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Alloy design for Al addition on microstructure and mechanical properties of Ni₃(Si,Ti) alloy

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ABSTRACT

The effect of Al addition on the microstructure and tensile properties of Ni₃(Si,Ti) alloys with an L1₂ ordered structure, which were fabricated through thermomechanical processing from arc-melted ingots, was investigated. Al was added to a Ni₃(Si,Ti) alloy by using two methods such that Al substituted for (1) only Ti and (2) both Ni and Ti along a Ni₃(Si,Ti)–Ni₃Al pseudo-binary line. In the case of the alloys prepared by the former method, the addition of more than 4 at.% Al resulted in a two-phase microstructure consisting of disordered fcc Ni solid solution dispersions in the L1₂ matrix, while in the case of the alloys prepared by the latter method, the addition of 4 at.% Al retained the L1₂ single-phase microstructure. In the case of the 4 at.% Al-added alloys, the room-temperature tensile properties were similar and independent of the alloying methods, whereas the high-temperature yield stress was higher in the alloys prepared by the latter method than in the case of the alloys prepared by the former method. These results suggest that a single-phase microstructure consisting of an entire L1₂ structure is favorable for obtaining high-temperature tensile properties.

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1. Introduction

An L12-type Ni3(Si,Ti) alloy, which has been developed by adding Ti to Ni₃Si [1], possesses many attractive properties such as those of high-temperature structural materials. For example, the Ni₃(Si,Ti) alloy shows a positive temperature dependence of yield strength and exhibits higher strength and peak temperature in the strength versus temperature curve than Ni₃Si [1]. Further, the relatively low density (\sim 8 g/mm³) and the good corrosion resistance of the Ni₃(Si,Ti) alloy are favorable properties for structural materials [2]. In addition, a high tensile ductility over a wide range of temperatures due to the prevention of the propensity for an intergranular fracture is a notable feature of the Ni₃(Si,Ti) alloy [1]. Such high tensile ductility of the Ni₃(Si,Ti) alloy is noteworthy because most intermetallic alloys have limited ductility, particularly at low temperatures. Recently, heavily cold-rolled thin sheets and foils of the Ni₃(Si,Ti) alloy were fabricated from conventional polycrystalline ingots via a repeated warm-rolling and annealing process [3]. The Ni₃(Si,Ti) thin sheet annealed at an intermediate temperature (e.g., around 900 K) exhibited an extremely high tensile strength and yield strength (more than 2 GPa) with a reasonable fracture strain at room temperature and was superior to the commercial nickel alloys and stainless steels below 873 K [3]. However, the tensile strength and the elongation of the annealed $Ni_3(Si,Ti)$ thin sheets were found to rapidly decrease at temperatures beyond 873 K with an increase in the temperature, and therefore, the improvement of the high-temperature mechanical properties is considerably desired.

Thus far, the improvement of high-temperature tensile properties for the Ni₃(Si,Ti) alloy has been attempted by means of an alloving technique [4–6]. Transition metals. Hf. Nb. Cr. Fe and Mn. were added to the Ni₃(Si,Ti) alloy at 1 at.% level below their solubility limits [4]. However, their addition was less effective in improving the high-temperature tensile properties of the Ni₃(Si,Ti) alloy. Further, 2 at.% of Al, Cr, and Mo metals was added to the Ni₃(Si,Ti) alloy [6]. The 2 at.% addition of Al was below the solubility limit in the Ni₃(Si,Ti) matrix; therefore, the new alloy retained the L1₂ single-phase microstructure. On the other hand, the 2 at.% addition of Cr was beyond the solubility limit in the Ni₃(Si,Ti) matrix and consequently resulted in a two-phase microstructure consisting of disordered fcc Ni solid solution dispersions in the L1₂ matrix. It was found that the addition of Al as well as Cr was slightly effective in improving both high-temperature ductility and oxidation resistance [6].

In the present study, we selected Al as an alloying element for the $Ni_3(Si,Ti)$ alloy because addition of Al is expected to yield a number of microstructural, mechanical, engineering, and economic advantages such as low density, low cost, the use of a ubiquitous element, high oxidation resistance, high corrosion resistance, and large solubility in the $L1_2$ matrix. In order to make Al largely soluble in the $Ni_3(Si,Ti)$ matrix, various alloying procedures were

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Table 1Chemical compositions, microstructures, and hardness of the alloys used in this study.

Group	Alloy	Ni (at.%)	Si (at.%)	Ti (at.%)	Al (at.%)	B (wt ppm)	Microstructure	Hardness (HV)
Group A	Base	79.5	11.0	9.5	_	50	L1 ₂	385
	#A2	79.5	11.0	7.5	2.0	50	L1 ₂	376
	#A4	79.5	11.0	5.5	4.0	50	$L1_2 + Ni_{SS}$	375
	#A6	79.5	11.0	3.5	6.0	50	$L1_2 + Ni_{SS}$	367
	#A8	79.5	11.0	1.5	8.0	50	$L1_2 + Ni_{SS}$	372
Group B	#B4	78.9	11.0	6.1	4.0	500	L1 ₂	389
	#B8	78.3	11.0	2.7	8.0	500	$L1_2 + Ni_{SS}$	391

taken into consideration. In this study, the microstructure, cold workability, and high-temperature tensile properties of the Aladded Ni₃(Si,Ti) alloys were investigated, and the alloy composition and the microstructure preferable for high-temperature structural materials were suggested.

2. Experimental procedure

The base composition of the Ni₃(Si,Ti) alloy used in this study was 79.5 at.% Ni, 11.0 at.% Si, and 9.5 at.% Ti doped with 50 wt ppm B. Boron doping of the Ni₃(Si,Ti) alloy is necessary to suppress the intergranular fracture due to environmental embrittlement at room temperature [7]. The quaternary element Al was added to the base alloy composition by using two methods: (1) Al substituted only for Ti and (2) Al substituted both for Ni and Ti. The samples prepared by the two methods are hereafter referred to as groups A and B, respectively (Table 1). In the case of group A, the addition of 2, 4, 6 and 8 at.% Al content was designed to reduce the amount of Ti that was harmful for the oxidation resistance of the Ni₃(Si,Ti) alloy [8]. In the case of group B, the addition of 4 and 8 at.% Al content was designed to reduce the amount of Ni and Ti, forming a continuous L12 solid solution along the pseudo-binary Ni₃(Si,Ti)-Ni₃Al line, as shown in Fig. 1. In the alloys of group B, an increased amount of 500 wt ppm B was added because a higher boron content was required in Ni₃Al than in Ni₃(Si,Ti) to suppress the intergranular fracture [9].

Alloy ingots were prepared by non-consumable arc melting in an argon gas atmosphere on a copper hearth. All the ingots were homogenized at 1323 K for 48 h in a vacuum and then cut into several plates with a thickness of approximately 10 mm using an

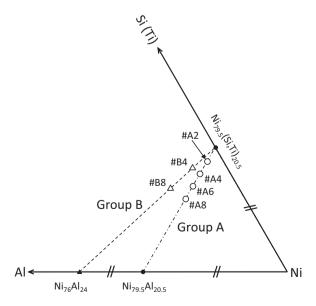


Fig. 1. Alloy compositions used in this study. Note that Al contents were changed along two pseudo-binary lines, $Ni_{79.5}(Si,Ti)_{20.5}-Ni_{79.5}Al_{20.5}$ (group A) and $Ni_{79.5}(Si,Ti)_{20.5}-Ni_{76}Al_{24}$ (group B).

electro-discharge machine (EDM). These plates were warm-rolled at 573 K and then annealed at 1273 K for 5 h in a vacuum. This procedure was repeated until the thickness of the plate became 2 mm. The sheets were finally annealed at 1273 K for 1 h in a vacuum before cold rolling. Cold rolling was conducted without intermediate annealing up to 90% reduction in thickness. Reduction in sheet thickness per rolling pass was approximately 0.1 mm both for warm and cold rolling. The thickness of the obtained cold-rolled sheets was approximately 0.2 mm. Microstructural observations were carried out by using a scanning electron microscope (SEM; JEOL, JSM-5600 operating at 15 kV) and a transmission electron microscope (TEM; JEOL, JEM-2000FX operating at 200 kV). Further, X-ray diffraction (XRD; RIGAKU, RINT2500HK) was performed to determine constituent phases and second-phase dispersions by using Ni-filtrated Cu K α radiation at an accelerated voltage of 30 kV mostly with a scan speed of 1°. In the XRD measurement, bulk materials of the alloys were used as the specimens.

Dog-bone-shaped tensile test pieces with a length of $10 \, \mathrm{mm}$, width of $4 \, \mathrm{mm}$, and thickness of $0.2 \, \mathrm{mm}$ were prepared by using the EDM. Room-temperature tensile tests were carried out in air by using the cold-rolled and fully recrystallized ($1173 \, \mathrm{K}$; $1 \, \mathrm{h}$) specimens. High-temperature tensile tests were conducted in a vacuum in a temperature range of room temperature to $973 \, \mathrm{K}$ by using the fully recrystallized specimens. A strain rate of $8.4 \times 10^{-5} \, \mathrm{s}^{-1}$ was used in both room-temperature and high-temperature tensile tests. After the tensile tests, the fracture surfaces of the test pieces were examined by using an SEM.

3. Results and discussions

3.1. Microstructure

Fig. 2 shows the SEM microstructures of the alloys in group A (i.e., alloys #A2, #A4, #A6 and #A8) after homogenization at 1323 K. Alloy #A2 as well as the base alloy (not shown here) showed an L12 single-phase microstructure, indicating that 2 at.% Al is soluble in the L12 matrix. On the other hand, the other alloys (i.e., alloys #A4, #A6 and #A8) exhibited a two-phase microstructure containing second-phase dispersions. At Al contents above 2 at.%, the volume fraction of the second-phase dispersions increased with an increase in the Al content. Bimodal microstructures were more or less observed in the alloys with a two-phase microstructure: eutectic-like two-phase microstructures consisting of L12 and Al (fcc) phases were mixed with a featureless L1₂ single-phase region. This eutectic-like microstructural feature may be explained such that the Al phase with Ni-rich content is formed as a primary phase and present in the dendritic core region during solidification and remains as a two-phase microstructure after homogenization.

Fig. 3 shows the XRD profiles of alloy #A4 with a two-phase microstructure, together with the alloys with a single-phase microstructure (i.e., the base alloy and alloy #B4 (see later)). The XRD profiles of all alloys shown in Fig. 3 were basically identical to each other, although the relative peak intensities of each diffraction were slightly different among the shown alloys because of the crystallographic textures of these alloys. The observed reflection

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