

Fabrication and characterization of bulk glassy $\text{Co}_{40}\text{Fe}_{22}\text{Ta}_8\text{B}_{30}$ alloy with high thermal stability and excellent soft magnetic properties

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Abstract

In this work, $\text{Co}_{40}\text{Fe}_{22}\text{Ta}_8\text{B}_{30}$ alloy as a new bulk metallic glass with a wide supercooled liquid region of 74 K and excellent soft magnetic properties was prepared by the powder metallurgy method. Glassy $\text{Co}_{40}\text{Fe}_{22}\text{Ta}_8\text{B}_{30}$ powders were obtained by ball milling of melt-spun glassy ribbons at a cryogenic temperature and subsequently consolidated by hot pressing into disk-shaped specimens 10 mm in diameter and 2 mm thick. It was found that the new glassy alloy exhibited the largest diameter compared with the other Co-based bulk metallic glasses produced in the well-known Co–Fe–Ta–B alloying system up to now. The influence of the consolidation time on the microstructure and magnetic properties of the bulk samples was investigated by X-ray diffraction, differential scanning calorimetry, vibrating sample magnetometry and Faraday magnetometry. The results indicate that the new alloy exhibits a long incubation time before crystallization upon annealing above the glass transition temperature: noticeably longer than for other known (Co,Fe)-based amorphous alloys. The glassy bulk sample consolidated for 600 s at 923 K had a relative density of 99.2%, a saturation magnetization of $46.6 \text{ A m}^2 \text{ kg}^{-1}$, a Curie temperature of 425 K and a low coercivity of 6 A m^{-1} . In addition, the new bulk glassy alloy exhibited ultra-high hardness of 13.48 GPa. The mechanisms by which the thermal stability and incubation time prior to crystallization increase are explained in accordance with pair correlation function analysis.

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Keywords: Bulk metallic glass; Magnetic properties; Pair correlation function

1. Introduction

Bulk metallic glasses (BMG) have attracted increasing interest owing to their excellent properties, such as ultra-high mechanical strength, high corrosion resistance, high electrochemical discharge capacity and soft magnetic properties [1–4]. Multicomponent BMG several millimeters thick have been produced by rapid solidification techniques such as copper mold casting of different alloy systems, e.g., Pd-based, Zr-based, Cu-based, Ni-based, Fe and Co-based alloys [3]. BMG with shapes and dimensions that cannot be produced by direct casting may be accessible by powder processing routes [5]. For this purpose, glassy bulk speci-

mens could be produced by the consolidation of glassy powders above the glass transition T_g in the supercooled liquid region (SLR), where the viscosity of the powders is significantly suppressed [5]. In order to achieve a high density product upon compaction, it is essential to allow enough consolidation time in the temperature range of the SLR, without crystallizing the sample. Therefore, a large width of SLR and long incubation time before crystallization are essential requirements for sintering.

Co-based BMG are well known for their excellent soft magnetic properties, such as very low coercivity and high magnetic permeability [1,6]. Almost 10 years ago, glassy $\text{Co}_{43}\text{Fe}_{20}\text{Ta}_{5.5}\text{B}_{31.5}$ rods 2 mm in diameter, with an ultra-high fracture strength of 5 GPa, low coercivity of 0.25 A m^{-1} and maximum magnetic permeability of 550,000, were produced by copper mold casting [1,7].

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Glassy $\text{Co}_{43}\text{Fe}_{20}\text{Ta}_{5.5}\text{B}_{31.5}$ alloy showed the maximum width of SLR (72 K) in the composition range 5–30 at.% Fe, 4–6 at.% Ta and 20–33 at.% B [1]. In contrast to its high glass-forming ability compared with other Co-based BMG, annealing of $\text{Co}_{43}\text{Fe}_{20}\text{Ta}_{5.5}\text{B}_{31.5}$ alloy in the SLR results in crystallization with very short incubation times, even when the annealing temperature is slightly above T_g . As an example, the $(\text{Co,Fe})_{21}\text{Ta}_2\text{B}_6$ phase precipitates upon annealing at 928 K for 60 s, i.e., just 18 K above the T_g [1].

In this work, $\text{Co}_{40}\text{Fe}_{22}\text{Ta}_8\text{B}_{30}$ BMG (i.e., with higher Ta content compared with $\text{Co}_{43}\text{Fe}_{20}\text{Ta}_{5.5}\text{B}_{31.5}$ alloy) was produced by the consolidation of glassy powders above the corresponding T_g . Previous results indicated that the mechanical alloying process was unsuccessful at producing fully glassy $\text{Co}_{40}\text{Fe}_{22}\text{Ta}_8\text{B}_{30}$ powders [8,9]. The unreacted boron inclusions, besides the small fraction of nanocrystalline α -(Fe, Co) phase, remained in the glassy matrix even after milling for 200 h [8,9]. In the current research, the glassy powders were obtained by controlled ball milling of melt-spun glassy ribbons with a wide SLR of 74 K and a significantly longer incubation time prior to crystallization compared with the $\text{Co}_{43}\text{Fe}_{20}\text{Ta}_{5.5}\text{B}_{31.5}$ alloy. Subsequent consolidation of the milled powders allowed bulk glassy samples 10 mm in diameter and 2 mm thick to be produced. Besides a very good thermal stability, the resulting samples possess a high relative density of 99.2%, low coercivity of 6 A m^{-1} and ultrahigh hardness of 13.48 GPa. To the present authors' knowledge, there are no reports on the fabrication of Co-based BMG more than 3 mm in diameter in the Co–Fe–Ta–B system with such excellent thermal stability and good soft magnetic properties.

2. Experimental procedure

A master alloy with nominal composition of $\text{Co}_{40}\text{Fe}_{22}\text{Ta}_8\text{B}_{30}$ (at.%) was prepared by arc-melting of pure Co (99.9%), Fe (99.95%), Ta (99.96%) and crystalline B (99.5%) under a Ti-gettered Ar atmosphere. Amorphous ribbons (width 3.5 mm, thickness 29 μm) were prepared under Ar flow by a single-roller Bühler melt spinner on a copper wheel rotating at 41 m s^{-1} tangential velocity. The ball milling of the ribbons was performed on a planetary ball mill (Retsch PM4000), using hardened steel balls 10 mm in diameter and hardened steel vials with a capacity of 250 ml. Before the milling, the as-cast ribbons were cut into small pieces ($3.5 \times 10 \text{ mm}^2$). A typical charge of 4 g of ribbons was milled for 3 h at a ball-to-powder weight ratio of 15:1 and a rotation speed of 200 rpm. To avoid a strong temperature rise, the milling was interrupted at each 15 min of working, and the vials were cooled down for 20 min in a bath of liquid nitrogen. To minimize oxidation, the sample handling was carried out in a glove box under purified argon atmosphere ($<1 \text{ ppm O}_2$ and H_2O). The as-milled powders were sieved to $<100 \mu\text{m}$ and used for the subsequent consolidation. Disk-shaped samples 10 mm in diameter and 2 mm thick were prepared under

an argon atmosphere by consolidation of the powders, using a uniaxial hot pressing machine. The powders were placed under a pressure of 800 MPa, heated at a rate of 20 K min^{-1} to 923 K and kept at this temperature for different pressing times, then cooled down to ambient temperature using a forced flow of Ar gas.

The phase analysis of the samples was performed using X-ray diffraction (XRD) in reflection configuration, using Co K_α ($\lambda = 1.7889 \text{ \AA}$) radiation, as well as in transmission configuration, using a high-intensity, high-energy monochromatic synchrotron beam with an incident beam energy of 100 keV at the BW5 beamline at Hamburger Synchrotronstrahlungslabor, HASYLAB (Deutsches Elektronen-Synchrotron, DESY, Hamburg, Germany). The two-dimensional XRD patterns were integrated to the Q -space, where Q is the wave vector, using the FIT2D software package [10]. The integrated data were corrected for polarization, sample absorption, fluorescence contribution and inelastic scattering using the pair correlation function PDFgetX2 software [11]. The total structure factor $S(Q)$ was determined from the normalized elastically scattered intensity according to Faber–Ziman equation [12]. The pair distribution function $PDF(r)$ was calculated as

$$PDF = 1 + \frac{1}{2\pi^2\rho_0 r} \int_0^{Q_{\max}} Q(S(Q) - 1) \sin(Qr) dQ \quad (1)$$

where ρ_0 is the average atomic number density.

The microstructure was characterized by high-resolution transmission electron microscopy (HRTEM; Tecnai F30 operating at 300 kV) and scanning electron microscopy (SEM; Hitachi TM 1000). The thermal stability was investigated with a differential scanning calorimeter (NETZSCH DSC 404) at a 20 K min^{-1} heating rate under a flow of high-purity argon. The viscosity of the samples was measured by parallel plate rheometry using a Perkin-Elmer thermomechanical analyzer (TMA7) device operated at a heating rate of 20 K min^{-1} and under a static load of 2.6 N. The density of ribbons and bulk samples was measured by the Archimedes method, using a computer-controlled microbalance and glycol as the working liquid. The final values were obtained by averaging over 50 experimental values. The Vickers hardness was determined for the bulk samples using a computer-controlled Struers Duramin 5 Vickers hardness tester. The tests were performed using a typical diamond indenter in the form of a pyramid with a square base and an angle of 136° between opposite faces, applying a load of 1.96 N for 10 s.

For magnetic investigations, the M – H hysteresis loops were recorded by vibrating sample magnetometry (VSM) at ambient temperature. For the bulk samples, VSM measurements were carried out on rectangular specimens ($8 \times 4 \text{ mm}^2$) cut from the consolidated disks. The coercivity was measured using a Foerster Coercimat under an applied field high enough to saturate the samples. The variation in magnetization with temperature and the Curie temperature T_c of the samples were determined using a

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