



Letter

Preparation of strong and ductile pure titanium via two-step rapid sintering of TiH₂ powder

A B S T R A C T

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Titanium hydride

Pure-Ti

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The present work demonstrates the feasibility of preparing bulk-Ti, with high strength and good ductility, via spark plasma sintering of TiH₂ powders. The microstructure and mechanical properties of bulk titanium prepared under two different processing conditions, i.e. (i) Simultaneous dehydrogenation and sintering under one-step spark plasma sintering (SPS), and (ii) dehydrogenation in vacuum followed by Spark Plasma Sintering i.e. two-step SPS, are presented and discussed. The one-step sintering process, i.e. process (i), resulted in Ti specimens with very high tensile strength, but poor ductility. On the other hand, two-step sintering process, i.e. process (ii), resulted in bulk Ti with high strength and high ductility.

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1. Introduction

Titanium alloys have exceptional potential for expanded usage in various industries. However, initial high cost of Ti and the complex fabrication and subsequent machining process often limit their applications. However, powder metallurgy (P/M) processing has emerged as an efficient method with a possibility to produce titanium alloys with wide range of compositions, controlled properties, and near net shape characteristics. Particularly, P/M processing approach based on elemental powders are of particular interest due to relatively lower processing cost as well as its higher maneuverability with respect to materials, microstructural, and product designing [1,2]. However, the usage of pure-Ti powder poses several problems, such as high affinity toward oxygen, nitrogen, and carbon, sticking with vials and balls during blending and milling, and higher cost associated with preparing high-purity Ti powders. These factors lead to lack of compositional/microstructural control together with less than expected cost advantages [3,4]. In comparison to pure-Ti powder, use of titanium hydride powder has several advantages, including lower initial price, less impurity contents, and relative ease of handling during mechanical mixing/milling without encountering the problem of sticking due to its inherent brittleness, leading to higher powder yield, smaller particle size, and better compositional control. The fine-sized titanium hydride powder can be utilized to obtain sintered products with high density, fine grained microstructure, and relatively higher strength at a relatively lower sintering temperature [5,6].

Various techniques are used for the consolidation of powders [7–10]. Particularly, spark plasma sintering technique has several advantages over the conventional hot pressing, including higher heating/cooling rates and a greater applied mechanical load. Hence, SPS technique offers a possibility to achieve dehydrogenation and

consolidation of titanium-hydride powder in a comparatively shorter period of time, together with high density products with controlled microstructure and properties [11]. However, a very small amount of information is available in the literature on the TiH₂ based powder metallurgy technology and mechanical properties of bulk-Ti alloys prepared using titanium hydride powders [12–16, Table 2]. Therefore, it would be interesting to carry out a feasibility study about the microstructure and mechanical properties of bulk-Ti prepared via SPS of TiH₂ powder, to realize the full potential of elemental powder based P/M approach using titanium hydride powders. Hence, the present study is an initiative to evaluate the feasibility of fabricating high purity bulk-Ti via spark plasma sintering using titanium hydride powder as starting material. The microstructure and mechanical properties of the fabricated Ti compacts, thus prepared, are presented and discussed.

2. Materials and methods

TiH₂ powder (45 μm pass, ~99% pure, supplied by “Kojundo Chemical Laboratory Co. Ltd.” Japan) and pure-Ti (45 μm pass, 99.98% pure, Gas-atomized low oxygen titanium powder, supplied by “Osaka Titanium Technologies Co. Ltd”, Japan) were used as starting materials. The sintering was carried out using graphite die and punch under high vacuum atmosphere (~10⁻³ Pa) in DR.SINTER Spark Plasma Sintering machine (“Sumitomo Coal Mining Co. Ltd.”, Japan). The powders were consolidated using two different approaches: (i) One-step SPS process: the pressure and temperature were increased simultaneously from zero pressure to 50 MPa and room temperature to 1200 °C, respectively. The holding time at 1200 °C was 1 h. Subsequently, the furnace was cooled to room temperature. High purity Ti powder was also sintered under similar conditions. (ii) Two-step SPS process: The first step

Table 1
Calculated lattice parameters of Ti-1 and Ti-2 specimens.

Lattice parameters	Ti-1 specimen			Ti-2 specimen		
	a (Å)	c (Å)	c/a (Å)	a (Å)	c (Å)	c/a (Å)
α -phase	2.94977	4.6932	1.591	2.9471	4.6884	1.590
δ -phase	4.41	–	–	4.39	–	–
γ phase	–	–	–	4.235	4.623	1.091

involved dehydrogenation at 800 °C for 0.5 h without any applied external pressure for maximum dehydrogenation. In the second step, temperature and pressure were increased, simultaneously, from 800 °C to 1200 °C and zero pressure to 50 MPa, respectively. Subsequently, the powder mass was held at 1200 °C for 0.5 h followed by furnace cooling up to room temperature. After sintering, disc-shaped compacts with dimensions 15 mm (diameter) \times 5 mm (height) were obtained. Henceforth, in the subsequent discussion, the specimens prepared by one-step sintering of TiH₂, two-step sintering of TiH₂, and one-step sintering of pure Ti powders will be referred as Ti-1, Ti-2, and Ti-P specimens, respectively. Tensile tests were carried out at an initial strain rate of 5.6×10^{-4} using specimens with gauge size 3 mm \times 1 mm \times 1 mm. Average Vickers micro-hardness was obtained at a load of 980.7 mN (Hv0.1) and dwell time 10 s. Moreover, dehydrogenation temperature of TiH₂ was analyzed by TG/DTA (Thermogravimetric and differential thermal analysis) and at a constant heating rate of 20 K min⁻¹ under argon atmosphere.

3. Results and discussion

3.1. Initial powders

The results of XRD analysis of the initial powders are shown in Fig. 1a. For pure Ti and TiH₂ powders, hexagonal close-packed (hcp) and cubic phases were identified, respectively. The XRD pattern of the titanium hydride powder was similar to the standard data corresponding to the hydride powder containing 95.96 mass% titanium and 4.04 mass% hydrogen. Fig. 1 (b) and 1(c) shows DTA and TGA plots, respectively, obtained from as received TiH₂ powder. The presence of two overlapping endothermic events clearly suggests that the decomposition process occurs in two stages. The DTA plot shows one small peak (indicated by T₁) at 469 °C due to low enthalpy change followed by second large peak (indicated by T₂) at 536 °C due to comparatively larger enthalpy change. Thermo-gravimetric analysis confirms that dehydrogenation starts above approximately 440 °C and completes at approximately 700 °C (Fig. 1c). A first hand estimation from TG analysis shows

that approximately 3.53 mass% hydrogen was lost, which is slightly lower than the theoretical hydrogen content, i.e. 4.04 mass%, in the initial hydride powder. Similar findings have also been reported by other researchers [17,18].

3.2. Microstructural characteristics of the sintered bulk-Ti specimens

In Fig. 2 (a), the XRD profiles of Ti-1 and Ti-2 specimens are compared with the Ti-P specimen. The diffraction patterns of Ti-P specimen consist of hcp α -phase only. However, Ti-1 specimens indicated the presence of δ -phase (tetragonal hydrides) along with α -phase. Interestingly, the XRD profile of Ti-2 specimen exhibit the presence of primarily α -phase with minor amounts of δ and γ -phases. In the Ti-H system, following hydride phases are known to exist; (i) δ -phase (51.22–66.67 at.% hydrogen) with f.c.c. structure of CaF₂ type in which hydrogen atoms occupy tetrahedral sites randomly and (ii) the γ hydrides (1–2.9 at.% hydrogen) with f.c.t. structure and having c/a \sim 1.09 [19]. These results indicate that a significant quantity of hydrogen is retained in the Ti-1 specimen. In the Ti-2 specimen, the presence of γ -phase indicates the retention of relatively lower amounts of hydrogen as compared to that of Ti-1 specimen. Therefore, it is apparent that the two-step process is more effective in achieving sintered compacts with significantly small amounts of retained hydrogen. In titanium, for hydrogen concentrations lower than 20 at.%, in addition to the strong diffraction peaks from α phase, the weak diffraction peaks i.e. {111} {002} and {200} from the γ -phase and very weak diffraction peaks i.e. {111} from δ -phase was also reported by Nakamura et al. [19].

In addition to the above results, the relative amounts of hydrogen in the pure titanium compacts, prepared from titanium hydride powders, was also estimated by calculating lattice parameters using the XRD profiles (shown in Table 1). It can be observed that the c/a ratio of α -phase in Ti-1 specimen is larger than that calculated for Ti-2 specimen. In general, increasing the hydrogen concentration leads to an increase in c/a ratio and to a negligible increase in the specific lattice volume [20,21]. The calculated lattice parameters for δ and γ -phases were found in good agreement with the results presented by A. San-Martin et al. [20].

The Ti-P specimen exhibited lath-type microstructure with basket-weave type morphology, having lath-width \sim 20 μ m and lath-length in the range of 50–200 μ m (Fig. 2b). Generally, in Ti alloys, such microstructure is developed due to slow cooling from β -region. The α -phase begins to form below the beta transus, which is about 882 °C. The α -phase forms in laths, with a crystallographic relationship, i.e. most close-packed plane of α -phase parallel to a

Table 2
Comparison of mechanical properties of bulk-Ti prepared by TiH₂ powder.

Initial material	Method of preparation	Sintering temp. (K)	Vickers hardness (Hv)	YS (MPa)	UTS (MPa)	Elongation (%)	Ref.
TiH ₂ powder	Press & sinter	1200	N/A	N/A	N/A	N/A	10
Ti + TiH ₂ powder	Hot isostatic pressing	1020	50 \pm 2 HRB (Rockwell B)	265	310	N/A	11
TiH ₂ powder	Press & sinter	(1)900 (2)1200 (3)1300	170–225 290–320 320–340	N/A	N/A	N/A	12
TiH ₂ powder	Spark plasma sintering	(1)1000 (2)1050 (3)1100	360–420	N/A	N/A	N/A	13
Pure-Ti powder	P/M compact, annealed	N/A	N/A	370	480	18	19
Pure-Ti powder	One-step SPS	1200	180	280	365	24	Present study
TiH₂ powder	One-step SPS	1200	267	575	690	3.5	Present study
TiH₂ powder	Two-step SPS	1200	243	360	530	16	Present study

(N/A = data not reported).

The bold words/data is from present study and others are from literature. The present study was compared with the data available in the literature.

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