



## Enhancing dye-sensitized solar cell efficiency by anode surface treatments



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### ABSTRACT

In this study, titanium substrates treated with HF solution and KOH solution sequentially forming micro- and nano-structures were used for the fabrication of flexible dye-sensitized solar cells (DSSCs). After wet etching treatments, the titanium substrates were then exposed to the O<sub>2</sub> plasma treatment and further immersed in titanium tetrachloride (TiCl<sub>4</sub>) solution. The process conditions for producing a very thin TiO<sub>2</sub> blocking layer were studied, in order to avoid solar cell current leakage for increasing the solar cell efficiency. Subsequently, TiO<sub>2</sub> nanoparticles were spin-coated on Ti substrates with varied thickness. The dye-sensitized solar cells on the titanium substrates were subjected to simulate AM 1.5 G irradiation of 100 mW/cm<sup>2</sup> using backside illumination mode. Surface treatments of Ti substrate and TiO<sub>2</sub> anode were found to play a significant role in improving the efficiency of DSSC. The efficiencies of the backside illumination solar cells were raised from 4.6% to 7.8% by integrating these surface treatments.

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### 1. Introduction

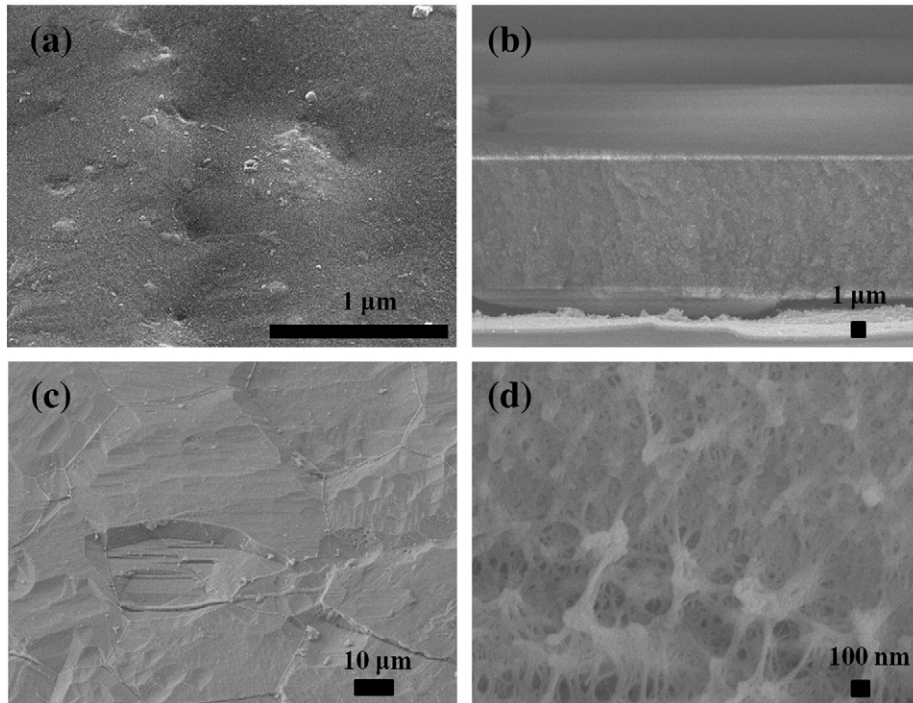
Significant progress has been recently achieved in developing dye-sensitized solar cells (DSSCs) with low cost and eco-friendly characteristics and high solar-power efficiencies (~15%) [1]. However, a typical DSSC is restricted by the heavy, rigid, and fragile transparent conductive oxide glass as the photoanode. Hence, flexible and lightweight substrates such as metal foil and plastics are required for roll to roll mass production, which can broaden the applications of DSSC. Nevertheless, the necking between TiO<sub>2</sub> nanoparticles remain poor by heating them up to 150 °C, an upper temperature limit for the plastic substrate to be thermally stable, resulting in low solar cell efficiency [2]. So many research groups have proposed the usage of metal substrates, in particular titanium (Ti) substrate, as the anode. Ti substrate is a good alternative to the plastic substrate due to its low sheet resistance, good flexibility, and superior corrosion resistance [3,4]. Grätzel et al. [5] reported highly efficient (7.2%) flexible DSSCs using a Ti foil substrate.

Aiming at improving the cell performance, previous researchers have proposed several one-dimension (1D) TiO<sub>2</sub> nanostructures to replace TiO<sub>2</sub> nanoparticles [6–8]. Those 1D nanostructures can transfer electrons

faster from dye molecules to the substrate for raising the collection efficiency. But, 1D TiO<sub>2</sub> nanostructures are not so densely packed leading to less dye adsorption. Wang et al. introduced nanotube morphology in conjunction with oxygen plasma, which compensated for less internal surface area and dye adsorption, and achieved a high power conversion efficiency (PCE) of 7.37% [9]. In addition, some other researchers investigated the effect of the underlayers between Ti substrate and TiO<sub>2</sub> film to enhance the electrical contact and suppress the dark current. For example, Yun et al. reported that the roughened substrates increased the PCE of DSSC on a stainless substrate from 4.3% to 5.7% [10]. Lee et al. revealed a PCE of 6.75% by introducing a sponge-like TiO<sub>2</sub> interfacial layer [11]. Tsai et al. obtained a high PCE of 7.1% by introducing an underlayer of TiO<sub>2</sub> network nanosheet [12]. Besides, some researchers sputtered Ti, Ni, Cu, and Al on the fluorine-doped tin oxide (FTO) to form a blocking layer and prevented the current leakage due to the recombination of electrons in FTO with I<sub>3</sub><sup>-</sup> in electrolyte [13,14].

Consequently, this study proposes various surface-treatment methods including TiCl<sub>4</sub> pre-treatments [5,15] and post-treatments [16–18], micro-nanostructures made by HF-KOH etching, and oxygen plasma pre-treatment. These surface treatments may act as the connection layer, the light scattering layer, and the blocking layer. This study further explores the effects of TiO<sub>2</sub> thickness and the O<sub>2</sub> plasma exposure time on the cell efficiency. After integrating these treatments using the optimized conditions, the PCE of the DSSC is dramatically improved.

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**Fig. 1.** (a) Top-view SEM image of untreated Ti substrate, (b) cross-section image of a 11.7  $\mu\text{m}$   $\text{TiO}_2$  film deposited on the untreated Ti substrate, and (c, d) top-view images of Ti substrates treated sequentially with HF and KOH.

## 2. Experimental details

### 2.1. Ti substrate surface treatment

Commercial Ti foil substrates (0.1 mm thickness, 99.9% purity, E-light Technology Inc., Taiwan) were etched by 1.8 wt.% HF for 2 min, followed by 3 M KOH at room temperature for 1 h. The substrates were then cleaned subsequently with de-ionized water, acetone, isopropanol and dried with  $\text{N}_2$ . Next, the cleaned substrates were exposed to  $\text{O}_2$  plasma at 30 W of power for various times. Chemical treatment was further performed by soaking the Ti foils in 40 mM  $\text{TiCl}_4$  aqueous solution at 70  $^\circ\text{C}$  for 30 min. Prior to immersing the treated foils in dye solvent,  $\text{TiCl}_4$  treatment was repeated again at the same condition.

### 2.2. Preparation of a $\text{TiO}_2$ film photoanode

The following steps were used to prepare  $\text{TiO}_2$  colloidal solution: (1) mix 6 g of  $\text{TiO}_2$ , Degussa P25 powder, with 10 g of de-ionized

water and stir the solution for 1 h with magnet. (2) Add 100  $\mu\text{l}$  of Triton X-100 (Panreac) and 200  $\mu\text{l}$  of acetyl acetone (Alfa Aesar) to  $\text{TiO}_2$  colloidal solution, stir it for 1 h, in order to disperse the solution; and (3) put the solution in an ultrasonic bath at a constant temperature until the solution was uniformly mixed. Then, the solution was spread onto Ti foil by spin coating with a  $\text{TiO}_2$  photoelectrode active area of 0.16  $\text{cm}^2$  ( $0.4 \times 0.4 \text{ cm}^2$ ). The sample was sintered at 450  $^\circ\text{C}$  for 30 min to remove excess solvent and dispersant in the film and to create good necking between  $\text{TiO}_2$  particles. After being cooled to 80  $^\circ\text{C}$ , the  $\text{TiO}_2$  electrode was immersed in a 0.5 mM N719 dye dissolved in the mixture of acetonitrile and *tert*-butyl alcohol (volume ratio: 1:1) and kept at room temperature for 24 h.

### 2.3. Fabrication and characterization of the DSSCs

The indium tin oxide/poly(ethylene terephthalate) (Pecell Technology Inc., Japan) counter electrode was coated with a layer of platinum by sputtering at a current of 40 mA for 20 s. The photoelectrode was assembled with the counter electrode separated by 60  $\mu\text{m}$  thick hot-melt spacers (SX1170-60, Solaronix). Then, the electrolyte, consisting of 0.1 M guanidinium thiocyanate, 0.03 M  $\text{I}_2$ , 0.6 M 1-butyl-3-methyl imidazolium iodide and 0.5 M 4-*tert*-butylpyridine in a mixture of acetonitrile and valeronitrile (volume ratio: 85:15), was injected into the solar cell. The photocurrent-voltage (*I*-*V*) characteristics of the DSSCs were measured using a solar simulator (Newport, Oriel class A, 91160A) under an irradiation of AM 1.5, 100  $\text{mW}/\text{cm}^2$ . The surface morphology was characterized by scanning electron microscopy (SEM, JEOLJSM-6700F) using 10 kV accelerating voltage. The IPCE spectra were measured by using the monochromatic light from a specially designed system made of a xenon lamp, a monochromator, and filters. A time-of-flight secondary ion mass spectrometry (TOF-SIMS, Cameca IMS-6f) with  $\text{Cs}^+$  ion gun operated at 15,385 eV was used for the O element depth profile analysis of the Ti substrate surface with and without the oxygen plasma treatment. The Ti substrate surface with and without the oxygen plasma treatment was also characterized by

**Table 1**

Photovoltaic performance of DSSCs using smooth and micro-/nano-structured Ti foil as photoanode followed by spin coating with  $\text{TiO}_2$  nanoparticles for different times.

Spin coating time	$\text{TiO}_2$ thickness ( $\mu\text{m}$ )	$V_{oc}$ (V)	$J_{sc}$ ( $\text{mA}/\text{cm}^2$ )	FF	$\eta$ (%)
<i>Smooth Ti foil</i>					
1	5.5	0.79	7.55	0.7	$4.2 \pm 0.05$
2	11.4	0.78	7.94	0.74	$4.6 \pm 0.02$
2 <sup>a</sup>	–	0.79	8.64	0.73	$5.0 \pm 0.04$
3	17	0.76	7.4	0.75	$4.2 \pm 0.03$
<i>Micro-nanostructures Ti foil</i>					
1	5.5	0.81	8.34	0.69	$4.7 \pm 0.03$
2	11.4	0.76	10.1	0.71	$5.4 \pm 0.04$
2 <sup>(a)</sup>	–	0.78	10.57	0.71	$5.9 \pm 0.05$
3	17	0.74	9.98	0.67	$4.9 \pm 0.02$

<sup>a</sup>  $\text{TiCl}_4$  post-treatment.

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