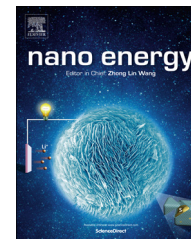


Available online at www.sciencedirect.com

ScienceDirect

journal homepage: www.elsevier.com/locate/nanoenergy

RAPID COMMUNICATION

Annealing-free efficient vacuum-deposited planar perovskite solar cells with evaporated fullerenes as electron-selective layers

Dewei Zhao^{a,b,*,1}, Weijun Ke^{a,1}, Corey R. Grice^a,
 Alexander J. Cimaroli^a, Xinxuan Tan^a, Mengjin Yang^b,
 Robert W. Collins^a, Hongmei Zhang^c, Kai Zhu^b, Yanfa Yan^{a,**}

^aDepartment of Physics and Astronomy and Wright Center for Photovoltaics Innovation and Commercialization, The University of Toledo, Toledo, OH 43606, United States

^bChemical and Materials Science Center, National Renewable Energy Laboratory, Golden, CO 80401, United States

^cInstitute of Advanced Materials and Key Laboratory for Organic Electronics & Information Displays, Nanjing University of Posts & Telecommunications, Nanjing 210023, PR China

Received 17 August 2015; received in revised form 25 October 2015; accepted 10 November 2015

KEYWORDS

Perovskite solar cells;
 Vacuum-deposition;
 Electron-selective layer;
 Hole blocking layer;
 C_{60} and C_{70} (fullerenes)

Abstract

We present efficient metal oxide-free and annealing-free planar perovskite solar cells with the regular cell structure using vacuum-deposited fullerenes C_{60} and C_{70} as the electron-selective layers and vacuum-processed perovskites as the light absorbers. The devices with an ultrathin C_{60} layer (5.5 nm) yielded an average power conversion efficiency of 14.3% and a maximum efficiency of 15.7%. The best-performing cell produced a steady-state efficiency of 14.6%. The high performance is attributed to the efficient blocking of holes and extraction of electrons by C_{60} due to a favorable energy level alignment between the C_{60} and the fluorine-doped tin oxide electrodes. With the realization of efficient cells, the annealing-free vacuum deposition of perovskite absorbers and C_{60} or C_{70} electron-selective layers and intermediate layers demonstrates its power for fabricating all-perovskite tandem solar cells.

© 2015 Published by Elsevier Ltd.

*Corresponding author at: Department of Physics and Astronomy and Wright Center for Photovoltaics Innovation and Commercialization, The University of Toledo, Toledo, OH 43606, United States.

**Corresponding author.

E-mail addresses: dewei_zhao@hotmail.com (D. Zhao), yanfa.yan@utoledo.edu (Y. Yan).

¹These authors contributed equally to this work.

<http://dx.doi.org/10.1016/j.nanoen.2015.11.008>

2211-2855/© 2015 Published by Elsevier Ltd.

Introduction

Organic-inorganic lead halide perovskite based solar cells have been extensively investigated over the past few years due to the excellent photovoltaic properties of perovskites such as high optical absorption coefficient, long charge

1 carrier lifetime, high carrier mobility, and band gap ten- 60
2 ability [1-5]. Moreover, perovskite solar cells can be made 61
3 by low cost solution process at low temperature, enab- 62
4 ling roll-to-roll manufacturing using flexible substrates. 63
5 Perovskite-based solar cells have been considered recently 64
6 as the most promising photovoltaic technology [6,7], with 65
7 their power conversion efficiencies (PCEs) rapidly rising 66
8 from 3.8% to a certified 20.1% [8-15]. 67

9 The high efficiencies of perovskite solar cells are largely 68
10 due to the high open circuit voltages (V_{oc} s), which are 69
11 attributed to the long carrier lifetime [1,2]. It has been 70
12 shown that the dominant defects do not create deep energy 71
13 levels in the band gaps of perovskite absorbers [3,4], 72
14 distinctively different from that seen in the common inor- 73
15 ganic solar cell materials. The V_{oc} deficit, defined by $E_g/$ 74
16 $q - V_{oc}$, of the record perovskite solar cells, is as small as 75
17 0.42 V, approaching that of the best single crystal GaAs thin 76
18 film solar cell [12]. Therefore, the V_{oc} of the high-efficiency 77
19 perovskite solar cells has nearly reached its theoretical 78
20 value. Unlike V_{oc} , the short-circuit current density (J_{sc}) of 79
21 the high-performing perovskite solar cells still has consider- 80
22 able room for improvement. Typically, high-efficiency per- 81
23 ovskite solar cells use approximately 350-450 nm thick 82
24 absorbers, which are not sufficiently thick to fully absorb 83
25 the incident light. Unfortunately, increasing the thickness of 84
26 the perovskite absorbers could only lead to lower efficien- 85
27 cies, indicating that intrinsic electronic property issues exist 86
28 for the low temperature processed polycrystalline perovs- 87
29 kite thin films. Fabricating tandem cells is a legitimate 88
30 option for further improving the overall device efficiency 89
31 and has recently attracted great attention. So far, most 90
32 perovskite tandem cells reported consist of perovskite solar 91
33 cells as the top subcells and inorganic solar cells such as 92
34 Si and Cu(In,Ga)Se₂ cells as the bottom subcells [16-18]. 93
35 Efficient all-perovskite tandem cells, a preferred approach, 94
36 have not been reported yet. Part of the challenge stems 95
37 from depositing intermediate layers without damaging the 96
38 existing underlying subcells. 97

39 The planar cell architecture is preferred for fabricating 98
40 tandem cells. A typical planar cell consists of a solution- 99
41 processed or vacuum-processed perovskite absorber layer 100
42 sandwiched between an electron-selective layer (ESL) and a 101
43 hole-selective layer (HSL). To avoid any damage to the 102
44 existing subcell, the ESL or HSL and the perovskite absorber 103
45 layer should be deposited preferably by vacuum processes 104
46 and without additional annealing. Furthermore, compared 105
47 to solution processes, vacuum processes are expected to 106
48 result in thin films with better uniformity, smoother sur- 107
49 faces, complete coverage, and more accurately contro- 108
50 lled thicknesses. However, to date efficient annealing-free 109
51 vacuum-processed perovskite solar cells have not been 110
52 reported. The current high-performing perovskite cells 111
53 generally use either high-temperature processed metal 112
54 oxide layers such as TiO₂ [19,20], solution-processed low 113
55 temperature materials such as ZnO [21], SnO₂ [22], and 114
56 Cs₂CO₃ [23], or organic materials such as [6,6]-phenyl-C₆₁- 115
57 butyric acid methyl ester (PC₆₀BM) and C₆₀ as the ESL [24]. 116
58 The high temperature and/or solution processes could 117
59 potentially either thermally decompose the existing per- 118
60 ovskites or dissolve the underlying subcells and there- 119
61 fore would be incapable of producing high-efficiency 120
62 tandem cells. 121

63 Vacuum-processed perovskite-based solar cells have been 122
64 demonstrated by several groups, and the devices have 123
65 exhibited promising efficiencies in both regular and inverted 124
66 architectures [25-31]. The first successful vapor-deposited 125
67 perovskite solar cells with 15% efficiency were presented in 126
68 a regular structure and used high temperature processed 127
69 compact TiO₂ as the ESL [25]. Most high-efficiency vacuum- 128
70 processed perovskite based solar cells used the inverted 129
71 structure [26,27,31], among which the best reported cells 130
72 with the highest PCE of 16.5% employed solution-processed 131
73 poly(3,4-ethylenedioxythiophene):poly(styrene sulfonate) 132
74 (PEDOT:PSS) and various polymers as the HSLs [28]. Other 133
75 groups have also reported efficient cells using vacuum- 134
76 processed perovskite absorbers [26,27,30,31]. Likewise, 135
77 these cells used either high temperature processed compact 136
78 TiO₂ or solution-processed HSL/ESLs. 137

79 Here, we demonstrate efficient annealing-free vacuum- 138
80 processed perovskite solar cells using thermal evaporation 139
81 of both the fullerene C₆₀ as the ESL and the perovskite as 140
82 the light absorber. We used fullerene C₆₀, because it can 141
83 function as an ESL [32] and can also be thermally evapo- 142
84 rated [33,34]. Thermally evaporated C₆₀ films exhibit good 143
85 electronic properties without post-deposition annealing 144
86 [35]. The devices with an ultrathin C₆₀ layer (5.5 nm) 145
87 showed an average PCE of 14.3% and a maximum PCE of 146
88 15.7%. The best-performing cell produced a steady-state 147
89 efficiency of 14.6%. Our detailed device characterization 148
90 revealed that the high performance of our annealing-free 149
91 vacuum-processed perovskite cells resulted from the use of 150
92 an ultrathin C₆₀ layer which effectively transfers the 151
93 electrons and blocks the holes due to the favorable energy 152
94 level alignment between C₆₀ and fluorine-doped tin oxide 153
95 (FTO) electrodes. With the realization of efficient cells, the 154
96 annealing-free vacuum deposition of perovskite absorbers 155
97 and C₆₀ or C₇₀ electron-selective layers and intermediate 156
98 layers demonstrates its power for fabricating all-perovskite 157
99 tandem solar cells. 100

101 Experimental section 102

103 Perovskite synthesis 104

105 The deposition of the perovskite films was carried out using 106
107 a thermal evaporator integrated with a glove box. CH₃NH₃I 108
109 and PbI₂ were loaded into their individual crucibles. Two 109
110 quartz crystal microbalances are placed close to each 110
111 crucible, monitoring the rate of each evaporation source. 111
112 In our approach, it is difficult to calibrate and monitor the 112
113 CH₃NH₃I rate; therefore, it was preferable to monitor the 113
114 CH₃NH₃I vapor phase pressure instead. After the crucibles 114
115 are filled with CH₃NH₃I and PbI₂, the chamber is evacuated 115
116 to a base pressure of 3×10^{-7} Torr. The PbI₂ crucible is 116
117 heated until its deposition rate reaches our target value. 117
118 Meanwhile, the CH₃NH₃I crucible is heated until the vacuum 118
119 pressure is 5×10^{-5} Torr. For an optimal deposition condi- 119
120 tion of the perovskite film, the CH₃NH₃I vapor pressure is 120
121 kept at 5×10^{-5} Torr by controlling the temperature of 121
122 CH₃NH₃I crucible and the deposition rate of PbI₂ was kept at 122
123 0.75 Å/s, a procedure which achieves the best crystalline 123
124 perovskite films. The samples for TRPL measurement were 124
125 coated by poly(methyl methacrylate) (PMMA). 125

Download English Version:

<https://daneshyari.com/en/article/1557312>

Download Persian Version:

<https://daneshyari.com/article/1557312>

[Daneshyari.com](https://daneshyari.com)