



Effect of the cooling rate on microstructure and hardness of MAR-M247 Ni-based superalloy

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ABSTRACT

The effect of cooling rate on the microstructure of MAR-M247 Ni-based superalloy was investigated via physical simulation of the casting process. Solidification experiments with cooling rates in the range of 0.25–10 K/s showed smooth temperature profiles with measured cooling rates matching the set values. The MAR-M247 showed cellular (0.25 K/s) and dendritic (1, 5 and 10 K/s) microstructures. Microconstituents also varied with cooling rates: γ/γ' matrix with carbides and γ/γ' eutectic at 0.25 K/s, γ/γ' matrix with carbides at 1 K/s, and γ/γ' matrix with carbides and γ/MC eutectic at 5 and 10 K/s. Moreover, the secondary dendritic arm spacing decreased and the hardness increased with the increase in the cooling rates.

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1. Introduction

Over the last three decades, significant progress has been made in the development of cast Ni-based superalloys as well as in the casting technology to obtain tailored microstructures. Among the various cast alloys, MAR-M247 has been widely used for high temperature applications in turbine rotor blisks and stators. It is a Ni-based, vacuum melted, cast superalloy with high volume fraction of γ' $\text{Ni}_3(\text{Al}, \text{Ti})$ ($\approx 62\%$) and high ($\text{Ta} + \text{W} + \text{Mo}$) refractory element content (14 wt.%) [1]. The alloy was developed in the early 1970s by Danasi et al. at Martin Metals Corporation [1] and presents good castability, exceptional high temperature strength, corrosion and oxidation resistance. It is commonly produced by directional solidification to improve creep rupture strength.

Most fundamental studies on the microstructural evolution of MAR-M247 alloy are related to directional solidification [2–6], while studies focused on conventional and investment casting processes are very limited [7]. The present work aims at covering this gap by investigating the effect of cooling rate on the microstructure of MAR-M247 Ni-based superalloy via physical simulation of the casting process. It has been shown that the dendrite spacing during directional solidification could be correlated with the cooling rate R in various Ni-based alloys, including MAR-M247 [2]. The same authors [4] have demonstrated that the microporosity depends on the hydrogen content and growth parameter, and that there is an optimum withdrawal velocity which leads to a minimum porosity and closely spaced dendrite arms. Moreover, Chen et al. [3] reported that the carbide volume fraction and average size increased with

increasing the growth rate. Nathal et al. [5,7] on the other hand, investigated the influence of chemical composition on the microstructure evolution. In the investment cast alloy, the γ' volume fraction and the mean particle size increased, while the amount of carbides present decreased with Co content [7]. In the single crystal alloy, substitution of Ni for Co caused an increase in γ' solvus temperature and in $\gamma-\gamma'$ lattice mismatch, as well as the precipitation of W-rich phases. On the other hand, substituting Ni for Ta led to large reduction in γ' solvus temperature, γ' volume fraction, and $\gamma-\gamma'$ lattice mismatch [5].

It should be finally noted that physical simulation has not been widely used to study the casting of Ni-based superalloys. Heck [8] studied the castability of directionally-solidified IN792 and CM247LC, while Robino [9] investigated the solidification and welding metallurgy of thermo-span alloy. In addition, Sekhar et al. [10,11] reported the solidification morphology and semisolid deformation in both equiaxed and directionally-solidified RENE-108 superalloy. Finally, Knorovsky et al. [12] used Gleeble simulation along with DTA analysis to develop the solidification diagram of IN 718 alloy. However, there is no information available in the open literature on the physical simulation of casting of MAR-M247.

In this article, a detailed study on the effect of cooling rate on the microstructure during casting is carried out via physical simulation route. The variation of microconstituents and dendritic arm spacing is compared and is correlated with the hardness.

2. Materials and methods

The material used in this investigation was a MAR-M247 superalloy of chemical composition (wt.%): C = 0.15, Cr = 8.0, Mo = 0.6, Ti = 1.0, Co = 10.0, W = 10.0, Ta = 3.0, Al = 5.5, Hf = 1.5, B = 0.015, Zr = 0.03

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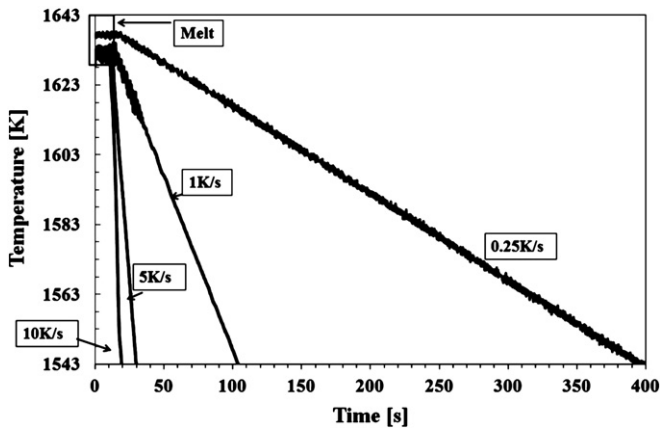


Fig. 1. Temperature vs. time curves recorded during solidification experiments.

and Ni=bal, supplied by Precicast Bilbao. Specimens with diameter of 10 mm and 121.5 mm in length were machined from larger ingots. A thin wire thermocouple (*R*-type) was welded to the midsection of each specimen to control the temperature during the solidification experiment. The liquid metal zone was contained in a quartz tube with a slot of 3.2 mm, which was put onto each specimen with the thermocouple in the slot. Thermocouple wires were fixed to the surface of the quartz tube by cement. Melting/solidification experiments were performed in the argon atmosphere using a GLEEBLE 3800 system for physical simulation.

MAR-M247 samples were heated at a rate of 40 K/s to 1513 K (1240 °C). In the temperature range of 1513–1593 K, the heating rate was reduced to 2 K/s. When 1593 K is reached, the temperature was manually increased very slowly up to ~1633 K until an evidence of melting was observed. The specimen was held at this temperature for 10 s and thereafter cooled down by 100 K at four different cooling rates, 0.25, 1, 5, and 10 K/s. These cooling rates were carefully chosen and correspond to the expected cooling rates in different parts of the mold during investment casting of MAR-M247 to manufacture nozzle guide vanes [13]. The experimental cooling rate was estimated as a slope of the temperature–time curve corresponding to the solidification stage. The samples were compressed in a stroke controlled manner to suppress formation of shrinkage porosity. Microhardness of the material was measured using SHIMADZU HMV-2 microhardness tester equipped with Vickers diamond indenter. The load of 300 g was applied for 15 s. Ten measurements were carried out in each sample.

A midsection of the melted zone was cut out for further microstructural studies. They were polished to the mirror-like surface finish using colloidal silica at the final stage and etched using Glyceregia (15 ml HCl, 10 ml glycerol and 5 ml HNO₃). The microstructure formed during solidification in the temperature-controlled area (the area under the weld point of thermocouple) was studied in scanning electron microscope EVO MA 15 at an operating voltage of 18 kV.

3. Results and discussion

The temperature–time plots for different cooling rates are presented in Fig. 1. They show that the cooling rate was very well controlled at

cooling rates of 0.25, 1, and 5 K/s. Some instabilities appeared at the highest cooling rate of 10 K/s, but this did not affect the average cooling rate. So, the experimental average cooling rates matched the set ones (Table 1). This indicates that the casting process with precisely controlled cooling rates can be successfully reproduced in the Gleeble thermomechanical simulator.

The cooling rate, R [K/s], is a product of the temperature gradient, G [K/mm] and the solidification rate, V [mm/s]. In order to estimate the solidification rate, additional experiments were carried out, in which the temperature gradients were measured. Two *R*-type thermocouples were welded to the surface of cylindrical specimens. One thermocouple was welded to the midsection and the other 4 mm aside. Temperature gradients, defined as the temperature difference measured by these thermocouples divided by the distance between thermocouples, were calculated from the temperature–time plots recorded by the thermocouples. As indicated in Table 1, they were ~1 K/mm for the cooling rates of 0.25 and 1 K/s and increased to ~5 K/mm for the cooling rates of 5 and 10 K/s. The difference is due to the geometrical variations of the copper grips used to hold the sample. For low cooling rates, partial contact copper grips were used with a contact length with the specimen of 12 mm, while full contact copper grips with a contact length with the specimen of 30.5 mm were used for high cooling rates. Hence, the solidification rates in the midsection of samples were calculated to be 0.25, 1, 1 and 2 mm/s, respectively (Table 1).

The initial microstructure of the MAR-M247 alloy consisted of γ Ni-rich solid solution containing a dispersion of γ' precipitates, discrete metal carbide (MC) particles, and γ/γ' eutectic pools as depicted in Fig. 2a. After solidification, the microstructure varied with the cooling rate. Common observation was the absence of pores, except from central shrinkage, whose size decreased with the solidification rate.

The sample cooled at 0.25 K/s showed cellular morphology with the same microstructural constituents as the as-cast sample: γ/γ' matrix, metal carbides and γ/γ' eutectic (Fig. 2b). The sample solidified at a cooling rate of 1 K/s showed an obvious transition from cellular to dendritic growth (Fig. 2c). The dendritic structure was well defined through formation of the secondary arms. These secondary dendrites result from capillary instabilities at the dendrite tip. These perturbations are small at first and then grow at the expense of the smaller arms in a process similar to Ostwald ripening. The driving force for this coarsening is the increase in interfacial energy at the tips of the smaller arms due to the higher curvature there [14]. The microstructure comprised only γ/γ' matrix and carbides, without the presence of γ/γ' eutectic. Microstructure of the samples solidified at 5 and 10 K/s was similar and contained primary dendrites, carbides and γ/MC eutectic pools resulting from the reaction $L \rightarrow \gamma + \text{MC}$ in which liquid transforms into two solid phases, γ and MC (Fig. 2d). The presence of the eutectic microstructure indicates that the undercooling of the last to solidify liquid and its composition fulfilled the conditions for eutectic solidification inside the coupled-growth zone (CGZ). CGZ is defined as a range of alloy composition and interface temperature (or velocity) where eutectic growth prevails upon solidification [15–17]. The observed microconstituents are summarized in Table 1 as a function of the cooling rate. The samples cooled at 1 and 5 K/s yielded the same solidification rate but differed in microstructure. This behavior was due to the differences in thermal gradient which affects the dendrite spacing and the microstructure evolution in the same manner as the solidification rate. The

Table 1

Nominal and measured cooling rates and their influence on the temperature gradient, solidification rate, microconstituents and microhardness of MAR-M247 alloy.

Cooling rate, R [K/s]		Temperature gradient, G [K/mm]	Solidification rate, V [mm/s]	Constituent phases	Hardness
Set	Measured				
0.25	0.250	1	0.25	$\gamma/\gamma' + \text{MC} + \gamma/\gamma'$ eutectic	403.2
1.00	1.001	1	1	$\gamma/\gamma' + \text{MC}$	432.0
5.00	5.026	5	1	$\gamma/\gamma' + \text{MC} + \gamma/\text{MC}$ eutectic	447.0
10.00	11.285	5	2	$\gamma/\gamma' + \text{MC} + \gamma/\text{MC}$ eutectic	449.2

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