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Suppression of T_c in Co-doped (Cu_{0.5}Tl_{0.5})Ba₂Ca₂Cu_{3-x}Co_xO_{10- δ} superconductor

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1. Introduction

Since the discovery of oxide superconductors the exact mechanism of superconductivity at microscopic level has not been completely explored yet. In cuprate high temperature superconductors (HTSCs) MBa₂Ca_{n-1}Cu_nO_{$2n+4-\delta$} [n=2-5 and M=Y, Tl, Hg, Bi, Cu, (Cu,Tl)] superconductor families, the two major constituents of unit cell are $MBa_2O_{4-\delta}$ a charge reservoir layer and $nCuO_2$ conducting planes [1]. The key structural unit in the oxide superconductors is the conducting CuO₂ planes, which is considered to be responsible for the superconductivity [2-5]. The charge reservoir layer supplies carriers to the conducting $nCuO_2$ planes and these carriers in CuO₂ planes become the source of superconductivity. The carriers in CuO₂ planes can be perturbed by directly doping at the planner sites. Many research groups working in the field of high T_c superconductivity focus their attention on the CuO₂ planes; they substitute different elements such as Co, Ni, Fe and Zn at Cu sites in CuO₂ planes in order to study the response of the superconductor material to a disorder cre-

ABSTRACT

The most possible sources of suppression of superconducting properties (i.e., zero resistivity $T_c(R=0)$, magnitude of diamagnetism, etc.) in cobalt-doped (Cu_{0.5}Tl_{0.5})Ba₂Ca₂Cu_{3-x}Co_xO_{10- $\delta}$} (x=0, 0.05, 0.1, 0.5) superconductor have been explored. It has been observed from these studies that Co-doping at Cu²⁺ sites in the CuO₂ planes somehow localizes the carriers at Co³⁺ sites and decreases the density of mobile carriers in the superconducting CuO₂ planes. We doped Li at Tl sites in Cu_{0.5}Tl_{0.5}Ba₂O_{4- δ} charge reservoir to obtain an efficient carriers doping (Cu_{0.5}Tl_{0.25}Li_{0.25})Ba₂O_{4- δ} charge reservoir layer of (Cu_{0.5}Tl_{0.25}Li_{0.25})Ba₂Ca₂Cu_{3-x}Co_xO_{10- δ} compound to replenish the density of mobile carriers in the CuO₂ planes; since alkali metals are known to lose their outermost "s-orbital" electron easily, which can be supplied to the superconducting CuO₂ planes. The increased free carrier density in the superconducting CuO₂ planes enhances the superconducting properties of the final compound.

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ated by these ions in CuO₂ planes [6–14]. In these studies, it was found that doping of such impurities leads to a decrease in the critical temperature " T_c ". The proposed reasons for the T_c suppression after the substitution of impurity atoms in CuO₂ planes are magnetic pair breaking, potential scattering of carriers, localization of carriers and suppression of effective pairing interaction [15–17]. The magnetic ions Co, Ni and Fe produce the magnetic pair breaking and localization of carriers due the interaction between the Cooper pairs and the magnetic spins of the impurity atoms [18-20]. The Cu_{1-x}Tl_xBa₂Ca₂Cu₃O_{10- δ} (x = 0.4-0.8) superconductor is a promising material in cuprate family of superconductors due to its low anisotropy and the ability of carrying high critical current [21,22]. In this article, we substituted Co at Cu in CuO₂ planar sites in $(Cu_{0.5}Tl_{0.5})Ba_2Ca_2Cu_3O_{10-\delta}$ superconductor to observe the response of these materials to the substitution of magnetic impurity in CuO₂ planes. This material has Tl³⁺/Cu²⁺ atoms in the charge reservoir layer; therefore, the charge transfer mechanism to CuO₂ planes is different from that of other cuprates, which has only trivalent atoms in their charge reservoir layer [23-25]. Therefore, we expected some different results as compared to the other superconductor materials after Co-doping. However, our experiments on Co-doped ($Cu_{0.5}Tl_{0.5}$)Ba₂Ca₂Cu_{3-x}Co_xO_{10- δ} superconductors have suggested that the mobile free carriers are most likely to be localized at Co³⁺ sites in CuO₂ planes due to (3+) oxidation state of cobalt [26]. These studies have also suggested that if we replenish the free carriers by some other means to CuO₂ planes, superconduc-

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Table 1 Fraction of different phases present in the $Cu_{0.5}Tl_{0.5}Ba_2Ca_2Cu_{3-x}Co_xO_{10-\delta}$ (*x* = 0, 0.05, 0.1, 0.5) superconductor.

Sr. no.	Co-concentration	% of 1223	% of 1234	% of 1212	a-axis (Å)	c-axis (Å)
1	0	95	4	1	3.98	15.02
2	0.05	94	4	2	4.03	15.08
3	0.1	94	5	1	4.04	15.32
4	0.5	95	4	1	3.89	15.88

tivity can be restored in Co-doped Cu_{0.5}Tl_{0.5}Ba₂Ca₂Cu_{3-x}Co_xO_{10- δ} material. Therefore, we have also substituted Li at Tl sites in the (Cu_{0.5}Tl_{0.5})Ba₂O_{4- δ} charge reservoir layer to obtain an efficient carriers doping (Cu_{0.5}Tl_{0.25}Li_{0.25})Ba₂O_{4- δ} charge reservoir layer of (Cu_{0.5}Tl_{0.25}Li_{0.25})Ba₂Ca₂Cu_{3-x}Co_xO_{10- δ} compound. The objective of Li-doping was to supply required number of mobile carriers to CuO₂ planes, which would have been depleted due to the localization of carriers at Co³⁺ in CuO₂ planes; the supply of excess carriers could restore the superconductivity in Co-doped samples [27,28].

2. Experimental

The samples were prepared by solid-state reaction method accomplished in two steps. In the first step $Cu_{0.5}Ba_2Ca_2Cu_{3-x}Co_xO_{10-\delta}$ (x=0, 0.05, 0.1, 0.5) and (Cu_{0.5}Li_{0.25})Ba₂Ca₂Cu_{2.5}Co_{0.5}O_{10-δ} precursor materials were synthesized by using Ba(NO₃)₂, CaCO₃, Cu(CN), Co(NO₃)₃ and LiCl as starting compounds. These compounds were mixed in appropriate ratios and ground for about an hour in an agate mortar and pestle. The ground material was loaded in quartz boat and fired in chamber furnace at 850°C for 24h followed by furnace cooling to room temperature. The precursor material was then mixed with Tl_2O_2 and ground for about an hour. Thallium mixed material was then palletized under 3.2 tonnes/cm² pressure and the pellets were wrapped in a gold capsule. Pellet containing gold capsule was heat-treated at 850°C for 10min and quenched to room temperature to get $(Cu_{0.5}TI_{0.5})Ba_2Ca_2Cu_{3-x}Co_xO_{10-\delta}$ (x = 0, 0.05, 0.1, 0.5) and $(Cu_{0.5}Tl_{0.25}Li_{0.25})Ba_2Ca_2Cu_{3-x}Co_xO_{10-\delta}$ as final reactants compositions. The resistivity of these samples was measured by four probe method and ac-susceptibility by using SR530 Lock-in Amplifier at a frequency of 270 Hz with H_{AC} = 5.69 mOe of primary coil at various temperatures. The bar shaped samples of dimension $2 \text{ mm} \times 10 \text{ mm} \times 2.5 \text{ mm}$ were used for resistivity and ac-susceptibility measurements. The structure of material was determined by using X-ray diffraction scan (D/Max IIIC Rigaku) with a CuK α source of wavelength 1.54056 Å and cell parameters were determined with the help computer software program (i.e., Check cell).

3. Results and discussion

The X-ray diffraction (XRD) scans of $Cu_{0.5}Tl_{0.5}Ba_2Ca_2Cu_{3-x}Co_xO_{10-\delta}$ (x=0, 0.05, 0.1, 0.5) {CuTI-1223} samples prepared at 850 °C are shown in Fig. 1. Most of the diffraction lines are indexed according to tetragonal structure following *P4/mmm* space group. The inclusions of derivative phases such as Cu–TI–Ba–Ca–Cu–Co–O_{12- δ} {CuTI-1234} and Cu–TI–Ba–Ca–Cu–Co–O_{8- δ} {CuTI-1212} were also observed in the XRD scans. The relative percentages of these phases with Co-doping concentrations are given in Table 1, which were calculated by using the following relations [29].

% of CuTl-1223 phase

$$= \frac{\sum I(CuTI-1223)}{\sum I(CuTI-1223) + \sum I(CuTI-1234) + \sum I(CuTI-1212)} \times 100$$

% of CuTl-1234 phase

$$= \frac{\sum I(CuTI-1234)}{\sum I(CuTI-1223) + \sum I(CuTI-1234) + \sum I(CuTI-1212)} \times 100$$

% of CuTl-1212 phase

$$= \frac{\sum I(\text{CuTI-1212})}{\sum I(\text{CuTI-1223}) + \sum I(\text{CuTI-1234}) + \sum I(\text{CuTI-1212})} \times 100$$

'*I*' is the intensity of the phases present in the final compound.

The lengths of 'a'- and 'c'-axes increase with the increase in Codoping at Cu sites in CuO_2 planes. The dominant phase in all these samples is CuTl-1223, while the other phases in minute quantity are due to the impurities arising during the preparation processes of these samples [10]. Some additional phases such as CuTl-1234 and CuTl-1212 are observed along with the desired CuTl-1223 phase. We have taken XRD spectra with a very slow scan speed of 1°/min to observe even small amount of impurities.

The resistivity measurements of Co-doped $(Cu_{0.5}Tl_{0.5})Ba_2Ca_2Cu_{3-x}Co_xO_{10-\delta}$ (x=0, 0.05, 0.1, 0.5) samples with magnetic ac-susceptibility measurements in inset are shown in Fig. 2. A metallic variation of resistivity from room temperature down to onset of the superconductivity is the salient feature of these samples. The zero resistivity critical temperature { $T_c(R=0)$ } is observed around 114, 104, 84 and 80K for Co-doping concentration of x=0, 0.05, 0.1, 0.5, respectively. The increased concentration of Co ($x \ge 0.5$) shifts the transition temperature below 77 K. The magnetic ac-susceptibility curves of these samples



Fig. 1. The X-ray diffraction (XRD) scans of $(Cu_{0.5}Tl_{0.5})Ba_2Ca_2Cu_{3-x}Co_xO_{10-\delta}$ (x = 0, 0.05, 0.1, 0.5) superconductors and "*" & "." denote the CuTl-1234 and CuTl-1212 phases, respectively.

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