

Auger electron spectroscopy analysis of chromium depletion in a model Ni–16Cr–9Fe alloy oxidized at 950 °C

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A complete chromium depletion profile with its depth resolution in the nanometre range was obtained on a Ni–16Cr–9Fe model alloy after a 10 h oxidation at 950 °C in air. Auger electron spectroscopy coupled to ion sputtering and energy-dispersive X-ray spectroscopy were used to measure the chromium depletion profile. The chromium content at the oxide/alloy interface is below 0.5 wt.%. This exceptionally low Cr content is in agreement with Wagner's model for binary alloys.

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High-temperature oxidation of industrially relevant alloys, containing typically between 16 and 30 wt.% Cr, results in the formation of a protective chromium-rich oxide scale. One of the major types of damage induced in the substrate by the growth of chromium-rich oxides is chromium depletion, which was theoretically assessed by Wagner [1]. This depletion is known to be high in all face-centred cubic (fcc) alloys [2,3] due to the relatively low Cr diffusion coefficient. Mechanisms of such depletion have been extensively studied in the 800–1000 °C temperature range using model binary Ni–Cr alloys [4–9] and energy-dispersive X-ray spectrometry (EDX) on polished cross-sections at a micrometre scale. These results have allowed experimental Cr depletion profiles to be established and the chromium diffusion coefficient to consequently be determined. The Cr concentration at the oxide/alloy interface has been shown to be frequently close to half of the bulk Cr content of the alloy and never lower than 5 wt.%. Attempts to extrapolate a Cr concentration profile between the area of EDX validity and the exact oxide/alloy interface position have also resulted in an interfacial chromium concentration in the range of 10 wt.% [10]. Consequently, the spatial resolution of EDX performed

on polished cross-sections appears to be a limitation, and the real chromium content close to the oxide/alloy interface cannot be assessed by this widely used spectrometry, in particular in the range between 0 and 1.5 μm for a typical 15 keV accelerating voltage.

In this paper, we report a complete chromium depletion profile obtained in a Ni–16Cr–9Fe model alloy after a 10 h oxidation at 950 °C in air with special emphasis on the first tens/hundreds of nanometres beneath the oxide/alloy interface, as analysed by auger electron spectroscopy (AES) coupled with ion sputtering. We discuss this result in the frame of a classical Wagner's analytical solution for binary alloys [1].

Experiments were performed on cylindrical platelets ($\varphi = 18$ mm, $h = 2$ mm) of Ni–16Cr–9Fe model alloy with impurity concentrations lower than 10 ppm (LECO analysis of C, S, O and N). Sample surfaces were mechanically polished using SiC papers down to grade 1200 and diamond paste down to 0.25 μm. No chemical attack was performed.

High-temperature oxidations were performed at 950 °C for either 10 or 100 h. Three different oxidation procedures were used (Table 1): first, an oxidation in a standard furnace with a very slow cooling rate (EDX01); second, a controlled oxidation in a thermogravimetric system (TGA02); and finally, an oxidation followed by a liquid nitrogen quenching (AES03). The

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Table 1. Experimental conditions of oxidation and type of analysis performed.

Sample	Oxidation conditions	Cooling	Analyses
EDX01	10 h, 950 °C, synthetic air with 20% O ₂	Very slow cooling, ≈10 h, thermal inertia of furnace	SEM/EDX on the transverse cross-section
TGA02	100 h, 950 °C, He-20% O ₂	Controlled cooling, 4 °C min ⁻¹	TGA
AES03	10 h, 950 °C, synthetic air with 20% O ₂	Quenching in liquid nitrogen, some seconds	AES coupled with Ar ⁺ ion sputtering and AES on the transverse polished section

sample EDX01 was analyzed by scanning electron microscopy (SEM) and EDX on its cross-section (Fig. 1a), oxidation kinetics was measured on the TGA02 sample by thermo-gravimetric analysis (TGA) and the AES03 sample was specifically prepared for two types of AES analyses (Fig. 1b and c).

SEM/EDX observations were performed using a JEOL JSM 6400 microscope equipped with a PGT analyser at 15 keV, with a 0.3 nA beam current. AES was performed with a Thermo VG Thetaprobe spectrometer using a field emission electron gun operating at 10 kV accelerating voltage, with a 5 nA beam current and a 45° tilt, resulting in a 150 nm effective spot size. Spectra were recorded on a hemispherical analyser in fixed retard ratio mode. Quantitative analysis by AES was performed measuring in direct spectra the ratio of the Auger peak height to its background close to the peak for the transitions Cr L₃M_{2,3}V at 529 eV, Fe L₃M_{2,3}M_{2,3} at 598 eV and Ni L₃VV at 848 eV. Analyses were carried out on the both AES03 sample and pure metal standards. Concentrations were calculated from intensities using the method described by Shimizu [11], which takes into account the variation in the backscattering coefficient and the mean free path of the Auger electrons as a function of composition [12]. Depth profiles were

performed, using either combined steps of Ar⁺ sputtering and AES analyses or AES spectra recorded directly on the cross-section of the sample. For the ion sputter depth profile, sputtering sequences were carried out using 3 keV argon ions onto a 1 mm² rastered area with a 19 μA cm⁻² current density and an incidence angle of 45°. Under these conditions, the conversion of sputtering time scale into depth scale was done considering a constant sputtering rate of 1.65 nm min⁻¹, as experimentally measured on an Ni–5Cr model alloy and supposed to be three times lower in the oxide, i.e. during the first 1200 s of sputtering. It is worth noting that the AES signal comes from a depth of the order of 5 nm and that the sputtering procedure allows removal of the alloy by layers of approximately one nanometre close to the interface up to dozens or hundreds of nanometres far from the interface.

A complete chromium concentration profile was obtained by assembling an SEM/EDX profile in the area of its validity and an AES depth profile acquired from the oxide/alloy interface down to 2 μm from the oxide/alloy interface. The quality of the assembly procedure was validated by AES analysis of the cross-section, i.e. with 150 nm lateral resolution.

SEM/EDX concentration profiles were measured on an EDX01 specimen cross-section starting from the external oxide, through the chromium-depleted layer towards the bulk. The SEM micrograph (Fig. 1a) shows a representative cross-section of Ni–Cr–Fe alloy after a 10 h oxidation at 950 °C with a 3.5 μm thick chromium-rich oxide. The white arrow indicates the position where the EDX elemental Cr, Fe and Ni profiles were measured. The results of the EDX analyses (Fig. 2a) cover a distance that is longer than that represented by the white arrow in Figure 1a in order to reach the bulk chromium concentration. The chromium profile indicates strong chromium depletion down to a depth of approximately 15 μm, with an apparent minimum of 5 wt.% at a depth of 1.8 μm beneath the oxide/alloy interface. The presence of the chromium-rich oxide layer in fact induces an apparent increase in chromium content between 0 and 1.8 μm. The chromium detected in this part of the EDX profile comes from both the alloy and the oxide because of the size of the EDX interaction volume and interface irregularities.

In order to obtain a real Cr concentration profile on the first 2 μm close to the oxide/alloy interface, AES depth profiling was thought up. The main difficulty to be overcome is the presence of the thick oxide layer in the surface, which must be removed in order to use AES analysis coupled with ion sputtering (sputtering of this thick oxide would otherwise result in extremely high roughness, typically of the order of 1/10 of the

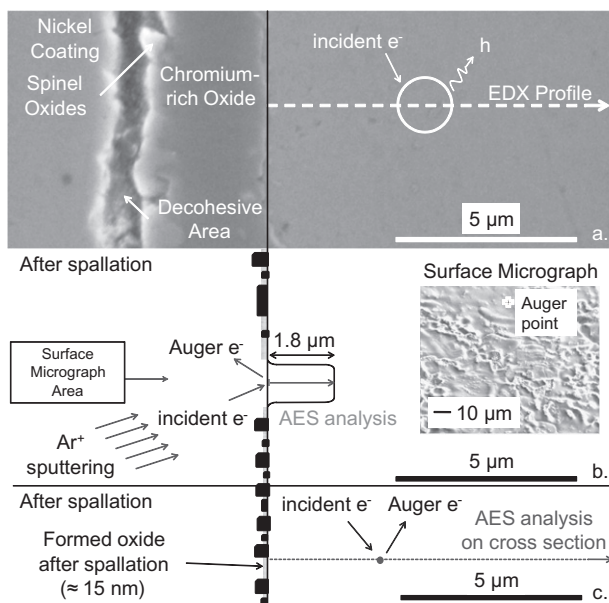


Figure 1. Description of the experimental analysis. (a) SEM/EDX analysis of the cross-section of EDX01. (b) Depth profiling by AES analysis coupled with ion sputtering of AES03 and the top view of the alloy surface after quenching showing the position of Auger point analysis. (c) AES analysis of the AES03 cross-section.

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