

Available online at www.sciencedirect.com

SciVerse ScienceDirect

E**≣≋₹**S

Journal of the European Ceramic Society 32 (2012) 2705-2709

www.elsevier.com/locate/jeurceramsoc

Short communication

Spark plasma sintered β-phase silicon nitride with Sr and Ca as a sintering aid for load bearing medical applications

Maria Pettersson^{a,*}, Zohreh Pakdaman^a, Håkan Engqvist^a, Yi Liu^b, Zhijian Shen^b, Erik Östhols^c

^a Applied Material Science, Uppsala University, Ångström Laboratory, Lägerhyddsvägen 1, 75121 Uppsala, Sweden

^b Department of Materials and Environmental Chemistry, Stockholm University, Arrhenius Laboratory, 10691 Stockholm, Sweden

^c Sandvik Tooling, 126 80 Stockholm, Sweden

Available online 23 January 2012

Abstract

Due to its inherent good physical and chemical properties silicon nitride has high potential to be used for load bearing implants. However, the standard sintering additives alumina and rare earth oxides are limiting the biocompatibility of the material. The aim of the current project is to exchange the additives for more biologically beneficial additives. Spark plasma sintered silicon nitride was manufactured with strontium or calcium as sintering aids. The ability of forming high strength β -phase microstructure silicon nitride was investigated. Powders were prepared with 10 and 30 wt.% glass phase and sintered at 1600, 1650, 1700 and 1750 °C. X-ray diffraction demonstrated compositions with 10 wt.% glass phase with strontium as sintering aid to yield larger amount of β -phase. The highest amount of β -phase (96% of the crystalline structure) was obtained using SPS for strontium-doped silicon nitride at sintering temperature 1750 °C, resulting in the highest fracture toughness, 4.2 MPa m^{1/2}. © 2012 Elsevier Ltd. All rights reserved.

Keywords: Spark plasma sintering; Si₃N₄; Strontium; Calcium; Biomedical applications

1. Introduction

During the last decades silicon nitride (Si₃N₄) has been a subject of investigations for biomedical applications, especially for implants requiring high mechanical strength.^{1–3} In load bearing applications such as hip and knee joints, the property profile that Si₃N₄ can offer, i.e. high wear resistance, low friction, high fracture toughness, low density and non-toxic chemical composition, has a great potential compared to current implant materials. In particular β-phase Si₃N₄ can obtain high fracture toughness, due to that it easily grows elongated grains, which can increase the material crack propagation resistance.^{4,5} In addition to the mechanical properties, the chemical elements, silicon and nitrogen, are promising for use in biological applications.^{6,7} Silicon nitride ceramics are today used as spinal implants, and recently a femoral head was implanted in a hip replacement.^{3,8}

To obtain a dense Si_3N_4 , sintering aids are used (to enable liquid phase sintering), since pure Si_3N_4 dissociates before

* Corresponding author. *E-mail address:* maria.pettersson@angstrom.uu.se (M. Pettersson).

0955-2219/\$ - see front matter © 2012 Elsevier Ltd. All rights reserved. doi:10.1016/j.jeurceramsoc.2011.12.027 complete densification.^{4,9} Commonly, aluminum oxide and various rare earth or lanthanide oxides, such as yttrium oxide and magnesia, are used. In this study two combinations of sintering aids are investigated, Si_3N_4 plus strontium (Sr) and Si_3N_4 plus calcium (Ca). Both sintering aids are of interest for biomedical applications, Sr for its ability to increase bone formation, and decrease bone resorption^{10,11} and Ca because it is the main element in hydroxyapatite and has successfully been used as a sintering aid for SiAION previously.^{12–14} A comparatively large amount of sintering aid was used, 10 and 30 wt.%, with the aim of acquiring some of the properties of bioglass, since bioglass has a high bone bioactivity.¹⁵

Spark plasma sintering (SPS) was used to densify the Si_3N_4 based materials. An advantage with SPS, compared to for example hot pressing, is the short sintering times. Another advantage with SPS compared to conventional sintering methods, is that lower temperatures can be used for sintering, which, in combination with the short sintering times, allow for smaller grain sizes.^{16,17}

During the sintering process, sintering aids react with added silica (SiO₂) and creates an oxide melt. A transformation from α -to β -Si₃N₄ can take place, at high temperature α -phase dissolves

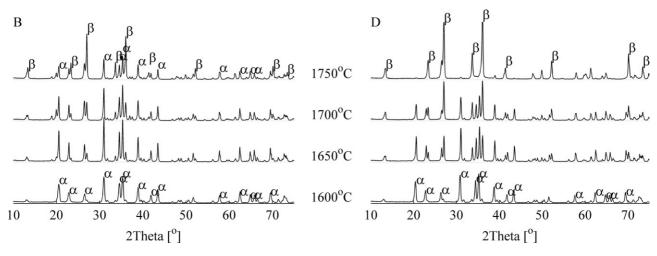


Fig. 1. XRD-spectra for sintered samples from powders B and D, at sintering temperatures 1600–1750 °C.

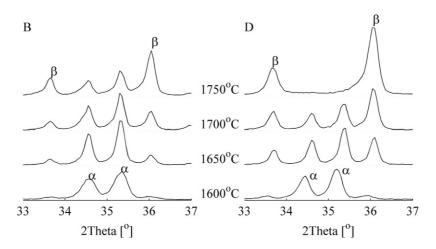


Fig. 2. XRD-spectra for sintered samples from powders B and D show more narrow spectra from Fig. 1. It demonstrates how the α - and β -phase changes with increased sintering temperatures, from 1600 °C to 1750 °C.

in the liquid phase, and $\beta\mbox{-phase}$ grains are precipitated according to

$$[\alpha-\mathrm{Si}_3\mathrm{N}_4] + [\mathrm{SiO}_2] + [\mathrm{M}_x\mathrm{O}_y]$$

$$\rightarrow [\beta-\mathrm{Si}_3\mathrm{N}_4] + [\mathrm{M}-\mathrm{Si}-\mathrm{O}-\mathrm{N}], \qquad (1)$$

where M is the metal used as sintering aid, here Sr or Ca. Around the β grains the liquid phase solidifies into M–Si–O–N glass phase. 18

The aim in this study is to form dense β -Si₃N₄ with high fracture toughness using biologically resorbable sintering aids. In this short communication the microstructure and mechanical

Table 1
Powder composition and theoretical amount of glass phase.

properties are investigated. The biocompatibility and bioactivity are subject for further research.

2. Experimental

Powders were prepared and evaluated according to the following procedure. A slurry of α -Si₃N₄ (>95%, SN-E10, UBE America Inc., USA), dispersing agent (Dispex A40), SiO₂ (Quartz, Carl Roth GmbH + Co. KG, Germany), CaCO₃ (Merck KGaA, Germany) or SrO₃ (Merck KGaA, Germany) and deionized water were milled in a vibration mill, with Si₃N₄ based milling media. After wet sieving, the slurries were freeze granulated. The slurry was sprayed into liquid nitrogen, and the

Powder	α -Si ₃ N ₄ [wt.%]	SiO ₂ [wt.%]	CaCO ₃ [wt.%]	SrCO ₃ [wt.%]	Glass phase [%]
A	69	26	5	0	30
В	89	9	2	0	10
С	65	21	0	14	30
D	89	7	0	4	10

Download English Version:

https://daneshyari.com/en/article/1475068

Download Persian Version:

https://daneshyari.com/article/1475068

Daneshyari.com