



High quality factor of $(\text{Zn}_{0.6}\text{Mg}_{0.4})_{1.918}\text{GeO}_{3.918}$ microwave dielectrics

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Abstract

The dependence of the microwave dielectric properties on the structural characteristics of $(\text{Zn}_{1-x}\text{Mg}_x)_{1.918}\text{GeO}_{3.918}$ ($0.2 \leq x \leq 0.6$) was investigated. With increasing Mg^{2+} substitution up to $x=0.4$, the single-phase rhombohedral structure was detected, however, the secondary phase of Mg_2GeO_4 with orthorhombic structure was observed for further substitution of Mg^{2+} . The quality factor (Qf) was dependent on the structural symmetry, in particular the tetrahedral distortion. $(\text{Zn}_{0.6}\text{Mg}_{0.4})_{1.918}\text{GeO}_{3.918}$ showed the highest Qf value throughout the entire compositional range, which could be attributed to the high symmetry of the crystal structure with equilateral hexagonal rings composed of GeO_4 and $(\text{Zn,Mg})\text{GeO}_4$ tetrahedra. The dielectric constant (K) and the temperature coefficient of resonant frequency (TCF) of the specimens changed only slightly up to $x=0.4$, while at $x=0.5$ and 0.6 , they changed remarkably due to the formation of the secondary phase of Mg_2GeO_4 .

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1. Introduction

Various types of dielectric materials have been investigated to meet the requirements for the components and devices working at microwave and millimeter-wave frequencies due to the remarkable progress in wireless communications. For these applications, dielectric materials should have a low dielectric constant (K) to minimize cross-coupling with conductors, a high quality factor (Qf) to increase the selectivity, and a near-zero temperature coefficient of resonant frequency (TCF) to ensure stability of the frequency against temperature changes [1].

Several dielectric ceramics with low K and high Qf value, such as Al_2O_3 [2], $\text{Mg}_4\text{Nb}_2\text{O}_9$ [3], $\text{Ca}_3\text{Sn}_2\text{SiO}_9$ [4], Mg_2SiO_4 [5], and Zn_2SiO_4 [6] have received attention with respect to applications as ceramic substrate materials of microwave integrated circuits (MIC) to minimize the cross-coupling with conductors and to shorten the time for electronic signal transmission.

Recently, Zn_2SiO_4 (willemite)-based ceramics [6–9] were widely investigated because of their high Qf values, as shown in

Table 1. However, the data exhibit a lack of reproducibility and low Qf values of Zn_2SiO_4 were also reported, which could be improved by Mg^{2+} substitution (see Table 1). According to Li et al. [7], the Qf value of Zn_2SiO_4 ceramic can be improved by the substitution of 0.4 mol of Mg^{2+} for Zn^{2+} owing to the suppression of secondary-phase formation. However, Song et al. [8] reported that $(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{SiO}_4$ ceramics with single, pure rhombohedral phase showed high Qf values, even though these ceramics were prepared by the same solid-state reaction method. This result indicated that the reason for the high Qf value of Mg^{2+} substituted willemite ceramics cannot be explained by only extrinsic factors, such as the relative density, grain size, and percentage of the secondary phase, but that intrinsic factors, such as the structural characteristics, degree of crystal structural ordering, and lattice distortions have to be included [10–12]. However, the relationships between the Qf value and the intrinsic factors based on crystal structural characteristics were not reported yet for willemite-type ceramics.

It has been reported [13] that Zn_2SiO_4 has a rhombohedral structure belonging to space group $R\bar{3}$, and Mg_2SiO_4 has an orthorhombic structure belonging to space group $Pmnb$. The unit cell of Zn_2SiO_4 consist of two ZnO_4 tetrahedra and one SiO_4 tetrahedron connected by a shared vertex, and Mg_2SiO_4

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Table 1
Microwave dielectric properties of $(\text{Zn},\text{Mg})_2\text{SiO}_4$ prepared by solid-state reaction.

Composition	K	Qf (GHz)	TCF (ppm/°C)	Ref.
Zn_2SiO_4	6.6	219,000	−61	[6]
Zn_2SiO_4	6.5	34,700	−52	[7]
$(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{SiO}_4$	6.1	148,740	−54	
Zn_2SiO_4	6.7	35,330	−58	[8]
$(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{SiO}_4$	6.6	95,650	−60	
$\text{Zn}_{1.8}\text{SiO}_{3.8}$	6.6	147,000	−22	[9]

is composed of two MgO_4 octahedra connected to one SiO_4 tetrahedron by a shared vertex and edge. Similarly, Zn_2GeO_4 and Mg_2GeO_4 also have same crystal structures like the silicates Zn_2SiO_4 , Mg_2SiO_4 . With the substitution of Mg^{2+} for Zn^{2+} in Zn_2GeO_4 , the coexistence of the rhombohedral and orthorhombic phase was reported at the 0.5 mol of Mg^{2+} in Zn_2GeO_4 [14,15]. From these structural considerations, it could be expected that the Qf value of Zn_2GeO_4 would be improved by the substitution of Mg^{2+} for Zn^{2+} . Therefore, the investigation of the relationships between Qf value and crystal structural characteristics of Mg^{2+} substituted Zn_2GeO_4 may clarify the reason for the high Qf value of these substitutional oxides, as well as promote the development of new willemite-type ceramics with high Qf value.

In this study, the effects of Mg^{2+} substitution on the microwave dielectric properties of $(\text{Zn}_{1-x}\text{Mg}_x)_{1.918}\text{GeO}_{3.918}$ ($0.2 \leq x \leq 0.6$) were investigated, based on crystal structural characteristics by means of Rietveld refinements of X-ray diffraction data and analysis of Raman spectra.

2. Experimental procedure

$(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{GeO}_4$ and $(\text{Zn}_{1-x}\text{Mg}_x)_{1.918}\text{GeO}_{3.918}$ ($0.2 \leq x \leq 0.6$) were synthesized by the solid-state reaction method. High purity oxide powders of ZnO (99%), MgO (99.9%), and GeO_2 (99.9%) were mixed in stoichiometric ratios and the mixed powders were milled with ZrO_2 balls for 24 h in ethanol. The powders were calcined at 1000 °C for 3 h and the calcined powders were re-milled for 24 h, dried, and then sieved through 250 meshes. Finally, the powders were isostatically pressed into pellets under a pressure of 1500 kg/cm². These pellets were sintered at 1300 °C for 4 h in air.

The density of the sintered specimens was measured by the Archimedes method. Scanning electron microscopy (BSEM, JSM-7500F, JEOL, Japan), energy dispersive X-ray spectrometry (EDS, JSM-7500F, JEOL, Japan), high-resolution transmission electron microscopy (HR-TEM, JEM-2100F, JEOL, Japan) and X-ray powder diffraction analysis (XRD, D/Max-2500V/PC, RIGAKU, Japan) were used to evaluate the crystalline phases and microstructures of the sintered specimens.

The crystal structural characteristics of the specimens were determined by Rietveld refinements of the XRD patterns using the FullProf program [16]. Raman spectra of $(\text{Zn}_{1-x}\text{Mg}_x)_{1.918}\text{GeO}_{3.918}$ were obtained with a Micro-Raman spectrometer (Dispersive

Raman, SENTERRA Raman, Bruker Corporation, Germany) using a He–Ne laser (633 nm, 10 mW).

The microwave dielectric properties (K , Qf) were measured by the Hakki and Colema method [17] with the TE_{011} mode at 12 GHz, and the TCF was measured by the cavity method [18] in a chamber over a temperature range from 25 °C to 80 °C.

3. Results and discussion

3.1. Preparation of single phase $(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{GeO}_4$ with rhombohedral structure

$(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{GeO}_4$ was prepared by conventional solid-state reaction and their XRD patterns are given in Fig. 1. For $(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{GeO}_4$, the ZnO secondary phase was detected, as well as Zn_2GeO_4 (JCPDS no. 80-0580) with rhombohedral structure as the main phase. According to Nguyen et al. [9], the complete interaction between SiO_2 and ZnO in Zn_2SiO_4 is difficult because of the stable crystal structure of crystalline SiO_2 , which induces the formation of ZnO as secondary phase in Zn_2SiO_4 , along with a degenerated Qf value. Similarly, GeO_2 also has the same crystal structure like SiO_2 and is non-reactive toward other oxide [19], which caused the formation of ZnO as secondary phase in $(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{GeO}_4$ sintered at 1300 °C for 4 h.

In order to remove the secondary ZnO phase, the A-site mole fraction of $(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{GeO}_4$ was reduced from 2 to 1.918. It was reported that the quality factor (Qf) of willemite ceramics could be improved by the elimination of the secondary ZnO phase in Zn_2SiO_4 [9]. As shown in Fig. 1, the secondary phase of ZnO was clearly removed and single phase rhombohedral zinc germanate (JCPDS no. 80-0580) was confirmed for $(\text{Zn}_{0.6}\text{Mg}_{0.4})_{1.918}\text{GeO}_{3.918}$ sintered at 1300 °C for 4 h. Furthermore, the rhombohedral structure could be observed by high-resolution TEM (HRTEM) and a spot pattern of the selected-area electron diffraction (SAED) is shown in Fig. 2. HRTEM images showed the rhombohedral crystal structure along the c -axis and the corresponding (001) lattice planes exhibited smaller spacing of 0.69 nm than that reported for Zn_2GeO_4 (0.724 nm) [19] due to the smaller ionic size of

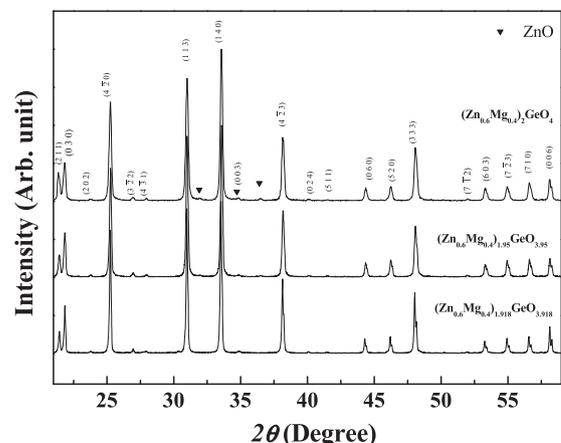


Fig. 1. XRD patterns of $(\text{Zn}_{0.6}\text{Mg}_{0.4})_2\text{GeO}_4$, $(\text{Zn}_{0.6}\text{Mg}_{0.4})_{1.95}\text{GeO}_{3.95}$, and $(\text{Zn}_{0.6}\text{Mg}_{0.4})_{1.918}\text{GeO}_{3.918}$ sintered at 1300 °C for 4 h.

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