Optical Materials 32 (2010) 998-1001

Contents lists available at ScienceDirect

Optical Materials

journal homepage: www.elsevier.com/locate/optmat

Near-infrared quantum cutting in Ce³⁺, Yb³⁺ co-doped YBO₃ phosphors by cooperative energy transfer

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ARTICLE INFO

Article history: Received 4 August 2009 Received in revised form 18 November 2009 Accepted 31 January 2010 Available online 7 March 2010

Keywords: Quantum cutting Cooperative energy transfer Downconversion Ce³⁺, Yb³⁺ co-doped YBO₃

ABSTRACT

An efficient near-infrared (NIR) quantum cutting (QC) in Ce^{3+} , Yb^{3+} co-doped YBO₃ phosphors has been demonstrated, which involves the emission of two low-energy NIR photons (around 973 nm) from an absorbed ultra-violet (UV) photon at 358 nm via a cooperative energy transfer (CET) from Ce^{3+} to Yb^{3+} ions. Yb^{3+} concentration dependent quantum efficiency has been calculated and the maximum efficiency approaches up to 175% before reaching the critical concentration quenching threshold. The development of NIR QC Ce^{3+} , Yb^{3+} co-doped phosphors may open up a new approach to achieve high efficiency silicon-based solar cells by means of downconversion.

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1. Introduction

The high energy of the vacuum ultra-violet (VUV) photons makes it theoretically possible to generate two visible photons for a single absorbed VUV photon with quantum efficiency (QE) close to 200%. This kind of two-photon luminescence phenomenon is called quantum cutting (QC). The visible quantum cutting has received considerable attentions due to its wide application in mercury-free fluorescent lamps and plasma display panels [1].

Interestingly, quantum cutting materials also show promising applications in silicon-based solar cells [2–11]. One major energy loss in Si solar cells is thermalization, which is expected to be considerably reduced if the absorbed ultra-violet (UV)/blue photon (300 nm < λ < 500 nm) can be cut into two near-infrared (NIR) photons and then absorbed by Si (λ_{abs} < 1100 nm). This scheme can possibly be realized by using a rare earth doped quantum cutting phosphor as a downconversion convertor in front of solar cell panels.

Trupke et al. proposed a method to realize multiple electronhole pair generation per incident photon by downconversion mechanism [12]. From then on, the Tb^{3+} , Yb^{3+} co-doped QC phosphors had been investigated extensively in recent years [2–11]. Vergeer et al. reported QC by cooperative energy transfer (CET) in $Yb_xY_{0.99 - x}PO_4$:Tb³⁺ phosphors [2], which showed an optimal QE of 188% with the doping concentration of Yb^{3+} as high as 99%.

Hitherto, though the physics of CET allows for the efficient QC and the theoretical QE is high [10], the visible emission of Tb^{3+} remains strong and the NIR luminescence of Yb^{3+} is still inevitably weak. These two phenomena resulted from the following reasons. One is the actual QE is low, another is the excitation of the Tb^{3+} is inefficient for its intra-4f forbidden transitions, i.e., low absorption cross sections, typically in the order of 10^{-21} cm² [8], the other is the excitation is fixed at individual wavelengths such as 308 nm, 360 nm and 489 nm. Furthermore, the upconversion process of Tb^{3+} excited by energy transfer from Yb^{3+} (excited by NIR 980 nm laser) in Tb^{3+} , Yb^{3+} co-doped system [9], that is so-called back energy transfer, is efficient. All these reasons mentioned above will reduce the usefulness of Tb^{3+} , Yb^{3+} co-doped QC phosphors in silicon-based so-lar cells.

The Ce³⁺, Yb³⁺ co-doped QC phosphors may be superior to Tb³⁺, Yb³⁺ co-doped ones in silicon-based solar cells. First, the absorption of Ce³⁺ is originated from the allowed electric-dipole transition from the 4f ground state to the 5d excited one, which results in a very high absorption cross section in an order of 10^{-18} cm² in the UV region [8]. Second, the excitation band of Ce³⁺ is broad and the excitation wavelength of Ce³⁺ can be tuned by the host materials, which means that high QE may be obtained by choosing appropriate host materials. Moreover, the back energy





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^{0925-3467/\$ -} see front matter \odot 2010 Elsevier B.V. All rights reserved. doi:10.1016/j.optmat.2010.01.040

transfer from Yb³⁺ to Ce³⁺ is small. Though the qualities of Ce³⁺, Yb³⁺ co-doped QC phosphors exceed those of Tb³⁺, Yb³⁺ co-doped ones, the investigation on the Ce³⁺, Yb³⁺ co-doped QC phosphors is very scarce [8].

In this work, we demonstrate the NIR QC in Ce³⁺, Yb³⁺ co-doped YBO₃ phosphors by CET for the first time. The NIR emission around 973 nm, originated from Yb³⁺: ${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}$ transition, is observed under the excitation of Ce³⁺: 4f \rightarrow 5d absorption (326–377 nm), owing to the CET from one Ce³⁺ ion to two nearest neighboring Yb³⁺ ions simultaneously. The dependence of Yb³⁺ concentration on the visible and NIR emissions, decay lifetime of Ce³⁺, and QE from the QC phosphors have been investigated. The highest QE approaches 175% before reaching the critical concentration quenching threshold. These results show the development of NIR QC Ce³⁺, Yb³⁺ co-doped phosphors may open up a new approach to achieve high efficiency silicon-based solar cells by means of downconversion.

2. Experimental

Powder samples of YBO₃:Ce³⁺, Yb³⁺ were synthesized by conventional solid-state reaction method. Y₂O₃ (99.99%), Ce(NO₃)₃ (A.R.), Yb₂O₃ (99.99%), HBO₃ (A.R.) and HNO₃ (A.R.) were used as raw materials. Yb₂O₃ was first dissolved in dilute HNO₃ to prepare Yb(NO₃)₃ solution (1 M). The mixtures of Y₂O₃, HBO₃ (10% excess), Ce(NO₃)₃ (1 M) and Yb(NO₃)₃ were ground and then heated at 500 °C for 2 h. After cooling to room temperature, the powders were reground and then heated at 900 °C for another 2 h. This process repeated at 1200 °C to make sure that the reaction has been completed and pure YBO₃:Ce³⁺, Yb³⁺ has been obtained. The optimum doping concentration of Ce³⁺ and heating temperature were 1% and 1200 °C, respectively. In this paper, the doping concentration varied from 0%, 0.2%, 0.4%, 1%, 2%, 3%, 4% to 6%.

X-ray diffraction was carried out on a Philips X'Pert PRO SUPER X-ray diffraction apparatus with Cu K α radiation (λ = 0.154056 nm). The emission spectra, excitation spectra, and decay curves were recorded by using a FLS920 spectrofluorometer (Edinburgh Instruments). Emission and excitation measurements were performed by using a 450 W Xe lamp as the excitation source. Decay curves measurements were obtained by using a nanosecond flashlamp (nF900) as the excitation source. All the measurements were operated at room temperature.



Fig. 1. XRD patterns of the YBO₃:Ce³⁺ (1%), Yb³⁺ (x%) (x = 0, 0.2, 1, 3, and 6) and the reference data of JCPDS card Nos. 74–1929 for pure YBO₃.

3. Results and discussion

Fig. 1 shows the XRD patterns of YBO₃:Ce³⁺ (1%), Yb³⁺ (x%) samples and the reference data of JCPDS Card No. 74-1929 for pure YBO₃. As indexed in the figure, all diffraction peaks match well with those of pure hexagonal YBO₃ structure with lattice constants of a = 0.3778 nm and c = 0.8814 nm. The stronger diffraction peaks appear at 27.2°, 34.1°, and 49.8°, corresponding to the (1 0 0), (1 0 2), and (1 0 4) planes of YBO₃ crystals, respectively.

Luminescence measurements of Ce³⁺ and Yb³⁺ and decay curves of Ce³⁺ in YBO₃:Ce³⁺, Yb³⁺ phosphors were investigated and gave a convincing evidence for the presence of Ce³⁺ to Yb³⁺ energy transfer. Fig. 2 gives the photoluminescence (PL) and photoluminescence excitation (PLE) spectra of the samples with different Yb³⁺ doping concentrations. By monitoring the 5d to 4f transition of Ce³⁺ at 414 nm and ${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}$ transition of Yb³⁺ at 973 nm, an intense and broad band was observed between 326 and 377 nm in the PLE spectra (Fig. 2a), which assigned to the 4f to 5d transition of Ce³⁺. The observation of the Ce³⁺ 4f to 5d band in the excitation spectrum of Yb³⁺ shows that a CET process from Ce³⁺ to Yb³⁺ is theoretically possible.

Excitation at 358 nm light gave rise to emissions from both Ce³⁺ and Yb³⁺ ions, as shown in Fig. 2b. The broad bands center at 388 nm and 414 nm in the visible region of 370–480 nm have been assigned to the transition from the lowest 5d level to ${}^{2}F_{5/2}$ and ${}^{2}F_{7/2}$ of Ce³⁺. The excited electronic configuration of Ce³⁺ is 5d¹. The 5d electron of Ce³⁺ is not shielded from the surroundings and thus has a strong interaction with the neighboring anion ligands in the compounds. As a result, the excitation band of Ce³⁺ is broad and the excitation wavelength of Ce³⁺ can be tuned by the host materials. Moreover, it is worthy to note that a clear emission centered at 973 nm was also observed, corresponding to the ${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}$ transition of Yb³⁺ ions. To prove the energy transfer from Ce³⁺ to Yb³⁺, the Yb³⁺ single doped YBO₃ sample were also excited by 358 nm, however, we did not observe any peaks in rang from 900 nm to 1100 nm.

These results indicate that the emission of two NIR photons per absorbed one UV photon is possible with the Ce³⁺–Yb³⁺ dual ions via a CET process. Fig. 3 illustrates the schematic energy levels with transitions which may be involved in the CET process from one



Fig. 2. (a) Photoluminescence excitation spectra of YBO₃:Ce³⁺ (1%) sample $(\lambda_{em} = 414 \text{ nm}, \text{Ce}^{3+}: 5d \rightarrow 4f \text{ transition})$ and YBO₃:Ce³⁺ (1%), Yb³⁺ (3%) sample $(\lambda_{em} = 973 \text{ nm}, Yb^{3+}: {}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2} \text{ transition})$; (b) Visible photoluminescence spectra of YBO₃:Ce³⁺(1%), Yb³⁺ (3%) samples with different Yb³⁺ concentrations and NIR photoluminescence spectra of YBO₃:Ce³⁺ (1%), Yb³⁺ (3%) sample ($\lambda_{ex} = 358 \text{ nm}, Ce^{3+}: 4f \rightarrow 5d \text{ transition})$.

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