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## One/two-photon cationic polymerization in visible and near infrared ranges using two-branched sulfonium salts as efficient photoacid generators



PIGMENTS

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## ABSTRACT

This paper describes the synthesis and photochemical behavior of two two-branched sulfonium-based photoacid generators (PAGs). The branched structure influences the optical properties of the PAGs by increasing absorption wavelengths and molar extinction coefficients as well as enhancing acid production under one/two-photon irradiation. The two-branched structures collectively showed twofold higher H<sup>+</sup> production abilities and improved two-photon absorption cross-sections compared with those in the corresponding mono-branched PAGs. Photochemical mechanisms were investigated through UV–Vis spectra analysis, molecular orbital calculations, and fluorescence analysis. These novel two-branched Dra-A type sulfonium-based photoinitiating systems are efficient cationic photoinitiators (epoxide conversion = 70%–85% and vinyl conversion > 75%) even under conditions with low initiator concentration (1 wt%) and curing light intensities (5 mW cm<sup>-2</sup>) upon exposure to visible light-emitting diodes (e.g., 405 nm). These two-photon fabricated microstructures also exhibit potential as cationic two-photon initiators.

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### 1. Introduction

Two-photon absorption (2 PA) is a nonlinear optical process that involves simultaneous absorption of two photons of identical or different frequencies and leads to excitation of a molecule from the ground state to the excited state [1]. 2 PA-related processes, such as nonlinear transmission and quadratically intensity-dependent properties, are utilized in optoelectronic devices, including 3D optical data storage systems [2]. The process can also be adopted in triggering subsequent events, such as fluorescence or chemical reactions [3,4] in biological imaging, photodynamic therapy [5], and 3D photolithography [6–9]. In the last two decades, 2 PA crosssections ( $\delta$ ) for organic 2 PA chromophores have been considerably improved. To maximize the  $\delta$  of chromophores, researchers have developed various strategies, including extension of conjugation length and dimensionality [10] as well as introduction of electron donating/accepting groups to create "push-pull" systems [11]. For example, symmetrically substituted  $D-\pi-D$  and  $D-\pi-A-\pi-D$  chromophores are effective designs for increasing  $\delta$  [12].

Photoacid generators (PAGs) constitute an important class of cationic photoinitiators and can be strategically used in the field of photoresists [13–15]. Applications of PAGs have been extensively developed in various research domains, such as microelectronics, photosensitized resins, and 3D microfabrication [16-21]. Considering the advantages of PAGs, scholars have become increasingly interested in designing new PAGs that can be excited with longwavelength light by two-photon modes [18,22–24]. Perry et al. [25] fabricated a sulfonium-based PAG by introducing a bis [(diarylamino)styryl] benzene derivative; the products exhibited high quantum yield for acid generation ( $\Phi_{\rm H}^+$  ~ 0.5) and a large 2 PA crosssection ( $\delta$ ) ( $\delta_{max} = 690$  GM). In this substance, the sulfonium salt groups were symmetrically associated with one benzene ring of the arylamino moieties. Belfield et al. [18] developed a 1,8-bis(4'-styryl) fluorene with two para-substituted diphenyl-sulfonium groups at both extremities of the PAGs. In our previous studies, we investigated two materials, namely, 4-ethoxystilbene [3] or 4-N,Ndiphenylaminostilbene [24,26,27], whose strong electron-donating group and elongated D- $\pi$ -A conjugation red shifted the absorbance of PAGs and enlarged the 2 PA cross-section to ca. 600 GM.

Although several new photoinitiators have been developed for



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the photopolymerization under various LEDs [28–33], developing novel PAGs sensitive to two-photon excitation is an essential step toward promising applications. The multibranch strategy is a relevant tool used to modulate linear and nonlinear absorption properties without drastic effects on the energy position of electronic transitions [34–36]. The para-to-meta substitution strategy is also an effective technique used to adjust the absorption wavelength and influence photolysis efficiency in sulfonium salt PAGs [24].

In this study, we designed and prepared four two-branched D- $\pi$ -A typed sulfonium-based PAGs (Bi-Para and Bi-Meta; Scheme 1). These PAGs maintain an excellent transparency in a large portion of the visible and near-infrared regions and are thus suitable for one/two-photon polymerization. The fabricated PAGs could be a potential cationic one/two-photon photoinitiator, given their outstanding efficiency or acid generation ability.

#### 2. Experimental section

#### 2.1. Materials

All the chemicals used for synthesis were purchased from Sinopharm Chemical Reagent Co., Ltd., TCI, Alfa Aesar, or J&K Chemical. The reagents were utilized without further purification unless otherwise specified. All the solvents employed for photophysical measurements were Fluka spectroscopic grade. The monomer cyclohexene oxide (CHO, 98%) and triethyleneglycol divinyl ether (DVE-3, 98%) were purchased from Sigma-Aldrich from the highest purity available and used as received without further purification.

#### 2.2. General instrumentations

Proton and carbon nuclear magnetic resonance spectra (<sup>1</sup>H, <sup>13</sup>C NMR) were recorded on a Bruker Avance 400 (400 MHz) spectrometer. Chemical shifts were reported in parts per million (ppm) downfield from the Me<sub>4</sub>Si resonance, which was used as the internal standard when recording NMR spectra. Mass spectra were obtained on a Micromass GCTTM. The molecular weights of the photopolymers were determined using a GPC instrument equipped with MZ-Gel SD plus columns (HR series 3, 4, and 5E), with THF as eluent at a flow rate of 1 mL min<sup>-1</sup> and a Waters 410 differential refractometer detector. The melting point of molecules were determined using Melting Point Apparatus (Nikon-Ti, Co. Ltd. Japan). IR data of molecules were measured with a Thermo-Nicolet iS5 instrument infrared spectrometer.

UV–Vis spectra were recorded on a Mapada UV-6300 spectrophotometer. The photodecomposition process in a solution of acetonitrile and CHO were monitored by UV–Vis spectra under a light-emitting diode (LED) point curing (Uvata, Shanghai). irradiation of 405 nm (0.4 mW cm<sup>-2</sup>). Steady-state fluorescence spectra are collected from a Hitachi-2700 spectrofluorometer. Emission spectra are spectrally corrected, and fluorescence quantum yields include the correction due to solvent refractive index are determined relative to quinine bisulfate in 0.05 M sulfuric acid ( $\Phi_{\rm fl} = 0.52$ ) [3].

Quantum yields for acid generation were measured under irradiation at 405 nm LED. All irradiated PAGs dissolved in acetonitrile were previously  $N_2$  degassed. The progress of the photoreaction was monitored via UV–Vis absorption spectroscopy. The absorbance at the excitation wavelength must be higher than 2.5 before assuming a total absorption of the incident photons. The dose rates were maintained sufficiently small such that the changes in absorbance were lower than 10%. Rhodamine B (RhB) was used as



Scheme 1. Molecular structures of the precursor and PAGs.

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