



Effect of Al₂O₃ insertion on the electrical properties of SrTiO₃ thin films: A comparison between Al₂O₃-doped SrTiO₃ and SrTiO₃/Al₂O₃/SrTiO₃ sandwich structure



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ABSTRACT

The effect of Al₂O₃ insertion on the electrical properties of SrTiO₃ films is systemically investigated in metal–insulator–metal (MIM) capacitor because SrTiO₃ films with a high dielectric constant generally suffer from high leakage current problem caused by grain boundaries and a narrow band gap. To find an effective Al₂O₃ insertion method, Al₂O₃ is inserted into SrTiO₃ thin films by two different ways. The first method is doping of Al₂O₃ in SrTiO₃ thin films and the second method is sandwiching a nanometer-thick Al₂O₃ layer between SrTiO₃ thin films. With respect to leakage blocking properties, the leakage current of Al₂O₃-doped SrTiO₃ films is effectively reduced when the SrTiO₃ film becomes amorphous by doping. In case of the SrTiO₃/Al₂O₃/SrTiO₃ structure, an Al₂O₃ layer with a thickness of more than 1.19 nm effectively acts as a leakage current blocking layer without SrTiO₃ amorphization. Moreover, the degradation of the dielectric properties of Al₂O₃-doped SrTiO₃ films is more severe, caused by structural degradation, than of SrTiO₃/Al₂O₃/SrTiO₃ structured films. Therefore, compared with Al₂O₃-doped SrTiO₃, a more than two times higher value (~45) of the dielectric constant can be obtained in the SrTiO₃/Al₂O₃/SrTiO₃ structured films with a similar leakage current density of 10⁻⁷ A/cm².

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1. Introduction

Perovskite strontium titanate (SrTiO₃) thin films have been widely investigated in view of electronic applications such as ferroelectric film capacitors [1], tunable resonant circuits [2], non-volatile memories [3], optoelectronics [4] and, especially, high dielectric capacitors in dynamic random access memories [5–7] owing to their excellent dielectric properties. Excellent dielectric properties of SrTiO₃ films can be achieved if these films have a perovskite crystal structure. However, dielectric thin films with a high crystallinity generally have a lot of grain boundaries that act as conduction paths of leakage current. Therefore, SrTiO₃ thin films with a high dielectric constant suffer from high leakage current owing to internal grain boundaries and/or cracks [8] and, moreover, the narrow band gap of SrTiO₃ (about 3.3 eV) could also cause a high

leakage current [9]. This leakage current problem challenges the practical application of SrTiO₃ films in microelectronic industry, especially with regard to MIM applications.

To improve the leakage current properties of high-*k* dielectrics, the introduction of various combinations of current barrier materials such as Al₂O₃, La₂O₃, and SiO₂ have been proposed in forms of stacked structures, intermixing structures, sandwich structures, and others [10–13]. Generally, the method of reducing the leakage current by the insertion of a current barrier material may accompany a problem of decreasing the capacitance, because leakage current barrier materials have relatively low dielectric constants. Therefore, it is necessary to investigate the variation of the electrical properties of a high-*k* material upon insertion of a current barrier material for various insertion methods. However, most of the researches have focused on the effect of an individual specific structure resulting in a lack of a comparative study between different structure combinations of high-*k* materials and current barrier material systems.

In this paper, the effect of alumina (Al₂O₃) insertion on the electrical properties of SrTiO₃ films was systemically investigated. To examine the variation of the electrical properties of Al₂O₃-added

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SrTiO₃ films, Al₂O₃ was introduced into SrTiO₃ thin films by two different ways. The first method for reducing leakage current is doping of Al₂O₃ in SrTiO₃ thin films. Since Al₂O₃ films have excellent leakage current properties, it is anticipated that Al₂O₃-doped SrTiO₃ films will exhibit a reduced leakage current. The second method is the insertion of an Al₂O₃ layer with a thickness of the order of a few nanometer in between SrTiO₃ thin films (sandwiching the Al₂O₃ layer between SrTiO₃) as a leakage current blocking layer. It is expected that the thin Al₂O₃ layer with amorphous structure might act as a leakage current blocking layer in a crystalline SrTiO₃ thin film because Al₂O₃ has a high crystallization temperature (>900 °C) and a large band gap (about 8.7 eV) with a corresponding low leakage current [14].

2. Experimental details

Al₂O₃-added SrTiO₃ thin films were deposited on a SrRuO₃ seed layer using plasma-enhanced atomic layer deposition (PEALD) adopting a supercycle concept [15]. Before the Al₂O₃-added SrTiO₃ films were deposited, the SrRuO₃ seed layer was prepared by a 2.7 nm SrO layer on a 30-nm-thick bottom electrode of RuO₂ and post-annealing, as explained in an earlier report [16]. The deposition of SrO layers on top of RuO₂/SiO₂/Si substrates by PEALD was followed by 10 min of annealing for a reaction of SrO and RuO₂ to form a SrRuO₃ layer under O₂ ambient at a temperature of 600 °C. Next, two types of Al₂O₃-added SrTiO₃ thin films were formed on the SrRuO₃ seed layer at a deposition temperature of 225 °C and a pressure of 3 Torr. The used precursors were Sr(DPM)₂ [Sr(C₁₁H₁₉O₂)₂] and TTIP [Ti(O_i-C₃H₇)₄] for SrTiO₃ deposition and TMA [Al(CH₃)₃] for Al₂O₃ addition. Oxygen plasma was used as an oxidant. Sr(DPM)₂ was dissolved in butyl acetate (0.2 M) and supplied to the reaction chamber by a liquid delivery system. TTIP vapor was carried in argon gas through a bubbler at 60 °C and TMA vapor was supplied without carrier gas to the reaction chamber while maintaining a vessel temperature of 30 °C. At a converged regime, the saturated growth rates of SrO, TiO₂, and Al₂O₃ films were 0.054, 0.036, and 0.11 nm/cycle, respectively. One supercycle for SrTiO₃ deposition was consisted of six TiO₂ cycles and seven SrO cycles in order to obtain a stoichiometric SrTiO₃ films (Sr:Ti = 1:1.03).

To investigate the impact of the structure of the leakage current barrier material on the electrical properties of Al₂O₃-added SrTiO₃ films, Al₂O₃ was inserted in two different ways, as shown Fig. 1 and Table 1. The first method is doping of Al₂O₃ in SrTiO₃ thin films (Al₂O₃-doped SrTiO₃) and the second method is sandwiching an Al₂O₃ layer between bottom and top SrTiO₃ thin films (SrTiO₃/Al₂O₃/SrTiO₃, sandwich). Applying the first method, a cycle for Al₂O₃ was periodically inserted during the SrTiO₃ deposition by varying the number of Al₂O₃ cycles (denoted by 'n') from 0 to 21. Concerning the second method, the process sequence was as follows. First, 10-nm-thick SrTiO₃ films were deposited on SrRuO₃ seed layers. Then, the Al₂O₃ layers were deposited on top of the SrTiO₃ films in a continuous manner by varying the number of deposition cycles ('n') from 5 to 20 (corresponding to a thickness range of about 0.6–2.4 nm). After that, 5-nm-thick SrTiO₃ films were deposited on the Al₂O₃ blocking layers; consequently, the sandwich structure of SrTiO₃/Al₂O₃/SrTiO₃ films was fabricated. For both methods, the number of supercycles (denoted by 'm') for the deposition of SrTiO₃ was fixed to 21 (one supercycle consisted of six TiO₂ and seven SrO cycles.) which resulted in 15-nm-thick stoichiometric SrTiO₃ films [15]. For crystallization, the as-deposited films underwent a rapid thermal annealing at 600 °C for 10 min under N₂ ambient. The film thickness was measured at a wavelength of 632.5 nm using an ellipsometer (Gaertner L116C) and X-ray diffraction (XRD, Rigaku) analysis using Cu Kα radiation (λ = 1.5405 Å) was carried out to investigate the crystal structures

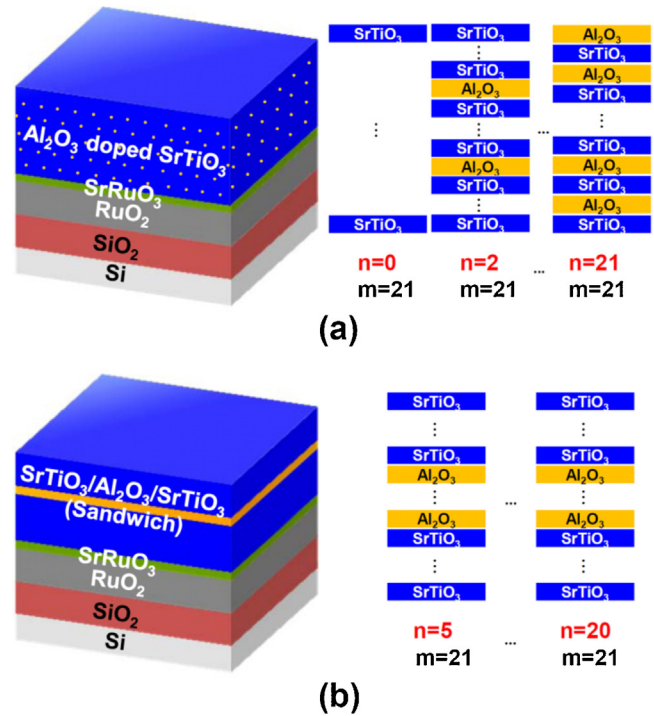


Fig. 1. Schematic diagrams of two different types of Al₂O₃-added SrTiO₃ films and the corresponding PEALD cycle sequences; (a) doping of Al₂O₃ in SrTiO₃ films (Al₂O₃-doped SrTiO₃), (b) sandwiching an Al₂O₃ layer between a bottom and a top SrTiO₃ film (SrTiO₃/Al₂O₃/SrTiO₃). The letter *n* and *m* denotes the number of Al₂O₃ cycles and SrTiO₃ supercycles, respectively.

of the Al₂O₃-added SrTiO₃ thin films. To examine the electrical properties, 100-nm-thick Pt dots deposited by sputtering with a diameter of 160 μm were used as a top electrode and the dielectric constant was measured using a C–V analyzer (Keithley 590) at a frequency of 1 MHz.

3. Results and discussion

Fig. 2 demonstrates the variation of the total thickness of Al₂O₃-added SrTiO₃ films as a function of the number of Al₂O₃ cycles while keeping fixed the number of SrTiO₃ supercycles at 21. The total thickness linearly increased with the number of Al₂O₃ cycles and the deposition rates of Al₂O₃ in SrTiO₃ films obtained from the slope of the graph were very similar for direct doping and the sandwich structure (0.121 nm/cycle for doping and 0.119 nm/cycle for the sandwich structure). This means that the deposited

Table 1

A detailed summary of PEALD cycle sequences for Al₂O₃ added SrTiO₃ thin films.

Film structure	# of Al ₂ O ₃ unitcycles	# of SrTiO ₃ supercycles	PEALD cycle sequences ^a
SrTiO ₃ thin films	<i>n</i> = 0	<i>m</i> = 21	STO*21
Al ₂ O ₃ -doped SrTiO ₃ thin films	<i>n</i> = 2	<i>m</i> = 21	(STO*7)-(AO*1)-(STO*7)-(AO*1)-(STO*7)
	<i>n</i> = 4	<i>m</i> = 21	{[(STO*4)-(AO*1)]*4}-(STO*5)
	<i>n</i> = 7	<i>m</i> = 21	{[(STO*3)-(AO*1)]*7}
	<i>n</i> = 10	<i>m</i> = 21	{[(STO*2)-(AO*1)]*10}-(STO*1)
	<i>n</i> = 21	<i>m</i> = 21	{[(STO*1)-(AO*1)]*21}
SrTiO ₃ /Al ₂ O ₃ /SrTiO ₃ sandwiched thin films	<i>n</i> = 5	<i>m</i> = 21	(STO*14)-(AO*5)-(STO*7)
	<i>n</i> = 10	<i>m</i> = 21	(STO*14)-(AO*10)-(STO*7)
	<i>n</i> = 15	<i>m</i> = 21	(STO*14)-(AO*15)-(STO*7)
	<i>n</i> = 20	<i>m</i> = 21	(STO*14)-(AO*20)-(STO*7)

^aSTO: one supercycle for SrTiO₃ deposition; AO: one cycle for Al₂O₃ deposition.

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