus our campaign to introduce the acid functionality on 3-methoxyphenyl

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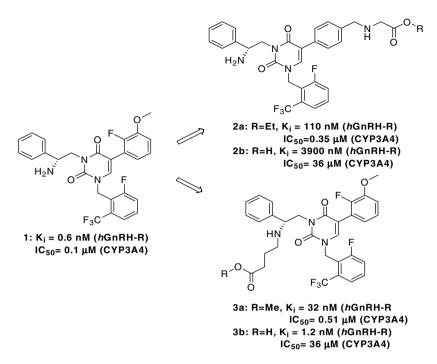
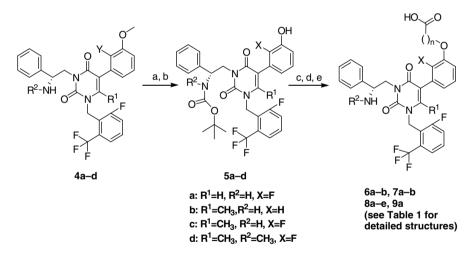
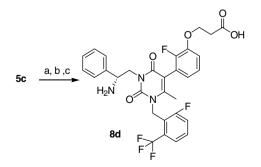


Figure 1. Structures and biological activities of previously reported compounds 1-3.



Scheme 1. Reagents and condition: (a) BBr₃, DCM, -78 °C; (b) Boc₂O, DCM, Et₃N; (c) K₂CO₃, DMF, Br(CH₂)_nCO₂Et; (d) LiOH, THF/water; (e) TFA/DCM.



Scheme 2. Reagents and condition: (a) K_2CO_3 , DMF, BrCH₂CH₂CH₂OH, 50 °C; (b) cat. RuCl₃, NaIO₄, DCM, MeCN, H₂O; (c) TFA/DCM.

position of compound **1** initialized with a long alkyl-acid such as pentanoic acid (**6a**), which yielded moderate but encouraging

activity (K_i = 34 nM); the longer acid **6b** did not improve the activity. Both compounds, as we expected, did not exhibit significant CYP3A4 inhibition. To search for an improvement on GnRH activity, we turned our attention to modify on more potent analogs (4b-d). Indeed, compound 7a, 6-methyluracil analog based on **4b**, was much more potent ($K_i = 2.1 \text{ nM}$) than that of the nonmethyl analog 6a. However, shortening the chain length (7b and **7c**) decreased the potency slightly. Historically, addition of a fluoro group to the 3-methoxyphenyl ring of **4b** further enhances the GnRH activity. Therefore, compounds 8a-e were prepared accordingly. Enhancement of the potency by fluoro group was not clearly observed in the binding assay, but was well displayed in the functional assay which measures the ability of a compound to inhibit GnRH-stimulated [³H] inositol phosphate hydrolysis. Overall, fluoro analogs were about 5-10 times more potent than the corresponding non-fluoro analogs (such as 8a and 8b vs 7a and 7b). Because of their low possibility of CYP3A4 inhibition and potent GnRH antagonistic activity, pharmacokinetic studies of several

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