



# Structural, electrical, optical and magnetic properties of p-type $\text{Cu}(\text{Cr}_{1-x}\text{Mn}_x)\text{O}_2$ thin films prepared by pulsed laser deposition



Xiaoshan Zhou, Fangting Lin\*, Wangzhou Shi, Aiyun Liu

Key Laboratory of Optoelectronic Material and Device, Department of Physics, Shanghai Normal University, Shanghai 200234, PR China

## ARTICLE INFO

### Article history:

Received 29 March 2014  
Received in revised form 4 June 2014  
Accepted 18 June 2014  
Available online 27 June 2014

### Keywords:

Semiconductors  
Magnetic films and multilayers  
Microstructure  
Electrical transport  
Optical properties  
Magnetization

## ABSTRACT

$\text{Cu}(\text{Cr}_{1-x}\text{Mn}_x)\text{O}_2$  delafossite thin films ( $0 \leq x \leq 15$  at.%), with a *c*-axis quasi-epitaxial orientation, were prepared by pulsed laser deposition. The effects of Mn content on microstructure and physical properties were investigated. Both the variation of lattice constant *c* and X-ray photoelectron spectroscopy reveal that the Mn ions substitute for  $\text{Cr}^{3+}$  as  $\text{Mn}^{3+}$  and  $\text{Mn}^{4+}$ . The proportion of  $\text{Mn}^{4+}$  is gradually increased with the Mn doping, leading to the changes of hole density and mobility. As a result, the p-type conductivity is decreased at first and then increased. The conduction mechanism is thermal activation between 130 and 300 K. It is indicated that the PLD method and the  $\text{Mn}^{3+}\text{--O--Mn}^{4+}$  and  $\text{Mn}^{3+}\text{--O--Cr}^{3+}$  double-exchange interactions are favorable to the high hole density and mobility, respectively. Meanwhile, the double-exchange interactions produce the near-room-temperature ferromagnetism. The saturation magnetization and Curie temperature are gradually increased with the Mn content. For optical properties, all thin films have a comparatively high transmittance for visible light, with the highest value 70% for  $x = 15$  at.% at the wavelength of 750 nm. As increasing the Mn content, the transparency and direct optical bandgap exhibit the similar trend as the hole density. With the relatively balanced electrical, magnetic and optical properties, these pulsed-laser-deposition-grown  $\text{Cu}(\text{Cr}_{1-x}\text{Mn}_x)\text{O}_2$  thin films are expected to become a promising p-type transparent diluted magnetic semiconductor material.

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## 1. Introduction

In the semiconductor industry there is an uninterrupted evolution towards higher and higher integration which results in capability increase of devices and dimension reduction of elements. Hence numerous researches on multifunctional materials, e.g. diluted magnetic semiconductors (DMSs), ferroelectric semiconductors, ferroelectric photovoltaic materials and multiferroic materials, have been motivated in both technological and scientific fields. Oxide-based DMSs exhibit unique magnetic, magneto-electrical and magneto-optical effects as well as high transparency in the visible region. They can be used in spintronic devices and transparent optoelectronic devices, such as spin-valve transistors, spin light-emitting diodes, nonvolatile logic devices, transparent electrodes in flat panel displays, and window layers in solar cells.

Especially, p-type oxide-based DMSs have substantially aroused the researchers' enthusiasm, because theoretical predictions [1–3] pointed out that ferromagnetic coupling between magnetic ions could be formed more easily in p-type semiconductors due to hole carriers serving as a medium. In such a way, some key issues exist-

ing in the present DMSs are expected to be alleviated, e.g. small magnetization and low Curie temperature ( $T_C$ ).

As one typical p-type transparent conducting oxide (TCO),  $\text{CuCrO}_2$  delafossite with the  $\text{ABO}_2$  structure is considered as an excellent candidate for the matrix semiconductor. Its conductivity ( $\sigma = 220 \text{ S cm}^{-1}$ ) is the highest one among all p-type TCOs [4], which is beneficial to the magnetic exchange mechanism [1–3]. Low-temperature ferromagnetism has been successively achieved in  $\text{Cu}(\text{Cr}_{1-x}\text{M}_x)\text{O}_2$  ( $\text{M} = \text{Mn}$  [5], Ni [6,7], Al [8], Rh [9], Co [10]) ceramics. In 2011, Wang et al. [11] synthesized  $\text{Cu}(\text{Cr}_{1-x}\text{Mn}_x)\text{O}_2$  thin films by chemical solution deposition and obtained room-temperature ferromagnetism which was ascribed to the double-exchange (DE) interaction between  $\text{Mn}^{3+}$  and  $\text{Cr}^{3+}$ . However, the highest transmittance in the visible range, about 65%, was not satisfactory, and the resistivity ( $\rho$ ) increased rapidly with the Mn content. The lowest  $\rho$  value from un-doped  $\text{CuCrO}_2$  film was  $74 \text{ } \Omega \text{ cm}$ , whereas those of Mn-doped ones were even about four orders of magnitude higher.

In our work, a physical method, pulsed laser deposition (PLD), was adopted to prepare *c*-oriented Mn-doped  $\text{CuCrO}_2$  thin films. The near-room-temperature ferromagnetism is achieved, and the highest transmittance for visible light is improved by 5% compared with Ref. [11]. More important, the p-type conductivity is

\* Corresponding author. Tel.: +86 21 64322726; fax: +86 21 64328894.

E-mail address: [nounou7@163.com](mailto:nounou7@163.com) (F. Lin).

remarkably enhanced. In comparison with the similar compositions in Ref. [11], the  $\sigma$  values of our un-doped and Mn-doped  $\text{CuCrO}_2$  films are respectively about two and three orders of magnitude higher. Our thin films possess relatively balanced electrical, optical and magnetic properties, showing promise for applications in multifunctional devices.

## 2. Experimental

$\text{Cu}(\text{Cr}_{1-x}\text{Mn}_x)\text{O}_2$  thin films ( $x = 0, 5, 10$  and  $15$  at.%) were grown on  $\text{Al}_2\text{O}_3(001)$  substrates by PLD (KrF excimer laser,  $\lambda = 248$  nm). The targets were prepared by the solid-state reaction technique. High-purity (99.99%)  $\text{CuO}$ ,  $\text{Cr}_2\text{O}_3$  and  $\text{MnO}_2$  powders based on the stoichiometric compositions were mixed, ground and calcined at  $1050^\circ\text{C}$  for 24 h. Subsequently, the powders were pressed and sintered at  $1200^\circ\text{C}$  for 48 h. The substrates with a size of  $1.5 \times 1.5$  cm<sup>2</sup> were cleaned with ethanol and acetone in an ultrasonic cleanser and rinsed with deionized water. The substrate was set parallel to the target at a distance of 6 cm. In order to obtain uniform thin films, both the target and substrate were rotated at a speed of 6 rpm during the deposition. Thin films were deposited at  $750^\circ\text{C}$  for 15 min in oxygen atmosphere with a pressure of 5 Pa. The base pressure was lower than  $1 \times 10^{-4}$  Pa. The pulse energy and repetition rate were fixed at 200 mJ and 8 Hz, respectively.

Crystal structure was analyzed by X-ray diffraction (XRD) on a Bruker D8 Focus diffractometer with  $\text{Cu K}\alpha$  radiation. The thickness of thin films was determined by a Bruker DektakXT Stylus Profiler. Valence states of ions were verified by X-ray photoelectron spectroscopy (XPS) using a Perkin-Elmer PHI 5000C ESCA system with Al  $\text{K}\alpha$  radiation. The DC resistivity was measured with a four-point probe configuration. The carrier type, density and mobility were determined at room temperature by a Lake Shore 7600 hall measurement system (HMS). The optical transmission spectra were recorded using an Aquila NKD-8000 spectrophotometer. The magnetization vs. magnetic field ( $M-H$ ) curves and the magnetization vs. temperature ( $M-T$ ) curves were measured on a Quantum Design PPMS-9T (EC-II) physical property measurement system (PPMS), with an applied magnetic field parallel to the film plane.

## 3. Results and analysis

### 3.1. Structural properties

#### 3.1.1. XRD characterization

Fig. 1 gives the XRD patterns for  $\text{Cu}(\text{Cr}_{1-x}\text{Mn}_x)\text{O}_2$  thin films. There is no structural difference among four samples. All films are crystallized in a single 3R- $\text{CuCrO}_2$  delafossite structure (JCPDF No. 89-0539). No evidence for impurity formation (e.g. manganese oxides) or phase separation is found. Therefore the Mn ions seem to have been incorporated into the host lattice, substituting for  $\text{Cr}^{3+}$ . Moreover, only (003), (006), (009) and (0012) diffraction peaks from  $\text{CuCrO}_2$  delafossite are observed. It means that these

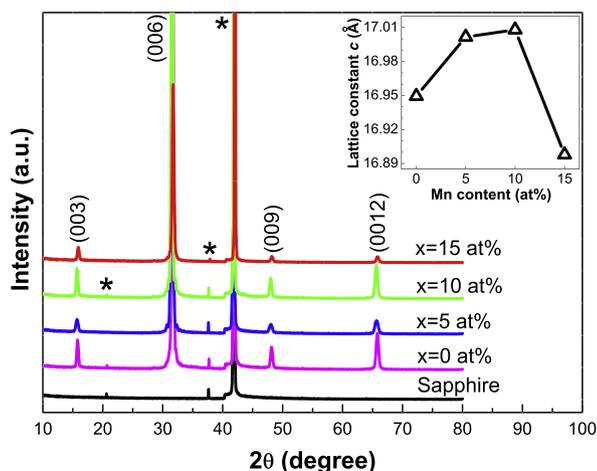


Fig. 1. XRD patterns of  $\text{Cu}(\text{Cr}_{1-x}\text{Mn}_x)\text{O}_2$  thin films on sapphire substrates. The peaks marked by an asterisk (\*) correspond to the substrate. The inset shows the composition dependence of lattice constant  $c$ .

films have a quasi-epitaxial growth orientation along the  $c$  axis. The Full-Width at Half-Maximum (FWHM) of (006) peaks ranges between  $0.32^\circ$  and  $0.43^\circ$ .

The calculated lattice constant  $c$  is plotted in the inset of Fig. 1, in accordance with 3R- $\text{CuCrO}_2$  (JCPDS No. 89-0539). With the Mn addition, the  $c$  value is increased at first; and then its increase slows down obviously; finally, it is remarkably decreased instead. According to Ref. [12], the ionic radii in Coordination (VI) are 0.615, 0.830, 0.645 and 0.530 Å, respectively, for  $\text{Cr}^{3+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Mn}^{3+}$  and  $\text{Mn}^{4+}$ . Firstly, the existence of  $\text{Mn}^{2+}$  can be excluded. If the Mn valence were 2+, the increase of  $c$  would be much greater owing to the much larger size of  $\text{Mn}^{2+}$  than that of  $\text{Cr}^{3+}$ . Moreover, our ceramic targets were prepared using  $\text{MnO}_2$  as raw material and sintered in air for a long time and the corresponding thin films were grown in oxygen atmosphere. During such a preparation process,  $\text{Mn}^{2+}$  is very unlikely to form, but  $\text{Mn}^{4+}$  is likely to exist. Combined with the variation in lattice constant, the Mn ions are deduced to replace  $\text{Cr}^{3+}$  in the mixed valence state:  $\text{Mn}^{3+}$  and  $\text{Mn}^{4+}$ . At  $x = 5$  at.%, the  $\text{Mn}^{3+}$  ions are dominant over  $\text{Mn}^{4+}$ , leading to a 0.31% increase in  $c$  because the ionic radius of  $\text{Mn}^{3+}$  is larger than that of  $\text{Cr}^{3+}$ . As  $x$  is raised to 10 at.%, the proportion of  $\text{Mn}^{4+}$  ions is enhanced, and therefore  $c$  is increased by only 0.04% due to the smaller size of  $\text{Mn}^{4+}$  compared with  $\text{Cr}^{3+}$ . When  $x$  reaches 15 at.%, the proportion of  $\text{Mn}^{4+}$  is further enhanced. As a result, the  $c$  value no longer increases but decreases instead, even 0.30% lower than that of un-doped  $\text{CuCrO}_2$ .

#### 3.1.2. XPS characterization

The XPS measurements were carried out to determine the valence states of ions. All thin films show similar results and the  $x = 15$  at.% one is taken for example here as presented in Fig. 2. The typical Cu 2P XPS spectrum is observed in Fig. 2(a). Its line shape with no clear satellite is quite similar to that of  $\text{Cu}_2\text{O}$  ( $\text{Cu}^+$ ). Hence the Cu ions are concluded to be in the monovalent state [5,13]. From Fig. 2(b), it can be seen that the Cr 2p<sub>3/2</sub> peak is located at 576.1 eV and the energy difference between the Cr

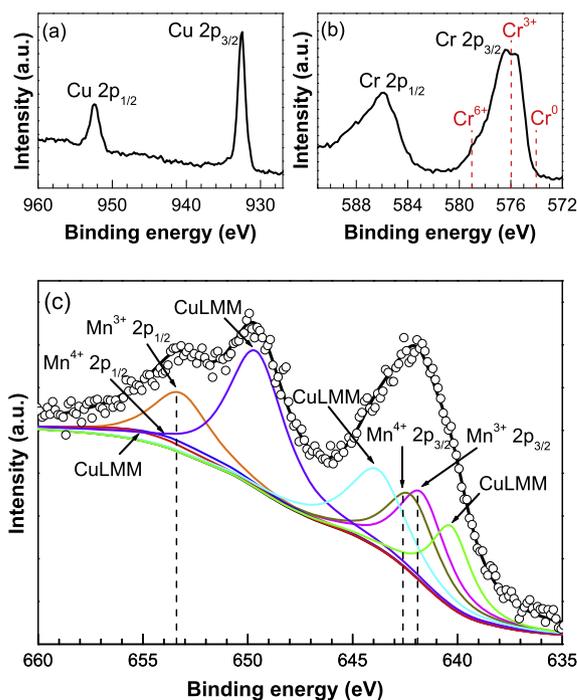


Fig. 2. XPS spectra of  $\text{Cu}(\text{Cr}_{0.85}\text{Mn}_{0.15})\text{O}_2$  thin film: (a) Cu 2p XPS spectrum, (b) Cr 2p XPS spectrum, and (c) Mn 2p XPS spectrum where points and solid lines represent the experimental and fitting data, respectively.

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