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Intensification of CO₂ Stripping from Amine Solutions by Ultrasonic

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Abstract

Ultrasound was introduced to enhance CO₂ stripping from loaded amine solutions in this work. The effects of ultrasound on desorption of CO₂ from loaded amine solutions, i.e. MEA (Monoethanolamine) and MDEA (Methyldiethanolamine), at various temperature, amine concentration, CO₂ loading and energy input of ultrasound were investigated at ambient pressure. Energy consumption of CO₂ stripping assisted by ultrasound was estimated. The data are useful as a platform for further ultrasonic related research in CO₂ capture plant.

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Keywords: Ultrasound; CO₂ stripping enhancement; Amine solutions

1. Introduction

One of the main challenges for post combustion CO₂ capture is the large energy consumption in the regeneration unit. This contributes to high capture costs and is considered to be one of the main economical challenges for large scale post combustion CO₂ capture. Ultrasound can be used for medical imaging, detection, measurement and cleaning, and also for enhancement of mass transfer rates in biological and other industrial field [1-4].

In the present work, ultrasound was introduced to enhance CO₂ stripping from loaded amine solutions.

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Ultrasound leads to cavitation and nucleation in the liquid and thus the formation of bubbles. Once formed, it could be relatively easy for bubbles to grow as more gas diffuses to the bubbles and becomes part of the bubble. In this way ultrasound makes it easy for gas to escape in the form of bubbles. Using ultrasound is expected to contribute to a significant reduction of the CO₂ loading in lean amine absorbent without increasing the specific energy input in reboiler (kg steam/kg CO₂ desorbed). Therefore a reduction of equipment sizes is expected, and the enhanced desorption at lower temperatures which will lead to a reduction of the CO₂ capture cost. Furthermore the problems related to degradation of amine solvents may be reduced.

In this work, the effects of ultrasound on desorption of CO₂ from loaded amine solutions (MEA (Monoethanolamine) and MDEA (Methyldiethanolamine)) at various values of parameters (temperature, amine concentration, CO₂ loading, energy input) were investigated at ambient pressure.

Nomenclature

α CO₂ loading in amine solution, (mol CO₂ / mol amine)

MEA Monoethanolamine

MDEA Methyldiethanolamine

2. Experimental Section

2.1. Experimental set-up

A semi-batch rig, seen in figure 1, was employed to investigate the effect of ultrasound on desorption of CO₂ from amine solutions. In the rig, the loading of amine solution is a batch operation and CO₂ outlet is continuous. 50mL CO₂ loaded solution in the flask is heated by an oil bath to a desired temperature, and then treated by an ultrasonic processor (Hielscher, UP200Ht) with a frequency 26 kHz. The desorbed CO₂ from the solution passes through a condenser and a buffer/scrubbing bottle, where vapours of water and amine are removed, and the gas temperature drops to room temperature. Then the degassed CO₂ flow rate is measured by a digital mass flow meter (Top-Trak, Sierra 820). All these data including degassing flow rate, liquid temperature, room temperature, and the gas pressure in the flask are recorded by a data logger and collected by a computer. Lastly, the degassed CO₂ is collected into a gas bag which can be used to verify measurement of the rate of CO₂ desorption. Liquid analysis can be conducted by the method from Weiland and Trass [5] for determining the CO₂ loading before and after the experiment. The rig is easy to operate, it has good accuracy and repeatability, low solution requirement implying quick preparation of runs thus enabling many runs. Its operation is low cost, and results can be obtained quickly.

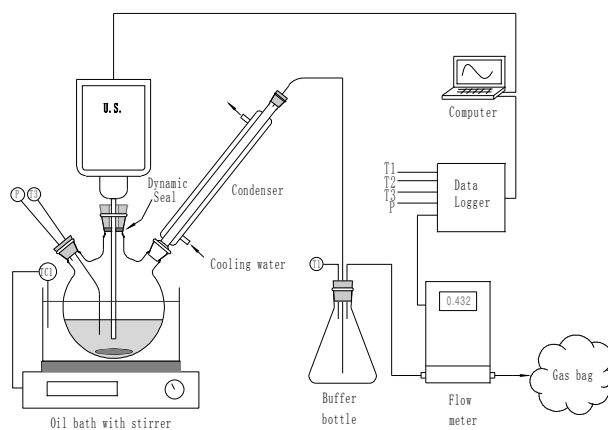


Figure 1. Semi-batch apparatus for ultrasound assisted stripping of CO₂.

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