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DFT B3-LYP/3-21G geometry optimisation and effective charge values calculations for azodiazaphenanthrenes and acylaminodiazaphenanthrenes

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Abstract

For four azodiazaphenanthrenes 1–4 and three acylaminodiazaphenanthrenes 5–7 the geometry was optimised and their effective charge and dipole moment values were calculated using DFT B3-LYP/3-21G method. For 5–7 the results have been compared with those obtained by AM1 method. The UV experimental values of 1–4 are presented. With the use of DFT B3-LYP/6-31G** optimised geometry the simulation of UV spectra of 5–7 by AM1 and ZINDO/S methods was made and correlations with experimental UV values have been performed.

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Keywords: Calculations; Effective charge; Geometry optimisation; Wavenumber

1. Introduction

In a continuation of our study concerning calculations and spectral properties of diazaphenanthrene derivatives, such as amino- [1], formyl- [2] and methyldiazaphenanthrenes [3], as well as their quaternary salts with haloalkanes [4], the present paper deals with four azodiazaphenanthrenes 1–4 and three acylaminodiazaphenanthrenes 5–7 derived from parent 1,5- and 4,6-diazaphenanthrenes (DAP) 8 and 9.

DAPs and their derivatives are a topic of our research. DAPs are interesting for their reactivity [5,6]; due to the presence of nitrogen atoms they may be oxidized and quaternized. DAP *N*-oxides and quaternary salts undergo a variety of reactions, for example *N*-oxides form ylides [7]; quaternary salts afford tetracyclic products [8] or alkylbenzonaphthyridones [9]. Various quaternary salts of DAPs have been obtained [10–12], some of them are

* Corresponding author. *E-mail address:* w.sliwa@ajd.czest.pl (W. Śliwa). precursors of 1,3-dipoles in cycloaddition reactions [13]. AminoDAPs upon diazotization and coupling reactions afford azoDAPs [14], they also in the Skraup procedure yield pyridoDAPs [15], and with aldehydes the tetracyclic products [16,17]. DAPs form complexes with metal ions [18] and are interesting for their antibacterial [19] and enzyme enhancing [20] activities.

In the present work the geometry of 1–7 has been optimised and their effective charge values and dipole moments calculated using DFT B3-LYP/3-21G method.

For 5–7 the geometry, effective charge values and dipole moments calculated by DFT B3-LYP/3-21G method (next referred to as DFT) have been compared with those calculated by AM1 method [1]. The UV spectral data for 1–4 are presented and differences of their experimental wavenumber values as compared with those of parent daps 8 and 9 are given.

Using DFT B3-LYP/6-31G^{**} optimised geometry the UV spectra of 5–7 have been calculated by semiempirical AM1 and ZINDO/S methods and correlations with experimental UV data have been made.

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Table 1 Bond lengths and angles for 1–7 calculated by DFT B3-LYP/3-21G method

	1		2		3		4		5		6		7
Bond length (Å)													
N1–C2	1.336	N1–C2	1.334	C1–C2	1.382	C1–C2	1.382	N1–C2	1.336	N1–C2	1.334	C1–C2	1.382
C2–C3	1.408	C2-C3	1.406	C2–C3	1.410	C2–C3	1.410	C2–C3	1.404	C2–C3	1.407	C2–C3	1.411
C3-C4	1.384	C3-C4	1.382	C3-N4	1.333	C3-N4	1.333	C3–C4	1.383	C3–C4	1.381	C3-N4	1.333
C4–C4a	1.408	C4–C4a	1.423	C4–C4a	1.364	C4–C4a	1.364	C4–C4a	1.407	C4–C4a	1.409	C4–C4a	1.363
C4a–C10b	1.420	C4a-C10b	1.423	C4a-C10b	1.419	C4a-C10b	1.418	C4a-C10b	1.423	C4a-C10b	1.423	C4a-C10b	1.418
C10b-N1	1 358	C10b-N1	1 359	C10b-C1	1 414	C10b-C1	1 419	C10b-N1	1 365	C10b-N1	1 360	C10b-C1	1 415
C4a-N5	1 392	C4a-N5	1 359	C4a-C5	1 436	C4a-C5	1 436	C4a-N5	1 389	C4a-N5	1 391	C4a-C5	1 436
N5-C6	1 309	N5-C6	1 307	C5-N6	1 306	C5-N6	1 306	N5-C6	1 307	N5-C6	1 306	C5-N6	1 308
C6-C6a	1 442	C6-C6a	1 440	N6-C6a	1 386	N6–C6a	1 386	C6–C6a	1 440	C6-C6a	1 440	N6-C6a	1 390
C6a-C10a	1 422	C6a_C10a	1 427	C_{6a} - C_{10a}	1 428	C_{6a} - C_{10a}	1 428	C6a-C10a	1 425	C6a-C10a	1 427	C_{6a} - C_{10a}	1 426
C10a-C10b	1 441	C10a-C10b	1 453	C10a-C10b	1 454	C10a-C10b	1 454	C10a-C10b	1 453	C10a-C10b	1 456	C10a-C10b	1 453
C6a_C7	1 409	C6a-C7	1 408	C_{6a} - C_{7}	1 429	C6a-C7	1 428	C_{6a} - C_{7}	1 407	C_{6a} - C_{7}	1 409	C_{6a} - C_{7}	1 453
C7_C8	1 385	C7-C8	1 384	C7-C8	1 392	C7_C8	1 391	C7-C8	1 381	C7-C8	1 380	C7-C8	1 389
$C_{8}^{-}C_{9}^{-}$	1 402	C8-C9	1 396	C8-C9	1 398	C8-C9	1 399	C8-C9	1 400	C8-C9	1 402	C8-C9	1.303
C9-C10	1 307	C_{9} C_{10}	1 398	$C_{9}-C_{10}$	1 385	$C_{9}-C_{10}$	1 385	$C_{9}-C_{10}$	1.400	C9-C10	1 302	$C_{9}-C_{10}$	1 382
C_{10}	1.357	C_{10} C_{10}	1.370	C_{10} C_{10}	1.303	C_{10} C_{10}	1.505	C_{10} C_{10}	1.402	C_{10} C_{10}	1.372	C_{10} C_{10}	1.302
C10-N11	1 438	C10-N11	1 408	C7_N11	1 413	C7_N11	1.416	C10-N11	1 388	$C10_{2}$ N11	1.416	C7_N11	1.415
N11_N12	1 274	N11_N12	1 315	N11_N12	1 298	N11_N12	1 297	N11_C12	1 305	N11_C12	1 301	N11_C12	1 396
N12_C13	1 432	N12_C13	1.313	N12_C13	1.290	N12_C13	1.207	C12-013	1.375	C12-013	1.371	C12-C13	1.570
$C_{13} C_{14}$	1.405	$C_{13} C_{14}$	1.375	$C_{13} C_{14}$	1 403	C13 C14	1.400	C12-013	1.521	C12_C14	1.250	C12-C13	1.317
C13 - C14 C14 $C15$	1 38/	C13 = C14 C14 $C15$	1.421	C13 = C14 C14 $C15$	1 383	C13 = C14 C14 $C15$	1 / 1 3	012-014	1.521	C12 = C14 C14 $C15$	1.497	012-014	1.237
C14-C15	1.304	C14 - C15 C15 - C16	1.425	C15_C16	1.303	C14-C15	1 360			C15_C16	1 306		
C15-C10	1.413	C16 C162	1.303	C15-C10	1 /10	C16 C162	1.309			C16 C17	1 307		
C10-C17	1 292	C16a $C17$	1.431	C10-C17	1.419	C16a $C17$	1.420			C17_C18	1.397		
C17 - C18 C18 $C13$	1.303	C10a = C17	1 220	C17 = C13	1.300	C10a = C17	1.420			C17 = C10	1 202		
C16 N10	1.404	C17 = C18 C18 $C10$	1.300	C16_N10	1.407	C17 = C18 C18 $C10$	1.370			C10 - C19	1.392		
C10-IN19	1.3/1	C10 - C19	1.412	C10-IN19	1.309	$C_{10} = C_{19}$	1.412			C19-C14	1.400		
		$C_{19} = C_{20}$	1.362			$C_{19} = C_{20}$	1.300						
		$C_{20} = C_{20a}$	1.414			$C_{20} = C_{20a}$	1.422						
		$C_{20a} = C_{15}$	1.445			$C_{20a} = C_{10a}$	1.457						
		$C_{20a} = C_{10a}$	1.420			$C_{20a} = C_{13}$	1 272						
		021	1.554			021	1.373						
Angle (°)													
N1-C2-C3	123.097	N1-C2-C3	122.838	C1C2C3	119.213	C1C2C3	119.203	N1-C2-C3	122.764	N1-C2-C3	123.309	C1C2C3	119.283
C2-C3-C4	118.670	C2-C3-C4	118.168	C2-C3-N4	122.957	C2-C3-N4	122.971	C2-C3-C4	118.282	C2-C3-C4	118.174	C2-C3-N4	122.921
C3–C4–C4a	119.416	C3C4C4a	119.882	C3–N4–C4a	117.949	C3–N4–C4a	117.949	C3-C4-C4a	119.964	C3–C4–C4a	119.679	C3–N4–C4a	117.908
C4C4aC10b	118.150	C4-C4a-C10b	118.669	N4-C4a-C10b	123.387	N4-C4a-C10b	123.358	C4-C4a-C10b	118.742	C4-C4a-C10b	118.716	N4-C4a-C10b	123.468
C4a-C10b-N1	121.882	C4a-C10b-N1	120.182	C4a-C10b-C1	116.879	C4a-C10b-C1	116.918	C4a-C10b-N1	120.030	C4a-C10b-N1	120.608	C4a-C10b-C1	116.888
C10b-N1-C2	118.733	C10b-N1-C2	120.261	C10b-C1-C2	119.615	C10b-C1-C2	119.601	C10b-N1-C2	120.217	C10b-N1-C2	119.507	C10b-C1-C2	119.531
C4a-N5-C6	117.868	C4a–N5–C6	117.598	C4a-C5-N6	123.963	C4a-C5-N6	123.943	C4a–N5–C6	117.494	C4a–N5–C6	117.613	C4a-C5-N6	123.991
N5-C6-C6a	124.974	N5-C6-C6a	124.928	C5-N6-C6a	119.584	C5-N6-C6a	119.513	N5-C6-C6a	125.159	N5-C6-C6a	124.873	C5–N6–C6a	119.306
C6-C6a-C10a	117.947	C6-C6a-C10a	118.960	N6-C6a-C10a	122.063	N6-C6a-C10a	122.176	C6-C6a-C10a	118.906	C6-C6a-C10a	118.951	N6-C6a-C10a	122.289
C6a-C10a-C10b	117.715	C6a-C10a-C10b	116.614	C6a-C10a-C10b	117.509	C6a-C10a-C10b	117.468	C6a-C10a-C10b	116.387	C6a-C10a-C10b	116.683	C6a-C10a-C10b	117.563
C10a–C10b–C4a	118.785	C10a–C10b–C4a	118.716	C10a–C10b–C4a	118.593	C10a–C10b–C4a	118.575	C10a–C10b–C4a	119.095	C10a–C10b–C4a	118.405	C10a–C10b–C4a	118.558
C10b-C4a-N5	122.577	C10b-C4a-N5	120.927	C10b-C4a-C5	118.115	C10b-C4a-C5	118.137	C10b-C4a-N5	122.956	C10b-C4a-N5	123.384	C10b-C4a-C5	118.281
C6a-C7-C8	119.478	C6a–C7–C8	119.952	C6a-C7-C8	119.133	C6a-C7-C8	119.380	C6a-C7-C8	119.051	C6a-C7-C8	119.948	C6a-C7-C8	118.955
С7-С8-С9	120.582	C7–C8–C9	119.919	C7–C8–C9	121.340	C7–C8–C9	121.141	C7–C8–C9	120.956	C7–C8–C9	119.554	C7–C8–C9	121.289

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