

# Sodium alginate-assisted exfoliation of MoS<sub>2</sub> and its reinforcement in polymer nanocomposites



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## ABSTRACT

In this work, molybdenum disulfide (MoS<sub>2</sub>) nanosheets were facilely prepared by direct exfoliation of MoS<sub>2</sub> in aqueous media with the assistance of sodium alginate (SA). Transmission electron microscopy (TEM), X-ray diffraction (XRD) and Raman spectra results showed that the raw MoS<sub>2</sub> was successfully exfoliated into few-layer MoS<sub>2</sub> nanosheets (SA-MoS<sub>2</sub>). FTIR and thermal gravimetric analysis (TGA) investigations showed that the obtained MoS<sub>2</sub> nanosheets were modified by SA after exfoliation and improved dispersion in water were achieved. The obtained SA-MoS<sub>2</sub> nanosheets were employed to reinforce the water-soluble polymer SA. No obvious macroscopic phase separation could be found from the SA/SA-MoS<sub>2</sub> films. Dynamic mechanical analysis (DMA) results showed that almost 9 times enhancement for the storage modulus of SA was achieved with the incorporation of only 0.5 wt% of SA-MoS<sub>2</sub>, and the thermal stability of SA was also found improved with the addition of SA-MoS<sub>2</sub> according to the thermal gravimetric analysis TGA results.

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## 1. Introduction

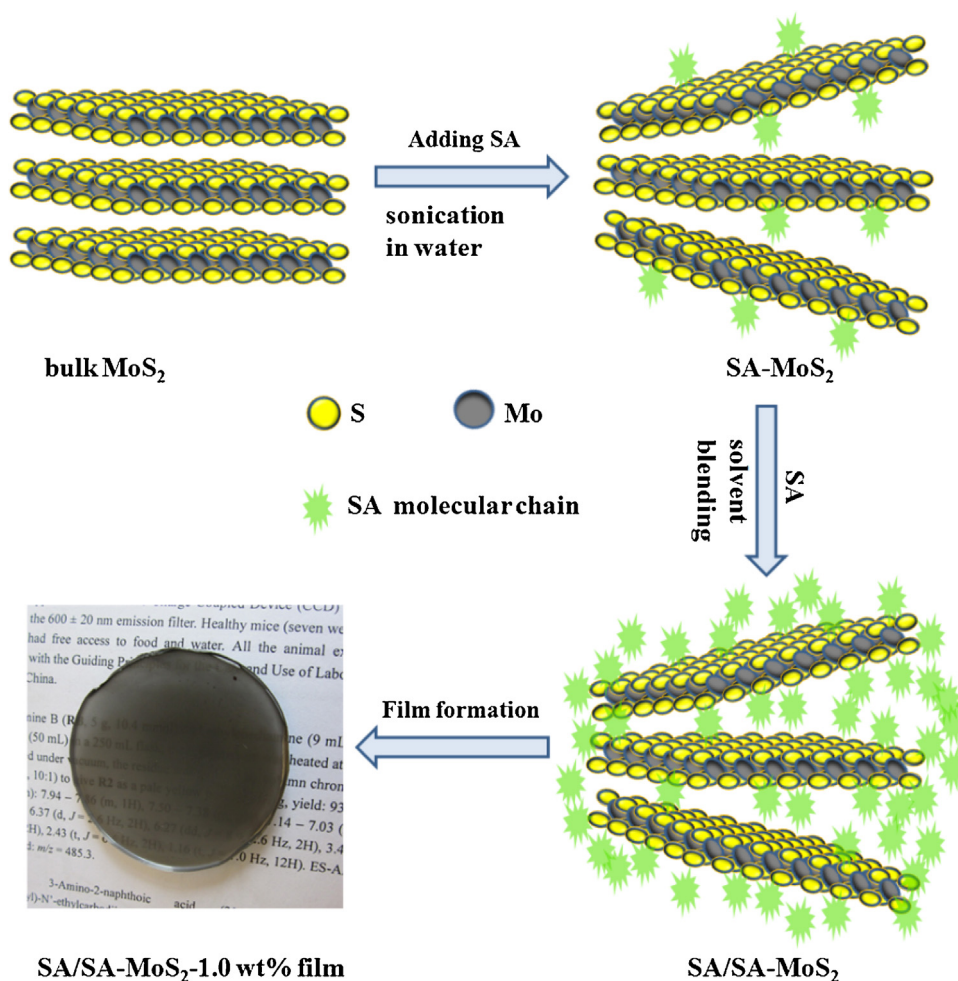
Due to the excellent anisotropic physical properties, ultra-thin structure, and quantum size effects (Guo, Duan, Zhou, & Zhu, 2014; Kisku & Swain, 2012; Layek & Nandi, 2013; Young, Kinloch, Gong, & Novoselov, 2012), graphene has been drawing great attentions to serve as nanofillers in polymer composites. However, the electrical conductivity of graphene based polymer nanocomposites restricts their use in some electrical and electronic applications such as transmission lines, power electronics and capacitors. Similar to graphene, MoS<sub>2</sub> has a layered structure that hexagonal Mo layers are sandwiched between two hexagonal S layers (Huang, Chen et al., 2013; Zhang, Wu, Guo, & Lou, 2012). The MoS<sub>2</sub> nanosheets have interesting optical and electrical properties (Coleman et al., 2011) which have been capitalized in many research fields such as photochemistry, materials science (Zhou, Liu, Wen, Hu, & Gui, 2014), nanoelectronics (Sanchez et al., 2006) and catalysis (Radisavljevic, Radenovic, Brivio, Giacometti, & Kis, 2011; Shi et al., 2012; Yang et al., 2013; Zhiyuan et al., 2011). Same to graphene, MoS<sub>2</sub> nanosheets also possess excellent mechanical properties and high storage modulus. But different

from graphene, the MoS<sub>2</sub> nanosheets are nonconductive which is quite beneficial as a reinforcer for polymer nanocomposites in some special fields. Recently, a general trend has emerged for using MoS<sub>2</sub> modified polymer, such as MoS<sub>2</sub>/PVA nanocomposites (Jiang et al., 2014) and MoS<sub>2</sub>/chitosan nanocomposites (Feng et al., 2014). The polymer nanocomposites prepared with MoS<sub>2</sub> showed enhanced mechanical properties in varying degrees compared to the neat polymers, suggesting the promising potential as a nanofiller of MoS<sub>2</sub> nanosheets. Importantly, those MoS<sub>2</sub> reinforced polymers preserved their electrical insulation well.

MoS<sub>2</sub> nanosheets can be prepared through several different ways such as mechanical exfoliation (Splendiani et al., 2010), chemical vapor deposition (CVD) on substrates (Jeon et al., 2015), chemical exfoliation through intercalation (Chou et al., 2013; Luo et al., 2014), etc. Mechanical exfoliation is an effective method to prepare few-layers MoS<sub>2</sub> nanosheets. Novoselov et al. used the method to prepare nanosheets from various bulk materials to single layer (Novoselov et al., 2005). However, the yield is low and the method is time consuming. CVD has been used to synthesize high quality MoS<sub>2</sub> nanosheets in a controllable manner with different compositions. Lee et al. used MoO<sub>3</sub> and S as the precursors to prepare MoS<sub>2</sub> nanosheets under the condition of heating (Lee et al., 2012). But the method is not easy to scale-up. Chemical exfoliation is a simple method to prepare few-layers MoS<sub>2</sub> nanosheets. The mostly used chemical exfoliation method involves lithium inter-

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**Scheme 1.** Illustration of the preparation process of the SA/SA-MoS<sub>2</sub> nanocomposites.

calation using *n*-butyllithium (*n*-BuLi) as a lithiation agent, which was invented by Joensen (Ashori, 2014; Gee, Frindt, Joensen, & Morrison, 1986), etc. One disadvantage about this method is the lithium intercalation rate is low and often needs more than 40 h and a high temperature to prepare MoS<sub>2</sub> nanosheets (Ashori, 2014; Tang & Alavi, 2011).

Recently, the liquid-phase exfoliation of MoS<sub>2</sub> assisted with green polymer attracted many attentions. Because it not only can produce the well-stripped nanosheets, but also graft some hydrophilic polymer chains at the same time, which can facilitate introducing into many water soluble polymers and achieved homogeneously dispersion in them (O'Neill, Khan, & Coleman, 2012). For example, by taking alkali lignin as a surfactant, Lu et al. directly exfoliated MoS<sub>2</sub> into single-layer and few-layer nanosheets, which possessed abundant hydrophilic groups on its surface (Liu, Zhao et al., 2015; Liu, Zhou et al., 2015). SA is a natural polysaccharide derived from brown sea algae, which is a linear polyanionic copolymer composed of (1–4)-linked-D-mannuronic acid (M) and guluronic acid (G) residues (Pawar & Edgar, 2012; Yang, Xie, & He, 2011). As a hydrophilic biopolymer with –COOH and –OH in each unit, SA is favorable for its intercalation into MoS<sub>2</sub> layers. Besides, SA was demonstrated as a dispersing agent and stabilizer for graphene oxide in aqueous solution (Nie et al., 2015). Furthermore, SA possesses many advantages including sustainability, chemical flexibility, human and eco-friendliness. However, as a polymer material, neat SA displays some unsatisfactory properties such as weak mechanical strength, low thermal stability and

loss of structural integrity, which limits its applications in many fields (Rani, Mishra, & Sen, 2013). Compositing with nanofillers was proved to be an efficient way to enhance SA (Wan, Chen, Xiong, Guo, & Luo, 2014).

In this article, hydrophilic polymer SA was employed to assisting exfoliates MoS<sub>2</sub> in aqueous media. Morphology and structure characterizations showed that the MoS<sub>2</sub> was successfully exfoliated and functionalized at the same time. Taking SA as a model water soluble polymer, the prepared SA-MoS<sub>2</sub> was employed to reinforce SA (Scheme 1). DMA results showed that the storage modulus of SA/SA-MoS<sub>2</sub> was almost 9 times higher compared to neat SA. The thermal stability of SA was found improved as well.

## 2. Experimental section

### 2.1. Materials

MoS<sub>2</sub> (99% purity) was purchased from Sigma-Aldrich Chemicals Inc (USA). Sodium alginate (SA) was purchased from Adamas-beta (China) and used without further purification.

### 2.2. Exfoliation of MoS<sub>2</sub>

The preparation of the MoS<sub>2</sub> dispersion was described as follows: 1 g SA was dissolved in 100 mL deionized water at 50 °C in glass vial with stirring to prepare SA aqueous solution for 2 h. Then, 0.1 g MoS<sub>2</sub> powder was put into the solution upon ultra-

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