

Letter

The crystal structure and magnetic properties of the
 $\text{GdCr}_x\text{Al}_{12-x}$ ($x = 3.5$ and 4.0) intermetallicsYu. Verbovitsky^{a,b}, K. Łątka^{b,*}, A.W. Pacyna^c, K. Tomala^b^a Department of Inorganic Chemistry, Ivan Franko National University of Lviv, Kyryla and Mefodiya 6, 79005 Lviv, Ukraine^b Marian Smoluchowski Institute of Physics, Jagiellonian University, Reymonta 4, 30-059 Kraków, Poland^c Henryk Niewodniczański Institute of Nuclear Physics, Polish Academy of Sciences, Radzikowskiego 152, 31-342 Kraków, Poland

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Abstract

The crystal structure of the $\text{GdCr}_x\text{Al}_{12-x}$ (with $x = 3.5$ and 4.0) intermetallic compound was determined by X-ray powder diffraction using the Rietveld method. The investigated compound crystallizes with ThMn_{12} structure type (space group $I4/mmm$, Pearson symbol $tI26$). Magnetic measurements carried out for the title compound point to the antiferromagnetic–paramagnetic transition observed at $T_N = 6.50$ and 6.75 K for compositions with $x = 3.5$ and 4 , respectively.

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1. Introduction

Among the R–Cr–Al (R = Sc, Y and rare earths) ternary systems the isothermal sections were investigated in the partial or whole concentration region with R = Sc, Y, La, Ce, Gd, Dy, Yb [1–7]. Additionally, other alloys of these systems were investigated for the existence of the isostructural ternary compounds, which belong to the $\text{CeCr}_2\text{Al}_{20}$, $\text{Ho}_6\text{Mo}_4\text{Al}_{43}$ and CeMn_4Al_8 structure types [8–10]. Magnetic properties of some selected intermetallics of these systems were also studied [11,12].

In this paper, we report the results of crystal structure investigations of the $\text{GdCr}_x\text{Al}_{12-x}$ (with $x = 3.5$ and 4.0) intermetallic compound using the Rietveld method together with the results of magnetic measurements.

2. Experimental details

Samples were prepared by arc-melting of initial components under high purity argon on a water-cooled copper hearth. Starting materials were used in the form of pieces of high purity metals (>99.9 wt.%). The samples were remelted three times for a better homogeneity. Ingots were wrapped with tantalum foil and afterwards sealed in evacuated quartz tubes and annealed at 500°C for 3 months. After heat treatment, the samples were quenched by submerging the silica tubes in cold water.

The crystal structures of ternary compounds were studied by means of X-ray powder diffraction (Siemens D500 diffractometer, Cu $K\alpha$ -radiation, scanning parameters: 2θ region 10 – 145° , step scan 0.05° , counting time per step 10 s). All the crystal structure calculations were performed by means of the Rietveld method using the FullProf programme [13].

The magnetic susceptibility and magnetization of polycrystalline samples were determined by means of SQUID (Quantum Design) magnetometer, a Cahn RG automatic electro-balance and/or ac/dc magnetic measurements with Lake Shore 7225 apparatus in the temperature range from 2 to 300 K.

3. Results and discussion

The structure of the $\text{GdCr}_x\text{Al}_{12-x}$ (for $x = 3.5$ and 4.0) intermetallic compound annealed at 500°C was refined from X-ray powder diffraction data using the Rietveld method. Pseudo-Voigt profile shape function was used. The background was refined with a polynomial function. The title ternary intermetallic compound crystallizes in tetragonal space group $I4/mmm$ (Pearson symbol $tI26$), with ThMn_{12} structure type (or CeMn_4Al_8 type with $x = 4.0$) [14]. The results of the crystal structure determination for the $\text{GdCr}_x\text{Al}_{12-x}$ with $x = 3.5$ and $x = 4.0$ are summarized in Table 1. Results of the Rietveld profile refinement of the GdCr_4Al_8 X-ray diffraction data are presented in Fig. 1. Projections of the $\text{GdCr}_x\text{Al}_{12-x}$ structure on the xy plane and coordination polyhedra for the atoms are given in Fig. 2. Interatomic distances and coordination numbers for atoms in the $\text{GdCr}_x\text{Al}_{12-x}$ are presented in Table 2. These dis-

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Table 1

Results of the crystal structure determination of the $\text{GdCr}_x\text{Al}_{12-x}$ ($x=4.0\text{--}3.5$) compound

$\text{GdCr}_x\text{Al}_{12-x}$	$x=3.5$	$x=4.0$
Lattice parameters		
a (Å)	8.9900 (6)	8.9683(7)
c (Å)	5.1229 (4)	5.1252(5)
V (Å ³)	414.04 (5)	412.23(6)
Reliability factors		
R_B (%)	8.77	8.45
R_p (%)	3.31	3.39
Atom parameters		
	Gd in 2(a) 0 0 0	Gd in 2(a) 0 0 0
B_{iso} (Å ²)	0.68(10)	0.85(8)
	0.87(3)Cr + 0.13(3)Al	Cr in 8(f) 1/4 1/4 1/4
	in 8(f) 1/4 1/4 1/4	
B_{iso} (Å ²)	0.67(13)	1.05(10)
	Al1 in 8(i) x 0 0	Al1 in 8(i) x 0 0
	$x=0.3530(8)$	$x=0.3537(8)$
B_{iso} (Å ²)	0.89(18)	0.46(18)
	Al2 in 8(j) x 1/2 0	Al2 in 8(j) x 1/2 0
	$x=0.2791(8)$	$x=0.2795(9)$
B_{iso} (Å ²)	0.64(18)	0.95(20)

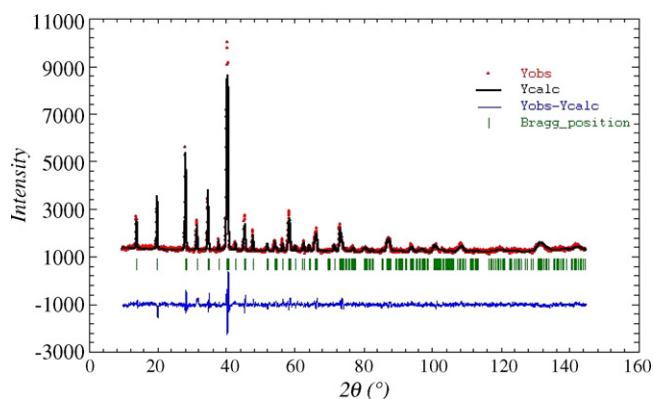
Fig. 1. Rietveld profile refinement of X-ray diffraction data for the GdCr_4Al_8 .

Table 2

Interatomic distances (δ^*) and coordination numbers (CN) for atoms in the $\text{GdCr}_x\text{Al}_{12-x}$ intermetallics

Atoms	δ (Å)	CN	Atoms	δ (Å)	CN
$\text{GdCr}_x\text{Al}_{12-x}$ ($x=3.5$)			$\text{GdCr}_x\text{Al}_{12-x}$ ($x=4.0$)		
Gd		20	Gd		20
–4Al1	3.174		–4Al1	3.172	
–8Al2	3.241		–8Al2	3.237	
–8M	3.427		–8Cr	3.420	
M		12	Cr		12
–2Cr	2.561		–2Cr	2.563	
–4Al2	2.600		–4Al2	2.596	
–4Al1	2.748		–4Al1	2.745	
–2Gd	3.427		–2Gd	3.420	
Al1		14	Al1		14
–Al1	2.642		–Al1	2.624	
–4M	2.748		–4Cr	2.745	
–2Al2	2.824		–2Al2	2.827	
–2Al2	2.836		–2Al2	2.829	
–4Al1	3.170		–4Al1	3.164	
–Gd	3.174		–Gd	3.172	
Al2		12	Al2		12
–4M	2.600		–4Cr	2.596	
–2Al2	2.808		–2Al2	2.796	
–2Al1	2.824		–2Al1	2.827	
–2Al1	2.836		–2Al1	2.829	
–2Gd	3.241		–2Gd	3.237	

* Standard deviations ≤ 0.001 Å. M = 0.87(3)Cr + 0.13(3)Al.

tances are close to the sum of the atomic radii of the respective atoms.

SQUID magnetic measurements indicate that the $\text{GdCr}_x\text{Al}_{12-x}$ with $x=3.5$ orders antiferromagnetically at Néel temperature $T_N=6.50(5)$ K, presenting modified Curie–Weiss behaviour above T_N (Fig. 3a) in the form $\chi_\sigma = \chi_0 + C/(T - \Theta_p)$, where χ_0 is a temperature independent factor, C the Curie constant and Θ_p is the paramagnetic Curie temperature. The negative paramagnetic $\Theta_p = -5.9$ K Curie temperatures obtained for this composition of $\text{GdCr}_x\text{Al}_{12-x}$ compound points to a dominant antiferromagnetic exchange interaction among gadolinium magnetic moments. The effective magnetic moment

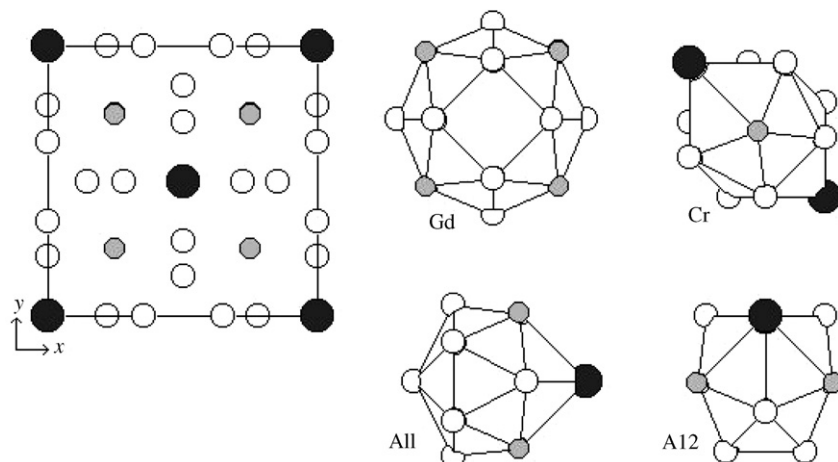


Fig. 2. Projections of the GdCr_4Al_8 structure on the xy plane and coordination polyhedra of the atoms. Black circles indicate Gd atoms, grey filled circles are Cr atoms, and Al atoms are marked by white circles.

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