

Short communication

Facile synthesis of Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures and their application in lithium storage

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ARTICLE INFO

Article history:

Received 2 April 2014

Received in revised form 8 July 2014

Accepted 22 July 2014

Available online 2 August 2014

Keywords:

Zn-doped SnO₂

Dendrites

Hierarchical architectures

Hydrothermal synthesis

Lithium storage capacity

ABSTRACT

Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures were successfully synthesized by a facile hydrothermal approach without the use of any surfactants or templates. The as-prepared samples were characterized by the X-ray diffraction (XRD), X-ray photoelectron spectroscopy (XPS), field emission scanning electron microscopy (FESEM), high-resolution transmission electron microscopy (HRTEM), and Raman spectroscopy. The observation of FESEM and HRTEM showed that Zn-doped SnO₂ hierarchical cube-like architectures were composed of numerous oriented dendrites. Each dendrite is assembled by a pronounced trunk with highly ordered branches distributing on the both sides. The as-prepared Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures were used as anode materials for Li-ion battery, and a stable capacity of 488.3 mA h g⁻¹ was achieved after 50 cycles. The results of electrochemical measurements indicated that the as-prepared Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures have potential application in Li-ion battery.

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1. Introduction

It is well known that the properties and applications of complex hierarchical architectures are strongly dependent on the size, morphology, shape, chemical composition and surface property [1–5]. Thus, great efforts have been devoted to synthesizing functional materials with complex hierarchical architectures [1–9]. The dendrite-built hierarchical architecture has been considered to be one of the most important categories because of the large surface area and good stability [4]. Nowadays, dendrite-built hierarchical architectures have attracted increasing attentions due to their unique properties in the fields of solar energy conversion, field electron emission, gas sensing and photocatalysis [6–9]. So far, a variety of dendrite-built hierarchical architectures, including Zn-doped NiO, CuSe, lead chalcogenides, Fe₂O₃, Cu-doped Fe₂O₃, Co, SnO₂, have been successfully synthesized for pursuing excellent properties [10–15].

SnO₂, as an n-type semiconductor with a band gap of 3.6 eV, has numerous applications in optoelectronic devices, dye-sensitized

solar cells, secondary lithium batteries, electrode materials, transistors, gas sensors, and catalyst supports [15–20]. In order to meet increasing demands under more complicated and harsh conditions, many investigations have been made on the modification of SnO₂, e.g., element doping and heterostructure constructing [19–26]. In the previous study [21], SnO₂ nanoparticles were synthesized by a molten-salt decomposition method. The product was used as anode materials for lithium ion battery and the specific capacity was 402 mA h g⁻¹ after 40 cycles. Liu et al. prepared SnO₂ hollow microspheres via a hydrothermal-annealing method and investigated the electrochemical property. The specific capacity was 374.2 mA h g⁻¹ after 100 cycles [22]. All the above experimental procedure was complicated and the results were not satisfied. Generally, doping is considered to be a simple and effective approach to enhance the properties of oxide semiconductors. Up to now, several types of Zn-doped SnO₂ nanostructures have been synthesized and exciting results have been obtained in the previous studies [23–27]. However, the design and synthesis of Zn-doped SnO₂ hierarchical architectures with desired morphology assembled by one-dimensional nanobuilding blocks are still a challenge. Therefore, one-step surfactant-free approach is strongly desirable for the synthesis of Zn-doped SnO₂ dendrite-built hierarchical architectures. Moreover, there are few studies focused on the synthesis and electrochemical properties of Zn-doped SnO₂ dendrite-built hierarchical architectures.

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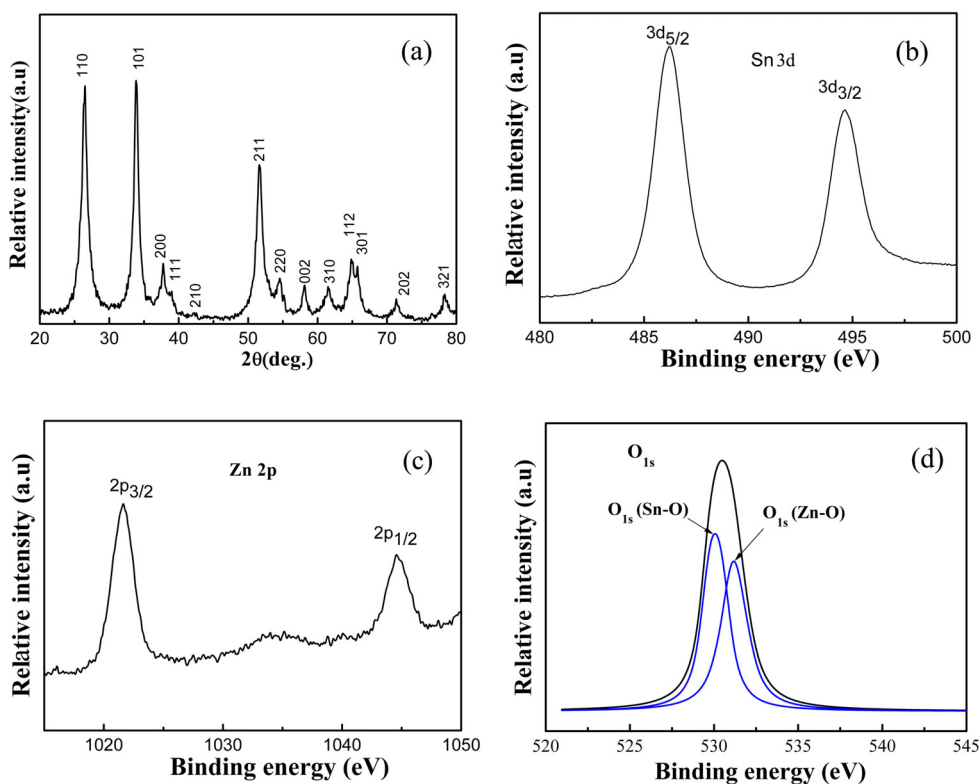


Fig. 1. (a) XRD pattern, and high-resolution XPS spectra of (b) Sn 3d and (c) Zn 2p and (d) O 1s of the as-prepared Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures.

Herein, we present a facile and reliable hydrothermal approach for preparing Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures without the use of any templates or surfactants. The electrochemical properties of the as-prepared samples used as anode materials for Li-ion battery were also investigated.

2. Experimental

2.1. Preparation of Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures

All the reagents were analytically pure (Shanghai Chemical Industrial Co.) and used as received. In a typical procedure, 1.2 g of SnCl₄·5H₂O and 0.168 g of Zn(NO₃)₂·6H₂O were dissolved in 60 mL distilled water. Then, 100 mL of NaOH aqueous solution (0.6 M) was added dropwise into the solution under vigorous stirring. After stirring for 30 min, the mixture was transferred into a Teflon-lined stainless steel autoclave, and maintained at 200 °C for 20 h. After the hydrothermal procedure, the autoclave was cooled naturally down to room temperature. The resultant products were centrifuged and washed with distilled water and ethanol for several times, followed by drying in vacuum at 60 °C for 12 h. For comparison, pure SnO₂ nanoparticles were prepared via the similar approach without using zinc source.

2.2. Characterization

X-ray diffraction (XRD) pattern was collected using a Rigaku ultimIII diffractometer equipped with Cu K α radiation ($\lambda = 0.15406$ nm). Scanning electron microscopy (SEM) images were taken on a Hitachi S-4800 field emission scanning electron microscope. High-resolution transmission electron microscopy (HRTEM) images were performed on a JEM 2010FEF transmission electron microscope. X-ray photoelectron spectroscopy (XPS)

measurement was carried out on a VG Multilab 2000X X-ray photoelectron spectrometer. Raman spectra were recorded with a Renishaw/invia Raman microscope at the room temperature with an Ar⁺ laser of 514 nm line as the excitation source. The nitrogen adsorption and desorption isotherms were measured at 77 K using a Quantachrome NOVA 2000e sorption analyzer after the sample was vacuum-dried at 473 K overnight.

2.3. Electrochemical measurements

Electrochemical measurements were carried out by using two-electrode cells with lithium metals as the counter and reference electrodes. The assembly of the lithium cells has been reported in the previous literature [28]. The electrodes were fabricated by compressing a mixture of the active material including Zn-doped SnO₂, conductive material (acetylene black), and binder (polytetrafluoroethylene) with a weight ratio of 82:8:10. The electrode was dried at 80 °C for 1 h and cut into a disk of 1.0 cm in diameter. The cells were assembled in a glove-box filled with pure argon. The electrolyte was 1 M LiPF₆ dissolved in a mixture of ethylene carbonate (EC), dimethyl carbonate (DMC) and ethylmethyl carbonate (EMC) (1:1:1 by weight). An appropriate amount of electrolyte was added before sealing. The newly assembled cells were measured by a galvanostatic discharge–charge method in a multichannel battery tester (model Land CT2001A) in the potential range of 0.005–2.5 V (vs. Li⁺/Li). Typically, a working electrode of 1 cm² was prepared with the active material mass loading of 1.95 mg.

3. Results and discussion

3.1. Phase structure

Fig. 1a shows the XRD pattern of Zn-doped SnO₂ dendrite-built hierarchical cube-like architectures. All the diffraction peaks can be

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