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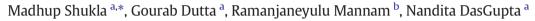
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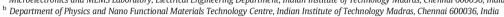


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## Electrical properties of reactive-ion-sputtered Al<sub>2</sub>O<sub>3</sub> on 4H-SiC







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#### ABSTRACT

Al $_2$ O $_3$  was deposited on n-type 4H-SiC by reactive-ion-sputtering (RIS) at room temperature using aluminum target and oxygen as a reactant gas. Post deposition oxygen annealing was carried out at a temperature of 1100 °C. Metal-oxide-semiconductor (MOS) test structures were fabricated on 4H-SiC using RIS-Al $_2$ O $_3$  as gate dielectric. The C-V characteristics reveal a significant reduction in flat band voltage for oxygen annealed RIS-Al $_2$ O $_3$  samples ( $V_{fb} > 10$  V), suggesting a reduction in negative oxide charge after oxygen annealing. Oxygen annealed RIS-Al $_2$ O $_3$  samples also showed significant improvement in I-V characteristics compared to as-deposited RIS-Al $_2$ O $_3$  samples. A systematic analysis was carried out to investigate the leakage current mechanisms present in oxygen annealed RIS-Al $_2$ O $_3$  on 4H-SiC at higher gate electric field and at different operating temperature. For measurement temperature (T > 303 K, Fowler-Nordheim (FN) tunneling was found to be the dominant leakage mechanism and for higher temperature (T > 303 K), a combination of FN tunneling and Poole-Frenkel (PF) emission was confirmed. The improvement in I-V characteristics of oxygen annealed RIS-Al $_2$ O $_3$ /4H-SiC MOS devices is attributed to large effective barrier height ( $\Phi_B = 2.53$  eV) at Al $_2$ O $_3$ /SiC interface, due to the formation of an interfacial SiO $_2$  layer during oxygen annealing, as confirmed from X-ray Photoelectron Spectroscopy results. Further improvement in C-V characteristics for oxygen annealed RIS-Al $_2$ O $_3$ /4H-SiC MOS devices was observed after forming gas annealing at 400 °C.

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#### 1. Introduction

Silicon Carbide (SiC) has unique physical and electrical properties like wider band gap (~3 eV), higher thermal conductivity (~4.9 W/cm), higher breakdown electric field (~3 MV/cm) and high electron saturation drift velocity ( $\sim 2.0 \times 10^7$  cm/s) [1], making it an excellent candidate for high temperature and high frequency applications. Along with this, SiC is the only compound semiconductor on which thermally stable silicon dioxide (SiO<sub>2</sub>) can be grown [2]. This has led to the development of SiC based metal-oxide-semiconductor (MOS) technology, particularly in the field of power electronics. However, the commercialization of SiC MOS devices with SiO<sub>2</sub> as gate dielectric remains a challenge because of low channel mobility due to high SiO<sub>2</sub>/SiC interface states near the conduction band edge of SiC [3]. Furthermore, as the dielectric constant of SiO<sub>2</sub> is low compared to that of 4H-SiC ( $k_{SiC} = 10$ ), the electric field in SiO<sub>2</sub> is 2.5 times higher than that in SiC resulting in a premature dielectric breakdown [4]. This leads to serious reliability issues for SiO<sub>2</sub>-SiC MOS devices as well as, underutilization of high breakdown electric field of SiC, which is an asset of SiC for high power applications. These limitations of SiO<sub>2</sub> have prompted researchers to investigate the use of high-k dielectrics on SiC. Among several high-k dielectrics, Al<sub>2</sub>O<sub>3</sub> has generated a lot of interest due to its excellent properties such as high dielectric constant ( $k_{Al2O3} \sim 10$ ), large band gap ( $\sim 6.2$  eV), chemical inertness, excellent conduction band and valence band offset with 4H-SiC [5].

In the past, several techniques such as, atomic-layer-deposition (ALD, using either ozone [5] or  $H_2O$  [6] as an oxidant), nitric acid oxidation [7] and RF-sputtering (using  $Al_2O_3$  target at a deposition temperature of 220 °C) [8], have been tried to deposit  $Al_2O_3$  on 4H-SiC. High temperature annealing of  $Al_2O_3$  on 4H-SiC at different temperatures (800 °C to 1100 °C) [5,9,10,11] and in various gas ambient ( $N_2/O_2/Ar$ ) [5,10,11] has demonstrated reduction in capacitance-voltage (C-V) hysteresis and flat band voltage ( $V_{fb}$ ) by trap passivation.

In this work, for the first time we report on the electrical characteristics of  $Al_2O_3$  deposited by reactive-ion-sputtering (RIS) on 4H-SiC at room temperature, using Aluminum (Al) as target. Our earlier work has shown RIS deposited  $Al_2O_3$  as a promising gate dielectric for Gallium Nitride (GaN) based metal-insulator-semiconductor (MIS) devices [12]. RIS deposition technique provides several advantages such as low-cost, uniform deposition at lower temperatures and no formation of any by-products and contaminants. Annealing in different gas ambient was carried out in order to improve bulk and interface properties of RIS- $Al_2O_3$  on 4H-SiC. Significant improvement in gate leakage current was observed after oxygen annealing. Analysis of gate leakage current in oxygen annealed RIS- $Al_2O_3$  on 4H-SiC revealed the presence of Fowler–Nordheim (FN) tunneling for below room temperature operation and a combination of FN tunneling and Poole-Frenkel (PF) emission for above room temperature operation. RIS- $Al_2O_3$  with subsequent

<sup>\*</sup> Corresponding author.

E-mail address: madhup.iit@gmail.com (M. Shukla).

oxygen annealing showed better/comparable gate leakage performance as compared to  $Al_2O_3$  deposited by other methods or in stacked configuration ( $Al_2O_3/SiO_2$  stack) on 4H-SiC. Further improvement in C-V characteristics of oxygen annealed RIS-Al $_2O_3$  was observed by annealing it in forming gas (5%  $H_2$  and 95%  $N_2$ ) ambient. X-Ray photoelectron spectroscopy (XPS) studies were carried out to analyze any change in compositional ratio at bulk/interfacial oxide due to annealing.

#### 2. Experimental details

300 µm thick, n-type (Nitrogen doped) 4H-SiC with a doping concentration of about  $1 \times 10^{19}$  cm<sup>-3</sup>, has been used in this work as bulk substrate to fabricate MOS capacitor structures. Three sets of devices were fabricated, designated as S<sub>0</sub> (MOS capacitors with as-deposited RIS-Al<sub>2</sub>O<sub>3</sub> as gate dielectric), S<sub>1</sub> (MOS capacitors with oxygen annealed RIS-Al<sub>2</sub>O<sub>3</sub> as gate dielectric) and S<sub>2</sub> (MOS capacitors with oxygen + forming gas annealed RIS-Al<sub>2</sub>O<sub>3</sub> as gate dielectric). The samples were cleaned by organic solvents and piranha solution to remove organic impurities. Then standard Radio Corporation of America (RCA) [13] cleaning was performed on the samples. The cleaned 4H-SiC samples were immediately loaded in the RF sputtering chamber. Al<sub>2</sub>O<sub>3</sub> was deposited at room temperature, using 99.999% pure Al as target. The RF sputtering chamber was initially evacuated to a base pressure of  $5 \times 10^{-7}$  mbar. Then, Ar and O<sub>2</sub> gasses were introduced into the chamber at a flow rate of 30 sccm and 5 sccm respectively. A process pressure of  $1.2 \times 10^{-2}$  mbar was maintained during sputter deposition. RF input power was kept at 100 W. The sputtered Al reacted with O<sub>2</sub> in the plasma to form Al<sub>2</sub>O<sub>3</sub> and get deposited on 4H-SiC surface. This Al<sub>2</sub>O<sub>3</sub> is designated as RIS-Al<sub>2</sub>O<sub>3</sub>. The rate of deposition of RIS-Al<sub>2</sub>O<sub>3</sub> on 4H-SiC was ~1.5 nm/min. After the deposition, samples S<sub>1</sub> and S<sub>2</sub> underwent annealing in O<sub>2</sub> ambient at a temperature of 1100 °C for around 2 h. Sample S<sub>2</sub> underwent an additional annealing in forming gas at 400 °C for another 2 h. Subsequently, Al was deposited over  $Al_2O_3$ , for all the three samples ( $S_0$ ,  $S_1$  and  $S_2$ ), by thermal evaporation in vacuum and then patterned by photolithographic technique to form an array of circular metal gate contacts of 40 µm diameter. Finally, back aluminum metallization was carried out after back oxide etching by buffered hydro fluoric acid (BHF). Multi-frequency C-V and current-voltage-temperature (I-V-T) measurements in dark were performed using Agilent's B1500A semiconductor device analyzer and Cascade Microtech Summit 12000AP thermal chuck probe station. The I-V measurements were carried out at different measurement temperatures (T) from 223 K to 393 K to obtain the gate current density ( $J_G$ ) vs. gate electric field  $(E_G)$  plot given by Eqs. (1) and (2).

$$J_G = I_G/A \tag{1}$$

and

$$E_G = (V_G - \Phi_{ms})/EOT. \tag{2}$$

Here A,  $\Phi_{ms}$  and EOT are gate area, metal–semiconductor work function difference and equivalent oxide thickness respectively. In order to study the film composition of oxygen annealed RIS-Al $_2$ O $_3$  on 4H-SiC, XPS measurements were carried out using SPECS with PHIBOS 100 energy analyzer (SPECS GmbH, Germany). Mg K $\alpha$  radiations were used to excite electrons.

#### 3. Results and discussion

The physical thickness of as-deposited RIS-Al $_2O_3$  and oxygen annealed RIS-Al $_2O_3$  as measured by a spectroscopic ellipsometer (J.A. Woollam EC 400) were around 18 nm and 22 nm respectively. The dielectric constant of as-deposited RIS-Al $_2O_3$  was calculated to be around 5.5 which is comparable to the previous reported result on sputtered-Al $_2O_3$  [14] for similar oxide thickness. From the value of the

accumulation capacitance and physical thickness (~22 nm) of the gate oxide, the dielectric constant and *EOT* of oxygen annealed RIS-Al<sub>2</sub>O<sub>3</sub> of sample S<sub>1</sub> was calculated to be around 5.1 and 17 nm respectively. The higher physical thickness and lower value of dielectric constant of oxygen annealed RIS-Al<sub>2</sub>O<sub>3</sub> in comparison with as-deposited RIS-Al<sub>2</sub>O<sub>3</sub> is attributed to the growth of interfacial SiO<sub>2</sub> layer during oxygen annealing as discussed later. The I-V and C-V measurements were first carried out at room temperature for both samples S<sub>0</sub> and S<sub>1</sub>. As can be seen from Fig. 1,  $J_G$  for as-deposited RIS-Al<sub>2</sub>O<sub>3</sub> of sample S<sub>0</sub> is  $4.67 \times 10^{-7}$  A/cm<sup>2</sup> at a low gate electric field of 3 MV/cm which is reasonably good, however the breakdown field strength is low. Post deposition annealing in oxygen ambient (sample S<sub>1</sub>) further improves the I-V characteristics in terms of leakage current and dielectric breakdown field strength (9 MV/cm) as shown in Fig. 1.

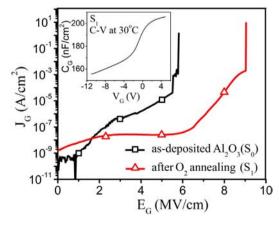
Inset of Fig. 1 shows the C-V characteristics of sample  $S_1$  with a flat band voltage  $(V_{fb})$  of 1.95 V. However, the C-V curve of  $S_0$  (not shown) could not reach accumulation capacitance because of its large positive flat band voltage  $(V_{fb} > 10 \text{ V})$ . In order to reach accumulation for  $S_0$ , a much large positive gate voltage had to be applied, leading to oxide breakdown. This large value of positive  $V_{fb}$  is due to the presence of large negative fixed oxide charge density  $(Q_f)$  in as-deposited RIS-Al<sub>2</sub>O<sub>3</sub>. In earlier reports [15,16], the negative  $Q_f$  in Al<sub>2</sub>O<sub>3</sub> has been attributed to intrinsic and stable oxide defects such as oxygen interstitials [17,18]. Oxygen annealing on RIS-Al<sub>2</sub>O<sub>3</sub> improves the C-V characteristics (by reducing negative  $Q_f$  in RIS-Al<sub>2</sub>O<sub>3</sub>) as well as improves its gate leakage current. Investigation of conduction mechanisms, affecting the gate leakage current in oxygen annealed RIS-Al<sub>2</sub>O<sub>3</sub> on 4H-SiC was carried out in order to understand the quality of Al<sub>2</sub>O<sub>3</sub>/SiC interface and oxide trap energy levels.

## 3.1. Study of leakage current conduction mechanism in oxygen annealed RIS-Al<sub>2</sub>O<sub>3</sub> on 4H-SiC

In order to study the gate leakage current through oxygen annealed RIS-Al<sub>2</sub>O<sub>3</sub> at moderate-to-high electric fields, various leakage current mechanisms were explored.  $J_G$  vs.  $E_G$  characteristics of sample S<sub>1</sub> at different T (223 K  $\leq T \leq$  393 K) are shown in Fig. 2(a). Presence of FN tunneling has been previously reported in Al<sub>2</sub>O<sub>3</sub> on 4H-SiC [6]. Temperature independent FN tunneling occurs in thicker oxides and at higher  $E_G$  [19] and is given by Eq. (3) [20].

$$J_G = AE_G^2 \exp\left[\frac{-B}{E_G}\right] \tag{3}$$

where, 
$$B = (8\pi \sqrt{2m_n^*(q\phi_B)^3})/3qh$$
.



**Fig. 1.** Room temperature  $J_G$  vs.  $E_G$  characteristics of sample  $S_0$  and  $S_1$ . Inset: C-V characteristics of sample  $S_1$ .

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