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# Chemical vapor deposition of ultra-thin molybdenum dioxide nanosheets



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## ABSTRACT

We report the growth of ultra-thin molybdenum dioxide nanosheets on SiO<sub>2</sub>/Si substrate via chemical vapor deposition using molybdenum trioxide and sublimated sulfur as precursors. The thicknesses of the obtained MoO<sub>2</sub> nanosheets show notable dependence on the baking temperature of the sulfur precursor. At sulfur temperature of 90 °C, the obtained nanosheets can be 5.5 nm thin, more than one order of magnitude thinner than that previously reported, in a narrow scatter ranging from 5.5 to 11.5 nm. Two-probe electrical measurements show that the as-prepared ultrathin MoO<sub>2</sub> nanosheets preserve a high electrical conductivity of 3600 S/cm with thermal stability up to 200 °C. Above 250 °C, metallic MoO<sub>2</sub> nanosheets are oxidized into insulating MoO<sub>3</sub> flakes in air.

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## 1. Introduction

Two dimensional (2D) semiconducting molybdenum dichalcogenides have attracted intensive interest for their atomic layered structure with excellent electrical properties. However, electrical contact to these 2D semiconductors remains a great challenge for the Fermi level pinning effect at the metal–semiconductor junction [1]. Molybdenum oxides (MoO<sub>x</sub>), as a material of high work function material, has been demonstrated to exhibit promising contact to MoS<sub>2</sub> or WSe<sub>2</sub> based field-effect transistors and diodes because of the efficient hole-injection and lower degree of interface Fermi-level pinning [2]. Moreover, an enhanced electrical connection is also observed in organic light-emitting diodes when using ultrathin layer of molybdenum dioxide (MoO<sub>2</sub>) with thicknesses ranging from 0.25 to 10 nm as buffer layer [3]. Although MoO<sub>x</sub> is a promising electrode material, the preparation of highly conductive 2D MoO<sub>x</sub> with thickness less than 10 nm has not been achieved.

Previously, MoO<sub>2</sub> nanosheets have been prepared via a hydrothermal method [4]. But the complex nanostructure and rich defects of MoO<sub>2</sub> resulting from hydrothermal reduction would severely limit its application in devices. Chemical vapor deposition (CVD), as a well-established method to grow high-quality 2D crystals, has been successfully employed to synthesize MoO<sub>2</sub> or

MoO<sub>3</sub> flakes [5–8]. Reduced by sulfur vapor, MoO<sub>3</sub> powder can be thermally evaporated and deposited as MoO<sub>2</sub> flakes on the Si/SiO<sub>2</sub> substrate [9]. Through this method, Wang et al. have prepared high-quality MoO<sub>2</sub> microplates as templates for the growth of highly crystalline MoS<sub>2</sub> layers [5]. But these MoO<sub>2</sub> microplates are commonly thicker than 100 nm, far from the desired thickness less than 10 nm. Nanosheets prepared by Hao show thickness down to 15 nm but in very poor quality as indicated by the Raman spectrum where several typical peaks of the MoO<sub>2</sub> crystalline are even absent [6].

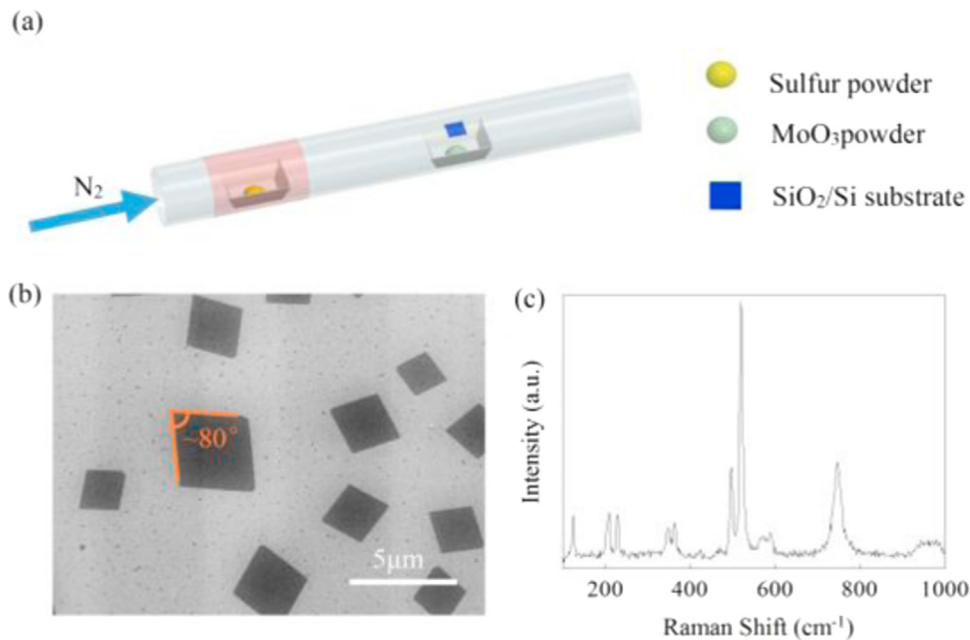
Here we report CVD growth of MoO<sub>2</sub> nanosheets with both high crystallinity and ultrathin thickness down to 5.5 nm. Moderate sublimation of sulfur precursor is crucial for the growth of high-quality MoO<sub>2</sub> nanosheets. The obtained MoO<sub>2</sub> flakes preserve high electrical conductivity with heat resistance up to 200 °C. These outstanding performances indicate MoO<sub>2</sub> to be a promising candidate in the area of nanoelectronics.

## 2. Materials and methods

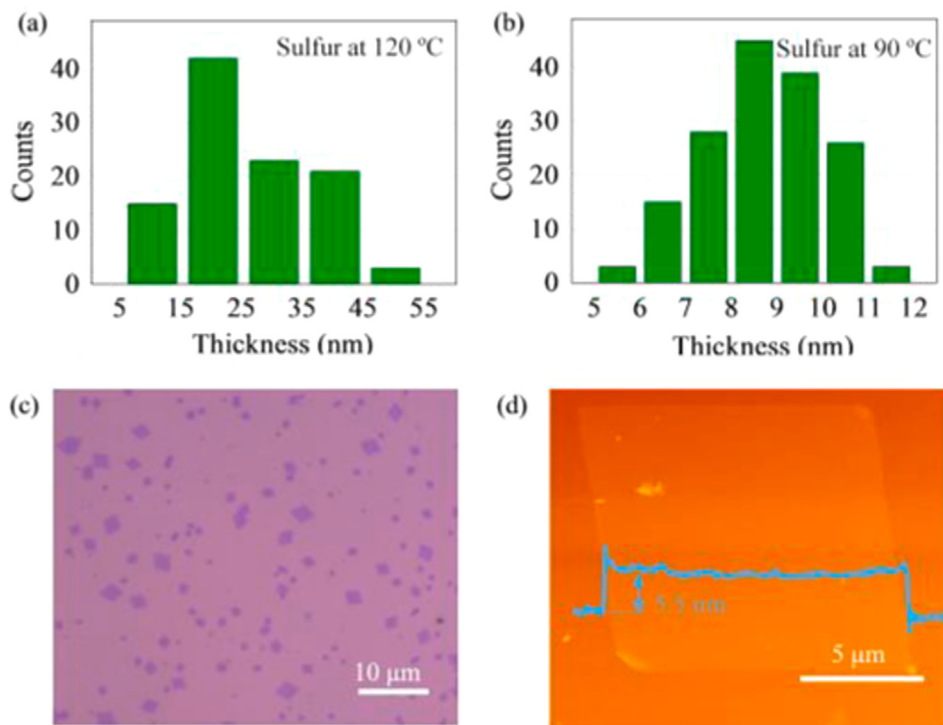
The ultrathin MoO<sub>2</sub> flakes were synthesized on substrate of Si with 285 nm of SiO<sub>2</sub> via CVD process. As illustrated in Fig. 1a, the SiO<sub>2</sub>/Si substrate cleaned by piranha solution (sulfuric acid:hydrogen peroxide=3:1) was loaded into the center of a 2-inch furnace and placed face-down above a ceramic crucible with 10 mg of MoO<sub>3</sub> powder (Ourchem, 99.99%) inside. Another

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**Fig. 1.** Growth and characterization of ultrathin MoO<sub>2</sub> nanosheets. (a) Schematic diagram of the CVD system for MoO<sub>2</sub> synthesis. The temperature of sulfur powder is controlled by a heating belt (red region). (b) SEM image of MoO<sub>2</sub> nanosheets grown on SiO<sub>2</sub>/Si substrate. (c) Typical Raman spectrum of a MoO<sub>2</sub> nanosheet. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)



**Fig. 2.** Dependence of flake thickness on the temperature of sulfur. (a, b) Thickness distribution of MoO<sub>2</sub> flakes grown at a sulfur temperature of 120 °C (a) and 90 °C (b), respectively. (c) Optical microscopy image of the MoO<sub>2</sub> flakes grown on SiO<sub>2</sub>/Si substrate at sulfur temperature of 90 °C. (d) AFM image of an ultrathin MoO<sub>2</sub> nanosheet and the height profile.

crucible with 100 mg of sulfur powder (Alfa Aesar, 99.5%) was placed upstream. The furnace was heated to 750 °C in 30 min with a flow of 10 sccm nitrogen gas, and the temperature of sulfur was controlled at desired temperature by a heating belt. After growth at 750 °C for 10 min, the furnace was cooled slowly down to 600 °C without feedback. Finally, the N<sub>2</sub> flow was set to 100 sccm

before the sulfur source and the furnace were cooled to ambient temperature rapidly. The Raman spectra was conducted by a Renishaw Raman spectrometer with a 532 nm solid-state laser. The devices were fabricated by standard electron-beam lithography, thermal evaporation of ~60 nm Au electrodes, and final lift-off. The voltage was loaded on the devices and the current was

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