



ELSEVIER

Contents lists available at ScienceDirect

Comptes Rendus Chimie

www.sciencedirect.com



Preliminary communication/Communication

# Synthesis of 2-(3-amino-2-oxoindolin-3-yl)-3-hydroxynaphthalene-1,4-dione derivatives *via* a one-pot, three-component reaction under catalyst-free conditions



Fengfeng Che, Yongsheng Wang, Tianhua Shen, Xiaoxia An, Qingbao Song\*

College of Chemical Engineering, Zhejiang University of Technology, Hangzhou 310014, PR China

## ARTICLE INFO

## Article history:

Received 3 September 2014

Accepted after revision 22 September 2014

Available online 25 March 2015

## Keywords:

2-(3-Amino-2-oxoindolin-3-yl)-3-hydroxynaphthalene-1,4-dione derivatives  
Ammonium acetate  
Catalyst-free  
2-Hydroxy-1,4-naphthoquinone  
One-pot

## ABSTRACT

A concise, facile and straightforward synthetic method has been described for the synthesis of 2-(3-amino-2-oxoindolin-3-yl)-3-hydroxynaphthalene-1,4-dione derivatives by a one-pot, three-component reaction of isatins, 2-hydroxy-1,4-naphthoquinone and ammonium acetate under catalyst-free conditions in ethanol. This protocol becomes highly efficient due to its mild reaction conditions, operational simplicity, and overall good to excellent yields (80–99%).

© 2014 Académie des sciences. Published by Elsevier Masson SAS. All rights reserved.

## 1. Introduction

Multicomponent reactions (MCRs) can be defined as special types of organic reactions in which three or more starting materials are employed in an atom-economic, convergent, efficient and timesaving manner to obtain a final product in a one-step procedure.

The 3-amino-2-oxindole moiety has drawn great attention of chemists and pharmacologists in the past few decades [1], owing to its potency and wide spectrum of biological activities, like anti-fungal [2], anti-microbial [3], anti-tubercular [4], and anti-tumor [5] functions. Besides, the 1,4-naphthoquinone unit also displays a fascinating array of biological applications, such as anti-cancer [6], anti-platelet [7], and radical scavenging [8] activities.

Therefore, developing a single molecular framework possessing biologically versatile oxindole and naphthoquinone units can be particularly promising as well as strongly desired, since they may create new medicinal properties or enhance biological activities [9,10].

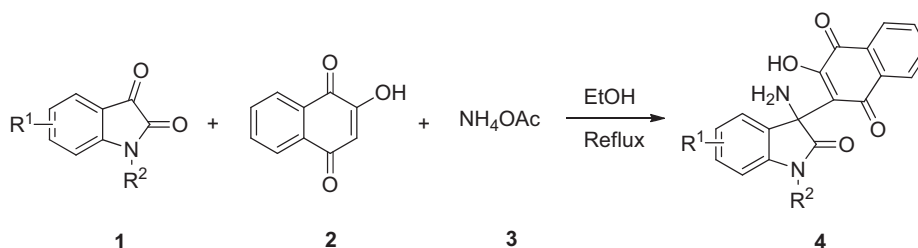
As far as we know, there is no report available in the literature describing the synthesis of 3-amino-2-oxindole derivatives with 1,4-naphthoquinone fragments [11–14]. All these facts have prompted us to design a novel and highly practical approach that allows convenient formation of a wide range of 2-(3-amino-2-oxoindolin-3-yl)-3-hydroxynaphthalene-1,4-dione derivatives incorporating the above-mentioned two moieties (Scheme 1).

## 2. Results and discussion

Initially, we employed isatin **1a**, 2-hydroxy-1,4-naphthoquinone **2** and ammonium acetate **3** as a model reaction to establish the feasibility of the strategy and optimize the reaction conditions. To our delight, when the

\* Corresponding author.

E-mail address: qbsong@zjut.edu.cn (Q. Song).



Scheme 1. Synthesis of 2-(3-amino-2-oxindolin-3-yl)-3-hydroxynaphthalene-1,4-dione derivatives.

reaction was attempted without a catalyst, it was found that the highest yield of product was obtained after 2 h (Table 1, entry 6). Using ammonium hydroxide instead of ammonium acetate with 20 mol% HOAc resulted in a lower yield (Table 1, entry 5). Then we investigated the amount of ammonium acetate; the results indicated that 1.5 equiv of  $\text{NH}_4\text{OAc}$  was the best choice for completing the reaction, since decreasing the amount of  $\text{NH}_4\text{OAc}$  led to a lower yield (90%), while the use of excessive  $\text{NH}_4\text{OAc}$  had no impact either on the product yield (99%) or on the rate of the reaction (Table 1, entries 6–8). Next, various solvents, such as MeOH, *i*-PrOH,  $\text{H}_2\text{O}$ , DCM,  $\text{CH}_3\text{CN}$  and THF were examined and EtOH emerged as the solvent of choice in terms of product yield (Table 1, entries 6,9–14).

Encouraged by the efficiency of the reaction protocol described above, the scope and specificity of this protocol were further investigated under the optimal reaction conditions (Table 1, entry 5). Thus, a broad range of structurally diverse isatins afforded the corresponding products in good to excellent yields; the results are depicted in Table 2. From these results, we could see that isatins **1** bearing substituents at the 4-position gained a lower yield due to the steric hindrance (Table 2, entries 2 and 3). Substrates having electron-withdrawing or

Table 1  
Optimization of reaction conditions for the synthesis of **4a**<sup>a</sup>.

Entry	Catalyst <sup>b</sup>	Solvent	Time (h)	Temp (°C)	Yield <sup>f</sup> (%)
1	TsOH·H <sub>2</sub> O	EtOH	2	78	80
2	HOAc	EtOH	2	78	93
3	HCl	EtOH	2	78	70
4	$\text{Cs}_2\text{CO}_3$	EtOH	2	78	78
5	HOAc	EtOH	2.5	78	82 <sup>c</sup>
6	–	EtOH	2	78	99
7	–	EtOH	2	78	90 <sup>d</sup>
8	–	EtOH	2	78	99 <sup>e</sup>
9	–	MeOH	1.5	64	95
10	–	<i>i</i> -PrOH	2	82	88
11	–	$\text{H}_2\text{O}$	1.5	90	70
12	–	DCM	2	39	60
13	–	$\text{CH}_3\text{CN}$	3	80	94
14	–	THF	4	65	83

<sup>a</sup> Reactions were performed with isatin **1a** (1 mmol), 2-hydroxy-1,4-naphthoquinone **2** (1 mmol) and ammonium acetate **3** (1.5 mmol) in the specified solvent (5 mL) under reflux conditions.

<sup>b</sup> Using 20 mol% catalyst.

<sup>c</sup> Ammonium acetate was replaced by ammonium hydroxide (1.5 equiv).

<sup>d</sup> Ammonium acetate (1.3 equiv) used.

<sup>e</sup> Ammonium acetate (1.7 equiv) used.

<sup>f</sup> Isolated yields.

Table 2  
Preparation of 2-(3-amino-2-oxindolin-3-yl)-3-hydroxynaphthalene-1,4-dione derivatives **4**<sup>a</sup>.

Entry	R <sup>1</sup>	R <sup>2</sup>	Product	Time (h)	Yield <sup>b</sup> (%)
1	H	H	<b>4a</b>	2	99
2	4-Br	H	<b>4b</b>	3	86
3	4,7-diCl	H	<b>4c</b>	4	83
4	5-F	H	<b>4d</b>	2	99
5	5-Cl	H	<b>4e</b>	2	97
6	5-Br	H	<b>4f</b>	2	93
7	5-I	H	<b>4g</b>	2	90
8	5-CH <sub>3</sub>	H	<b>4h</b>	2	96
9	5-OCH <sub>3</sub>	H	<b>4i</b>	2	96
10	5,7-diCH <sub>3</sub>	H	<b>4j</b>	4	83
11	6-Cl	H	<b>4k</b>	2	97
12	7-CF <sub>3</sub>	H	<b>4l</b>	2	81
13	H	Bn	<b>4m</b>	3	80

<sup>a</sup> Reactions were conducted with **1** (1 mmol), **2** (1 mmol) and **3** (1.5 mmol) in EtOH (5 mL) under reflux for the indicated time.

<sup>b</sup> Isolated yields.

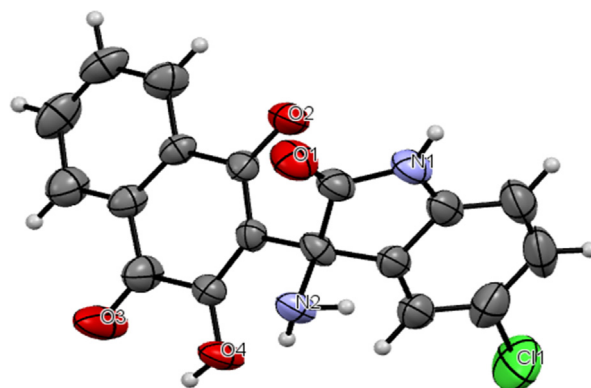


Fig. 1. (Color online.) ORTEP drawing of **4e**<sup>11</sup>.

electron-donating groups at the 5-position proceeded smoothly to give the desired compounds in excellent yields (Table 2, entries 4–9). A slight decrease in the yield was also observed when the 1-position (Table 2, entry 13) or 7-position was substituted (Table 2, entries 3, 10 and 12).

Finally, the structural elucidation was unequivocally determined by single-crystal X-ray analysis of **4e** (Fig. 1).

<sup>11</sup> CCDC-1003513 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

Download English Version:

<https://daneshyari.com/en/article/170585>

Download Persian Version:

<https://daneshyari.com/article/170585>

[Daneshyari.com](https://daneshyari.com)