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# The influence of SiC particle size and volume fraction on the thermal conductivity of spark plasma sintered UO<sub>2</sub>–SiC composites



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#### ABSTRACT

This study examines the influence of Silicon Carbide (SiC) particle addition on thermal conductivity of  $UO_2$ –SiC composite pellets.  $UO_2$  powder and  $\beta$ -SiC particles of different sizes and of different volume fractions were mechanically mixed and sintered at 1350–1450 °C for 5 min by Spark Plasma Sintering (SPS). The particle size (0.6–55  $\mu$ m diameter) and volume fraction (5–20%) of SiC were systematically varied to investigate their influence on the resulting  $UO_2$ –SiC composite pellet microstructure and the thermal properties. It was found that SiC particle size less than 16.9  $\mu$ m with larger volume fraction is more effective for improving the thermal conductivity of the fuel pellets. Scanning Electron Microscopy examination revealed micro-cracking and interfacial debonding in the composites containing larger size SiC particles (16.9 and 55  $\mu$ m) which resulted in reduced thermal conductivity. For the  $UO_2$ –SiC composite pellets containing 1  $\mu$ m diameter SiC particles, the thermal conductivity increased almost linearly with volume fraction of particles. However, the addition of a larger volume fraction of SiC reduces the amount of heavy metal in the composite pellet and therefore a higher U-235 enrichment is necessary to compensate for the heavy metal loss. The experimental thermal conductivity values of the  $UO_2$ –SiC composite pellets are in good agreement with the theoretical values based on the available model in the literature.

# 1. Introduction

The primary ceramic fuel used in nuclear reactors, uranium dioxide ( $UO_2$ ), has a low thermal conductivity (7.7 W/m K at 300 K [1]) which results in a decrease in the energy output due to low heat transfer efficiency from fuel to coolant. The introduction of high thermal conductivity fuel pellets enables a nuclear reactor to produce more thermal energy while maintaining plant safety due to lower pellet centerline temperature and thermal gradient, resulting in a lower level of fission gas release and thermal cracking.

In recent years, Spark Plasma Sintering (SPS) has evolved as a promising sintering technique for rapid fabrication of UO<sub>2</sub> pellets of required shape and size [2]. In our previous research [3] it has been shown that enhanced thermal conductivity of UO<sub>2</sub>–10 vol%–SiC composite fuel pellets can be fabricated by the Spark Plasma Sintering (SPS) technique. In that study, SPS provided higher density composites, better interfacial contact, and reduced chemical reaction between UO<sub>2</sub> and SiC particles, compared to conventional sintering. SPS pellets also revealed a thermal conductivity increase of up to 62.1% at 900 °C compared to the literature value of UO<sub>2</sub> [1]. Because of its unique and superior properties such as high thermal

conductivity, low neutron cross section, high melting point, and great chemical stability, Silicon Carbide (SiC) was chosen as the secondary phase in the  $UO_2$  matrix to form heat conducting paths in the ceramic composites.

The influence of particle size and volume fraction of second phase particles on the effective thermal conductivity of a composite and its association with the interfacial thermal resistance have been well documented in literature. Hanada et al. [4] determined that diamond particle size and volume fraction had a significant influence on the effective thermal conductivity of copper-diamond composites. They found that the composites containing larger diamond particles up to 7.7 µm diameter and smaller volume fraction near 1% show higher thermal conductivity. Bai et al. [5] fabricated MoSi<sub>2</sub>-SiC composites containing 10, 20, and 30 vol% of 0.5 μm and 100 nm SiC particles and found that the composite showed lower thermal conductivity with decreasing SiC particle size and increasing volume fraction due to the interfacial thermal resistance. Chu et al. [6] sintered Cu-Carbon Nanotube (CNT) composites and found that these composites had much lower thermal conductivity than calculated thermal conductivity value based on the rule of mixture which ignored the interfacial resistance. In general, it is found that ceramic composites containing larger particles (low surface to volume ratio) reduce the interfacial thermal resistance, and hence, potentially increase the effective thermal conductivity of ceramic composites.

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In the current study, a series of  $UO_2$ –SiC composite fuel pellets with different sizes and volume fractions of SiC particles was fabricated using the SPS technique. The thermal conductivities of these composites were measured and compared to the values determined from theoretical formulations available in the literature. During the fabrication process, the sintering parameters such as hold time, up/down ramp rate, and pressure were kept constant so as to investigate only the effects of SiC particle size and volume fraction on the resulting thermal conductivity of the composite pellets.

# 2. Experimental

# 2.1. Fabrication of UO<sub>2</sub>–SiC composite pellets

UO<sub>2</sub>-SiC pellet fabrication procedure using SPS was described in detail in our recent publication [3]. Therefore, only a brief description is provided here. The uranium dioxide  $(UO_{2.11})$  powder was obtained from AREVA NP, Richland, WA and the SiC powder was obtained from Superior Graphite, Inc., Chicago, IL. The SiC powder was reported to have a high purity (>98%), a surface area from gas absorption method [7] of 1.46 m<sup>2</sup>/g, a particle density range of 3.1- $3.2 \text{ g/cm}^3$ , and particle mean diameters of 0.6, 1.0, 9.0, 16.9, and 55 µm. Reported major impurities are listed in Table 1. The UO<sub>2</sub> and SiC powders were mixed in a ceramic vial with stainless steel balls and a blending aid, 2,3-dihydroperfluoropentane, and blended in a SPEX 8000 shaker for 1 h. The blended mixture was then dried up in a hood for at least 6 h to decompose the blending aid. It was assumed that the blending aid was all decomposed because the weight change between unblended powders and dried mixture was less than 0.13%. For each mixing run the SiC mean particle size and the volume fraction of SiC powder in the mixture with UO<sub>2</sub> were varied as shown in Table 2 to investigate their effect on the thermal conductivity of the resulting UO2-SiC composite pellet. The SiC particle sizes used in this study are most widely used in various applications and are also available as high purity (>98%) powders, and hence were obvious choice in our study. SiC particles with 1 µm size at 5, 10, 15, and 20 vol% were chosen to fabricate UO2-SiC composite pellets. This volume range was selected because 20 vol% is the maximum range where the available models are valid (to be described in the next section) so that we can compare the resulting thermal conductivity of the composites with model prediction. Also, excessive addition of SiC in nuclear fuel is unrealistic due to the exorbitant cost and stringent regulation of U-235 enrichment process which is necessary to compensate reduced fissile isotope in UO<sub>2</sub>-SiC composite fuel. SiC particles dispersed in UO<sub>2</sub> powders were then sintered using a Dr. Sinter® SPS-1030 system at 1350 °C and 1450 °C for 5 min in a vacuum ( $\sim$ 30 m Torr). The ramp up/down rate and mechanical pressure at the maximum sintering temperature were held constant at 100 °C/min and 36 MPa, respectively.

The heat treatment procedure described in ASTM C 1430-07 was conducted on the sintered composite pellet to reduce  $UO_{2+X}$  to stoichiometric  $UO_{2,0}$ , which is known to have the optimum thermal properties [8]. This process is also required to produce similar

**Table 1**Reported major impurities and their fractions in SiC powder.

Element	Maximum fraction
Carbon (free)	1.0 wt%
Oxygen	1.0 wt%
Nitrogen	0.4 wt%
Silicon (metallic)	0.03 wt%
Iron	800 ppm
Calcium	400 ppm

O/U ratio in all fabricated UO<sub>2</sub>–SiC composites. The ramp up/down rate and maximum temperature were set at 2.6 °C/min and 800 °C, respectively. The heat treatment was performed in a Lindberg® alumina tube furnace using  $4\%H_2$ –N $_2$  gas with a dew point maintained at 35 °C.

# 2.2. Characterization methods

The weight of each pellet in air and water was measured and the average density was calculated from three weight measurements per pellet using the Archimedes principle. The measured density of the composite was then compared with theoretical density obtained from the rule of mixture [9].

$$\rho_c = \rho_{UO_2}(1 - V_p) + \rho_{SiC}V_p \tag{1}$$

where  $\rho_{UO_2}$ ,  $\rho_{SiC}$ , and  $V_p$  are the densities of  $UO_2$  and SiC, and the SiC volume fraction, respectively.

The microstructure of the fabricated composite pellets were observed using a scanning electron microscope (SEM, JEOL JSM-6335F). The pellets were metallographically polished with successively smaller grit SiC abrasive paper and finally with 0.06  $\mu m$  colloidal silica. The surface was thermally etched at 1340 °C in Ar atmosphere for 4 h to reveal the grain boundaries of UO $_2$  matrix in the composite pellets.

The measurement of thermal diffusivity was carried out at 100, 500, and 900 °C using a laser flash instrument (AnterFlash-line\*3000) with a Xenon discharge pulse for 1  $\mu$ s duration. Three measurements were performed at each temperature on each pellet and the average diffusivity was obtained. The specific heat capacity of UO<sub>2</sub>–SiC composite pellet was calculated using the Neumann–Kopp rule [10] i.e.,

$$C_p = C_{UO_2}(1 - f_p) + C_{SiC}f_p$$
 (2)

where  $C_{UO_2}$ ,  $C_{SiC}$ , and  $f_p$  are theoretical specific heat capacities of UO<sub>2</sub> and SiC, and weight fraction of SiC particles, respectively, at a specific temperature.  $C_{UO_2}$  and  $C_{SiC}$  at 100 °C, 500 °C, and 900 °C are listed in Table 3. The thermal conductivity, K, of composite pellets was then determined from the relation.

$$K = DC_p \rho_c \tag{3}$$

where D and  $\rho_c$  are the thermal diffusivity and density of the composite, respectively.

### 3. Results and discussion

# 3.1. Size effect of SiC particles on $UO_2$ –5 vol% SiC composite properties

The micro-morphologies and thermal properties of  $\rm UO_2-5-vol\%SiC$  composite fuel pellets containing SiC particles with five

**Table 2**Details of SiC particle size, volume fraction, and sintering conditions in the SPS.<sup>a</sup>

SiC particle mean diamter (µm)	SiC volume fraction (%)	Maximum sintering temperature (°C)	% TD of the composite pellet ± SD
0.6	5	1350	95.25 ± 0.24
1	5	1350	95.27 ± 0.3
1	5	1450	96.81 ± 0.39
1	10	1450	96.63 ± 0.35
1	15	1450	$95.14 \pm 0.23$
1	20	1450	94.41 ± 0.3
9	5	1350	95.15 ± 0.09
16.9	5	1350	94.75 ± 0.17
55	5	1350	95.1 ± 0.13

<sup>&</sup>lt;sup>a</sup> Hold time = 5 min; ramp up/down rate = 100 °C/min; pressure = 36 MPa.

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