



Thermal and mechanical properties of Yb&Mg co-doped InFeZnO₄



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ABSTRACT

In_{0.4}Yb_{0.6}FeZnO₄ was confirmed to possess the lowest thermal conductivity in Yb-doped InFeZnO₄. To further reduce the thermal conductivity, In_{0.4}Yb_{0.6}FeZn_{1-x}Mg_xO₄ ($x = 0, 0.1, 0.2, \dots, 0.8, 0.9, 1.0$) ceramics were synthesized by solid state reaction at 1300 °C for 30 h. XRD results indicated that single phase In_{0.4}Yb_{0.6}FeZn_{1-x}Mg_xO₄ ceramics were successfully synthesized with layered YbFe₂O₄-type crystal structure in the hexagonal $R\bar{3}m$ space group. It was found that the thermal conductivity of In_{0.4}Yb_{0.6}FeZn_{0.5}Mg_{0.5}O₄ ($x = 0.5$) reached a minimum value of 1.13 W m⁻¹ K⁻¹ at 1000 °C. The investigations on the thermal stability, thermal expansion coefficient, abradable property and erosion wear resistance indicate that In_{0.4}Yb_{0.6}FeZn_{0.5}Mg_{0.5}O₄ ceramics are promising thermal barrier coatings.

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1. Introduction

Thermal barrier coatings (TBCs) are widely used to protect hot sections of metallic components in advanced turbine engines, which can increase the inlet temperature with a consequent improvement of the efficiency, or reduce the requirements for cooling systems [1]. One of the most common TBC material is 8 wt% yttria stabilized zirconia (8YSZ), as it has low thermal conductivity (2.1 W m⁻¹ K⁻¹), relatively high thermal expansion coefficient (11×10^{-6} K⁻¹) and chemical inertness in combustion atmospheres. However the phase transition of YSZ leads to instability in long-term service over 1200 °C [2]. Therefore, an urgent demand for alternative TBC materials is raised. Researches on the rare-earth zirconates, Ln₂Zr₂O₇ (Ln = rare-earth elements) have revealed the promising thermo-physical properties, such as thermal conductivity values range from 1.1 to 1.2 W m⁻¹ K⁻¹, which is much lower than that of YSZ [3]. However, thermal expansion coefficients of Ln₂Zr₂O₇ ceramics are lower than those of metallic substrates, resulting in high thermal stress in TBCs applications [4].

Generally, the selection of TBC material is restricted by several requirements: low thermal conductivity, high melting temperature, high thermal expansion coefficient and phase stability at elevated temperature [5,6]. Recently, a new compound, InFeZnO₄, was discovered as a promising TBC. InFeZnO₄ possesses low thermal conductivity (1.36 W m⁻¹ K⁻¹ at 1200 °C), high thermal

expansion coefficient (11.7×10^{-6} K⁻¹ at 1200 °C) and no phase transition until 1400 °C [7,8], these outstanding properties meet the requirement of TBCs. Investigations of In_{1-x}Yb_xFeZnO₄ ($x = 0, 0.1, 0.2, \dots, 0.8, 0.9, 1.0$) show that the thermal conductivity of In_{0.4}Yb_{0.6}FeZnO₄ ($x = 0.6$) could achieve a minimum of 1.19 W m⁻¹ K⁻¹ at 1000 °C [9], indicating that the thermal conductivity could be reduced by introducing point defects through Yb doping. In this paper, we selected In_{0.4}Yb_{0.6}FeZnO₄ as matrix and doped Zn sites with different Mg fractions, In_{0.4}Yb_{0.6}FeZn_{1-x}Mg_xO₄ ($x = 0, 0.1, 0.2, \dots, 0.8, 0.9, 1.0$) ceramics were synthesized and the thermal transport properties were investigated. It was found that the thermal conductivity of In_{0.4}Yb_{0.6}FeZn_{0.5}Mg_{0.5}O₄ ($x = 0.5$) reached the minimum value of 1.13 W m⁻¹ K⁻¹ at 1000 °C, which is ~5% lower than that of In_{0.4}Yb_{0.6}FeZnO₄ and ~22% lower than that of InFeZnO₄. We found that there is no transformation of structure and composition occurred up to 1350 °C and the crystal structure (YbFe₂O₄ type) maintained after annealing treatment at 1350 °C for 100 h. The thermal expansion coefficient increased along with the elevating temperature and reached the value of 11.63×10^{-6} K⁻¹ at 1350 °C, which is higher than that of Sm₂Zr₂O₇, a promising coating material. The friction coefficients of In_{0.4}Yb_{0.6}FeZn_{0.5}Mg_{0.5}O₄ at room temperature, 400 °C and 800 °C are lower than those of 8YSZ. Rockwell hardness of In_{0.4}Yb_{0.6}FeZn_{0.5}Mg_{0.5}O₄ (73HR_{45Y}) was lower than that of 8YSZ (87HR_{45Y}). Erosion rate of In_{0.4}Yb_{0.6}FeZn_{0.5}Mg_{0.5}O₄ decreased with increasing impact angle and temperature. Under the experimental condition of (30°, 60°, 90°) × 20 °C and 60° × (20 °C, 400 °C, 800 °C), the erosion rates varied between 2.339×10^{-4} cm³ g⁻¹ and 36.486×10^{-4} cm³ g⁻¹.

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2. Experimental

2.1. Synthesis

$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.5}\text{Mg}_{0.5}\text{O}_4$ samples ($x = 0, 0.1, 0.2, \dots, 0.8, 0.9, 1.0$) were prepared by means of solid-state synthesis using In_2O_3 (99.99%), Yb_2O_3 (99.99%), Fe_2O_3 (99.9%), ZnO (99.99%) and MgO (99.99%) as starting powders. All powders were calcined at 600 °C for 6 h in air to remove the possibly absorbed water. Then the starting powders were weighed in mole ratio of $\text{In}_2\text{O}_3:\text{Yb}_2\text{O}_3:\text{Fe}_2\text{O}_3:\text{ZnO}:\text{MgO} = 0.4:0.6:1:2(1-x):2x$, dissolved in absolute ethyl alcohol and fully mixed by planetary ball mill (QM-3SP4) at 400 r/min for 10 h. The mixtures were dried in oven at 110 °C for 2 h. The obtained powders were cold pressed in a $\Phi 15$ mm steel die under uniaxial pressure of 250 MPa, and then sintered in BeO crucibles at 1300 °C under air for 30 h.

2.2. X-ray diffraction and scanning electron microscopy

The phase structure of synthesized samples was measured by X-ray diffraction (XRD, Rigaku D/Max 2200 PC, Cu α). The fracture morphology and composition of $\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.5}\text{Mg}_{0.5}\text{O}_4$ samples ($x = 0, 0.1, 0.2, \dots, 0.8, 0.9, 1.0$) were analyzed by scanning electron microscope (SEM, CamScan 3400) and electron probe micro-analyzer (EPMA) with energy dispersive spectroscopy (EDS), respectively.

2.3. Thermal conductivity

Thermal conductivity (λ) was calculated by Eq. (1) with thermal diffusivity (α), specific heat capacity (C_p) and density (ρ).

$$\lambda = C_p \cdot \alpha \cdot \rho \quad (1)$$

Thermal diffusivity (α) was measured using a laser-flash apparatus (Netzsch LFA 427) in atmosphere of argon at 25 °C–1000 °C. Dimensions of the samples were about 12.7 mm in diameter and about 1.5 mm in thickness. The specific heat capacities (C_p) were calculated by the Neumann-Kopp rule [10] using heat capacities of In_2O_3 , Yb_2O_3 , Fe_2O_3 , ZnO and MgO obtained from the thermodynamic database [11]. The uncertainty of the thermal conductivity is estimated to be within 10%, comprising uncertainties of 3% for the thermal diffusivity, 5% for the specific heat, and 3% for the sample

density. As the sintered samples were not fully dense, the porosities (Φ) were calculated by experimental densities (ρ) and theoretical densities (ρ_0) using Eq. (2). Experimental densities were determined by Archimedes' method and theoretical densities were calculated with lattice parameters obtained from XRD patterns and relative molecular mass of compounds. The measured thermal conductivities were extrapolated to the fully dense case by Eq. (3), where λ_0 is thermal conductivity of fully-dense sample and λ is measured thermal conductivity.

$$\Phi = 1 - \frac{\rho}{\rho_0} \quad (2)$$

$$\frac{\lambda}{\lambda_0} = 1 - \frac{4}{3}\Phi \quad (3)$$

2.4. Thermo-physical properties

Thermal stability of $\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.5}\text{Mg}_{0.5}\text{O}_4$ was investigated by differential scanning calorimetry (DSC) up to 1400 °C with heating rate of 15 °C/min in argon atmosphere. The high temperature annealing treatment was carried out under the condition of 1350 °C \times 100 h. Thermal expansion coefficients (TECs) of the bulk material was measured using putter method and was calculated by Eq. (4) with the original length of the cylinder bulk material (l_0), temperature increment (dT) and length increment (dl). The sample size is 4 mm \times 4 mm \times 25 mm.

$$\text{TECs} = \frac{1}{l_0} \frac{dl}{dT} \quad (4)$$

2.5. Mechanical properties

2.5.1. Abradable property

Abradable property of $\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.5}\text{Mg}_{0.5}\text{O}_4$ was characterized by friction coefficient and Rockwell hardness. The friction coefficients were measured by SRV-4 high-temperature friction and wear test machine for 20 min at room temperature, 400 °C and 800 °C, respectively, using $\Phi 24$ mm \times 7.88 mm sample. The Rockwell hardness of $\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.5}\text{Mg}_{0.5}\text{O}_4$ was measured using HSRD-45 electric surface Rockwell hardness test machine under 45 kg pressure and HR_{45Y} staff gauge. 8YSZ was selected as a reference material in these two experiments.

2.5.2. Erosion wear resistance

Erosion wear resistance of $\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.5}\text{Mg}_{0.5}\text{O}_4$ material is

Table 1

Experimental densities (ρ), theoretical densities (ρ_0) and porosities (Φ) of $\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{1-x}\text{Mg}_x\text{O}_4$ samples.

Sample	ρ	ρ_0	Φ
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZnO}_4$	5.574	6.610	15.64%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.9}\text{Mg}_{0.1}\text{O}_4$	5.729	6.508	11.97%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.8}\text{Mg}_{0.2}\text{O}_4$	5.339	6.417	16.79%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.7}\text{Mg}_{0.3}\text{O}_4$	5.092	6.336	19.63%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.6}\text{Mg}_{0.4}\text{O}_4$	5.599	6.245	10.35%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.5}\text{Mg}_{0.5}\text{O}_4$	5.488	6.153	10.81%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.4}\text{Mg}_{0.6}\text{O}_4$	5.189	6.091	14.82%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.3}\text{Mg}_{0.7}\text{O}_4$	5.083	6.020	15.57%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.2}\text{Mg}_{0.8}\text{O}_4$	4.856	5.942	18.27%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{0.1}\text{Mg}_{0.9}\text{O}_4$	4.828	5.861	17.62%
$\text{In}_{0.4}\text{Yb}_{0.6}\text{FeMgO}_4$	4.898	5.783	15.30%

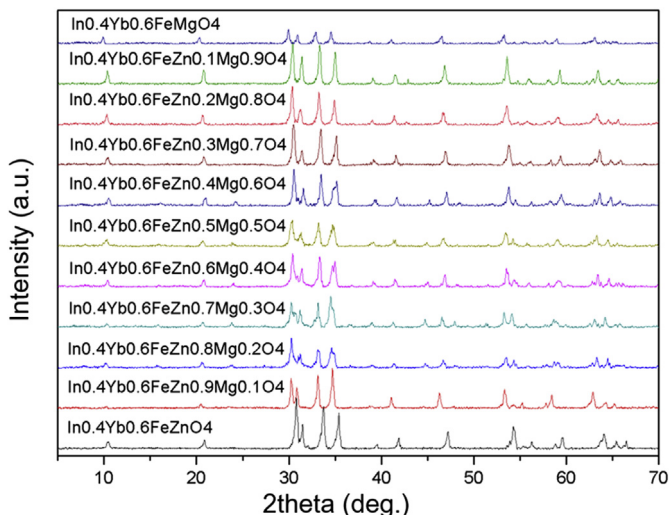


Fig. 1. Powder XRD patterns of $\text{In}_{0.4}\text{Yb}_{0.6}\text{FeZn}_{1-x}\text{Mg}_x\text{O}_4$.

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