



# Oxygen transfer model development based on activated sludge and clean water in diffused aerated cylindrical tanks



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## HIGHLIGHTS

- The volumetric oxygen mass transfer  $k_L a$  was measured under different operational conditions.
- Experiments in clean water and with activated sludge were done.
- The experimental results were used to develop a high fit empirical model.
- The airflow rate was the main factor affecting the  $k_L a$ .

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## ABSTRACT

The oxygen mass transfer  $k_L a$  is generally studied under non-reactive conditions, leaving out the most fundamental operational condition in activated sludge processes (ASPs). Existing oxygen transfer models, used in wastewater treatment plant design and optimizations, have therefore a major shortcoming. More accurate  $k_L a$  models lead to improved system analysis and knowledge. This work studied the volumetric oxygen mass transfer  $k_L a$  in an ASP, under varying operational conditions. An empirical correlation for  $k_L a$  versus nine studied variables (tank volume ( $V_t$ ), height ( $H_t$ ), diameter ( $D_t$ ), surface area ( $A_t$ ), airflow rate ( $Q_a$ ), diffusers surface area ( $A_d$ ) and depth ( $h_d$ ), bubble size ( $d_b$ ) and dynamic viscosity ( $\mu$ ) for clean water ( $k_{L,a,CW}$ ) and for activated sludge ( $k_{L,a,AS}$ ) in a diffused aerated cylindrical batch reactor is created. The experimental results were used to develop a high fit empirical model for  $k_{L,a,AS}$  ( $R^2 = 0.96$ ) and  $k_{L,a,CW}$  ( $R^2 = 0.95$ ). The following equations were obtained ( $k_L a$  in  $s^{-1}$ ):

$$\frac{D_t^2 k_{L,a,CW}}{D} = 0.030 Re^{1.718} Fr^{-0.709} \left(\frac{d_b}{h_d}\right)^{-0.291} \left(\frac{H_t}{D_t}\right)^{-0.554} \left(\frac{A_d}{A_t}\right)^{0.135} \left(\frac{D_t}{h_d}\right)^{0.321} \left(\frac{H_t}{h_d}\right)^{0.086} \left(\frac{V_t}{A_d^{1.5}}\right)^{-0.017}$$

$$\frac{D_t^2 k_{L,a,AS}}{D} = 0.060 Re^{1.906} Fr^{-0.631} \left(\frac{d_b}{h_d}\right)^{-0.23} \left(\frac{H_t}{D_t}\right)^{-0.120} \left(\frac{A_d}{A_t}\right)^{0.326} \left(\frac{D_t}{h_d}\right)^{0.164} \left(\frac{H_t}{h_d}\right)^{0.173} \left(\frac{V_t}{A_d^{1.5}}\right)^{-0.01}$$

The Reynolds ( $Re = \frac{v_l}{\nu} = \frac{Q_a \rho}{D_t \mu}$ ) and the (adapted) Froude number ( $Fr = \frac{v}{\sqrt{lg}} = \frac{Q_a}{\sqrt{D_t^3 g}}$ ) were used. The coefficients for clean water and activated sludge varied up to 66% for the same base model but show similar trends and effects for different hydrodynamic, physicochemical and geometrical parameters. The airflow rate was the main factor affecting both  $k_{L,a,AS}$  and  $k_{L,a,CW}$ . Next were diffusers depth and bubble size. Airflow rate and diffusers surface area had a significantly larger impact in the presence of biomass, since it promotes bubble distribution, mixing of the solution and an improved oxygen transfer, therefore demonstrating the need for an adapted model for ASPs.

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## 1. Introduction

The oxygen mass transfer,  $k_L a$ , in wastewater is often studied under nonreactive conditions, even though it is known that biomass significantly impacts the oxygen transfer in activated sludge systems (ASPs) [1–3]. Existing oxygen transfer models, used among

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## Nomenclature

$\alpha$	alpha factor (–)	$k_{La}$	volumetric mass transfer coefficient ( $\text{h}^{-1}$ , unless otherwise specified)
$\mu$	dynamic viscosity (kg/m/s)	$k_{LaAS}$	volumetric mass transfer coefficient in activated sludge tank ( $\text{h}^{-1}$ , unless otherwise specified)
$\nu$	kinematic viscosity ( $\text{m}^2/\text{s}$ )	$k_{LaCW}$	volumetric mass transfer coefficient in clean water tank ( $\text{h}^{-1}$ , unless otherwise specified)
$\rho$	density ( $\text{kg}/\text{m}^3$ )	$L$	length of the tank (m)
$\theta$	temperature correction factor (–)	MLSS	mixed liquor suspended solids (mg/l)
A/A/O	anaerobic–anoxic–oxic	OTR	oxygen transfer rate ( $\text{mgO}_2/\text{l/d}$ )
$A_d$	total coverage area of the diffusers ( $\text{m}^2$ )	OUR	oxygen up-take rate by microorganisms ( $\text{mgO}_2/\text{l/d}$ )
$A_t$	total area of the tank ( $\text{m}^2$ )	$Q$	volumetric wastewater flow rate ( $\text{m}^3/\text{s}$ unless otherwise stated)
AS(P)	activated sludge (process)	$Q_a$	airflow rate ( $\text{m}^3/\text{s}$ )
$B_t$	width of the tank (m)	SRT	sludge retention time (d)
$C_{O_2}$	concentration of dissolved oxygen ( $\text{mgO}_2/\text{l}$ )	$T$	temperature ( $^\circ\text{C}$ )
$C_{O_2}^*$	oxygen saturation concentration ( $\text{mgO}_2/\text{l}$ )	TIC	Theil's inequality coefficient (–)
$D$	diffusion coefficient ( $\text{m}^2/\text{s}$ )	$V$	working volume reactor ( $\text{m}^3$ )
$d_b$	bubble diameter (m)	$y_{\text{calc}}$	calculated values
$d_{eq}$	equivalent bubble diameter (m)	$y_{\text{exp}}$	data points obtained through the experiments
$D_t$	tank diameter (m)		
DO	dissolved oxygen (mg/l)		
$Fr$	Froude number (–)		
$g$	acceleration due to gravity ( $=9.8 \text{ m/s}^2$ )		
$H_t$	height of tank (m)		
$h_d$	diffuser submergence (m)		
HRT	hydraulic retention time (h)		

**Table 1**  
Empirical correlations for  $k_{La}$  prediction for diffused aeration systems.

Empirical correlation	Reference
$\frac{L^2 k_{La}}{D} = 0.033 Re^{1.46} Fr^{-0.49} \left(\frac{d_b}{h}\right)^{-0.73} \left(\frac{H_t}{L}\right)^{-1.77} \left(\frac{A_d}{A_t}\right)^{-0.24}$	[12]
$k_{La} = 6.86 d_B^{-1.3} \nu^{0.93} \left(\frac{W}{H}\right)^{-0.49} \left(\frac{H_t}{L}\right)^{1.63} \left(\frac{p_h^{0.074} - p_b^{0.44}}{p_h - 1}\right)$	[34]
$k_{La} = 49 Re \left(\frac{\nu}{g}\right)^{0.5} \left(\frac{D}{D_t}\right) \left(\frac{A_d}{h_d^2}\right)^{0.72}$	[10]
$\frac{k_{La}}{\nu} \left(\frac{\nu^2}{g}\right)^{1/3} = 7.77 \times 10^{-5} \left(\frac{A_d}{A_t}\right)^{0.24} \left(\frac{A_b}{A_d}\right)^{-0.15} \left(\frac{D_t}{h}\right)^{0.13}$	[6]

others in wastewater treatment plant (WWTP) optimizations, have therefore a major shortcoming as they are not based on the most fundamental operational condition in ASPs. Often oxygen mass transfer is measured to check the performance of ASPs before start-up of the WWTP or during design of a new one. These tests are mainly done in clean water, following the ASCE [4] and NFEN [5] standard. This leads to significant inaccuracies in oxygenation performance prediction of the full-scale system, since many factors in the wastewater affect the oxygen transfer [6]. Physicochemical (solution composition, biomass, viscosity, pH, TSS, dissolved oxygen (DO), etc.), geometrical (aerator submergence, length and width of the tank, total tank area, bubble diameter, diffusers total coverage area, reactor's working volume, etc.) and dynamical parameters (airflow rate, water density, surface tension, kinematic viscosity, airflow velocity, etc.) and aerator type contribute to aeration and oxygen mass transfer all to a different extent depending on the wastewater type, treatment system and equipment used. Refining the oxygen mass transfer prediction will lead to an improved optimized system, meaning reduced costs and increased effectiveness of ASPs and even WWTPs. This remark counts particularly for medium-sized plants, as operational inspections are more difficult to accomplish systematically [6].

Diffused aeration (subsurface or submerged bubble aeration) is defined as the injection of air or oxygen enriched air under pressure below a liquid surface [7]. Air is blown into the water by means of diffusers or mechanical agitators. This contribution deals with submerged fine pore diffusers ( $d_b < 5 \text{ mm}$ ). These release air via porous media or nozzles at increased depths [8]. The fine-pore

**Table 2**  
Ranges of the operational variables and derived (non-dimensional) variables.

Variable	Name	Unit	Range or value
$Q_a$	Airflow rate	l/min	0.24–0.60
$h_d$	Diffusers depth	m	0.01–0.24
$V_t$	Tank volume	l	2.7–9.3
$H_t$	Tank height	m	0.13–0.53
$D_t$	Tank diameter	m	0.10–0.30
$A_t$	Tank surface area	$\text{m}^2$	$1.8 \times 10^{-3} - 7.4 \times 10^{-3}$
$d_b$	Bubble diameter	m	0.005
$A_d$	Diffusers surface area	$\text{m}^2$	$7 \times 10^{-4}$
MLSS	Mixed liquor Suspended Solids	mg/l	918–2543
$\mu_{CW}$	Dynamic viscosity CW	kg/ms	$1.1 \times 10^{-3} - 9.8 \times 10^{-4}$
$\mu_{AS}$	Dynamic viscosity AS	kg/ms	$1.4 \times 10^{-3} - 1.9 \times 10^{-3}$
$\rho$	Density	$\text{kg}/\text{m}^3$	998
$D$	Oxygen diffusion coefficient	$\text{m}^2/\text{s}$	$1.86 \times 10^{-9}$
$Re_{CW}$	Reynolds CW		13–105
$Re_{AS}$	Reynolds AS		14–54
$Fr$	Froude		$2.6 \times 10^{-5} - 1.1 \times 10^{-3}$
$d_b/h$			0.021–0.76
$H_t/D_t$			0.44–3.8
$A_d/A_t$			0.01–0.096
$D_t/h_d$			0.57–46
$H_t/h_d$			2–20
$V_t/A_d^{1.5}$			145–494

aerators are different from coarse bubble aerators, owing to their flow regime. The former has low interfacial gas velocities, hence induces low flow regimes at gas–liquid interfaces. Due to this flow regime (and the added decrease through scaling and fouling over time) the alpha factor ( $\alpha$ , [9]) is substantially lower compared to the wide variety of aeration systems, leading to a lowered  $k_{La}$ . In contrast coarse-bubbles produce greater velocity gradients at the gas–liquid interface [8]. However the  $k_{La}$  of coarse bubble aerators is about six times smaller in contrast to the fine bubble  $k_{La}$ . Nonetheless when demanding an equal  $k_{La}$ , 3 times more diffusers are needed for fine bubble diffusers as these have the highest  $A_d$ , but demand a lower airflow rate, hence a higher power consumption (at 10 m tank depth +40%) [10]. Diffusers are still the most

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