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Journal of the European Ceramic Society 31 (2011) 183–188

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Effect of processing on the microstructure and crystalline phase composition of wood derived porous SiC ceramics

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Received 28 October 2009; received in revised form 2 August 2010; accepted 7 August 2010 Available online 15 September 2010

Abstract

Pine (*Pinus silvestris*) wood samples were dried and impregnated with a SiO_2 sol from a sol-gel process. The impregnation involved a two step process in a custom-made apparatus. Impregnated samples were dried and pyrolised at $500\,^{\circ}$ C under an oxygen-free atmosphere. SiC synthesis was performed in a high-temperature furnace in an argon atmosphere at a temperature of $1600\,^{\circ}$ C for 2, 4 and 8 h. The samples were investigated with X-ray diffraction (XRD) and field emission scanning electron microscopy (FE-SEM). The changes in the SiC synthesis time at the maximum temperature lead to changes in the microstructure and crystalline phase composition. An increase in the synthesis time opens up the possibility to produce mainly α -SiC crystalline modification containing porous SiC ceramics.

Keywords: A. Sol-gel process; B. Whiskers; Electron microscopy D. SiC; Wood

1. Introduction

SiC ceramics with a wood-like microstructure, in which wood is used as the carbon precursor, represent a relatively new research area in the 21st century. One of the most promising applications is for solar energy absorbers in volumetric solar receivers that require porous ceramics with open porosity, excellent solar energy absoption and high thermal conductivity. It is well known that SiC ceramics, having larger grain sizes and the addition of α -SiC, produce higher thermal conductivity values

In the recent years, several technologies have been developed for making porous SiC ceramics with a wood-like microstructure. The evaluation of the previously published research, with attention to the crystalline phase composition of wood-like SiC ceramics, allows summarising the present capabilities. The use of reactive infiltration with Si-containing melts at 1500–1600 °C for 1–2 h only forms the β -SiC crystalline phase 2,3 or β -SiC and a small amount of α -SiC. The reactive silicon vapour infiltration at 1600 °C for 4–8 h only produces

the $\beta\text{-SiC}$ crystalline phase. 5,6 However, heating at $1550\,^{\circ}\text{C}$ for $4\text{--}16\,\text{h}$ forms a small amount of $\alpha\text{-SiC}$ that decreases by extending the reaction time. 7 Obtaining wood-like SiC ceramics using SiO2 sol impregnation with the following carbothermal reactions in the temperature ranges from 1300 to $1600\,^{\circ}\text{C}$ for $30\,\text{min}$ to $4\,\text{h}$, $\beta\text{-SiC}^{8-10}$ and, as reported in some publications, additionally a small amount of $\alpha\text{-SiC}^{11-16}$ (somewhere described as stacking faults of cubic lattice 17,14,18) are detected.

In the synthesis of SiC-containing ceramics via other technologies based on carbothermal reactions, the α -SiC crystalline phase is not determined even at so high-temperature conditions as 1800 °C for 30 min¹⁹ using pulse current sintering and 2000 °C for 30 min²⁰ using reaction sintering.

In the current research, the possibility of obtaining low-temperature cubic $\beta\text{-SiC}$ crystalline modification, low-temperature cubic $\beta\text{-SiC}$ and high-temperature $\alpha\text{-SiC}$ crystalline modification composite, and mainly high-temperature $\alpha\text{-SiC}$ crystalline modification containing porous SiC ceramics was shown. Besides the changes in the crystalline modification, also changes in the microstructure of the ceramics and whiskers formed in the pores of the samples were demonstrated. The thickness of the average SiC layer versus time was measured and discussed, comparing to the previous results available in the literature.

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2. Experimental procedure

2.1. Material preparation

Pine wood (*Pinus silvestris*) samples were prepared with dimensions of $20\,\mathrm{mm}\times20\,\mathrm{mm}\times5\,\mathrm{mm}$ (axial). Wood samples were placed in a self-made impregnation vessel, evacuated for 5 min at 520 mbar, and then showered with a SiO₂ sol and delayed for 5 min before increasing the pressure to atmospheric conditions. Samples with the SiO₂ sol were compressed at 30 MPa for 5 min in a custom-made hydraulic isostatic press, drained and then dried in an oven at $105\,^{\circ}\mathrm{C}$ for 24 h. Impregnation cycles were repeated 3 times to achieve the necessary SiO₂ content in the sample. Pyrolysis was performed in an oxygen-free atmosphere for 30 min at $500\,^{\circ}\mathrm{C}$ with the heating and cooling rate $120\,^{\circ}\mathrm{C/h}$.

More details are available in the previously published papers. 1,21

2.2. Synthesis of silicon carbide

The synthesis of SiC was performed in a modified high-temperature furnace (LHT 04/17, Nabertherm (Germany)). To maintain an inert atmosphere during the SiC synthesis, argon was introduced in the furnace. The temperature was raised to $1600\,^{\circ}$ C at $300\,^{\circ}$ C/h, and the samples were held for 2, 4 and 8 h.

2.3. Characterisation

X-ray diffraction (XRD) was applied to identify the crystalline phases over a 2θ range of 15–75 in a powder X-ray diffractometer (PANalytical X'Pert Pro, The Netherlands), using Cu K α radiation at 40 kV and 30 mA.

The morphology of the microstructure was observed from the fracture surface of the samples using a field emission scanning electron microscope (FE-SEM) (Tescan Mira/LMU, Czech Republic) at 25 kV and a distance of 7 mm.

3. Results and discussion

3.1. XRD analysis

Fig. 1 shows XRD patterns of pine wood, impregnated 3 times, and held heated at $1600\,^{\circ}$ C. Holding the samples for 2 h at $1600\,^{\circ}$ C (Fig. 1a), mainly a 3C-SiC crystalline modification is formed, with maxima at 2θ 35.84°, 41.85°, 60.23° and 72.00°. The maximum at 2θ 33.85° (marked with *) is characteristic for the stacking faults in the 3C-SiC crystalline lattice. The maximum at 2θ 22.16° represents SiO₂ cristobalite and suggests an incomplete reaction between SiO₂ and biocarbon (C_B) at $1600\,^{\circ}$ C for 2 h. Cristobalite is no longer present after heating at $1600\,^{\circ}$ C for 4 h (Fig. 1b), and the sample represents a 3C-SiC crystalline phase. A small amount of 2H-SiC is present, as is revealed by the diffraction peaks at 2θ 38.45° and 64.85°. Upon the heating for 8 h, the 12R-SiC crystalline phase begins to form (Fig. 1c). Then the phase composition includes 3C-SiC, 2H-SiC and 12R-SiC. The maxima characteristic for the 12R-SiC crystalline

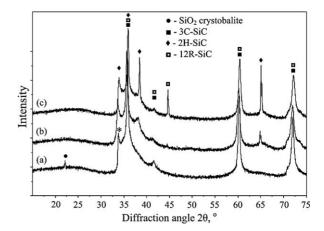


Fig. 1. XRD diffraction patterns of samples heated to $1600\,^{\circ}\text{C}$ for (a) $2\,\text{h}$, (b) $4\,\text{h}$ and (c) $8\,\text{h}$.

talline phase are located at 2θ 35.84°, 41.73° , 44.70° , 60.39° , and 72.17° . The diffraction peaks from the different phases overlap and are difficult to differentiate. It can be concluded that the main components are 2H-SiC and 12R-SiC as can be seen from the 80% intensity of the 2θ 38.45° and the 50% intensity of the 2θ 64.85° diffraction maxima for 2H-SiC, and from the 10% intensity of the 2θ 44.70° diffraction maximum for 12R-SiC. The amount of 3C-SiC decreases with time as shown by the shift of the diffraction maximum from 2θ 33.85° to 33.96° to form 2H-SiC. In all cases, the broad maxima indicate the formation of fine crystals.

A great amount of 2H-SiC and 12R-SiC is formed from the SiO_2 sol impregnation into the wood, as compared to the results reported in other publications. This can be explained by the close contact between the SiO_2 and biocarbon (C_B) in the pore walls from the vacuum/pressure impregnation, followed by the pyrolysis of the obtained SiO_2 gel/wood composite. In this case, a significant amount of SiO_2 was introduced into the wood cell wall, and SiO_2 was enclosed in the amorphous C_B matrix during pyrolysis. The formation of the high-temperature SiC crystalline phase at $1600\,^{\circ}C$ can be facilitated also by the presence of different impurities (e.g. Na, K, Ca, etc.) in the C_B matrix, which are retained in the wood derived SiC ceramics.

3.2. FE-SEM analysis

FE-SEM micrographs show the effect of holding time at $1600\,^{\circ}\text{C}$ on the microstructure. The samples held at $1600\,^{\circ}\text{C}$ for 2 h show a SiC layer on the pore walls (Fig. 2b). The non-reacted C_B is found deeper in the pore walls, between the SiC layers. The transverse direction shows a tubular SiC structure (Fig. 2c) and porous walls (Fig. 2d) with interconnected pores.

The grain size in the walls ranges from 100 to 200 nm.

The porous SiC ceramics formed at 1600 °C for 4 h have rounded tubular parallel pores, corresponding to the tracheids in the wood structure (Fig. 3(a)).

The microstructure of the ceramics has characteristics, similar to those of the wood microstructure. Openings are seen at some sites of the pore walls, formed from the pits present in pine wood.²³ The longitudinal pore size in SiC is $10-20 \,\mu m$,

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