



# Nondegenerate n-type doping phenomenon on molybdenum disulfide (MoS<sub>2</sub>) by zinc oxide (ZnO)



Dong-Ho Kang<sup>a,1</sup>, Seong-Taek Hong<sup>a,1</sup>, Aely Oh<sup>a</sup>, Seung-Hwan Kim<sup>b</sup>, Hyun-Yong Yu<sup>b,\*\*</sup>, Jin-Hong Park<sup>a,\*</sup>

<sup>a</sup> School of Electronic and Electrical Engineering, Sungkyunkwan University, Suwon 440-746, South Korea

<sup>b</sup> School of Electrical Engineering, Korea University, Seoul 136-701, South Korea

## ARTICLE INFO

### Article history:

Received 2 September 2015

Received in revised form 13 February 2016

Accepted 19 February 2016

Available online 23 February 2016

### Keywords:

A. Semiconductor

A. Electronic materials

A. Layered compounds

D. Electrical properties

D. Electronic structure

## ABSTRACT

In this paper, we have demonstrated nondegenerate n-type doping phenomenon of MoS<sub>2</sub> by ZnO. The ZnO doping effects were systematically investigated by Raman spectroscopy and electrical/optical measurements ( $I_D$ - $V_G$  with/without exposure to 520, 655, 785, and 850 nm laser sources). The ZnO doping improved the performance parameters of MoS<sub>2</sub>-based electronics ( $I_{on}$ ,  $\mu_{FE}$ ,  $n$ ) owing to reduction of the effective barrier height between the source and the MoS<sub>2</sub> channel. We also monitored the effects of ZnO doping during exposure to air; reduction in  $\Delta V_{TH}$  of about 75% was observed after 156 h. In addition, the optoelectronic performance of the MoS<sub>2</sub> photodetector was enhanced due to the reduction of the recombination rate of photogenerated carriers caused by ZnO doping. In our results, the highest photoresponsivity (about  $3.18 \times 10^3$  A/W) and detectivity ( $5.94 \times 10^{12}$  Jones) of the ZnO-doped photodetector were observed for 520 nm laser exposure.

© 2016 Elsevier Ltd. All rights reserved.

## 1. Introduction

Transition metal dichalcogenides (TMDs) with two-dimensional (2D) semiconducting layered structure, such as molybdenum disulfide (MoS<sub>2</sub>) and tungsten diselenide (WSe<sub>2</sub>), have been considered to be extremely promising materials for future transparent, flexible and wearable electronic/optoelectronic devices because of their exceptional electrical [1–3], optoelectronic [4] and mechanical properties [5]. Because of the excellent thickness scalability of TMDs as an atomic monolayer and van der Waals epitaxial structure without dangling bonds, TMD-based electronic devices have immunity to the short-channel effects; thus, they have been contemplated for future high-performance nanoelectronic applications [2,6]. In addition, TMD materials are expected to be suitable for high-performance optoelectronic devices (i.e., photodetectors and solar cells) because of their tunable band gap (between 1.2 and 1.8 eV, depending on layer thickness) and extremely high quantum efficiency [7,8].

To further improve the performance of TMD-based electronic/optoelectronic devices, it is necessary to reduce the contact resistance of metal–semiconductor junctions and to lower the recombination rate of carriers. However, such improvements cannot be carried out by using the ion implantation technique on TMD material because of the crystal damage of the 2D structure. For these reasons, recent work has focused on the developing of less disruptive doping techniques for application to TMD devices. Recently, Fang et al. reported p- and n-type doping technique on TMD devices, which involved the use of NO<sub>2</sub> molecules and potassium, respectively [9,10]. However, both p- and n-doping techniques produced very high levels of doping, making the TMD channel layers near metallic. Thus, development of lighter doping techniques for use with TMD devices is crucial for the optimization of TMD-based electronics/optoelectronics. Lin et al. and Park et al. reported low-level doping techniques using Cs<sub>2</sub>CO<sub>3</sub> ( $\sim 1 \times 10^{11}$  cm<sup>-2</sup> on MoS<sub>2</sub>) and DNA/M-DNA ( $\sim 6.4 \times 10^{10}$  cm<sup>-2</sup> on MoS<sub>2</sub>), respectively [11,12].

Here, we also reported the nondegenerate n-type doping technique on MoS<sub>2</sub> by zinc oxide (ZnO) film coating. The negative polar surface of the ZnO film was expected to reduce the hole carrier density in the MoS<sub>2</sub> channel layer, thereby causing an n-doping effect. We investigated the ZnO doping effects on MoS<sub>2</sub> by Raman spectroscopy and electrical measurements ( $I_D$ - $V_G$  and  $I_D$ - $V_D$ ). In addition, we also monitored the  $V_{TH}$  variations of ZnO-doped MoS<sub>2</sub> devices during 156 h of exposure to air. Finally, the

\* Corresponding author at: Sungkyunkwan University, Natural Sciences Campus, Cheoncheon-dong, Jangnan-gu, South Korea.

\*\* Corresponding author.

E-mail addresses: [jhpark9@skku.edu](mailto:jhpark9@skku.edu) (J.-H. Park), [yuhkyr@korea.ac.kr](mailto:yuhkyr@korea.ac.kr) (H.-Y. Yu).

<sup>1</sup> These authors contributed equally to this work.

effects of ZnO doping upon the photoresponsivity and detectivity performance of MoS<sub>2</sub>-based optoelectronic devices were observed by means of optical measurements performed under exposure to laser sources of wavelengths 520, 655, 785, and 850 nm.

## 2. Experiments

For the fabrication of back-gated MoS<sub>2</sub> devices (transistors and photodetectors), source/drain electrode regions were patterned (channel length and width are 5  $\mu\text{m}$ ) on MoS<sub>2</sub>/SiO<sub>2</sub>/Si samples by optical lithography, followed by Ti (10 nm) and Au (50 nm) deposition via e-beam evaporation. ZnO solutions were prepared by dissolving 1.1 g of zinc acetate dihydrate [Zn(CH<sub>3</sub>COO)<sub>2</sub>·2H<sub>2</sub>O] and 0.3 mL of monoethanolamine (MEA) in 8.96 mL of 2-methoxyethanol as the solute, stabilizer and solvent, respectively. The molar ratio of MEA to zinc acetate dihydrate was maintained at 1.0 and the concentration of zinc acetate dihydrate was 0.5 M. The solution was stirred at 60 °C for 2 h in air to obtain a transparent and homogeneous solution, followed by aging at room temperature for 5 days. For doping of MoS<sub>2</sub> devices, the ZnO solutions were dropped onto MoS<sub>2</sub> devices and spin coated at 3000 rpm for 15 s. Subsequently, the ZnO-doped MoS<sub>2</sub> devices were annealed by using a thermal furnace operated at 250 °C for 15 min in air ambient. The devices were electrically analyzed by using an HP4155A semiconductor parameter analyzer, and the carrier concentration ( $n$ ), field-effect mobility ( $\mu_{\text{FE}}$ ) and threshold voltage ( $V_{\text{TH}}$ ) were calculated by using the current–voltage ( $I_{\text{D}}-V_{\text{G}}$ ) data. To investigate the optoelectronic properties of the fabricated ZnO-doped MoS<sub>2</sub> devices, the  $I_{\text{D}}-V_{\text{G}}$  measurements were performed under both dark and illuminated conditions. Here, all drain currents ( $I_{\text{D}}$ ) were normalized by channel width ( $W$ ). The light sources were set up by using a dot laser of wavelength 520, 655, 785, and 850 nm and an optical power of 6 mW/cm<sup>2</sup>. Photoresponsivity ( $R$ ) and detectivity ( $D^*$ ) were calculated by using the acquired  $I_{\text{D}}-V_{\text{G}}$  curves.

## 3. Results and discussion

To verify the doping effects of ZnO film on MoS<sub>2</sub> film, we performed Raman spectroscopy on ZnO-doped MoS<sub>2</sub> samples exfoliated on a SiO<sub>2</sub>/Si substrate. Fig. 1a shows Raman spectra of undoped/ZnO-doped MoS<sub>2</sub> samples. In MoS<sub>2</sub> samples, the two conventional peaks ( $E_{12g}$  and  $A_{1g}$ ) were observed at about 380 and 406 cm<sup>-1</sup>, attributed to the in-plane ( $E_{12g}$  peak) and out-of-plane

( $A_{1g}$  peak) vibrations of MoS<sub>2</sub> [13]. We then extracted  $E_{12g}$  and  $A_{1g}$  peak positions for each undoped/doped MoS<sub>2</sub> sample (Fig. 1b). In the case of control samples of undoped MoS<sub>2</sub>, the  $E_{12g}$  and  $A_{1g}$  peak positions were located at 380–380.5 and 405.2–405.8 cm<sup>-1</sup>, respectively. After ZnO doping, those two peaks were red-shifted (379.4–380.2 cm<sup>-1</sup> for  $E_{12g}$  and 404.6–405.3 cm<sup>-1</sup> for  $A_{1g}$ ). These phenomena indicated that ZnO causes the softening of MoS<sub>2</sub> vibrational modes and consequently causes n-type doping effects in MoS<sub>2</sub> films [14]. The Raman shifts of the two peaks were 0.3–0.6 cm<sup>-1</sup>, indicating that the ZnO treatment produced low levels of doping, as desired.

To confirm the doping effects of ZnO on MoS<sub>2</sub> once again, we fabricated MoS<sub>2</sub>-based transistors and performed the electrical measurement ( $I_{\text{D}}-V_{\text{G}}$ ). Fig. 2a shows the three-dimensional schematic illustrations of ZnO-doped MoS<sub>2</sub> back-gated transistors and an energy band diagram of the metal (Ti)-undoped/doped MoS<sub>2</sub> junctions. Because the ZnO crystal structure is asymmetric wurtzite, it has ionic and polar structures consisting of oxygen- and zinc-terminated surfaces [15]. From these effects, the hole carriers further accumulated at the interface between ZnO and MoS<sub>2</sub>, thereby causing the n-doping phenomenon on the MoS<sub>2</sub> channel layer. As shown in Fig. 2a, these negative polar surface of ZnO was expected to be reduced the effective barrier height of the Ti-MoS<sub>2</sub> junction by shifting down of the MoS<sub>2</sub> energy band at the source–MoS<sub>2</sub> junction (Schottky barrier lowering effect:  $\Phi_{\text{Control,eff}} > \Phi_{\text{ZnO,eff}}$ ), eventually influencing the tunneling of electron carriers from the source metal to MoS<sub>2</sub>. This ZnO doping mechanism is similar to that of self-assembled monolayer (SAM) [16,17] and DNA/M-DNA doping [12]. As a result, a negative shift in threshold voltage ( $V_{\text{TH}}$ ) from −3.8 to −11.6 V and an on-current enhancement from  $1.6 \times 10^{-6}$  to  $8.27 \times 10^{-6}$  A/ $\mu\text{m}$  were observed in the ZnO-doped MoS<sub>2</sub> transistors (Fig. 2b). We then calculated field-effect mobility ( $\mu_{\text{FE}}$ ) and 2D sheet concentrations ( $n$ ) of undoped/ZnO-doped MoS<sub>2</sub> devices for investigating the changes in electronic performance caused by ZnO doping. Here, the field-effect mobility and concentration were extracted by using the following equations:  $\mu_{\text{FE}} = L/(WV_{\text{D}}C_{\text{OX}}) \times (\partial I_{\text{D}}/\partial V_{\text{G}})$  and  $n = I_{\text{D}}L/qW\mu_{\text{FE}}V_{\text{D}}$ , where  $q$  is the electron charge,  $L$  and  $W$  are the length and width of the channel, respectively, and  $C_{\text{OX}}$  is the gate oxide capacitance per unit area,  $\epsilon_{\text{OX}} \times \epsilon_0/t_{\text{OX}}$ . As shown in Fig. 2(c), the field-effect mobility value was increased from 12.8 to 29.3 cm<sup>2</sup>/V-sec after ZnO doping, indicating that the ZnO doping improved the field-effect mobility by increasing the carrier injection probability. This mobility improvement phenomenon by doping has also been reported

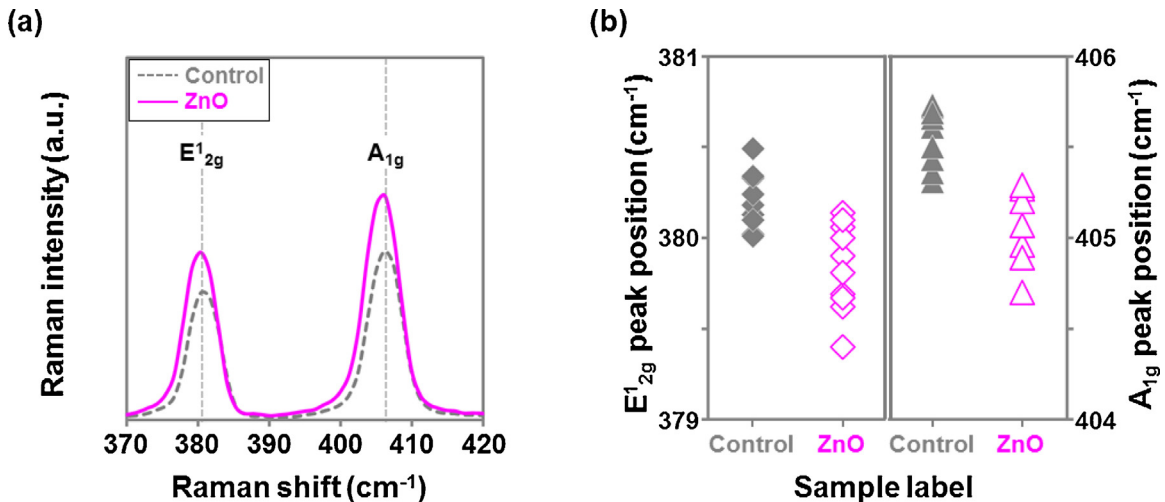


Fig. 1. (a) Raman spectra of undoped/ZnO-doped MoS<sub>2</sub> films. (b) Extracted  $E_{12g}$  and  $A_{1g}$  peak positions of undoped/ZnO-doped MoS<sub>2</sub> films.

Download English Version:

<https://daneshyari.com/en/article/1486973>

Download Persian Version:

<https://daneshyari.com/article/1486973>

[Daneshyari.com](https://daneshyari.com)