

Ab initio study of polymorphism in layered ternary carbide M_4AlC_3 ($M = V, Nb$ and Ta)

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The mechanical and thermodynamic stabilities of M_4AlC_3 ($M = V, Nb$ and Ta) and Ti_4AlN_3 polymorphs were investigated by means of the first-principles pseudopotential total energy method. All compounds satisfied the Born criteria for mechanical stability, but had different thermodynamic stabilities. Only Ta_4AlC_3 showed a possible polymorphic phase transformation driven by thermodynamic competition. The present theoretical results support the polymorphism of Ta_4AlC_3 in experimental synthesis and explain the underlying thermodynamic mechanism. Polymorphism was excluded from V_4AlC_3 , Nb_4AlC_3 and Ti_4AlN_3 .
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Layered ternary $M_{n+1}AX_n$ (where M is an early transition metal, A is an A-group element and X is carbon and/or nitrogen) ceramics are promising candidates for high-temperature structural applications. These ceramics crystallize in the space group $P6_3/mmc$ and contain two structural units alternately stacking along the c direction: the NaCl-type non-stoichiometric transition metal carbide or nitride slab, and the close-packed A-group atomic plane. Due to this layered crystal structure, these compounds exhibit an unusual combination of the properties of both carbides/nitrides and metals, such as low density, high elastic moduli, good high-temperature stability, high thermal and electrical conductivities, excellent thermal shock resistance, damage tolerance and micro-scale ductility at room temperature [1]. $M_{n+1}AX_n$ ceramics are classified into three groups with respect to the value of n , e.g. M_2AX for $n = 1$, M_3AX_2 for $n = 2$, and M_4AX_3 for $n = 3$. To date, approximately 50 M_2AX phases [2] and five M_3AX_2 compounds (Ti_3SiC_2 , Ti_3GeC_2 , Ti_3AlC_2 , Ti_3SnC_2 and Ta_3AlC_2) have been identified experimentally [3–7]. For M_4AX_3 phases, Ti_4AlN_3 , V_4AlC_{3-x} ($x \approx 0.31$),

Nb_4AlC_3 and Ta_4AlC_3 have been synthesized [8–12] in bulk form; and Ti_4SiC_3 and Ti_4GeC_3 have appeared in thin films synthesized by magnetron sputtering [13–15].

$M_{n+1}AX_n$ compounds show interesting polymorphism, which has been actively investigated in recent years [16–22]. For M_2AX and M_3AX_2 ceramics, polymorphism was traced to the A-group atom occupying different internal positions in the unit cell; the transition metal carbide or nitride slabs remained unchanged [16–21]. Polymorphic phase transformation was reported to involve the A-group atom sliding along the (0001) basal plane between two local minimum positions on the energy surface. For the M_4AX_3 compounds, on the other hand, polymorphism was only reported for Ta_4AlC_3 . Structural differences mainly reside within the Ta_4C_3 slab [12,21], as shown in Figure 1. The crystal structure of α - Ta_4AlC_3 shows a stacking sequence ABABACBCBC for transition metal Ta and Al atoms along the [0001] direction [21]; β - Ta_4AlC_3 has a stacking sequence ABABABABAB [12]. Other experimentally identified M_4AX_3 compounds, such as Ti_4AlN_3 , V_4AlC_3 , Nb_4AlC_3 , Ti_4SiC_3 and Ti_4GeC_3 , only show the α -type crystal structure.

Since transition metal atoms V and Nb have the same valence electrons as Ta, polymorphism would be expected for V_4AlC_3 and Nb_4AlC_3 too. However, Ta_4AlC_3 distinguished itself from the other compounds. The reason for this discrepancy is not clear and needs to be

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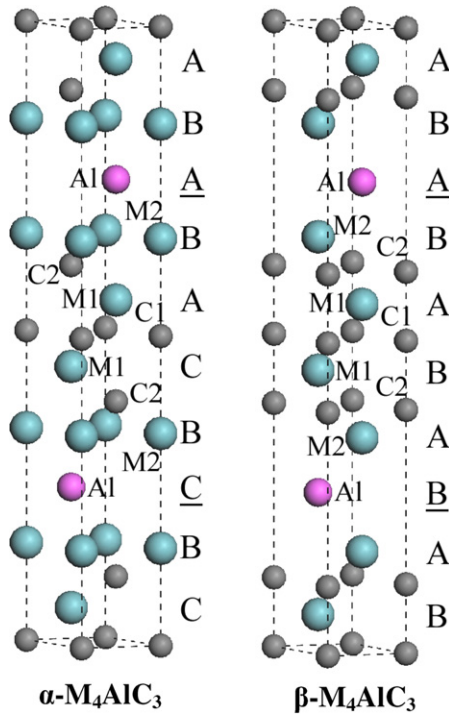


Figure 1. Crystal structures of M_4AlC_3 polymorphs.

explained. The results may contribute to a better understanding of the crystal structure and phase stability of ternary M_4AX_3 compounds. In the present paper, we studied the mechanical and thermodynamic stabilities of V_4AlC_3 , Nb_4AlC_3 , Ta_4AlC_3 and Ti_4AlN_3 polymorphs. The hypothetical crystal structures of β -type V_4AlC_3 , Nb_4AlC_3 and Ti_4AlN_3 were constructed with respect to that of β - Ta_4AlC_3 . Elastic coefficients and Gibbs free energy were calculated to study the mechanical and thermodynamic stabilities of the two possible polymorphic structures. The results showed that only Ta_4AlC_3 displayed a possible polymorphic phase transformation driven by thermodynamic competition.

Calculations were accomplished using the CASTEP code, in which the plane-wave pseudopotential total energy calculation was employed [23]. The plane-wave energy cutoff and the Brillouin zone sampling were fixed to 450 eV and $9 \times 9 \times 2$ special k -point meshes [24], respectively. Interactions of electrons with ion cores were represented by the Vanderbilt-type ultrasoft pseudopotential [25]. The electronic exchange-correlation energy was treated under the generalized gradient approximation (GGA-RPBE) [26]. The equilibrium crystal structures were fully optimized by independently modifying lattice parameters and internal atomic coordinates. The Broyden–Fletcher–Goldfarb–Shanno (BFGS) minimization scheme [27] was used to minimize the total energy and interatomic forces. The second-order elastic coefficients were determined by means of the linear fitting of the stress–strain relationship. The computational strategies can be referenced in our previous works, wherein crystal structure, elastic stiffness and interatomic force constants of layered ternary carbides and complex oxides such as $LaPO_4$ monazite, $Y_2Si_2O_7$ and $La_2Zr_2O_7$ pyrochlore were investigated [28–33].

The thermodynamic stabilities were investigated by comparing the free energy difference of the corresponding polymorphs. The most stable structure of any compound at a given temperature is the one with the lower Gibbs free energy, G , given by

$$G(p, T) = F(V, T) + pV, \quad (1)$$

where F is the Helmholtz free energy, and is the summation of vibrational free energy F_{vib} and perfect lattice energy $F_{perfect}$ ($F_{perfect} = U_0 + U_{el} - TS_{el}$). The U_0 and $U_{el} - TS_{el}$ are contributions from the lattice and electronic excitations, respectively. The vibrational free energy F_{vib} is calculated within the harmonic approximation:

$$F_{vib} = k_B T \sum_i \left(\frac{\hbar \omega_i}{2k_B T} + \ln \left(1 - e^{-\frac{\hbar \omega_i}{k_B T}} \right) \right), \quad (2)$$

where ω_i is the phonon frequency, T the absolute temperature, \hbar the Planck constant and k_B the Boltzmann constant. If the electronic excitations and pressure changes are neglected, the Gibbs free energy is given by $G = U_0 + F_{vib}$. Phonon frequencies were calculated via a direct force constant approach [34,35].

The mechanical stability of the polymorphs was investigated first. The fundamental basis for the study of the mechanical stability of a crystal lies on the formulation of stability criteria, a set of conditions that specify the critical internal strain or external stress under which the homogeneous crystal lattice becomes structurally unstable. Born and Huang systematically studied the lattice mechanical stability and formulated stability criteria in terms of the elastic constants c_{ij} . The Born criterion for a lattice to be mechanically stable is that the elastic energy density must be a positive definite quadratic function of strain. For a hexagonal crystal, only five elastic constants are independent, and in this case the mechanical stability criterion is [36]:

$$c_{11} + c_{12} - \frac{2c_{13}^2}{c_{33}} > 0, \quad c_{44} > 0, \quad \text{and} \quad c_{66} = \frac{c_{11} - c_{12}}{2} > 0. \quad (3)$$

The theoretical second-order elastic coefficients of studied M_4AX_3 compounds are listed in Table 1. All investigated polymorphs satisfy the mechanical stability conditions presented in Eq. (3). The results suggest that all phases are stable from the point of view of mechanical stability. Therefore, thermodynamic stability is considered in order to explain the absence of some polymorphic phases in experimental synthesis.

Table 1. Theoretical second-order elastic coefficients c_{ij} (in GPa) of Ti_4AlN_3 and M_4AlC_3 ($M = V, Nb$ and Ta) polymorphs

Compound	c_{11}	c_{12}	c_{13}	c_{33}	c_{44}	c_{66}
α - V_4AlC_3	435	121	105	384	168	157
β - V_4AlC_3	381	85	129	350	130	148
α - Nb_4AlC_3	413	124	135	328	161	144
β - Nb_4AlC_3	420	110	122	381	124	155
α - Ta_4AlC_3	496	154	181	417	200	171
β - Ta_4AlC_3	509	143	156	440	147	183
α - Ti_4AlN_3	387	96	96	352	153	145
β - Ti_4AlN_3	389	90	85	380	112	149

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