

Formation, densification, and selected mechanical properties of hot pressed Al_4SiC_4 , Al_4SiC_4 with 30 vol.% WC, and Al_4SiC_4 with 30 vol.% TiC

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Abstract

Powders of Al_4C_3 and SiC were combined by high-energy milling to produce Al_4SiC_4 , Al_4SiC_4 + 30 vol.% TiC, and Al_4SiC_4 + 30 vol.% WC. Five different temperatures were used to hot press the constituents. XRD, SEM, relative density, and hardness measurements showed that formation of single-phase Al_4SiC_4 occurred at 1450 °C and full densification (99%) was achieved at 1500 °C. Both of these temperatures are lower than previously reported. Adding TiC and WC increases hardness, while WC improves densification (99.5%).

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1. Introduction

Al_4SiC_4 is a low density (3.03 g/cm³), high melting temperature (>2000 °C) compound, characterized by superior oxidation resistance, and high compressive strength [1–5]. These desirable properties motivated several investigators to determine the material's high-temperature strength, thermal conductivity, temperature dependence of linear thermal expansion coefficient, heat capacity from 5.26 to 1047 K, temperature dependence of electrical resistivity, and equation of state [3,6–9]. Additional work has been performed on the synthesis, densification, microstructure, and mechanical properties of Al_4SiC_4 [5,10–16]. Some work has also been performed on the effects of C, AlN, and SiC additions to Al_4SiC_4 [17,18].

In all earlier work reported on Al_4SiC_4 , the sintering temperature was above 1600 °C, and many of these trials involved other phases mixed with the Al_4SiC_4 . In this study, we attempted to obtain high-density Al_4SiC_4 at lower sintering temperatures by hot pressing and by adding two different materials, WC and TiC, to the Al_4SiC_4 .

2. Experimental

2.1. Preparation of the starting powders

The Al_4SiC_4 starting powder for this study was prepared by mixing high-purity Al_4C_3 and SiC (99.8% purity) powders in a He-atmosphere. A Spex-8000 mixer/mill was used to comminute the powders in sealed, hardened steel vials with chrome steel milling media. The constituents were milled for 6 h, a milling time found to be suitable in earlier studies on processing TiB_2 – ZrB_2 and TiB_2 [19,20]. For some samples, the mechanically alloyed Al_4SiC_4 powder was then mixed with 30 vol.% TiC or 30 vol.% WC and milled for an additional 30 min. The second-phase additions were 99.5% pure, 1 μm average size for WC and 2 μm for TiC. All powder handling occurred in a He-atmosphere glove box to minimize oxygen contamination.

2.2. Consolidation

Powder consolidation was performed in a 75-ton Centorr hot press using graphite dies under a flowing argon atmosphere. All samples were pressed for 60 min at a pressure of 106 MPa. The dies were lined with boron nitride and graphite sheet as lubricants. Al_4SiC_4 samples were pressed at five different temperatures, 1200 °C, 1300 °C, 1400 °C, 1450 °C, and

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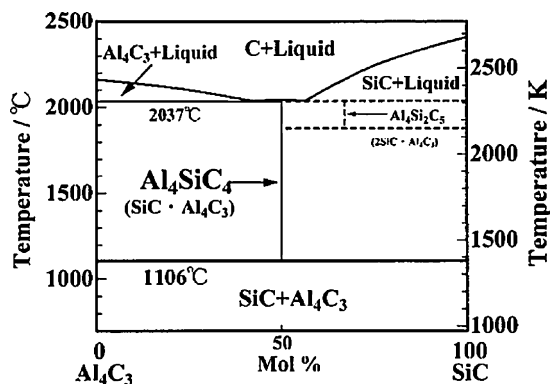


Fig. 1. The phase diagram for SiC–Al₄C₃ system [14].

1500 °C. Al₄SiC₄ + 30% WC and Al₄SiC₄ + 30% TiC specimens were pressed only at 1500 °C (Fig. 1).

2.3. Evaluation

XRD was used to identify the formation of Al₄SiC₄ phase in all specimens. The hot-pressed microstructures were analyzed by SEM and EDS. Density was determined by the Archimedes displacement method; the sintered specimen was weighed in air, and the sample was then lowered into water to measure its weight in water. Then the relative density of the sintered materials was calculated according to the following formula: Relative density (*D*) % = ρ_s/ρ_t where ρ_s is the sintered density (actual density), and ρ_t the theoretical density respectively. Hardness was measured at a load of 2000 g by Vickers microindentation with a Wilson-Tukon 2100B microhardness tester equipped with CCD image enhancement capability. The hardness values of the investigated materials were measured as the average of ten readings along the cross section surface of polished specimens.

3. Results and discussion

3.1. Formation, densification, and XRD analysis of the samples

Al₄SiC₄ can be obtained from the following reaction [21]. $\text{SiC} + \text{Al}_4\text{C}_3 \rightarrow \text{Al}_4\text{SiC}_4$ The standard Gibbs energy of formation for this reaction changes from positive to negative at 1106 °C. Therefore, formation of the ternary compound is favored at temperatures above 1106 °C, as can be seen from the phase diagram for SiC–Al₄C₃ system. Actually, the detection of Al₄SiC₄ starts at 1200 °C so this temperature was used as the first in a series of sintering temperatures [14,2].

The degree of densification achieved during hot-pressing of powders relates to several variables: powder particle size, powder handling atmosphere, milling conditions, sintering temperature, and hot pressing pressure. For example, a mixture of particle sizes can improve the fill ratio of a pressed compact. Also, milling under an inert atmosphere produces surfaces nearly free of oxides, which have high activity and are more readily sintered. High pressure during sintering generally increases densification [22–24].

Fig. 2 shows the XRD patterns of the Al₄SiC₄ sintered samples hot pressed for 1 h at five different temperatures (1200 °C, 1300 °C, 1400 °C, 1450 °C, and 1500 °C). Fig. 3 shows the XRD patterns for Al₄SiC₄, Al₄SiC₄ + 30% TiC, and Al₄SiC₄ + 30% WC samples sintered at 1500 °C for 1 h.

As shown in Fig. 2, the XRD patterns of the powders sintered at 1200 °C and 1300 °C clearly show the presence of four phases, SiC, Al₄C₃, the Al₄SiC₄ reaction product plus Fe from milling wear debris. Prior work on related materials indicates that the Fe content may be less than 1.5% [25], the intensity of the iron peak is strong due to the higher atomic number of the iron compared to the atomic number of the Al₄SiC₄. At these two temperatures, the two starting powder phases are dominant. At 1400 °C, however, the Al₄SiC₄ phase dominates, and at 1450 °C, and 1500 °C Al₄SiC₄ is the only non-ferrous crystalline phase detected. Taken together, these patterns show that Al₄C₃ and SiC powders can be fully reacted to Al₄SiC₄ by milling 6 h and hot pressing for 1 h at 106 MPa and 1450 °C.

The XRD patterns show hexagonal crystal structures for two detected phases of Al₄SiC₄ and WC and a cubic crystal structure for the TiC. As shown in Fig. 3 middle and 3 top, the intensities of Al₄SiC₄ peaks are low relative to those of TiC and WC, presumably due to the large differences in atomic weights of the elements in these three phases. Also, fewer peaks are observed in Fig. 3 middle than in Fig. 3 bottom and 3 top. This could be explained by the negative correlation between the crystal symmetry and the number of peaks. In support of this explanation, the cubic structure, which has the highest symmetry, was found to give rise to the minimum number of peaks [26].

These results indicate that hot pressing of Al₄C₃ and SiC with TiC or WC milled powders for 1 h under a pressure of 106 MPa at 1500 °C is adequate to produce two-phase Al₄SiC₄ + TiC or Al₄SiC₄ + WC with reasonably good phase purity.

3.2. Microstructure

Most of the phases were found to be smaller than 1 μm, except for WC, where abnormal grain growth was observed (Fig. 4c). This grain growth in the presence of carbon has been

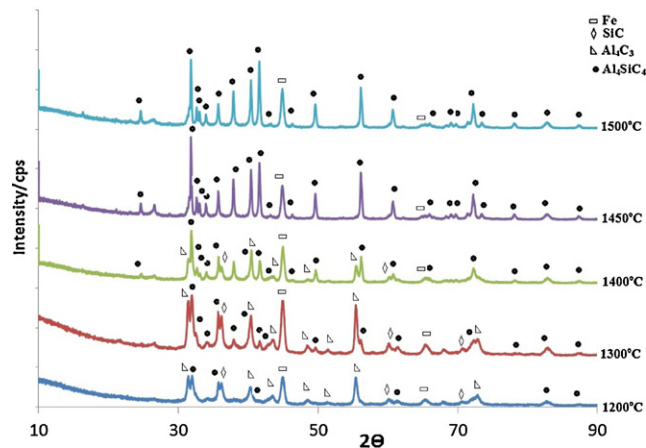


Fig. 2. XRD patterns of the various hot-pressed powders of Al₄SiC₄ at 1200 °C, 1300 °C, 1400 °C, 1450 °C, and 1500 °C for 1 h in flowing Ar. Fe is present in these samples from wear debris during milling.

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