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# Reactively sputtered TiO<sub>2</sub> layers on SnO<sub>2</sub>:F substrates: A Raman and surface photovoltage study

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# Abstract

DC reactively sputtered TiO<sub>2</sub> layers on SnO<sub>2</sub>:F substrates were investigated by Raman and surface photovoltage spectroscopy. The stoichiometry and layer thicknesses were investigated by elastic recoil detection analysis. The deposition temperature, the  $O_2/(O_2 + Ar)$  ratio and the deposition time were varied systematically. With increasing temperature, the layers become crystalline with the rutile modification dominating. Rutile phase preferentially forms on vertical facets of SnO<sub>2</sub> crystallites. Anatase phase starts to form during prolonged deposition and at lower  $O_2/(O_2 + Ar)$  ratios. The energy of the exponential absorption tails below the band gap, a measure of the defect density of the films, is determined by the deposition temperature and not by other parameters if the deposition temperature is relatively high, irrespective of the content of crystalline phases or the value of the band gap. Charge separation takes place at length scales significantly shorter than the layer thicknesses (diffusion length less than 6 nm). TiO<sub>2</sub> films sputtered at 380 °C show rectifying behaviour with a carbon contact. © 2007 Elsevier B.V. All rights reserved.

Keywords: Titanium oxide; Thin films; Direct current reactive sputtering; Surface photovoltage spectroscopy; Raman spectroscopy; Elastic recoil detection analysis

### 1. Introduction

TiO<sub>2</sub> layers are of great interest for very different applications in fields as gas sensors [1], photovoltaics [2], photocatalysis [3] or ultraviolet photosensing [4] with Pt/TiO<sub>2</sub> Schottky-diodes [5]. TiO<sub>2</sub> films can be prepared by various methods. One of them is the relatively simple reactive DC sputtering technique from elemental targets which allows, for example, an excellent control of stoichiometry under very pure conditions [6]. Reactively sputtered TiO<sub>2</sub> layers are also suitable for photocatalysis [7] or application in dye sensitized solar cells [8].

The structure, chemical bonds and electronic properties of reactively sputtered  $TiO_2$  layers depend on the sputter parameters as deposition temperature, total sputtering power, discharge power or oxygen content in the sputtering gas. The correlation of the electronic properties of reactively sputtered  $TiO_2$  layers with the morphology and the content of  $TiO_2$  phases (amorphous, rutile, brookite and anatase) is a major issue of the current

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research. For example, phase transitions from amorphous  $TiO_2$  to the anatase phase can be stimulated by the plasma parameters concomitant with substrate heating. It has been shown, that crystalline TiO<sub>2</sub> phases start to form at relatively low substrate temperatures (below 200 °C) during reactive DC sputtering [8]. For comparison, the phase transition from anatase to rutile proceeds typically at temperatures of the order of 600 °C under equilibrium conditions while the phase transformation of the surface region of anatase nanoparticles demands significantly higher temperatures [9]. Theoretical studies predict phase and shape transitions of TiO<sub>2</sub> nanoparticles controlled by surface chemistry [10]. For reactive sputtering, the impact of energetic particles, like sputtered atoms and ions from the plasma, additionally influences the chemical processes.

In this work, we correlate qualitatively structural with electronic properties of reactively DC sputtered  $TiO_2$  layers. The deposition temperature, the oxygen content in the sputtering gas, and the sputtering time were varied systematically in our experiments. The electrical quality of the sputtered  $TiO_2$  layers is demonstrated by current–voltage characteristics of  $SnO_2$ :F/TiO<sub>2</sub>/ graphite structures. Raman spectroscopy is applied to identify the phases of rutile and anatase in the  $TiO_2$  layers from the specific

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Raman modes [11]. Surface photovoltage spectroscopy (SPS) [12] is used to characterize the surface work function, the surface photovoltage amplitude, the optical band gap of  $\text{TiO}_2(E_g)$ , and the so-called Urbach-energy ( $E_t$ ) of the exponential absorption tails below the band gap [13]. The value of  $E_g$  amounts to 3.06 eV for the rutile phase [14]. For the anatase phase,  $E_g$  is increased by 0.2 eV in comparison to rutile [15]. It has to be noted that by SPS studies,  $E_t$  is only related to optical transitions into states from which charge separation is possible. It will be shown that different growth modes may play a role at different facets on the substrates surface and lead locally to preferential formation of anatase or rutile phases.

# 2. Experimental details

# 2.1. Deposition of $TiO_2$ layers by DC magnetron sputtering

The TiO<sub>2</sub> layers were deposited by reactive DC magnetron sputtering from a pure Ti target on glass substrates covered with SnO<sub>2</sub>:F (TEC8). Before deposition, the substrates (1×1 in.<sup>2</sup> size) were cleaned in C<sub>2</sub>H<sub>5</sub>OH:H<sub>2</sub>O=1:1 solution in an ultra-sonic bath and dried with N<sub>2</sub> gas. For the TiO<sub>2</sub> layer deposition, a home made sputtering and evaporation system (basic pressure  $5 \cdot 10^{-5}$  Pa) has been used. The Ti sputter target (3 in. diameter, 99.7% purity) was pre-sputtered for three minutes before reactive sputtering onto SnO<sub>2</sub>:F substrate layers was performed. The distance between the sputter target and the substrate was 66 mm for all depositions. The sputter plasma was generated at a constant power of 100 W (DC).

The deposition temperature, the oxygen content in the sputter gas, and the deposition time were systematically varied. The substrates were heated with a halogen lamp (up to 380 °C). The oxygen content in the gas flow was changed by variation of the  $O_2/(O_2 + Ar)$  ratio from 0.032 to 0.120 while the standard Ar flow was 48 sccm. The purities of the Ar and  $O_2$  gases were 99.999 and 99.99%, respectively. The pressure during the deposition was about 0.4 Pa. The deposition times varied between 5 and 60 min. The thickness of the sputtered TiO<sub>2</sub> layers has been changed over one order of magnitude.

#### 2.2. Sample characterization

The morphology of the sputtered  $\text{TiO}_2$  layers (lateral views and cross-sections) was investigated with a LEO 1530 (Gemini) *field emission scanning electron microscope*. The acceleration voltage was kept at 10 kV.

For thin layers, the ERDA (elastic recoil detection analysis) technique gives information about the areal density of the different atomic masses. The Berlin time-of flight ERDA setup at the Hahn-Meitner-Institute [16] was applied for the investigation of the stoichiometry and of the areal densities of Ti, O, Ar, Si, Fe, C, H in the sputtered TiO<sub>2</sub> layers. Areal density of Ti and corresponding TiO<sub>2</sub> thickness with sputter times for TiO<sub>2</sub> layers prepared at deposition temperature of  $380^{\circ}$ C and O<sub>2</sub>/(O<sub>2</sub>+Ar) ratio of 0.08 are given in Table 1. The thickness scales vary well with the deposition time. The areal density of Ti atoms, i.e. stoichiometry of TiO<sub>2</sub> was reached. The thickness

Table 1			
Areal density of Ti and	corresponding	TiO <sub>2</sub> thickness	with sputter time

Sputter time (min)	Areal density of TiO <sub>2</sub> (×10 <sup>17</sup> Ti/cm <sup>2</sup> )	TiO <sub>2</sub> thickness <sup>a</sup> (nm)
5	0.28	6
15	0.78	15
30	1.60	32
60	3.50	65

Deposition temperature and  $O_2/(O_2 + Ar)$  ratio for sputtered TiO<sub>2</sub> layers were 380 °C and 0.08, respectively.

<sup>a</sup> Assumed the density of  $TiO_2 = 4 \text{ g/cm}^3$  (rutile: 4.25 g/cm<sup>3</sup>, anatase: 3.89 g/cm<sup>3</sup>).

depends as well on  $O_2/(O_2 + Ar)$  ratio. For  $O_2/(O_2 + Ar)$  ratio of 0.04 and deposition time of 30 min, a 45 nm TiO<sub>2</sub> layer was obtained. The total concentration of Ar, C, Fe and Si atoms was less than 5% for the long sputter times and about 10% for the sputter time of 5 min. If assuming a density of TiO<sub>2</sub> equal to 4 g/ cm<sup>3</sup> (rutile: 4.25 g/cm<sup>3</sup>, anatase: 3.89 g/cm<sup>3</sup>), the thicknesses of the TiO<sub>2</sub> layers amount to 6, 15, 32 and 65 nm for sputter times of 5, 15, 30 and 60 min, respectively (deposition temperature 380 °C,  $O_2/(O_2 + Ar)$  ratio 0.08).

Some electrical properties of the sputtered  $TiO_2$  layers were checked by using small area  $SnO_2$ :F/TiO\_2/graphite structures (area about 4 mm<sup>2</sup>) which were manually fabricated with carbon ink. Current–voltage measurements were performed at room temperature in the dark with a source-measure unit (Keithley, SMU 237).

The anatase and rutile phases of  $TiO_2$  were identified by Raman spectroscopy. A Raman spectrometer with a HeNe laser (wavelength 632.8 nm) for excitation has been used. The spectrometer was calibrated with the phonon line of Si. The accumulation time was 300 s for all spectra. The spectra were normalized to the substrate (glass/SnO<sub>2</sub>:F).

The work function of the sputtered  $\text{TiO}_2$  layers was measured in air with a Kelvin probe (Besocke). Surface photovoltage spectra were obtained as the light-induced change of the work function. The SPS measurements were performed at room temperature. A quartz-prism monochromator (SPM2) with a Xe-lamp (1000 W) has been used for illumination (photon energies between 2.5 and 4.5 eV were used, the spectral resolution was better than 30 meV). Before starting the SPS measurements, the samples were kept in the dark until the value of the work function became stable. Values of the forbidden energy band gap and of the energy of the exponential defect tails below the band gap were obtained from the analysis of the surface photovoltage spectra.

### 3. Results and discussion

### 3.1. Morphology and electrical properties

Fig. 1 shows a typical cross-section of a  $TiO_2$  layer sputtered on  $SnO_2$ :F at 380 °C. The thicknesses of the  $SnO_2$ :F and  $TiO_2$ layers are about 500 and 50 nm, respectively. The large grains are related to the polycrystalline  $SnO_2$ :F substrate layer (thickness about 500 nm). The most striking feature of Fig. 1 is the presence of two distinct local regions A and B (marked by the circles in Fig. 1). Region A is characterized by an excellent  $TiO_2$  coating on  $SnO_2$  facets with angles of the order of 45°. The thickness of the Download English Version:

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