



# Magnetic and structural properties of $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$ nanoparticles

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## ABSTRACT

The structural and magnetic properties of barium hexaferrite nanoparticles ( $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$ ) with  $x = 0.0$ – $1.0$ , prepared by ball milling were investigated using XRD, TEM, and VSM. It was found that the particles and crystallites have similar mean size of  $\sim 41$  nm for all investigated samples. The saturation magnetization decreased slightly and nonlinearly with increasing  $x$ , and this was attributed to different preferential site occupation of Ga at low and high concentration ranges. The coercivity decreased slightly with increasing  $x$  for low concentrations of Ga ( $x \leq 0.2$ ), and then increased with increasing Ga concentration up to  $x = 1.0$ . This behavior of the coercivity was attributed to the change in the exchange coupling, which was confirmed by the variation of SFD, remanence ratio and Curie temperature with Ga concentration in the samples.

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## 1. Introduction

Barium hexaferrite with its stoichiometric chemical formula  $\text{BaFe}_{12}\text{O}_{19}$  is one of M-type hexaferrites (BaM). M-type hexaferrites have large saturation magnetization, high coercivity, high Curie temperature, large uniaxial magnetic anisotropy and excellent chemical stability. These materials have been intensively investigated due to their technological applications in permanent magnets, high-density magnetic recording media and microwave devices. Several techniques have been used to prepare hexaferrite particles including sol–gel method [1–3], citrate–nitrate gel combustion method [4], ammonium nitrate melt technique [5,6], mechano-combustion route [7], co-precipitation method [8], and microwave-induced hydrothermal reaction [9]. Several studies on hexaferrites in which Fe ions are substituted by different cations or combinations of cations have been carried out to develop materials with improved characteristics suitable for technological applications. Among the recently investigated substitutions are Mn–Co–Zr [10], Ni–Ti [11], Gd–Co [12], Co–Ti [13–15], Co–Zr [16], Zr–Zn [17], Mn and Ti [18].

The present work is concerned with the magnetic properties of BaM doped with gallium. The samples were prepared by ball milling, which had been recently used to prepare barium ferrite powders. This method is simple and useful for the production of powders consisting of fine particles smaller than the critical single

domain size. The magnetic, XRD and TEM data were analyzed in an attempt to explain the magnetic behavior of  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$ .

## 2. Experimental procedures

$\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$  powders with  $x = 0.0, 0.2, 0.4, 0.6, 0.8$  and  $1.0$  were prepared by ball milling. Metallic oxides ( $\text{Fe}_2\text{O}_3$ , and  $\text{Ga}_2\text{O}_3$ ) and barium carbonate ( $\text{BaCO}_3$ ) were used as starting materials. Mechanical alloying was carried out in a planetary ball mill (Fritsch Pulverisette 7) using a ball to powder ratio of 8:1. The milling was carried out for 24 h with an angular frequency of 250 rpm. After mechanical milling, the powders were pressed under a force of 50 kN into disks, 1 cm in diameter each. These disks were annealed in air at  $1100^\circ\text{C}$  for 2 h.

XRD analysis was carried out using Philips X'Pert PRO X-ray diffractometer (PW3040/60) with  $\text{CuK}\alpha$  radiation. XRD patterns for the samples examined were recorded in the range of  $2\theta$  between  $15^\circ$  and  $75^\circ$  with scanning step of  $0.017^\circ$ . A powder diffraction software package which includes the standards of the International Center for Diffraction Data (ICDD) was used to identify the observed structural phases. Transmission electron microscopy (TEM) using Zeiss EM10CR electron microscope operating at 80 kV was employed for estimating the particle sizes. TEM samples were prepared by suspending the powders in ethanol and shaking the suspension ultrasonically for several minutes. The suspension of the samples sintered at  $1100^\circ\text{C}$  was immediately light-brown colored, indicating the fine particle nature of the investigated samples. A drop of the suspension on a formvar-coated copper TEM grid was allowed to dry in air before starting the measurements. The magnetic measurements were carried out using a vibrating sample magnetometer (VSM MicroMag 3900, Princeton Measurements Corporation). The saturation magnetization ( $M_s$ ) was obtained from the extrapolation of the magnetization curve versus  $1/H$  to  $1/H = 0$ . The switching field distribution (SFD) for each sample was determined by differentiation and normalization of the DC demagnetization curve (DCD). The remanence coercivity ( $H_{cr}$ ) was determined from the normalized DCD curve.

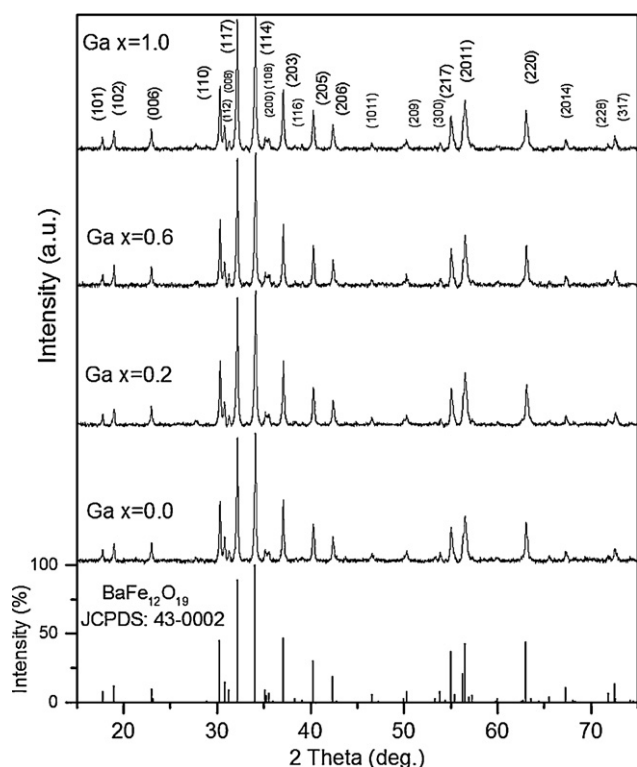
## 3. Results and discussion

Fig. 1 shows the XRD patterns of samples of  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$  along with the standard pattern for hexagonal barium ferrite

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**Fig. 1.** Standard JCPDS pattern for M-type hexagonal barium ferrite (file no.: 043-0002) and XRD patterns of  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$  with different doping concentration.

( $\text{BaFe}_{12}\text{O}_{19}$ ) with space group  $P6_3/mmc$  (JCPDS file no.: 043-0002) [19]. The figure shows that the XRD patterns for all samples exactly match the standard 043-0002 for hexagonal barium ferrite ( $\text{BaFe}_{12}\text{O}_{19}$ ) with no secondary phases, which indicates that  $\text{Ga}^{3+}$  ions diffused into the hexagonal structure forming a single hexagonal  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$  phase for all values of  $x$ . The lattice parameters  $a$  and  $c$  of this phase were calculated from the formula [20]:

$$\frac{1}{d_{hkl}^2} = \frac{4}{3} \left( \frac{h^2 + hk + k^2}{a^2} \right) + \frac{l^2}{c^2} \quad (1)$$

where  $d$  is the interplanar distance and  $h$ ,  $k$ , and  $l$  are Miller indices. The lattice parameters of Ga substituted barium hexaferrites (Table 1) show variations less than 0.1%. This is expected because the radius of  $\text{Ga}^{3+}$  ion (0.625 Å) is almost equal to that of  $\text{Fe}^{3+}$  (0.645 Å).

The average crystallite size was determined from the positions of the peaks (1 1 4), (2 1 7) and (2 2 0) using the well-known Scherrer formula [21]:

$$D = \frac{k\lambda}{\beta \cos \theta}, \quad (2)$$

where  $D$  is the crystallite size,  $k$  the Scherrer constant,  $\lambda$  the wavelength of radiation (1.54056 Å),  $\beta$  the peak width at half maximum measured in radians, and  $\theta$  the peak position. The average crys-

**Table 1**  
Lattice parameters and average crystallite sizes of  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$  measured by XRD.

$x$	$a$ (Å)	$c$ (Å)	$c/a$	Average crystallite size, $D$ (nm)
0.0	5.890	23.194	3.938	45
0.2	5.887	23.203	3.942	44
0.4	5.886	23.208	3.943	39
0.6	5.888	23.199	3.940	44
0.8	5.891	23.206	3.939	40
1.0	5.890	23.207	3.940	37

**Table 2**

Coersivity, saturation magnetization, remanence ratio  $M_r$  ( $M_r/M_s$ ) and Curie temperature for  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$ .

$x$	$H_c$ (kOe)	$M_s$ (emu/g)	$M_r$	$T_c$ (°C)
0.0	4.02	70.9	0.523	505
0.2	3.95	69.3	0.519	500
0.4	4.00	68.3	0.518	490
0.6	4.26	68.0	0.519	470
0.8	4.49	65.3	0.517	455
1.0	4.55	60.3	0.514	415

tallite size for the pure and doped samples (Table 1) ranges from 37 nm to 45 nm.

TEM images of representative samples are shown in Fig. 2. The average particle size for the pure sample is  $42 \pm 13$  nm, and for the sample with  $x = 1.0$  is  $41 \pm 13$  nm. These values are much smaller than the critical value of 460 nm reported by Rezlescu et al. [22] for single magnetic domain structure. Subsequently, we carried out magnetization measurements on the pure barium ferrite sample annealed at 1350 °C for 4 h and the results are shown in Fig. 3 together with those for the sample annealed at 1100 °C. The figure demonstrates that the magnetization of the sample annealed at 1100 °C increases slowly at low fields followed by a rapid increase at higher fields. This behavior is typical for randomly oriented single domain magnetic particles. In contrast, the initial magnetization for the pure sample annealed at 1350 °C shows a sharp increase at low fields then continues to increase slowly at higher fields. This behavior is typical for multi-domain particles. Further, attempts to suspend the powder of the sample annealed at 1350 °C in ethanol for TEM measurements failed, indicating that this powder consists of large particles. These observations indicate that ball milling method is suitable for preparing single domain magnetic nanoparticles with a narrow particle size distribution suitable for high-density recording applications.

Hysteresis loops for  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$  samples are measured as a function of applied magnetic field and the results are shown in Table 2. The magnetization for the non-substituted sample is characteristic of hard magnetic material with coercive field of about 4 kOe. This value of the coercivity agrees well with previous results on samples prepared by sol-gel method [23] and mechanical alloying method [24].

The influence of Ga content on the saturation magnetization and coercivity of  $\text{BaFe}_{12-x}\text{Ga}_x\text{O}_{19}$  is also shown in Fig. 4. The saturation magnetization decreases slowly for  $x = 0.0 - 0.6$  (decrease in  $M_s$  is 2.5%). For  $x \geq 0.6$ ,  $M_s$  drops faster, recording a reduction in  $M_s$  of 11% at  $x = 1.0$ . This decrease in  $M_s$  is still small compared with the reduction of 50% due to doping barium ferrites with  $\text{Cr}^{3+}$  [25]. The magnetic moment per formula for barium hexaferrite has been calculated from the sum of the magnetic moments of Fe ions in the different crystallographic sites as follows [26–28]:

$$\vec{m} = 2\vec{a} + 2\vec{b} + 12\vec{k} + \leftarrow 4f_1 + \leftarrow 4f_2 \quad (3)$$

The spin-down sites ( $4f_1$  and  $4f_2$ ) are occupied by two Fe ions each, whereas the spin-up sites  $2a$  and  $2b$  are occupied by one Fe ion each, and  $12k$  is occupied by six Fe ions. Based on this simple model, the magnetic moment of the pure phase is  $4 \times 5\mu_B = 20\mu_B$  per formula. Accordingly, replacement of spin-up Fe ions by non-magnetic Ga ions would result in a drop of the magnetic moment down to  $17\mu_B$  per formula (15% drop of the magnetization) for  $x = 0.6$ , and to  $15\mu_B$  per formula (25% drop of the magnetization) for  $x = 1.0$ . These values are significantly higher than the observed drop in magnetization, indicating that replacement of Fe ions by Ga occurs at both spin-up and spin-down sites, with a preference for replacement at the spin-up sites. The observed values of the drop in magnetization at  $x = 0.6$  and 1.0, and the behavior of the magnetization in this concentration range suggest that the fraction of Ga ions replacing

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