Effect of cesium and cerium substitution on the dielectric properties of CaCu$_3$Ti$_4$O$_{12}$ ceramics

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Abstract

In this study, CaCu$_3$Ti$_4$O$_{12}$ (CCTO) ceramics were doped with cesium and cerium atoms to possibly improve the electrical properties of these widely used ceramics. In all cases, pure phase perovskites were produced where cesium doping enhanced the grain growth and cerium doping produced grain growth inhibition. The cesium doping showed an improvement in loss tangent performance, in contrast to the cerium doping which showed a negative result. A high dielectric constant >15,000 with a dielectric loss lower than 0.06 was observed for cesium 2.0 mol% doped at high frequencies. These results were related to the change in microstructure and the properties of grain boundary after doping.
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1. Introduction

The high-K ceramic CaCu$_3$Ti$_4$O$_{12}$ (CCTO) first discovered by Subramanian et al. [1] exhibits a high dielectric constant over 10,000 at room temperature with temperature independence over the temperature range of ~100–400 K [2]. For ceramic and thin-film samples at room temperature, typical values of the loss tangent are about 0.2 at 10 kHz [1,3]. The crystal lattice structure of this ceramic is composed of titanate oxide crystals arranged in a cubic structure with an Im3 space group. Since the TiO$_6$ octahedra are tilted, there is a doubling of the perovskite-like lattice parameter which creates a square planar arrangement of the oxygen around the Cu^{2+} cations [4]. Due to these unusual properties, CCTO ceramics have an electrically heterogeneous structure with mobile charged species which show a Maxwell–Wagner relaxation behavior [5]. The internal interfaces in the polycrystalline CCTO also

give rise to a polarization of the semiconducting grains and insulating grain boundaries based on the internal barrier layer capacitor (IBLC) model [6–8]. There have been numerous reports of cation substitution on the CCTO lattice structure (substitution of Co, Zr, Fe, Ni, Sc and Nb on the B-site, and substitution of La on the A-site) [8–11]. In this report, we studied cation substitution on at both the A-site and B-site of CCTO by doping cesium (Cs) and cerium (Ce). Properties of the samples were investigated and reported.

2. Experimental

The CaCu$_3$Ti$_4$O$_{12}$ powders were prepared via a conventional solid state method. For the doping Ce and Cs studies, CeO$_2$ and Cs$_2$CO$_3$ powders were added to CCTO powder at the calcination stage in concentrations of 1.0 and 2.0 mol%. The samples were calcined at 900 °C for 2 h and sintered in air at 1100 °C for 4 h. Phase formation of the samples was investigated via an X-ray diffraction technique (XRD). The lattice parameter was calculated from the positions of the (2 2 0) and (4 2 2) X-ray diffraction peaks. Density of the ceramic samples was measured using the Archimedes method.

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Microstructure of the ceramics was characterized using a SEM (JEOL JSM-5910LV). Average grain sizes of the samples were determined by the intercept count method of ASTM E112, which is based on the number of the grain boundary intersections per unit length [8]. The dielectric constant and the loss tangent were measured with an Agilent 4284A LCR meter.

3. Results and discussion

Fig. 1 illustrates the XRD patterns of CCTO doped with Cs and Ce, which are similar to that of many previous works [3–7]. All samples exhibited a pure perovskite phase with cubic and no symmetry, no contaminating of raw materials or trace of Cs or Ce oxides. The increase in the mole fraction of Cs and Ce shows no evidence of a change in symmetry in the doped ceramics. For the Cs doped samples, the XRD peaks shifted to low angles with increasing Cs concentrations (inset (a) of Fig. 1). This result suggests that Cs doping produced a change in lattice parameter (a). Values of the lattice parameter of the Cs doped samples are shown in Fig. 2, where it increased with increasing Cs concentrations. This indicates that the Cs^{1+} entered into the lattice of CCTO. Comparing the radius of Cs^{1+} ($r_{Cs^{1+}} = 1.69$) to Ca^{2+} ($r_{Ca^{2+}} = 1.34$) and Cu^{2+} ($r_{Cu^{2+}} = 0.57$), the Cs^{1+} ions produce a larger lattice spacing as a substitute to Cu^{2+}. Thus, the increase in lattice parameter for the Cs doped samples (or the XRD peaks shifted in inset (a) of Fig. 1 is the result of substituting Ca^{2+} with Cs^{1+} [9,10]. For the Ce doping, the valance states of this ion are +3 and +4 with ionic radii of 1.06 and 0.97 Å, respectively. Since the ionic radii for Ca and Ti sites are 1.34 and 0.68 Å, respectively, and the lattice parameter for the Ce doped samples remained unchanged, this suggests that either the Ce ions did not enter into the lattices or did not change the average lattice spacing where they did enter. This evidence is similar to the work done by Kwon et al. [12].

The density and shrinkage values versus doping concentration of Ce and Cs are shown in the inset of Fig. 2. The general trend indicates that doping slightly increased the density and shrinkage value. However, the Ce doped samples showed a higher increase. This result indicates that Ce doping helped to

Fig. 2. Lattice constant of doped CCTO ceramics as a function of Cs and Ce concentrations. Inset shows density and shrinkage of doped CCTO ceramics versus Cs and Ce concentrations.

improve the densification of the doped CCTO ceramics. The microstructures of the doped samples are shown in Fig. 3. For the Cs doping, average grain size increased from ~20.4 μm for unmodified CCTO (Fig. 3(a)) to ~28.9 μm for the 2.0 mol% sample (Fig. 3(b)). In contrast to the Ce doping, a notable decrease in grain size was observed where the 2.0 mol% Ce sample had an average grain size of only 1.9 μm (Fig. 3(c)). This result implies that Cs doping promoted grain growth, while Ce doping inhibited grain growth. Furthermore, the reduced grain growth seen with the Ce ions gave rise to a segregation of the doping oxide which formed small scale secondary phases at the grain boundaries without being detected by XRD [10].

Plots of the dielectric properties as a function of temperature (Fig. 4(a)) for the pure CCTO ceramic (100 Hz–500 kHz) illustrated a high dielectric constant with temperature stability over the temperature range of ~27–60 °C of the pure CCTO ceramics, while a tangent loss was less than 0.01 over a range of 27–70 °C at 10 kHz (Fig. 5(a)). The dielectric constants as a function of temperature of the Cs (Fig. 4(b)) and Ce (Fig. 4(c)) doped CCTO ceramics (100 Hz–500 kHz), indicates that
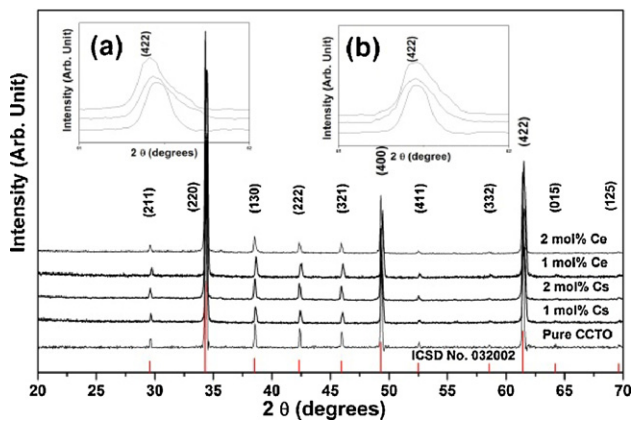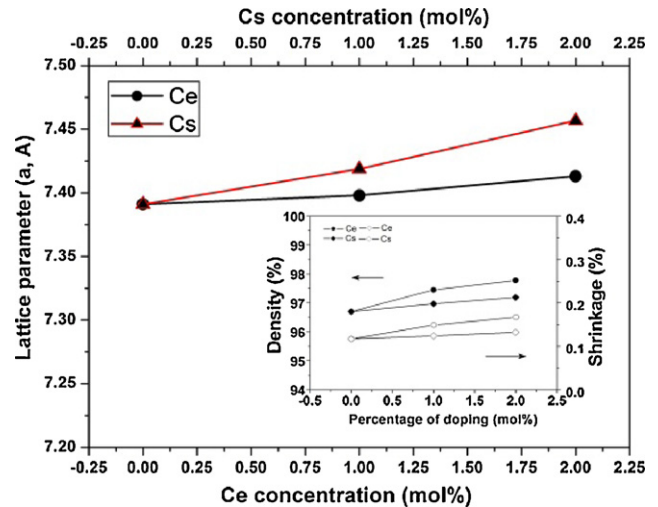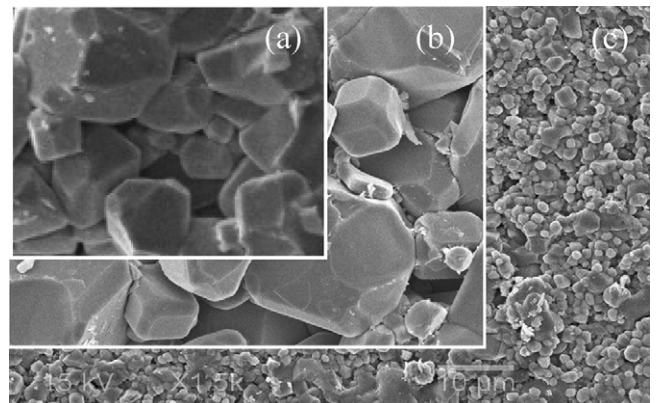

Fig. 1. XRD patterns of pure and doped CCTO ceramics. Insets (a) and (b) show XRD peaks of (4 2 2) planes for Cs and Ce doped CCTO ceramics, respectively.

Fig. 3. Fracture surfaces of selected samples: (a) Cs 2.0 mol% doped CCTO and (b) Ce 2.0 mol% doped CCTO.