

# Crystallinity and photocatalytic activity of TiO<sub>2</sub> films deposited by reactive sputtering with radio frequency substrate bias

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## Abstract

TiO<sub>2</sub> films with thicknesses of 400–460 nm were deposited on the unheated non-alkali glass by radio frequency (rf) reactive magnetron sputtering using a Ti metal target. Depositions were carried out using a 3-in. 1000 G magnetron cathode with various rf substrate bias voltages ( $V_{sb}$ , dc component of self bias) of 10–80 V under total gas pressure of 1.0 or 3.0 Pa. The oxygen flow ratio [O<sub>2</sub>/(O<sub>2</sub>+Ar)] and rf sputtering power were kept constant at 60% and 200 W, respectively. Photocatalytic activity on photoinduced oxidative decomposition of acetaldehyde (CH<sub>3</sub>CHO) of the TiO<sub>2</sub> films showed a clear tendency to decrease with the increase in the  $V_{sb}$  during the deposition. Most of the films consisted of the mixture of anatase and rutile polycrystalline portions. It was confirmed that the rutile phase content increased and anatase phase content decreased markedly with increasing  $V_{sb}$ , where the crystallinity of anatase phase was much higher than that of rutile phase.

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**Keywords:** Titanium dioxide; Photocatalyst; Reactive sputtering; rf bias; Anatase; Rutile

## 1. Introduction

Titanium dioxide (TiO<sub>2</sub>) is well known as a photocatalyst because of the strong oxidizing power of photogenerated holes [1]. When TiO<sub>2</sub> is illuminated by UV light with higher energy than the TiO<sub>2</sub> band gap, an inter-band transition can be induced, resulting in the generation of electron–hole pairs. Such excited electrons or holes can diffuse to the TiO<sub>2</sub> surface and generate various radicals or ions which can decompose organic compounds adsorbed on the TiO<sub>2</sub> surface [2]. Its photoinduced self-cleaning properties in particular have attracted considerable attention in various application fields. In recent years, photoinduced hydrophilicity has been observed on rutile and anatase TiO<sub>2</sub> surfaces, where the water contact angle decreased to less than 1° after UV illumination in air [3]. Some studies on the photocatalytic activities of TiO<sub>2</sub> have been reported, however, most of them have been carried out on anatase films synthesized by a wet sol–gel method using titanium alkoxide [4]. Compared with the conventional wet processes, sputter deposition of TiO<sub>2</sub> should be one of the most

promising techniques for large-area uniform coatings with a high packing density and strong adhesion [5]. Furthermore, dry processes including the sputtering method should have high potential for film growth [6–8] where the crystallographic structure or orientation can be controlled precisely and the basic study on the definitive structural factors dominating the chemical or physical properties can be expected. Some reports have been published on the sputter deposition of TiO<sub>2</sub> films [9], however, most of them focused on the film structure and detailed research on the photocatalytic properties of the sputter-deposited TiO<sub>2</sub> films has not been reported.

We have reported that crystal structure and photocatalytic activity of TiO<sub>2</sub> films deposited by reactive sputtering using various sputtering inert gases [10] or applying various magnetic field strengths or shapes at the magnetron cathode [11]. It was confirmed that the content of rutile phase markedly decreased and anatase content increased with increasing atomic mass of the sputtering gas where the crystallinity of anatase phase was much higher than that of rutile phase [10]. Furthermore, crystallinity and photocatalytic activity of TiO<sub>2</sub> films showed a clear tendency to decrease with the decrease in the magnetic field strength during the deposition [11]. Decrease in plasma impedance by increasing the magnetic field strength

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was confirmed to be effective in the deposition of high-performance photocatalytic TiO<sub>2</sub> films.

In this study, polycrystalline TiO<sub>2</sub> films were deposited by radio frequency (rf) magnetron sputtering under various rf substrate bias voltage under various total gas pressures. Rutile/anatase composition, crystallinity, photoelectron emission and photocatalytic decomposition properties of the films were analyzed in relation to the sputter deposition processes in order to investigate on the structural factors dominating the photocatalytic activities of the TiO<sub>2</sub> films.

## 2. Experimental details

TiO<sub>2</sub> films with thicknesses of 400–460 nm were deposited on the unheated non-alkali glass (AN100, Asahi Glass) by rf reactive magnetron sputtering (L-332F, ANELVA) using a Ti metal target (99.99%, Furuuchi Kagaku). Depositions were carried out under various rf substrate bias ( $V_{sb}$ : dc component of self bias at substrate holder) of 10–80 V. A magnetron cathode with the balanced magnetic filed shape and vertical maximum magnetic field strength at the target surface of 1000 G was used [11]. The total gas pressure ( $P_{tot}$ ) during the deposition was 1.0 Pa or 3.0 Pa. The oxygen flow ratio [ $O_2/(O_2+Ar)$ ] and rf sputtering power were kept constant at 60% and 200 W, respectively, at which the target surface was expected to be fully oxidized, and the reactive sputtering mode was “oxide mode” (i.e., the  $O_2$  flow was over the “transition region”) [12,13]. Water partial pressure of the residual gas was monitored using a quadrupole mass spectrometer (Transpector XPR2, Inficon) in order to guarantee high reproducibility of the film structures for all of the depositions.

Film thickness was measured using a Dektak<sup>3</sup> surface profiler (VEECO/Solan Tech.). The crystal structure of the films was analyzed by X-ray diffraction (XRD) with 40 kV, 20 mA  $CuK\alpha_1$  radiation (XRD-6000, Shimadzu). The work function and shallow electron energy level of the film surface were analyzed by UV photoelectron spectroscopy in air (ESA, AC-2, Riken Keiki). The photocatalytic decomposition of acetaldehyde ( $CH_3CHO$ ) was evaluated by measuring its concentration decay in a  $3 \times 10^5$  mm<sup>3</sup> quartz cell in which a  $50 \times 50$  mm<sup>2</sup> TiO<sub>2</sub>-coated glass sample was contained. The concentration of  $CH_3CHO$  was measured using a gas chromatograph (GC-8A, Shimadzu) before and after the UV illumination (black light lamp, 0.4 mW/cm<sup>2</sup>) with the maximum intensity centered at 352 nm.

## 3. Results and discussion

Fig. 1 shows XRD patterns of the TiO<sub>2</sub> films deposited on the non-alkali glass without substrate heating, with various  $V_{sb}$  of 16–50 V under  $P_{tot}$  of 1.0 Pa. The TiO<sub>2</sub> films deposited using  $V_{sb}$  from 16 V to 32 V showed mixture phases of anatase and rutile. On the other hand, the TiO<sub>2</sub> films deposited with  $V_{sb}$  from 35 V to 40 V showed rutile single phase, however the rutile (110) XRD peak of these films was much broader with lower intensity than those of the anatase TiO<sub>2</sub> films deposited with lower  $V_{sb}$ . The films deposited with  $V_{sb}$  higher than 50 V

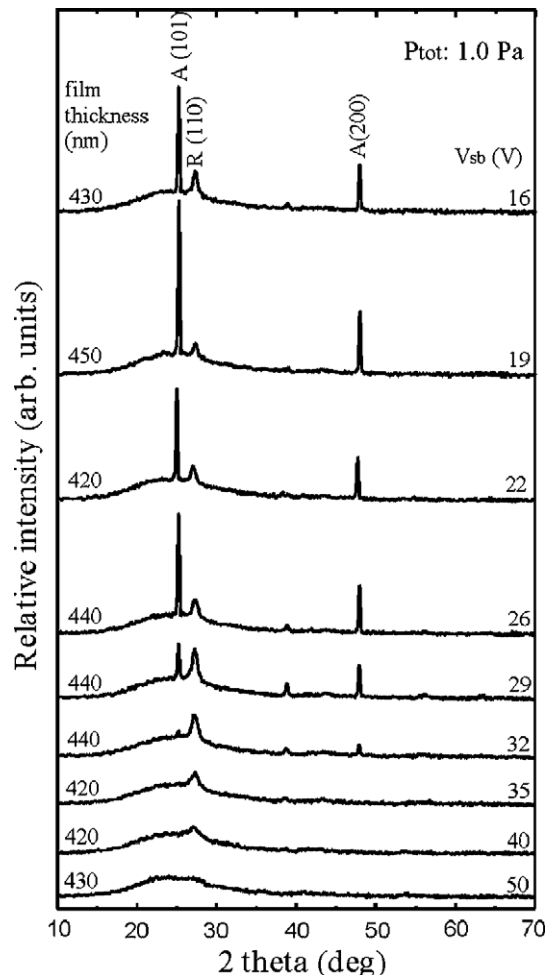


Fig. 1. XRD patterns of the TiO<sub>2</sub> films deposited on the nonalkali glass without substrate heating with various  $V_{sb}$  of 16–50 V under  $P_{tot}$  of 1.0 Pa.

showed amorphous structure. The TiO<sub>2</sub> films showed decrease in the anatase/rutile ratio and degradation of crystallinity with increasing  $V_{sb}$ . The TiO<sub>2</sub> film deposited with the small  $V_{sb}$  of 19 V showed higher crystallinity, compared with the other films. Such improvement in the crystallinity for the anatase polycrystalline films deposited using the small  $V_{sb}$  could be attributed to the  $Ar^+$  bombardment with a very low-energy, which could enhance surface migration at the growing film surface and hence improve the crystallinity of the films at RT.

Fig. 2 shows XRD patterns of the TiO<sub>2</sub> films deposited with various  $V_{sb}$  of 10–80 V under  $P_{tot}$  of 3.0 Pa. The XRD patterns showed that anatase/rutile ratio and crystallinity decreased markedly with increasing  $V_{sb}$ . The film deposited with  $V_{sb}$  of 10 V showed higher crystallinity compared with that of the other films, whereas the films deposited with  $V_{sb}$  higher than 50 V showed amorphous structure.

Fig. 3 shows the rough estimation of rutile content in the films calculated by the integrated intensity ratio of rutile (110)/{rutile (110)+anatase (101)} peaks for the TiO<sub>2</sub> films deposited with various  $V_{sb}$  under  $P_{tot}$  of 1.0 Pa and 3.0 Pa. The content of rutile markedly increased with increasing  $V_{sb}$ . Okimura et al. reported that the TiO<sub>2</sub> rutile films were successfully obtained on unheated substrates at  $P_{tot}$  of 0.27

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