

Available online at www.sciencedirect.com

SciVerse ScienceDirect

CERAMICSINTERNATIONAL

Ceramics International 39 (2013) S217-S220

www.elsevier.com/locate/ceramint

Dielectric and ferroelectric properties of BiFeO₃ ceramics sintered in different atmospheres

Hongyong Liu*, Yongping Pu, Xuan Shi, Qibin Yuan

School of Materials Science and Engineering, Shaanxi University of Science and Technology, Xi'an 710021, China

Available online 16 October 2012

Abstract

Pure BiFeO₃ powders were successfully synthesized by microwave-hydrothermal processing, and the pellets were sintered in different atmospheres. The properties of BiFeO₃ ceramic samples were characterized with X-ray diffraction (XRD), scanning electron microscopy (SEM), dielectric measurements (LCR) and polarization-field hysteresis loop. The effects of different atmospheres (including air, N_2 , O_2 and H_2) on ferroelectric properties of BiFeO₃ ceramics were studied in this paper. The results show that the BiFeO₃ ceramics sintered in H_2 and N_2 atmosphere have a single-phase rhombohedra distorted perovskite structure with no trace of other impurity phases. The dielectric properties of BiFeO₃ ceramics were greatly influenced by O_2 and H_2 atmospheres. The samples sintered in H_2 atmosphere have the highest dielectric constant. Polarization-field hysteresis loops measured at room temperature indicated the samples sintered in H_2 and H_3 atmospheres have higher spontaneous polarization and lower breakdown field, but the ferroelectric properties of samples sintered in H_3 atmospheres were poor.

© 2012 Elsevier Ltd and Techna Group S.r.l. All rights reserved.

Keywords: A. Microwave processing; C. Dielectric properties; C. Ferroelectric properties

1. Introduction

BiFeO₃ is a good candidate, with the crystallographic space group R3c, allows the existence of both antiferromagnetic and ferroelectric orders with very high transition temperatures [1]. It may give rise to an additional degree of freedom in providing an additional functionality in device designing [2]. So BiFeO₃ has been extensively investigated in the forms of bulk ceramics, single crystal and thin film owing to its high phase transition temperature and good multiferroic properties [3,4]. As for BiFeO₃ ceramics, their commercial applications are not yet successful due to difficulties in restricting the formation of impurity phase and pores as well as mitigating the serious leakage current-induced dielectric breakdown [5]. In order to improve the purity and electrical properties, some researchers attempt to use sol-gel method and spark plasma sintering to synthesis purity phase BiFeO₃ ceramics [6,7]. Recently it has been shown that rapid liquid phase sintering of BiFeO₃ can result in a high resistivity and polarization values of BiFeO₃ [8], but can also lead to high dielectric loss and more defects. Several research groups adopted the strategy of doping BiFeO₃ with different trivalent ions on A, B or both A and B site [9–11]. The doping has resulted in the reduction of the leakage current density and in the improvement of the ferroelectric properties to some extent, but it is still far from the practical applications. It is possible that the sintering atmosphere was a key parameter to improve the leakage current and structural inhomogenity.

Therefore, we aim in this work to prepare poreless, low-resistive single-phase multiferroic BiFeO₃ ceramics and study the effect of different atmospheres (including air, N₂, O₂, H₂) on the density and electric properties of BiFeO₃ ceramics, the optimal sintering atmosphere was determined by the analysis of density and electric properties of BiFeO₃ ceramics.

2. Experimental procedures

Analytical grade chemical bismuth nitrate $(Bi(NO_3)_3 \cdot 5H_2O)$ ($\geq 99.5\%$) and iron nitrate $(Fe(NO_3)_3 \cdot 9H_2O)$ ($\geq 99.5\%$) were used as raw materials. Sodium hydroxide, NaOH, was used as alkaline mineralizer. Equi-molar

^{*}Corresponding author. Tel.: +86 29 86168803; fax: +86 29 86168688. *E-mail address:* liusuanghua@163.com (H. Liu).

mixtures of bismuth nitrate and iron nitrate were dissolved in distilled water. NaOH was added dropwise to the above solution until pH=13. The resultant solution was poured into double-walled digestion vessels, microwave hydrothermal experiments were conducted in a temperature-controlled oven. The powder produced was washed thoroughly with distilled water and dried at $80\,^{\circ}$ C, and then heat-treated at $600\,^{\circ}$ C for 2 h. The BiFeO₃ ceramics were prepared using a conventional ceramic processing. The microwave hydrothermally synthesized BiFeO₃ powders were ball-milled with 1 wt% PVA and then dried. Pellets of 11 mm in diameter and 0.9 mm in thickness were uniaxially pressed at 60 MPa. The pellets were sintered in different atmospheres (including air, N₂, O₂ and H₂) at $800\,^{\circ}$ C.

The crystalline phase of the sintered samples was identified using an automated diffractmeter (D/max-₂₂00PC, RIGAKU, Japan). A scanning electron microscope (JSM-6460) was used to investigate the microstructure of BiFeO₃ ceramic sintered in different atmospheres. The dielectric data of BiFeO₃ ceramics were collected using the LCR meter (Agilent 4980A). Room temperature ferroelectric measurements were carried out using ferroelectric hysteresis loop tracer (Model:610E).

3. Results and discussion

Fig. 1 shows the X-ray diffraction patterns of BiFeO₃ ceramics. The XRD patterns reveal that the BiFeO₃ ceramics sintered in N₂ and H₂ atmospheres have a single-phase rhombohedra distorted perovskite structure. But trace of secondary phase was observed when samples sintered in O₂ atmospheres. Bernardo [12] has reported that BiFeO₃ was a metastable product, it seems to initiate its decomposition at the temperature above 800 °C and trace amounts of Bi₂Fe₄O₉ mullite-type phase can be observed at this temperature. Fig. 1 indicated that microwave hydrothermal synthesis was an effective method to synthesis pure BiFeO₃ powder and sintering in N₂ and H₂

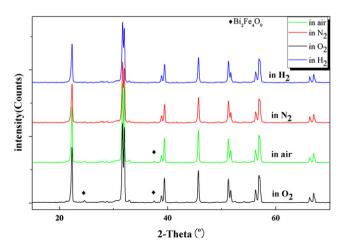


Fig. 1. X-ray diffraction pattern of $BiFeO_3$ ceramics sintered in different atmospheres.

atmospheres have no effect on the purity. The O_2 atmosphere has an influence on the purity, which is in good agreement with Bernardo's experimental results [12]. It indicated that sintering in O_2 atmosphere was favorable for the decomposition of BiFeO₃.

Fig. 2 shows the SEM micrographs of the as-sintered surface of BiFeO₃ ceramics. It can be seen that all grain samples were irregular in shape. The samples sintered in N₂ and H₂ atmospheres have a high densification, whereas, the densification was obviously poor when sintered in O₂ atmospheres, some distinct pores existed in the grain and grain boundary and several tiny whitish spots were observed immersed in a matrix of light gray grains. We thought that the tiny whitish spots were Bi₂Fe₄O₉ mullitetype phase. It indicates that the N2 and H2 atmospheres were beneficial to the sintering densification of BiFeO₃ ceramic, which could be explained that the concentration of oxygen vacancies, which was important for the transfer of mass and energy between reactants during sintering, was increased when sintered in N2 and H2 atmospheres, which resulted in the diffusion of ions was improved in sintering process.

However, the production of oxygen vacancies was inhibited when sintering in O_2 atmospheres, thus inhibiting the diffusion of ions, which led to the lower densification and the formation of secondary phase.

Fig. 3 shows the dielectric constant of the samples sintered in different atmospheres as a function of frequency at room temperatures. The dielectric loss for all samples are low at low frequencies up to 10 kHz and then show a rising trend above 10 kHz, the maximum value of dielectric loss is reached at the frequency of 1 MHz. Usually in BiFeO₃, oxygen deficiency is an inherent problem and space charge polarization is always present [4]. The high values of dielectric constant at lower frequency are explained on the basis of dipolar and space charge polarization due to the presence of inhomogeneities in the dielectric structure. However, at high frequencies the electronic polarization is effective and dipolar contribution becomes insignificant. The decrease in dielectric constant with increased frequency could be explained on the basis of dipole relaxation phenomenon, the inability of the electric dipoles to be in pace with the frequency of applied electric field at high frequency.

It can be seen in Fig. 3 that the samples sintered in O₂ atmosphere show higher dielectric constant and dielectric loss below the frequency of 0.1 MHz, but the dielectric constant and dielectric loss are the lowest above the frequency of 0.2 MHz. The samples sintered in H₂ atmosphere show the highest dielectric constant and dielectric loss values over the entire frequency range examined. The microstructures and impurity phases in BiFeO₃ samples sintered in O₂ atmosphere are some of the main reasons for dielectric properties of BiFeO₃ ceramic, the lower densification and the formation of secondary phase resulted in larger number of grain boundaries and the improvement of inhomogeneity, which act as scattering center for flow of

Download English Version:

https://daneshyari.com/en/article/1461643

Download Persian Version:

https://daneshyari.com/article/1461643

<u>Daneshyari.com</u>