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Microstructure and electrical properties of (Ba_{0.98}Ca_{0.02})(Ti_{0.94}Sn_{0.06})O₃modified Bi_{0.51}Na_{0.50}TiO₃ lead-free ceramics

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Abstract

 $(Ba_{0.98}Ca_{0.02})(Ti_{0.94}Sn_{0.06})O_3$ -modified $Bi_{0.51}Na_{0.5}TiO_3 [(1 - x)BNT-xBCTS]$ ceramics were prepared by the normal sintering. A stable solid solution is well formed between BNT and BCTS, and the morphotropic phase boundary of (1 - x)BNT-xBCTS ceramics is identified in the compositional range of $0.05 \le x \le 0.06$. The temperature dependence of the dielectric loss and the poling temperature dependence of the d_{33} value are used to determine the depolarization value (T_d) . The T_d value of these ceramics gradually decreases with increasing BCTS content, together with the gradual increase of the dielectric constant. An enhanced electrical behavior of $d_{33} \sim 170$ pC/N, $k_p \sim 32.8\%$, $P_r \sim 38.5 \mu$ C/cm², and $E_c \sim 34.3$ kV/cm is demonstrated for the ceramic with x = 0.06, which is double than that of pure BNT ceramic.

Keywords: C. Piezoelectric properties; Lead-free ceramics; Piezoelectric materials; Bi_{0.5}Na_{0.5}TiO₃; (Ba, Ca)(Ti, Sn)O₃

1. Introduction

Piezoelectric ceramics become one kind of very important smart materials because of their good dielectric, ferroelectric, pyroelectric, and piezoelectric properties, promising as a candidate material in the field of microelectronic and microelectromechanical devices. However, some environmental issues have arisen, and restricted some applications of lead-based piezoelectric ceramics because of their toxicity and high vapor pressure during processing [1]. In 1960, sodium bismuth titanate (Bi_{0.50}Na_{0.50}TiO₃, BNT) is first investigated by Smolenskii and Aganovskaya [2]. In the past of several years, considerable attention has been given to these BNT-based ceramics for the replacement of lead-based piezoelectric ceramics because of a dense microstructure and a relatively high piezoelectric behavior [3–9].

BNT has been recently considered as a promising candidate of lead-free piezoelectric ceramics because of its strong ferroelecricity, an electric field-induced strain, and a high Curie temperature of \sim 320 °C [3–9]. However, a high coercive field and a high conductivity become a serious barrier for the

development and application of BNT-based ceramics [3-9]. Some methods have been used to resolve these barriers of BNTbased ceramics. Among these methods, the formation of BNT solid solutions with other ferroelectrics is to help improve their electrical properties [10-19], for example, BNT-Ba(Cu_{1/2}W_{1/} 2)O₃ [10], BNT-K_{0.5}Bi_{0.5}TiO₃ [11], BNT-Bi_{0.5}K_{0.5}TiO₃-BaTiO₃ [12], and BNT-Bi(Mg_{2/3}Nb_{1/3})O₃ [13]. However, piezoelectric properties of BNT-based ceramics are relatively low [10–13]. Recently, Ca and Sn-modified BaTiO₃ ceramics have a high d_{33} value of >500 pC/N by constructing a phase boundary at room temperature [20,21]. It may be highly expected that an enhanced piezoelectric behavior is induced by forming the solid solution of BNT and BaTiO3 with Ca and Sn modification. However, there are few reports on the piezoelectric properties of BNT ceramics combined with (Ba_{0.98}Ca_{0.02})(- $Ti_{0.94}Sn_{0.06}O_3$ (BCTS). Moreover, the Bi excess can improve the electrical properties of BNT-based ceramics by compensating the loss of Bi during preparation [22-24]. In the present work, we tried to improve the electrical properties of BNT ceramics by introducing BCTS, and $(1 - x)Bi_{0.51}Na_{0.5}TiO_3 - x(Ba_{0.98}Ca_{0.02})$ $(Ti_{0.94}Sn_{0.06})O_3 [(1 - x)BNT - xBCTS]$ ceramics were prepared by a conventional solid reaction method. Effects of BCTS content on the electrical properties of (1 - x)BNT-xBCTSceramics were systematically investigated, and some related physical mechanisms are also studied.

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2. Experimental procedure

(1 - x)BNT-xBCTS piezoelectric ceramics with the composition of x = 0, 0.02, 0.04, 0.05, 0.06, 0.07, 0.08, and 0.10 were prepared by the solid state reaction route. $BaCO_3$ (99%), CaCO₃ (99.9%), TiO₂ (99%), SnO₂ (99%), Bi₂O₃ (99%), and Na₂CO₃ (99.8%) were used as raw materials of this work. These powders weighed according to the chemical formula of (1 - x)BNT - xBCTS materials. These raw powders were ball mixed with ZrO₂ balls for 24 h using ethanol as the medium. These mixtures were calcined at \sim 850 °C for 6 h, and then these calcined powders were mixed with a polyvinyl alcohol (PVA) binder solution and compacted into disk samples with a diameter of ~ 1.0 cm and a thickness of $\sim 1.0-1.3$ mm. All ceramics were sintered at \sim 1160 °C for 2 h in air after burning out the PVA binder at \sim 850 °C for 2 h. Silver pastes were fired at \sim 700 °C for 10 min on both sides of these samples as electrodes for electrical measurements. All samples were poled at ~25-180 °C in a silicone oil bath under a dc field of \sim 5.0 kV/mm for 20 min. The phase structure of these ceramics was measured by using X-ray diffraction (XRD) (DX1000, PR China). Scanning electron microscopy (SEM) was employed to study the surface morphologies of these ceramics. The dielectric behavior as a function of the measurement temperature of these ceramics was measured by using an LCR meter (HP 4980, Agilent, USA), and their piezoelectric constant d_{33} of the ceramics was measured using a piezo- d_{33} meter (ZJ-3A, China). The polarization versus electric field (P-E) hysteresis loops of the ceramics were measured using a Radiant Precision Workstation (USA).

3. Results and discussion

Fig. 1(a) plots the XRD patterns of (1 - x)BNT-xBCTS ceramics. A pure phase is observed for these ceramics, confirming that there is a stable solid solution between BNT and BCTS. Fig. 1(b) shows the expanded XRD patterns in the 2θ range of 45–48° of (1 - x)BNT-xBCTS ceramics. A single

peak of (2 0 2) is observed for these ceramics with x < 0.05, confirming the involvement of a rhombohedral symmetry [10,11,13]. The $(0\ 0\ 2)/(2\ 0\ 0)$ peak splitting gradually appears with increasing BCTS content, which is corresponding to a tetragonal symmetry. As a result, the MPB of rhombohedral and tetragonal phases is identified in the compositional range of 0.05 < x < 0.06 at room temperature. Moreover, the position of the diffraction peaks is shifted to a lower angle with increasing BCTS content, suggesting an expansion of the unit cell volume because of the part substitution of Ba²⁺/Ca²⁺ ($r_{Ba^{2+}} \sim 1.61 \text{ Å}$ and $r_{\text{Ca}^{2+}} \sim 1.34$ Å) and Sn⁴⁺ ($r_{\text{Sn}^{4+}} \sim 0.69$ Å) respective for the (Bi_{0.5}Na_{0.5})²⁺ ($r_{\text{Bi}^{3+}} \sim 1.40$ Å and $r_{\text{Na}^{+}} \sim 1.39$ Å) and Ti⁴⁺ $(r_{\text{T};4+} \sim 0.605 \text{ Å})$ sites. Fig. 2(a)–(d) shows the surface morphologies of (1 - x)BNT - xBCTS ceramics with x = 0, 0.04, 0.06, and 0.08, respectively. All the ceramics exhibit a dense microstructure, independent on the composition of these ceramics. Moreover, the average grain size of (1 - x)BNTxBCTS ceramics almost keeps unchanged with increasing BCTS content, and the inhomogeneous grain size may be attributed to the inhibition of the grain growth because of the introduction of BCTS.

In this work, two methods of the temperature dependence of the dielectric loss $(\tan \delta)$ and the poling temperature dependence of the d_{33} value are conducted to characterize the depolarization temperature (T_d) . Fig. 3(a) shows the temperature dependence of the tan δ of (1 - x)BNT - xBCTSceramics with different BCTS content, measured at 100 kHz. A peak for the tan δ value is observed for all ceramics, which has been identified as the T_d . Moreover, the tan δ peak position of (1 - x)BNT-xBCTS ceramics changes with increasing BCTS content, that is, the T_d value of (1 - x)BNT-xBCTS ceramics gradually decreases with increasing BCTS content, and approaches a room temperature for these ceramics with x > 0.07 because of the introduction of BCTS [25]. This result also confirms that the T_d near room temperature results in a low piezoelectric behavior of (1 - x)BNT-xBCTS ceramics in this work, which will be discussed later. In order to further confirm the T_d value of (1 - x)BNT-xBCTS ceramics, the



Fig. 1. (a) XRD patterns and (b) expanded XRD patterns of (1 - x)BNT-xBCTS ceramics as a function of x.

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