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## Renewable and Sustainable Energy Reviews





# Post-combustion carbon dioxide capture: Evolution towards utilization of nanomaterials

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#### ARTICLE INFO

#### Article history: Received 12 April 2011 Received in revised form 21 January 2012 Accepted 29 January 2012 Available online 20 March 2012

Keywords: Carbon dioxide Adsorption Absorption Post-combustion capture Nanomaterials

#### ABSTRACT

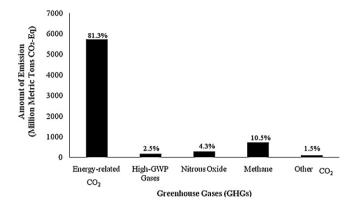
Carbon dioxide (CO<sub>2</sub>) is not the gas that gives the most severe global warming impact among the greenhouse gases (GHGs). However, its highest annual emission into the atmosphere makes it the most imperative anthropogenic GHG. This elevated emission is primarily coming from fossil fuel power plants. Hence, post-combustion CO<sub>2</sub> removal from power plants becomes crucial in global warming mitigation as it can be retrofitted directly into an existing plant. CO<sub>2</sub> removal technology nowadays is utilizing solvent-based sorbents, such as amine solutions and ionic liquids. Many extensive research works have been carrying out to improve the constraints of existing technology. In this paper, a general review on existing CO<sub>2</sub> removal technologies, existing research works on CO<sub>2</sub> removal sorbents was done. In conjunction with that, we will look into the potential and development of nanomaterials as CO<sub>2</sub> removal sorbents in the future. Nanomaterials have shown their potentials in CO<sub>2</sub> capture with its high surface area and adjustable properties and characteristics. Many limitations in existing technology were found improvable by nanomaterials.

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**Fig. 1.** GHGs emissions in United States by year 2008 [3]. High-GWP gases referred to high global warming potential gases, e.g. hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), and sulfur hexafluoride (SF<sub>6</sub>).

#### 1. Introduction

Carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>), nitrous oxide (N<sub>2</sub>O), hydrofluorocarbons (HFCs), perfluorocarbons (PFCs) and sulphur hexafluoride (SF<sub>6</sub>) have been listed in Kyoto Protocol 1998 as greenhouse gases (GHGs) [1]. CO<sub>2</sub> are the most important GHG because it emission is notably high compare to the others. CO<sub>2</sub> emission was recorded at 29.6 billion metric tons by year 2007 compared to 21.9 billion metric tons by year 1997 [2] and yet the figure is increasing from time to time. In United States, one of the highest CO<sub>2</sub> emission country, CO<sub>2</sub> emission is 81.3% of the total GHGs emitted [3]. By 2007, global CO<sub>2</sub> concentration hit 383 ppm which was 37% higher than pre-industrial period level [4]. Other GHGs concentrations are relatively low. For instance, concentration of CH<sub>4</sub> and N<sub>2</sub>O in the atmosphere were only 1774 part per billion (ppb) and 319 ppb, respectively, during 2005 [5]. Fig. 1 shows the comparison of the anthropogenic GHG emission in United States by year 2008. The data is reported in unit  $CO_2$ -equivalent ( $CO_2$ -eq), which is used to compare emission of different GHGs by counting their accumulated radiative forcing towards global warming effects over a given time period. Apparently, CO<sub>2</sub> gave highest impact to global warming among all the listed GHGs.

In nature, CO<sub>2</sub> will be absorbed by earth, either by weathering of rocks, photosynthesis of plants or ocean sinks by photosynthesis of marine plankton [6]. These natural sinks phenomena have balanced the natural source of CO<sub>2</sub> emission into the atmosphere over centuries. Excessive emission of CO2 since industrial era has made these natural removals became not sufficient anymore to maintain the CO<sub>2</sub> concentration in the atmosphere. CO<sub>2</sub> which was not being absorbed accumulated in the atmosphere and result in a drastic raise in CO<sub>2</sub> concentration. Every 3–5 gigatonnes of carbon will contribute to 1 ppm raise of CO<sub>2</sub> concentration in the atmosphere [7]. During the 1970s, CO<sub>2</sub> concentration in atmosphere increased by 1.3 ppm per year and this figure became 2.2 ppm per year by 2007 [4]. Fig. 2 shows the increasing trend of global CO2 emissions and its concentrations over years. Intergovernmental Panel of Climate Change (IPCC) predicted 2–3 °C of temperature increase from now [8] is dangerous, while Hansen et al. [9] argued that 1 °C rise of global temperature is the maximum tolerance for global warming to prevent the melt of the ice sheet and precious species extinction. To prevent global warming from endangering the world, climate models estimated that CO<sub>2</sub> concentration cannot exceed 450 ppm [9]. Scientist predicted the safe value for CO<sub>2</sub> concentration in atmosphere is 350 ppm while CO<sub>2</sub> concentration of the atmosphere is 383 ppm nowadays [4]. Hence, ideally, no more daily emission of CO<sub>2</sub> is allowed yet the live-long GHG in the atmosphere need to be removed.

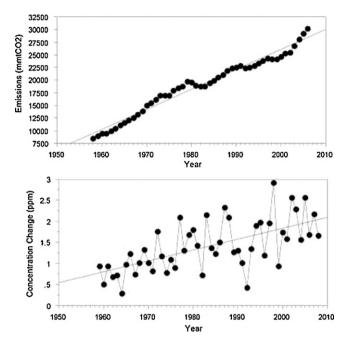
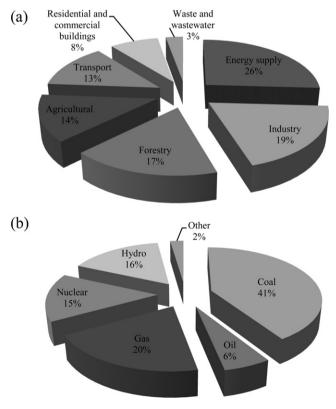


Fig. 2. (Top) Annual total global CO<sub>2</sub> emission; (bottom) annual change in CO<sub>2</sub> concentration [84].

#### 1.1. Main CO<sub>2</sub> sources

Energy supply sector contributed apparently higher emission (26%) compared to the other sectors in GHGs emission as shown in Fig. 3a. This sector is particularly referred to fossil fuel (including coal, natural gas and oil) power plants which are dominant in generating and supplying electricity (Fig. 3b). As second major  $\text{CO}_2$ 



**Fig. 3.** (a) Total anthropogenic GHG emissions from different sectors in 2004 (in terms of CO<sub>2</sub>-eq) [5]. (b) Total world electricity generation in 2006 [22].

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