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Full length article

Microstructures and mechanical properties of the age hardened Mg−4.2Y−2.5Nd−1Gd−0.6Zr (WE43) microalloyed with Zn

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Abstract

The effect of trace addition of 0.2 wt.% Zn on the microstructures and mechanical properties of the age-hardening Mg−4.2Y−2.5Nd−1Gd−0.6Zr (wt.%) (WE43) alloy has been investigated. As compared with the WE43 alloy after solid solution treatment at 525 °C, the block-like Zn−Zr phase was still observed in the WE43-0.2Zn alloy. However, the time for WE43-0.2Zn alloy to get peak hardness at 250 °C was two hours, a half earlier than that in WE43 alloy, meaning a accelerated age precipitation kinetics has been achieved due to the addition of 0.2 wt.% Zn. Microalloyed with 0.2 wt.% Zn enhanced the ultimate tensile strength (UTS) slightly and ductility significantly both in the solutionized and peak aged condition. The enhancement in strength and ductility is possible associated with the larger volume fraction of precipitation phases due to a reduction of the solubility of rare earth elements (RE) in the α-Mg matrix, the larger aspect ratio (length to width) of precipitates and a decrease in stacking fault energy by addition of Zn.

Keywords: Magnesium alloy; Microalloying; Microstructures; Mechanical properties

1. Introduction

The strong need for weight reduction of transportation vehicles for better fuel efficiency has indicated magnesium alloys as a good potential structural material for many structural applications due to their high specific strength, high stiffness and good damping capacity combined with quite low density [1,2]. There has been a rapid growth in interest in the development of higher strength, creep-resistant Mg alloys for the automotive and aerospace industries [3,4]. The most successful Mg alloys developed to date have been those based on the Mg−Y−Nd system, identified as WE54 and WE43 [5−7]. The strength of these alloys is achieved essentially via precipitation strengthening [8]. The characteristics of the precipitates in Mg−Y−Nd alloys, such as crystal structure, morphology, size, precipitation sequence and phase evolution have been much investigated. Depending on the ageing temperature, the precipitation sequence in WE alloys has been reported to involve formation of phases designated β″, β′, β_1 and β, which all four precipitate phases have been described to form as plates on prismatic planes of the Mg matrix phase [5,6,9−13].

J.F.Nie has discussed the predicted strengthening effect of precipitates of different shape and habit on the basal and {10−12} twinning deformation systems in Mg by the development of the Orowan equation. It revealed that whether

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particles are sheared or are shear resistant, and whether the deformation mode is basal slip or twinning, precipitate plates formed on prismatic planes of the Mg matrix phase provide the most effective barrier to gliding dislocations and propagating twins in the Mg matrix. It further suggest that a higher strength can be achieved if a high density of intrinsically strong, plate-shaped precipitates with prismatic and basal habit planes and of large aspect ratio can be developed in the microstructure by microalloying additions [8,14]. Zn is one of alloy elements with low cost and maximum solubility in solid Mg of 6.2 wt.%. The Mg alloys containing Zn possess excellent high mechanical properties [15]. Recently, a lot of work has been focused on Mg–RE alloys containing Zn elements, such as Mg–Y–Zn, Mg–Gd–Zn, Mg–Gd–Y–Zn alloys. It was reported that a certain amount of Zn added to Mg–Y, Mg–Gd and Mg–Y–Nd alloys improves the creep strength [16–20]. It also was reported that a certain amount of Zn added to Mg–Gd–Y alloys brought out the precipitations of long-period stacking ordered (LPSO) structures, which resulted in the decrease in volume fraction of the metastable phase in the aged condition [21]. However, Mg–Y–Nd alloys microalloyed with Zn have not been deeply investigated yet. In the present work, the effect of trace addition of 0.2 wt.% Zn on the microstructures and mechanical properties of the age-hardening WE43 alloy has been investigated.

2. Experimental procedures

Two Mg alloys with nominal compositions of Mg–4.2Y–2.5Nd–1Gd–0.6Zr (WE43) and Mg–4.2Y–2.5Nd–1Gd–0.2Zn–0.6Zr (WE43-0.2Zn) were prepared from high purity Mg (>99.95%), Y (>99%), Nd (99%), Gd (99%), Zn (99.9%) and a Mg–30Zr (wt.%) master alloy by melting in an electric resistance furnace at about 780 °C under protection with an anti-oxidizing flux. The melt was poured into a mild steel mold preheated to 200–300 °C. The detailed compositions of the obtained ingots were determined by inductively coupled plasma atomic emission spectroscopy (ICP-AES) and the results listed in Table 1. Specemens cut from the cast ingots were solution treated at 525 °C for 8 h, quenched into water and then subsequently aged at 250 °C for various periods of time.

Differential Thermal Analysis (DTA) has been performed on specimen of as-cast WE43 alloy, with a SETARAM-SETSYS Evolution 18 apparatus under protective pure argon atmosphere. Vickers hardness testing was performed using 500 g load and a holding time of 15 s. Not fewer than 10 measurements were taken in each alloy. The samples for tensile tests had a gauge length of 10 mm,width of 3.5 mm and thickness of 2.5 mm. Tensile tests were performed at roon temperature (RT) with an initial strain rate of 1.0×10^{-3} s^{-1}. At least three specimens were used for each condition to ensure the reproducibility of data and the average values were adopted.

Table 1
Analyzed chemical compositions of the investigated alloys (wt.%).

	Elements					
	Y	Nd	Gd	Zn	Zr	Mg
WE43	4.38	2.72	1.10	<0.01	0.56	Bal
WE43-0.2Zn	4.20	2.66	1.09	0.21	0.50	Bal

The constituent phases of the two alloys in different conditions were identified by X-ray diffraction (XRD) (Rigaku D/max 2500PC X-ray diffractometer) with Cu Kα radiation. Microstructures were observed by optical microscope (OM), scanning electron microscope (SEM, Philips XL30 ESEM-FEG/EDAX) and transmission electron microscope (TEM, JEM-2100F) operating at 200 kV. Samples in as-cast and ageing condition for optical microscopy were etched in a solution of 4 vol.% HNO_3 in ethanol after mechanical polishing. Samples in solution condition for optical microscopy were etched in a solution of 6 g picric acid, 40 ml acetic acid, 40 ml water and 100 ml ethanol. The mean grain size, d, was measured by the linear intercept method using the equation $d = 1.74L$, where L is the liner intercept grain size determined by optical microscopy. Compositions of phases were analyzed by energy dispersive X-ray spectrometry (EDS) in the SEM mode. Thin foil specimens for TEM were prepared by punching 3 mm diameter discs, followed by dimple grinding and Ar+ ion milling in a precision ion polishing system (PIPS, Gatan 691) operating at 4.5 kV accelerating voltage and ~8°incident angle.

3. Results

3.1. Characterization of the alloys in the as-cast state

Typical microstructures of WE43 alloy in the as-cast condition are shown in Fig. 1. It reveals that the as-cast of WE43 alloy consists mainly of the α-Mg phase (a solid solution of Mg containing Y, Nd and Gd) as the matrix and non-equilibrium eutectics which mainly aggregate at the grain boundaries, cuboid-shaped phase. Some zirconium cores are observable, which contribute to the grain refinement. The microstructure of the WE43-0.2Zn alloy is similar to that of the WE43 alloy. The average grain size of WE43 and WE43-0.2Zn alloy is about 94 μm and 106 μm, respectively. The composition of the eutectics (phase A) is Mg–6.41 at.% Nd–4.27 at.% Y–1.02 at.% Gd and the composition of the cuboid-shaped phase (phase B) is Mg–72.00 at.% Y–4.05 at.% Nd–5.07 at.% Gd, which are detected by EDS analyses in Fig. 1e and f. However, the composition of the eutectics is Mg–7.95 at.% Nd–4.83 at.% Y–1.18 at.% Gd–2.01 at.% Zn in the WE43-0.2Zn alloy, which is indicated that the microalloyed element Zn is partial dissolved in the eutectics. By XRD analysis (Fig. 2a), three phases are identified namely α-Mg, $Mg_{12}Nd$ and $Mg_{24}Y_5$.

3.2. Characterization of the solution-treated alloys

In order to determine its solution temperature, a preliminary DTA investigation (Fig. 2b) on a small quantity of