



Modeling the temperature dependence of supercritical gas adsorption on activated carbons, coals and shales



Pongtorn Charoensuppanimit^a, Sayeed A. Mohammad^a, Robert L. Robinson Jr.^a, Khaled A.M. Gasem^{a,b,*}

^a Oklahoma State University, School of Chemical Engineering, Stillwater, OK 74078, United States

^b University of Wyoming, Department of Chemical & Petroleum Engineering, Laramie, WY 82071, United States

ARTICLE INFO

Article history:

Received 25 October 2014

Received in revised form 18 December 2014

Accepted 18 December 2014

Available online 26 December 2014

Keywords:

Temperature dependence

Gas adsorption isotherm

Local-density model

CO₂ sequestration

Shale gas

Coalbed methane

ABSTRACT

The recovery of natural gas from unconventional gas reservoirs such as shales and coal beds has increased significantly over the past few years. Since a majority of gas in these reservoirs is in an adsorbed state, knowledge of adsorption behavior of gases is essential for reliable gas-in-place estimates of these reservoirs. Further, some of these reservoirs also offer the potential for carbon dioxide sequestration. The existence of geothermal gradients in such reservoirs can affect the amount of gas. Since the temperatures in a reservoir are generally in the near-critical or supercritical region of the adsorbed gases, an adsorption model should be capable of reliable predictions of temperature dependence of adsorption in this region. To date, limited studies exist in the literature for modeling of temperature dependence of *supercritical* gas adsorption. In this work, we present a modification of the simplified local density (SLD) model to improve predictions of the temperature dependence of supercritical gas adsorption. Activated carbons were chosen as the reference adsorbents in the development of the modifications to the model since adsorption data on activated carbons are much more plentiful than data on coals and shales, for which data at multiple temperatures are extremely limited. Since activated carbons are simpler structural analogs of other carbonaceous adsorbents such as coals and shales, a model for activated carbons should serve as a basis for extension to the more complex coals and shales.

To undertake this study, a database was compiled that includes adsorption data at several temperatures for 11 gases on 18 activated carbons containing about 2600 data points. The database also included adsorption isotherms measured on six coals and six shales at multiple temperatures with a total of 670 data points.

The SLD model was modified by introducing a new temperature-dependence expression in the model. The modified SLD model was tested for its efficacy in describing the temperature dependence of near-critical and supercritical gas adsorption on the adsorbents contained in the above database. Results indicate that the modified SLD model is effective in modeling supercritical and near-critical gas adsorption. Further, the model was generalized to predict the temperature dependence of adsorption as a function of adsorbate and adsorbent properties. The generalized model was tested with an extended dataset and was found capable of predicting the temperature dependence with an average absolute deviation of 5%, in general.

The modified SLD model was tested with adsorption data on coals and shales at multiple temperatures. The model appears capable of describing the temperature dependence of adsorption on coals and shales with reasonable accuracy. We note, however, that the development of a completely generalized model requires additional experimental data on coals and shales and further model testing.

© 2014 Elsevier B.V. All rights reserved.

1. Introduction

Shale and coalbed methane reservoirs have become an important source of natural gas, and a significant portion of the gas in these reservoirs exists in an adsorbed state. Thus, knowledge of gas adsorption behavior over a range of pressures and temperatures is required to

estimate the gas-in-place for these reservoirs. Further, some of these reservoirs can offer potential sites for CO₂ sequestration. Gases commonly encountered in these reservoirs are the natural gas components including carbon dioxide and nitrogen with the latter two being especially important in enhanced gas recovery and carbon dioxide sequestration. Since gas adsorption is temperature dependent, the presence of geothermal gradients in a reservoir affects the adsorption capacity of these gases. For example, in the Black Warrior basin the coal bed reservoir temperature varies from about 300 K to 325 K within the 0.3–1.8 km depth-range (Pashin and McIntyre, 2003). Thus, an accurate accounting for temperature dependence of gas adsorption is important

* Corresponding author at: Oklahoma State University, School of Chemical Engineering, Stillwater, OK 74078, United States.

E-mail address: gaseam@uwyo.edu (K.A.M. Gasem).

Table 1
Database for pure-gas adsorption on activated carbons, coals and shales.

Adsorbent	Adsorbate	Pressure range (MPa)	Temperature range (K)	BET surface area (m ² /g)	NDP*	Reference
<i>Dataset for activated carbons</i>						
PCB	CH ₄	0.1–3.7	296–480	1150	65	Ritter and Yang (1987)
	CO ₂	0.2–5.8				
	CO	0.1–1.3				
Norit-RB1	H ₂ S	0.3–6.7	294–350	1100	128	Van Der Vaart et al. (2000)
	CH ₄	0.1–0.8				
Nuxit-AI	CH ₄	0.01–0.6	293–363	1200	447	Szepesy (1963)
	C ₂ H ₂	0.001–0.1				
	C ₂ H ₆	0.01–0.7				
	C ₂ H ₄	0.01–0.6				
	C ₃ H ₈	0.01–0.7				
	C ₃ H ₆	0.01–0.8				
	nC ₄ H ₁₀	0.001–0.1				
Columbia Grade L	CH ₄	0.02–1.5	311–478	1152	272	Ray and Box (1950)
	N ₂	0.03–1.5				
	C ₂ H ₂	0.01–0.1				
	C ₂ H ₆	0.01–1.5				
	C ₂ H ₄	0.01–1.5				
	C ₃ H ₈	0.01–0.7				
	C ₃ H ₆	0.01–0.1				
BPL	CH ₄	0.01–3.8	213–301	988	233	Reich et al. (1980)
	CO ₂	0.003–3.84				
	C ₂ H ₆	0.001–1.71				
	C ₂ H ₄	0.001–1.70				
F30/470	CH ₄	0.44–6.0	303–383	993	154	Berlier and Frère (1997)
	N ₂	0.39–6.0				
F30/470	CO ₂	0.05–2.5	288–328	993	113	Frère and De Weireld (2002)
LAC	C ₃ H ₈	0.001–0.1	273–343	1011	102	Mofarahi et al. (2003)
	C ₃ H ₆					
Carbotech	CH ₄	0.006–3.4	300–318	885	32	Martin et al. (2011)
KT	CH ₄	0.6–3.4	300–318	668	31	Martin et al. (2011)
Maxsorb II	CH ₄	0.1–1.2	281–343	2768	126	Wang et al. (2010)
Templated carbon	CH ₄	0.05–4.6	263–303	1500	71	Chen et al. (2011)
	C ₂ H ₆	0.02–2.0				
Maxsorb III	CH ₄	0.06–2.4	263–303	3140	128	Loh et al. (2010)
Columbia Grade G	CH ₄	0.1–13.8	283–343	1157	141	Payne et al. (1968)
	nC ₄ H ₁₀	0.003–0.8				
	C ₃ H ₈	0.001–1.4				
JX101	C ₃ H ₈	0.001–1.4	283–313	1500	107	Qin Wu et al. (2005)
	CH ₄	0.01–1.0				
<i>External dataset for activated carbons used for validation</i>						
Coconut shell-derived carbon	CH ₄	0.1–9.3	235–333	3106	122	Zhou et al. (2000)
	N ₂	0.5–9.0				
F-400	CO ₂	0.14–9.4	303–318	850	116	Humayun and Tomasko (2000)
CNS-201	CH ₄	0.01–6.8	243–333	2000	62	Dundar et al. (2012)
CO-64	N ₂	0.01–6.6	153–298	1150	135	Dundar et al. (2012)
<i>Dataset for coals</i>						
Pocahontas	CH ₄	1.4–13.8	308–328	N/A*	20	Sakurovs et al. (2008)
	N ₂		298–328	20		
	CO ₂		308–328	20		
Illinois-6	CH ₄	14–13.8	308–328	N/A	20	Sakurovs et al. (2008)
	N ₂		298–328	20		
	CO ₂		308–328	20		
Beulah Zap	CH ₄	14–13.8	308–328	N/A	20	Sakurovs et al. (2008)
	N ₂		298–328	20		
	CO ₂		308–328	20		
Hulun Buir	CH ₄	2.5–19.1	308–328	N/A	41	Li et al. (2010)
	CO ₂	3.8–24.2			80	
Pingdingshan	CH ₄	3.0–19.7	308–328	N/A	43	Li et al. (2010)
	CO ₂	5.2–24.6			71	
Jingcheng	CH ₄	1.2–19.0	308–328	N/A	48	
	CO ₂	3.6–24.6			67	
<i>Dataset for shales</i>						
WIC7145	CH ₄	0.07–12.5	318–358	6.7	26	Rexer et al. (2014)
WIC7155		0.05–13.0		4.3	28	Rexer et al. (2014)
HAR7038		0.02–12.0		N/A	25	
HAR7060		0.06–13.5		N/A	14	
HAD7090		0.05–12.8		25.1	23	
HAD7119		0.07–13.8		21.0	29	
Total					3321	

* NDP is number of data points and N/A denotes not available.

Download English Version:

<https://daneshyari.com/en/article/1752983>

Download Persian Version:

<https://daneshyari.com/article/1752983>

[Daneshyari.com](https://daneshyari.com)