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# Dielectric and Magnetic properties of (1 - x)BiFeO<sub>3</sub>–xBa<sub>0.8</sub> Sr<sub>0.2</sub>TiO<sub>3</sub> ceramics

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#### **Abstract**

The polycrystalline samples of (1 - x)BiFeO<sub>3</sub>–xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> (x = 0, 0.1, 0.2, 0.25, 0.3, 0.4 and x = 1) were prepared by the conventional solid state reaction method. The effect of substitution in BiFeO<sub>3</sub> by Ba<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> on the structural, dielectric and magnetic properties was investigated. X-ray diffraction study showed that these compounds crystallized at room temperature in the rhombohedral distorted perovskite structure for  $x \le 0.3$  and in cubic one for x = 0.4. As Ba<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> content increases, the dielectric permittivity increases. This work suggests also that the Ba<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> substitution can enhance the magnetic response at room temperature. A remanent magnetization  $M_r$  and a coercive magnetic field  $H_C$  of about 0.971 emu/g and 2.616 kOe, respectively were obtained in specimen with composition x = 0.1 at room temperature.

Keywords: C. Dielectric properties; C. Magnetic properties; D. Perovskite; Ceramics

#### 1. Introduction

Multiferroic materials exhibit ferroelectric and magnetic properties simultaneously within a single phase. These materials are considered to offer potential in novel devices such as multi-state memory devices, transducers and sensors [1,2]. In the conventional mechanism of ferroelectricity in perovskite, an off-centring of B-site cations, requires the B site to have an empty d orbital, which is incompatible with magnetic ordering from partially filled d shells [3]. Thus, there are very few single phase multiferroic materials in nature. Bismuth ferrite BiFeO<sub>3</sub> is one of multiferroic materials [4]. BiFeO<sub>3</sub> is ferroelectric below  $T_{\rm C} \approx 830~{\rm ^{\circ}C}$  and G-type antiferromagnetic below  $T_{\rm N} \approx 370~{\rm ^{\circ}C}$ , with cycloidal spin magnetic arrangement [5]. BiFeO<sub>3</sub> crystallizes in a rhombohedral structure at room temperature with R3c space group [6]. The preparation of

BiFeO<sub>3</sub> in the bulk form without traces of impurities has been a difficult task. Sosnowska et al. [5] observed an impurity peak of Bi<sub>2</sub>Fe<sub>4</sub>O<sub>9</sub> and Tabares-Munoz et al. [7] that of Bi<sub>46</sub>Fe<sub>2</sub>O<sub>72</sub>. Therefore, BiFeO<sub>3</sub>–ABO<sub>3</sub> solid–solution systems such as PbTiO<sub>3</sub> [8], BaTiO<sub>3</sub> [9–11] and NaNbO<sub>3</sub> [12] and SrTiO<sub>3</sub> [13], have attracted great attention as a means to increase structural stability. Furthermore, BiFeO<sub>3</sub> ceramic is characterized by high leakage, small permittivity, high dielectric losses and multiple thermally activated relaxations in kHz range [14,15].

 $Ba_{0.8}Sr_{0.2}TiO_3$  is a prototype ferroelectric material with several excellent ferroelectric properties ( $T_C = 75$  °C, and  $\varepsilon_r \sim 11,500$ ) [16] better than  $BaTiO_3$  ( $T_C$ : 120 °C, and  $\varepsilon_r \sim 9000$ ) [17], the structure of  $Ba_{0.8}Sr_{0.2}TiO_3$  has tetragonal phase at room temperature [16], and is expected that both ferroelectricity and ferromagnetism still coexist in the compound formed when mixed with BiFeO<sub>3</sub>.

The aim of the present work is to study the structural, dielectric and magnetic properties of the (1 - x)BiFeO<sub>3</sub>–xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> ceramics with x = 0.1, 0.2, 0.3, 0.4 and 1.

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#### 2. Experimental

The (1 - x)BiFeO<sub>3</sub>-xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> (x = 0, 0.1, 0.2, 0.25,0.3, 0.4 and x = 1) ceramics were prepared using solid state reaction method. For compositions in the range  $0 \le x \le 0.4$ compositions, high purity Bi<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>, BaCO<sub>3</sub> and SrCO<sub>3</sub> powders were carefully weighed in stoichiometric proportions and thoroughly mixed in agate mortar for 2 h. Bismuth oxide was taken in 5% mole excess to compensate bismuth loss during sintering process. In order to obtain single phase samples, the powders were then pressed into discs and calcined rapidly at 600 °C for 1 h and later at 800 °C for 2 h with intermediate grindings. After calcination, samples were grounded for 2 h and pressed into pellets, then heated at 830 °C for 1 h in air with a high heating rate. For the composition x = 1. the appropriate mixture of powder was calcined at 1100 °C for 12 h. After being carefully milled, the powder was then pressed into pellets, and finally sintered at 1320 °C for 3 h.

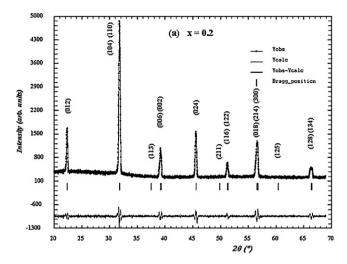
The crystal structures of the sintered samples were examined by an X-ray diffraction (XRD, XPERT-PRO) with Cu K $\alpha$  radiation ( $\lambda$  = 1.5406 Å), 0.02° scan step and 1 s/step counting time. The microstructure of the samples was examined by a scanning electron microscopy (Zeiss Ultra plus 40, Germany). The dielectric measurements were studied using LCR meter HP 4284A. The temperature and frequency ranges were between 30 and 600 °C and 100–1000 kHz, respectively. The magnetization hysteresis (M-H) loop was performed using vibrating sample magnetometer (NanoMOKE2 of Germany) superconducting quantum interference device (SQUID) at room temperature.

#### 3. Results and discussion

#### 3.1. Phase purity and microstructures

The X-ray diffraction (XRD) patterns were analyzed to confirm the phase purity, the symmetry and to calculate the lattice parameters for all compositions of the solid solution (1-x)BiFeO<sub>3</sub>–xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub>. A profile matching of the XRD spectra was made using the "Fullproof" software [18].

The analysis of the XRD patterns reveals that all the samples exhibit single phase characteristics with no trace of other impurity phases (e.g., Bi<sub>2</sub>Fe<sub>4</sub>O<sub>9</sub>, Bi<sub>46</sub>Fe<sub>2</sub>O<sub>72</sub>, etc.). The rhombohedral distorted perovskite structure of BiFeO<sub>3</sub> with R3c space group is conserved as xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> content increases up to x = 0.3 and at x = 0.4, it transforms into cubic structure (Pm3m). Fig. 1a and b shows the X-ray diffraction patterns of  $(BiFeO_3)_{0.8}$ - $(Ba_{0.8}Sr_{0.2}TiO_3)_{0.2}$  and  $(BiFeO_3)_{0.6}$ -(Ba<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub>)<sub>0.4</sub> ceramics respectively as examples. A similar phase transition from rhombohedral to cubic was observed at x = 0.3 for  $(1 - x)BiFeO_3 - xBaTiO_3$  [9,10]. The structure and lattice parameters of different compositions in the system (1 - x)BiFeO<sub>3</sub>-xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> are listed in Table 1. The lattice constant is not increasing monotonically with increase in x, but shows fluctuation. The ionic radii of  $Ba^{2+}(r(Ba^{2+}) = 1.36 \text{ Å})$  and  $Sr^{2+}$   $(r(Sr^{2+}) = 1.18 \text{ Å})$  are larger then that of Bi<sup>3+</sup>  $(r(Bi^{3+}) = 1.03 \text{ Å})$ , while the radius of  $Ti^{4+}$   $(r(Ti^{4+}) = 0.605 \text{ Å})$ is smaller than that of Fe<sup>3+</sup>  $(r(\text{Fe}^{3+}) = 0.645 \text{ Å})$  [19]. So



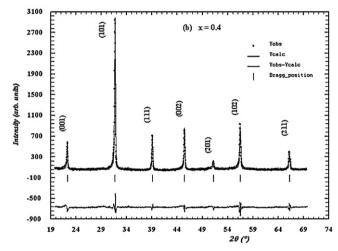


Fig. 1. XRD patterns for a rhombohedral and cubic ceramics samples of (1 - x)BiFeO<sub>3</sub>-xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> with (a) x = 0.2 and (b) x = 0.4.

substitution of Ba and Sr in A site will result in an increase in crystal lattice constant while it is the reverse for the Ti substitution on site B. So, incorporation of  $Ba_{0.8}Sr_{0.2}TiO_3$  is responsible for the fluctuation of the lattice constant.

The SEM micrographs of  $(1-x)BiFeO_3$ – $xBa_{0.8}Sr_{0.2}TiO_3$  samples with different compositions are shown in Fig. 2. Fig. 2a shows the surface of the specimen with composition x=0.1. The microstructure was heterogeneous with bimodal grain size distribution, consisting of large grains of  $\sim 8~\mu m$  and small grains of 3–4  $\mu m$ . We can notice also that the substitution of BiFeO<sub>3</sub> by Ba<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub> reduces the grain size and hence increases the volume fraction of grain boundaries and the

Table 1 Structure and lattice parameters of different compositions in the system (1-x)BiFeO<sub>3</sub>-xBa<sub>0.8</sub>Sr<sub>0.2</sub>TiO<sub>3</sub>.

Composition $(1 - x)$ BiFeO <sub>3</sub> – $x$ Ba <sub>0.8</sub> Sr <sub>0.2</sub> TiO <sub>3</sub>	Structure	Lattice parameters	
		a (Å)	α (°)
x = 0.1	Rhombohedral	5.638(2)	59.382(6)
x = 0.2	Rhombohedral	5.620(6)	60.157(1)
x = 0.3	Rhombohedral	5.619(7)	60.159(4)
x = 0.4	Cubic	3.987(9)	-

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