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Structures, and far-infrared and Raman spectra of $GdMn_{1-x}Co_xO_3$ (x = 0-1.0)

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Available online 8 April 2010	We report the structural and optical properties of the perovskite $GdMn_{1-x}Co_xO_3$ ($x=0-1.0$) compounds. Far-infrared (FIR) and Raman spectroscopy measurements were carried out at room temperature. The FIR spectra show four vibration modes at ~190, ~250, ~400, and ~600 cm ⁻¹ that correspond to external, torsional, bending, and stretching modes, respectively. It was found that the FIR active vibration energies
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Phonon modes Far-infrared spectroscopy exhibit splitting and shifting as the Co doping level increases. The Raman spectra reveal that the most intense spectral feature in the GdMnO₃ is at 612 cm^{-1} , which shifts to higher and lower energies, depending on the Co doping level.

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1. Introduction

The multiferroic materials which exhibit magnetism and the ferroelectric effect have drawn great attention due to their potential in emerging technological applications, as well as their exotic physical properties. The rare-earth compounds $RMnO_3$ (with R a rare earth element) have been numbered among these multiferroic materials since a very large magnetoelectric effect was discovered [1–3]. These RMnO₃ systems are interesting compounds for the study of the correlation between spin-phonon coupling and the magnetoelectric effect. This correlation can be examined by tuning the Mn³⁺ magnetic structure using different rare-earth and transition metal ions. The structures of the RMnO₃ compounds can be grouped into hexagonal structure for R = Ho–Lu and distorted orthorhombic structure for R = La - Dv. The hexagonal RMnO₃ compounds show both ferroelectric and magnetic order, but the corresponding ordering temperatures differ by an order of magnitude. GdMnO₃, TbMnO₃, and DyMnO₃ have comparable transition temperatures. TbMnO₃ and DyMnO₃ exhibit an incommensurate antiferromagnetic order below 40 K [1-6]. Cobalt is one of the typical transition metals that can significantly affect the magnetic and optical properties of the RMnO₃ compounds [7]. Infrared active phonons have been observed and investigated for $HoMn_{1-x}Co_{x}O_{3}$ [8]. Co doping shows a great effect on the structures and on the spin glass and spin states in $GdMn_{1-x}Co_xO_3$ for x < 0.5 [7]. In this paper, we report the crystal structures and active phonon

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modes of $GdMn_{1-x}Co_xO_3$ from x=0 to 1.0, with a step size of 0.1, using far-infrared (FIR) and Raman spectroscopy.

2. Experiments

Polycrystalline samples of $GdMn_{1-x}Co_xO_3$ (x=0-1.0) were synthesized by standard solid-state reaction. The appropriate amounts of Gd₂O₃, Mn₂O₃, and CoCO₃ were well mixed, pelletized, and sintered at 900 °C for 24 h in air. The final sintering was carried out at 1200 °C for 12 h. The structures of the GdMn_{1-x}Co_xO₃ (x = 0.2-0.9) compounds were characterized by using a Philips diffractometer with $CuK\alpha$ radiation. For FIR transmission experiments, the $GdMn_{1-x}Co_xO_3$ (x=0-1.0) powders were diluted in CsI, finely milled, and then pressed into pellets 1 mm in thickness and 10 mm in diameter. A pure CsI pellet was also prepared as a reference sample. The FIR transmission measurements were carried out employing a Bomen DA3.26 rapid scan interferometer with a detector that has a range of $10-700 \text{ cm}^{-1}$. The resolution and scan speed were 4 cm^{-1} and 0.2 cm^{-1} per second, respectively. Raman spectroscopy measurements were carried out using a JOBIN YVON HR800 Confocal Raman system with 632.8 nm diode laser excitation on a 300 lines/mm grating in the range of 100-2000 cm⁻¹. Both FIR and Raman spectra were obtained at room temperature.

3. Experimental results and discussion

The crystal structure of $GdMn_{1-x}Co_xO_3$ is schematically shown in Fig. 1. The calculated and observed X-ray diffraction patterns of the GdMn_{0.7}Co_{0.3}O₃ sample are shown in Fig. 2. The (Mn/Co)O₆ octahedron distorts with the various Co doping levels. The Rietveld refinement results show that $GdMn_{1-x}Co_xO_3$ is crystallized in the orthorhombic structure with space group Plmn. The lattice

7.5





Fig. 1. The crystal structure of $GdMn_{1-x}Co_xO_3$.

parameters of $GdMn_{1-x}Co_xO_3$ as a function of *x* are shown in Fig. 3. It can be seen that the lattice parameter *b* remains constant while the lattice parameters *a* and *c* increase with increasing Co content. This is an indication that the (Co/Mn)O₆ octahedron becomes more elongated along the *a*-axis with larger amounts of cobalt doping.

Fig. 4 shows the doping dependent far-infrared transmission spectra of GdMn_{1-x}Co_xO₃ in the range of 130–700 cm⁻¹. For the *Pbnm* space group of orthorhombic RMnO₃, a factor group analysis predicts 9B1u + 7B2u + 9B3u infrared-active modes and 7Ag + 7B1g + 5B2g + 5B3g Raman-active modes. These modes are associated with three phonon bands that correspond to the normal modes of the ideal cubic perovskite: the external mode is due to *R* and MnO₆, while the bending and stretching modes are mainly due to the oxygen vibration [6,9]. In FIR

transmission spectra of GdMnCoO₃ samples, the most active phonon modes at ~190 cm⁻¹, ~250 cm⁻¹, ~400 cm⁻¹ and ~600 cm⁻¹ are assigned to the external (E), torsional (T), bending (B), and stretching (S) modes, respectively. In Table 1, we list the value these four modes at various doping obtained from far-infrared measurement. There are some splits between the vibration bands. These results are similar to the FIR reflectivity spectrum observed from a GdMnO₃ sample [6,10]. The features of the phonon modes of GdMn_{0.5}Co_{0.5}O₃ (*R* = La, Nd, Dy, Ho, and Yb) [11]. The external band is at ~190 cm⁻¹, and weak bands are seen on both sides (~172 cm⁻¹, 210 cm⁻¹). The torsional band is at 290 cm⁻¹, and a weak band starts to develop at 272 cm⁻¹ when *x* = 0.5. Both the bending and the stretching bands broaden at ~400 cm⁻¹ and ~600 cm⁻¹.



Fig. 2. The calculated (solid line) and observed (symbols) XRD patterns for the GdMn_{0.7}Co_{0.3}O₃ sample. The bottom solid line shows the difference between them, and the short blue lines indicate the line positions of the standard.

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