



Magnetic and optical properties of WO₃/TiO₂ superlattice



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HIGHLIGHTS

- We found the magnetic phenomenon in the WO₃/TiO₂ thin film.
- The electrons transfer from TiO₂ and WO₃ induce the occurrence of ferromagnetism.
- The electrons transfer between TiO₂ and WO₃ shift the absorption region to visible light.

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ABSTRACT

The electronic structure and properties of WO₃/TiO₂ superlattice were investigated by the calculation and experiment. The O electrons transfer the spin angular momentum from the unoccupied Ti e_g states electrons to W e_g states electrons and let Ti and W electrons spin splitting, which were the reason of ferromagnetism occurrence, while the electrons transfer from the CB of TiO₂ to the CB of WO₃. Coupling TiO₂ and WO₃ greatly reduce the recombination rate of electron and hole and shift the absorption region to visible light because of the electrons transfer between TiO₂ and WO₃.

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1. Introduction

Superlattices are materials with narrow-band-gap and wide-band-gap, which can improve the effective electron quantity and activity [1–3]. Coupling nanometer TiO₂ with narrow band gap semiconductor (CdS, MgO, WO₃, Fe₂O₃) may apparently improve the active of TiO₂ and induce the ferromagnetism occurrence. Researchers [4–6] have reported that coupled TiO₂ with WO₃ (WO₃/TiO₂) can decrease the photo-induced hole-electrons recombination rate and the light absorption region transfer to visible light. Coey's group [7] discovers the high-temperature ferromagnetism of HfO₂ thin film and firstly proposes the anisotropic high-temperature d⁰ ferromagnetism. Yoon's group [8] finds that the oxygen vacancies surrounding Ti²⁺ and Ti³⁺ ions potentially can give rise to magnetism of TiO_{2-δ} films. The room-temperature ferromagnetism of SmCo₅/Co films are induced by strong interaction of Sm and Co [9]. The Fe₂O₃/TiO₂ films [10]

possess higher photocatalysis efficiency than pure TiO₂ films and the Fe₂O₃/TiO₂ composite particles [11] possess room-temperature ferromagnetism which comes mainly from Fe₂O₃ phase. Choi et al. [12] proved the existence of uniaxial magnetic anisotropy in epitaxial Fe/MgO films on GaAs (001).

However, the superlattice materials all have magnetic phases and it is of great significance to develop new type superlattice magnetic materials. Herein, we investigated the micro-structure, optical and magnetic properties of WO₃/TiO₂ thin films by the coupling method of experiment and calculation. We found the ferromagnetism of WO₃/TiO₂ thin films and explain the occurrence reason of ferromagnetism.

2. Calculations and experiments

The calculation method of pure TiO₂ and WO₃/TiO₂ superlattice are consistence with the calculation method described in Ref. [13], but the parameter setting is not exactly same. The CASTEP code was used to calculation, the electron wave function was expanded in plane waves up to cut-off energy of 450 eV. The Monkhorst-Pack scheme k-points grid sampling was set to be 4 × 4 × 5 for the

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Table 1
The band gap of anatase TiO₂ and WO₃ with different U values.

U (eV)	0.00	3.00	5.00	7.00	8.00	8.50	9.00
E _g of TiO ₂ (eV)	2.14	2.55	2.83	3.05	3.18	3.22	3.29
E _g of WO ₃ (eV)	1.48	2.08	2.47	2.71	2.84		

unit cell and $8 \times 8 \times 10$ for the supercell. As displayed in Table 1, the $U = 8.50$ eV for Ti 3d and $U = 7.00$ eV for W 5d were selected in subsequent calculations, because of the bandgap of TiO₂ and WO₃ were 3.22 eV and 2.71 eV, respectively, which is consistent with the experiment ones [14,15].

The $2 \times 2 \times 1$ superlattice of TiO₂ was built (Fig. 1a) for pure TiO₂ calculation. The lattice plane of TiO₂ and WO₃ were (002) and (101) respectively, in the WO₃/TiO₂ crystal model [6]. As shown in Fig. 1, the crystal model b, c and d were named W₄T₄, W₄T₆ and W₄T₈, respectively, because of the atom number of the W and Ti. The α , β , γ of W₄T₄, W₄T₆ and W₄T₈ are the same, and $\alpha = \beta = 90.0000^\circ$, $\gamma = 100.148^\circ$, but c of W₄T₄, W₄T₆ and W₄T₈ are different, $c = 1.48067$ nm for W₄T₄, $c = 1.83149$ nm for W₄T₆, $c = 2.26029$ nm for W₄T₈. All the models were geometrically optimized, then total density of states (TDOS) and partial density of states (PDOS) were calculated and analyzed.

The WO₃/TiO₂ thin films were prepared by magnetron sputtering method. At firstly, sputtering Ti on 3 sheets cleaned ultraviolet quartz glasses for 15 min, then the sheet glasses with coating were calcined at 500 °C for 0.5 h. Secondly, sputtering W on the glasses with coating for 5 min, 10 min and 15 min, respectively. Then the glasses were calcined at 500 °C for 0.5 h. The WO₃/TiO₂ thin films based on silicon slices were prepared by the same method. The WO₃/TiO₂ thin films were named W₅T₁₅, W₁₀T₁₅ and W₁₅T₁₅ because of the W and Ti sputtering time, respectively.

The crystallite phases of the samples were identified by XRD on X'pert Philips using Cu K α ($\lambda = 0.15406$ nm) radiation operating at 40 kV and 30 mA at a rate of 0.03°/s. The diffraction data were recorded between 15° and 80°. The magnetic properties of WO₃/TiO₂ were characterized by VSM (Versalab, Quantum DeSign, USA) at room temperature. The absorption spectra were determined by

spectroscopic analysis at 508 nm using a UV spectrometer (UV-1801, China).

3. Results and discussion

According to the crystal field theory, due to the hybridization of Ti–O and W–O, the Ti 3d and W 5d orbits split into two parts, the t_{2g} (d_{xy} , d_{xz} , d_{yz}) and e_g (d_{z^2} , $d_{x^2-y^2}$) states, meanwhile the O 2p orbit split into p_π and p_σ states. O 2p and the t_{2g} (d_{xz} , d_{yz}) of Ti 3d and W 5d devoted to the valence band (VB) (p_π devote to the top of valence band), while the conduction band (CB) was contributed by the d_{xy} and e_g of Ti 3d and W 5d (d_{xy} devote to the bottom of conduction band) [16]. Fig. 2 shows the TDOS and PDOS of TiO₂ and the TDOS and PDOS of WO₃/TiO₂, respectively. Comparing with pure TiO₂ (3.22 eV), the W₄T₄ (1.45 eV) and W₄T₆ (0.62 eV) have more narrow band gap, and the W₄T₈ has not obviously forbidden band. These illustrate that coupling TiO₂ with WO₃ can decrease the energy required for electron transfer and shift the light absorption region from ultraviolet to visible light.

As shown in Fig. 2a, there is no splitting between the spin-up and spin-down states, which confirms that TiO₂ has not magnetism. For the TDOS of WO₃/TiO₂ (Fig. 2b (I), c (I) and d (I)), there is a spin-split around the Fermi Level illustrating the existence of magnetism. For the PDOS of W 5d, O 2p and Ti 3d (Fig. 2b (II, III and IV), 2c (II, III and IV), 2d (II, III and IV)), there are also exchange splitting around the Fermi level between the spin-up and spin-down states, and the magnetic moment was mainly devoted by W and Ti atoms. The bottom of CB (d_{xy}) move to a low energy region and the top of VB (p_π) move to a high energy region with the increasing of TiO₂. This is the reason of why there is not obviously forbidden band in W₄T₈.

The XRD patterns for the pure and coextruded films are displayed in Fig. 3. The diffraction peaks of each sample can be indexed to anatase phase with space group $I4_1/amd$ (141) and the tungsten oxide phase with space group $P21/n$ and $P4/nmm$, which indicate that the W has not complete oxidation in W₁₀T₁₅ and W₁₅T₁₅ samples for calcined 0.5 h.

Researchers [17–19] found the ferromagnetism of TiO₂ films and indicated that the ferromagnetism of TiO₂ thin films were

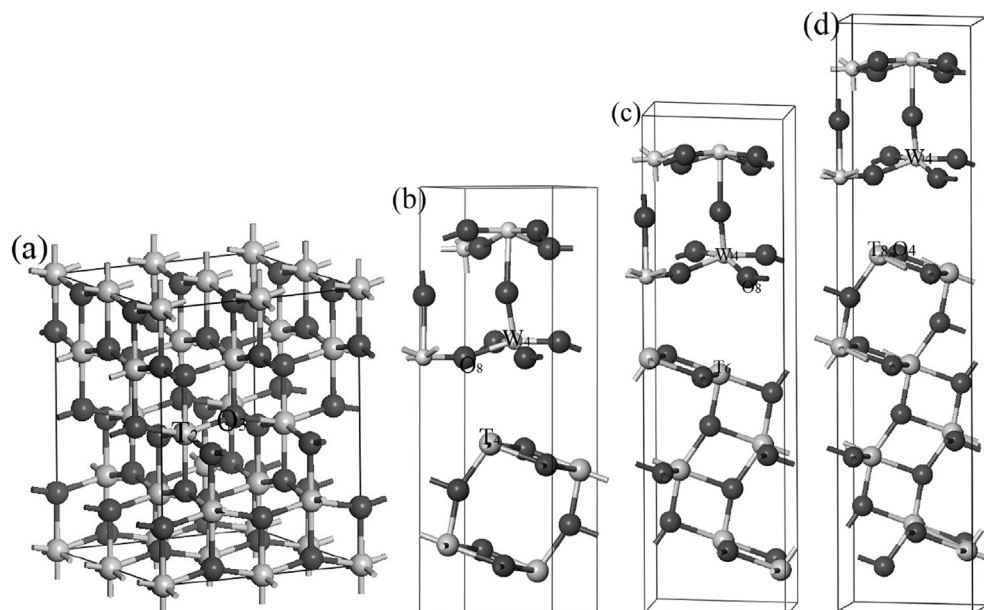


Fig. 1. The $2 \times 2 \times 1$ superlattice model of TiO₂(a) and superlattice crystal model of WO₃/TiO₂. The number ratio of W and Ti atoms were 4:4 (b), 4:6 (c), 4:8 (d).

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