



Thermodynamics at the nanoscale: A new approach to the investigation of unique physicochemical properties of nanomaterials

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ABSTRACT

An extension of the classic thermodynamics theory to nanometer scale has generated a new interdisciplinary theory – nanothermodynamics. It serves as a bridge between macroscopic and nanoscopic systems. Over the past decade, nanothermodynamics theories have developed rapidly owing to their critical role in investigating the size-dependent physicochemical properties of nanomaterials. This review examines up-to-date research results on this cutting-edge topic. The focus and emphasis are on the utilization of nanothermodynamics models to investigate the size-dependent thermal stability, magnetic properties, photoelectric behaviors, thermoelectric phenomena, mechanical properties, electrical properties, etc. of nanomaterials.

A range of properties have been studied with respect to the effects of size, dimensionality and composition through a quantitative nanothermodynamics model. It is found that (a) the size dependence of these properties can be universally reconciled to the effect of severe bond dangling; (b) for the same material size, the sequence of size effects on the properties, from strong to weak, is nanoparticles, nanowires and thin films; and (c) the composition effects on the properties of nanoalloys are substantial, having a nonlinear relationship. It also reveals that vacancy formation determined by the cohesive energy variation is one of the intrinsic factors which dominate the size-dependent physicochemical properties of nanomaterials.

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Contents

1. Introduction	4
1.1. Scope	4
1.2. Overview	4
1.3. Challenges	5
1.4. Objectives	5
2. Principles: atomic vibrational instability	5
2.1. Lindemann's melting criterion	5
2.2. Size-dependent melting thermodynamics of crystals	6
3. Interatomic potential and cohesive energy	7
3.1. Interatomic potential at the nanoscale	7
3.2. Current nanothermodynamics models of cohesive energy	7
3.3. A new model: size, dimensionality and composition dependency	7
3.4. Consistency among different nanothermodynamics models	8
3.5. Validity of new model	8

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4.	Thermal stability	8
4.1.	Cohesive energy E_c	8
4.1.1.	E_c of nanocrystals	8
4.1.2.	E_c of binary nanoalloys	9
4.2.	Solid–liquid transition temperature T_m	10
4.2.1.	T_m of isolated nanocrystals	10
4.2.2.	T_m of nanosized solid solution and eutectic binary alloys	10
4.2.3.	T_m of nanoparticles embedded in matrices	11
4.3.	Solid–vapor transition temperature	13
4.4.	Debye temperature	13
4.5.	Einstein temperature	14
4.6.	Melting enthalpy	14
4.7.	Size–temperature–composition phase diagrams	15
4.8.	Size–temperature–pressure phase diagrams	15
4.8.1.	Pentacene thin films	15
4.8.2.	Silver nanocrystals	17
4.8.3.	Carbon nanoparticles	18
5.	Magnetic properties	20
5.1.	Curie temperature	20
5.2.	Order–disorder transition temperature	20
6.	Photoelectric behaviors	21
6.1.	Bandgap energy	21
6.2.	Raman red shift	26
6.3.	Molar extinction coefficient	29
7.	Thermoelectric phenomena	30
7.1.	Thermal conductivity and diffusivity of nanocrystals	30
7.2.	Thermal conductivity of nanoporous and nanostructured bulk materials	31
8.	Mechanical properties	33
8.1.	Young's modulus	33
8.2.	Inverse Hall–Petch relationship	33
9.	Electrical properties	33
9.1.	Electrical conductivity	33
9.2.	Relative permittivity	34
10.	Other physicochemical properties	34
10.1.	Volume thermal expansion coefficient	34
10.2.	Mass density	34
10.3.	Catalytic activation energy	34
10.4.	Diffusion activation energy	34
10.5.	Vacancy formation energy	35
11.	Future work	35
	Acknowledgements	36
	References	36

List of symbols and abbreviations

a, c	lattice parameter
a_h	hole Bohr radius
A_c	a material constant
A_m	gram-atom (g-atom) surface area
b	bowing parameter
B	bulk modulus
BOLS	bond-order-length-strength
c_G	Gruneisen constant
c_i	coordination number-dependent reduction of bond length
C	a constant
C_p	heat capacity
C_v	equilibrium vacancy concentration
C_0	a size-independent pre-exponential coefficient
d	dimensionality of materials
d_{hkl}	inter-planar distance of (hkl)
D	thermal diffusivity

D_d	diffusion coefficient
DFT	density functional theory
D_0	a size-independent pre-exponential coefficient
e	electrical charge of one electron
E_a	first absorption peak energy
EAM	embedded-atom method
E_c	cohesive energy
E_{ca}	catalytic activation energy
E_d	diffusion activation energy
E_{exc}	spin–spin exchange interaction
E_g	bandgap energy
E_{lv}	lattice thermal vibrational energy
E_r	exciton Rydberg energy
E_v	vacancy formation energy
f	surface stress
$fwhm$	full width at half-maximum
FGS	fullerene-like graphite shells
G	Gibbs free energy

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