



# Humidity sensing properties of MoO<sub>3</sub>-NiO nanocomposite materials

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## Abstract

MoO<sub>3</sub>-NiO composites were prepared by a simple one-step electrospinning method and were characterized by X-ray diffraction (XRD), scanning electron microscope (SEM) and nitrogen adsorption-desorption test. Humidity sensing properties of MoO<sub>3</sub>-NiO composites with different mass ratios were studied, the best result was obtained for the sample with MoO<sub>3</sub>-NiO mass ratio of 0.5 (S1) under the same preparation condition. The impedance of the sensor changed more than three orders of magnitude when relative humidity ranged from 11% to 95%. The response and recovery time were about 2 s and 10 s respectively with maximum hysteresis of < 2%RH (relative humidity). The above findings suggest potential applications of the MoO<sub>3</sub>-NiO composite (S1) in high performance humidity sensors.

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## 1. Introduction

With the rapid development of the society, the demand for a high powered humidity sensor has become unprecedented. Particularly, the development of an excellent humidity sensing material is an important aspect in the creation of a high-powered humidity sensor. In the past decades, many kinds of materials including polymer [1–4], electrolytes [5–7], organic–inorganic hybrid composites [8–11], photonic crystal [12–14] and metal oxides or ceramic [15–20] have been explored to fabricate humidity sensing detectors. But each kind of material cannot completely meet all requirements desired for an excellent sensor such as high sensitivity, good linearity, quick response–recovery and long-term stability. Thus, various ways have been developed to improve the performance of a humidity sensor. One method which is widely used is the addition of additives or doping in a controlled quantity [21–22]. For example, Barnali Bej, and his co-workers enhanced the sensitive characteristics through mixing SiO<sub>2</sub> and NiO together to form a composite [23]. Hailong Yu and his co-workers enhanced the sensitive

properties through mixing ZnO and MoO<sub>3</sub> together to form a composite [24]. They mainly make use of improving specific surface area to enhance the sensitive characteristics by doping. But in our research we not only increased specific surface area but also offered more active sites to improve moisture properties through the NiO and MoO<sub>3</sub> mixed together.

Here MoO<sub>3</sub>-NiO composites were synthesized by electrospinning combined with a calcinations process. The humidity sensor was fabricated by coating MoO<sub>3</sub>-NiO powders on five pairs of Ag–Pd interdigitated electrodes. The humidity sensing properties of the materials were investigated. The results showed that the sensor based on MoO<sub>3</sub>-NiO composite had excellent impedance–RH linearity, narrow hysteresis and rapid response–recovery time, which exhibited a potential application in improving the humidity sensor performance.

## 2. Experimental

All chemicals (NiCl<sub>2</sub> · 6H<sub>2</sub>O, (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub> · 4H<sub>2</sub>O) used in the present work are AR grade from Tianjin Chemical Co, and used without further purification.

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MoO<sub>3</sub>-NiO composite NFs with four different contents (the mass ratio of MoO<sub>3</sub>-NiO is 0.5, 1, 2, 3) were prepared through a simple one-step electrospinning method. In a typical synthesis, 0.4 g of mixture of NiCl<sub>2</sub>·6H<sub>2</sub>O and (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub>·4H<sub>2</sub>O was dissolved in 13.2 g of mixed solvents of ethanol, N, N-dimethyl formamide (DMF) and Ammonia water (ethanol: DMF: Ammonia water = 1:1:1 by Mass) and stirred for 30 min at room temperature, then 1 g poly (vinylpyrrolidone) (PVP, Mw = 1,300,000) was added to the above solution. After vigorous stirring for 12 h, a clear solution was attained, which was used for the electrospinning preparation of precursor NFs. The electrospinning setup is similar with what reported before [25], which consists of three major components: a high-voltage power supply, a syringe pump and a spinneret setup. In electrospinning process, the collection distance between spinneret tip and collector is 12 cm and the applied steady-voltage is 20 kV. The feeding rate of solution is 0.6 ml/h. During the electrospinning, the solution jet solidified with accompanying evaporation of solvent and formed a nonwoven fibrous mat on the collector. After dried 12 h at room temperature under vacuum, the electrospinning fibers were annealed in a tube furnace with a rising rate of 1 °C/min from room temperature to 600 °C and kept for 2 h, then were self-cooled down to room temperature again, forming the final NFs. The resulting samples were marked as MoO<sub>3</sub>-NiO composites of different mass ratio (0.5, 1, 2, 3). We called these mixed samples as S1, S2, S3, S4 (according to the NiO mass ratio from more to less). We also synthesized the pure NiO and MoO<sub>3</sub> nanofibers in the same way as a comparison.

The samples were ground with deionized water in a weight ratio of 5:1 to form a paste which was then coated on a ceramic substrate with five pairs of Ag-Pd interdigitated electrodes to form the sensitive film. The finger width and the distance between each electrode were both 0.15 mm. The film was dried in oven at 60 °C for 6 h. At last, the humidity sensor was fabricated after aging in 100%RH with an alternating voltage of 1 V, 100 Hz for 24 h. A schematic image of this electrode is shown in Fig. 1.

NiO, MoO<sub>3</sub> and MoO<sub>3</sub>-NiO composites were characterized by X-ray diffraction (XRD) using a Shimadzu XRD-6000 diffractometer with Cu K $\alpha$  radiation ( $\lambda = 1.548 \text{ \AA}$ ). The surface morphology of the as-prepared samples was inspected using a JEOL JSM-7500 F field emission SEM. The electrical properties of the sensor were measured by a Precision Impedance Analyzer 6500B Serious of Wayne Kerr Electronics at the room temperature under different relative humidity (RH). The porosity of the samples was measured at 77k by the nitrogen adsorption-desorption isotherm and BJH methods on

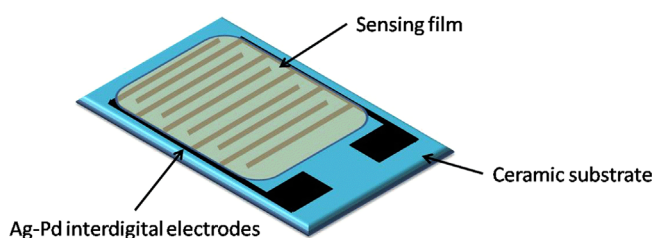


Fig. 1. Schematic drawing of the humidity sensor.

a Micromeritics ASAP 2020 M volumetric adsorption analyzer at the temperature of liquid nitrogen. The samples were put into six chambers with different RH at the constant temperature 25 °C and impedance changes were measured. The RH range of 11%–95% was obtained using saturated salt solutions as the humidity generation sources. The six chambers were the saturated salt solutions of LiCl, MgCl<sub>2</sub>, Mg(NO<sub>3</sub>)<sub>2</sub>, NaCl, KCl and KNO<sub>3</sub>, and their corresponding RH values were 11%, 33%, 54%, 75%, 85% and 95% RH respectively [25]. The voltage was set at AC 1 V, and the frequency range was from 20 Hz to 10 kHz in humidity studies.

### 3. Results and discussion

Powder X-ray diffraction analysis was performed on NiO, MoO<sub>3</sub> and MoO<sub>3</sub>-NiO composites (Fig. 2). The diffraction peaks (101) in the XRD patterns can be indexed to NiO (JCPDS 44-1159) and (110), (040), (021) are the reflection of MoO<sub>3</sub> (JCPDS 35-0609). No diffraction peaks from other impurities are seen. Moreover, the intensity of reflections (110), (040) (021) in the obtained samples get lower and lower along with the increase of NiO/MoO<sub>3</sub> mass ratio, peak intensity of (101) changes just in the opposite. In order to further explore the structure features, the obtained samples were measured by SEM. As shown in Fig. 3, the average diameters of the as-synthesized fibers are from 50 to 80 nm.

Fig. 4 shows the dependence of impedance versus the RH for NiO, MoO<sub>3</sub> and MoO<sub>3</sub>-NiO samples with different ratios operated under 1 V AC voltage with the frequency of 100 Hz [26]. All the samples exhibited very high impedance at low relative humidity. When the relative humidity increased, the impedance of MoO<sub>3</sub>-NiO samples declined dramatically, however, the impedance of pure MoO<sub>3</sub> and NiO samples is almost constant. On the whole, during the RH range of 11–95%, the sensor based on nanofiber MoO<sub>3</sub>-NiO (S1) shows the best linearity with impedance varying more than five orders of magnitude. Thus the MoO<sub>3</sub>-NiO (S1) sample is used in the following experiment.

The average pore diameter and BET surface area of pure MoO<sub>3</sub>, NiO and MoO<sub>3</sub>-NiO (S1) samples are listed in Table 1. The N<sub>2</sub> adsorption-desorption isotherms and pore size distribution of MoO<sub>3</sub>-NiO (S1) samples and pure MoO<sub>3</sub>, NiO nanofiber were shown in Fig. 5. We can see that MoO<sub>3</sub>-NiO

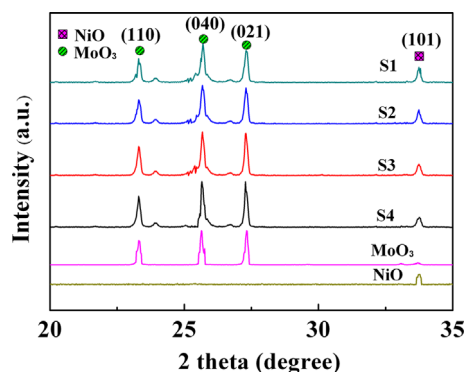


Fig. 2. XRD patterns of MoO<sub>3</sub>, NiO and MoO<sub>3</sub>-NiO composites.

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