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Evaluation of three methods for enriching H₂-producing cultures from anaerobic sludge

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Abstract

Hydrogen can be harvested from the microbial fermentation of organic substrates when methanogenesis is suppressed in an anaerobic digestion system. In this study three methods, heat-, acid- and alkaline-treatment, were used to suppress methanogenesis in mixed cultures and to enrich H_2 -producing inoculum. Highest H_2 yield of 2.00 mol- H_2 /mol-glucose was achieved with the heat-treated sludge, while lowest yield of 0.48 mol- H_2 /mol-glucose was obtained with the alkaline-treated sludge. A butyrate-type fermentation was found for both heat- and alkaline-treated sludge, while a mixed-type fermentation occurred for the acid-treated sludge. A model was established to describe the kinetics of H_2 production process and the yield coefficients of various products were estimated for the three cases with this model. The relationships among NADH/NAD⁺, oxidation–reduction potential and the H_2 partial pressure were established and the evolvement of NADH/NAD⁺ and oxidation–reduction potential in the fermentative process for the three cases was also evaluated. The comparative experimental results show that the heat-treatment method was better than the two others for enriching H_2 -producing inoculums from mixed anaerobic cultures. © 2006 Elsevier Inc. All rights reserved.

Keywords: Acid-treated; Alkaline-treated; Anaerobic sludge; Heat-treated; Hydrogen; NADH/NAD+

1. Introduction

Research on alternative energy sources has gained renewed interest, due to global awareness of accumulated carbon dioxide in the atmosphere as a potential cause of climate change [1]. Combustion of H₂ produces no greenhouse gases, and has a high-energy yield of 122 kJ/g, which is 2.75-fold greater than that of hydrocarbon fuels. Thus, utilization of H₂ as a clean energy source seems to be promising. The feasibility of applying acidogenesis of organic wastes to produce H₂ has been widely demonstrated at various laboratories [2–4]. Compared with photosynthetic bacteria, dark fermentative bacteria produce H₂ at a lower cost, because they do not need light provision and have simple requirements for microbial growth. In previous studies both pure cultures such as *Clostridium* sp. and mixed cultures of anaerobic bacteria, have been used to convert carbohydrates (e.g., glucose) to H₂ [5–8].

To scale-up biological H_2 production processes, getting a large amount of anaerobic H_2 -producing inoculums economi-

0141-0229/\$ - see front matter © 2006 Elsevier Inc. All rights reserved. doi:10.1016/j.enzmictec.2006.07.033 cally becomes crucial [7,9]. Previous studies have demonstrated that operation at a low pH and short sludge retention time, or seeding with heat-shocked sludge, is able to suppress the growth of methanogens and accordingly to enhance H₂ production. The acid-treated method and heated-shocked method have been, respectively, used in previous studies [7,8,10]. The effectiveness of pH control for enhancing H2 production have been demonstrated in batch tests [10]. The heat-treatment has been found to be an alternative for enriching H₂-producing inoculum from natural anaerobic cultures [7]. The heat- and acid-treated methods have been also compared for hydrogen production and a greater H₂ yield was observed in the heat treatment than that in the acid treatment [11]. Recently, it was found that H₂ production could also be enhanced from sewage sludge with alkaline pretreatment [12]. However, little information is available to evaluate the effectiveness of these three enrichment methods for H₂-producing cultures in parallel. Moreover, the evolvement of NADH/NAD⁺ and oxidation-reduction potential (ORP) with the H₂ partial pressure in the fermentation process has not been elucidated in these previous studies.

Therefore, in this study H₂-producing inoculum was enriched by using heat-, acid- and alkaline-treatment in parallel, and their effectiveness was evaluated and compared in a fermentor oper-

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ated in a batch mode. The profiles of H_2 production, substrate consumption and product formation were monitored for the three cases, and the kinetics of H_2 production was also explored. Moreover, the relationships between NADH/NAD⁺, ORP and the H_2 partial pressure, and the evolvement of NADH/NAD⁺ and ORP in the fermentation process for the three cases were investigated.

2. Materials and methods

2.1. Seed sludge

The anaerobic microflora used in this study was obtained from a full-scale upflow anaerobic sludge blanket reactor located in Benpu, China. This reactor is being operated for the treatment of soybean-processing wastewater and production of CH₄. Prior to use, the seed sludge was first washed five times with tap water, and was then sieved to remove stone, sand and other coarse matters. Thereafter, heat-, acid- and alkaline-treatments were, respectively, performed to inactive hydrogentrophic methanogens and to enrich H₂-producing bacteria. For the heat-treatment case the seed sludge was heated at 102 °C for 90 min [13]. For the acid-treatment, the sludge was adjusted to acidic pH (pH 3.0–4.0) with 0.1N HCl for 24 h, and was then adjusted back to pH 7.0 with the addition of 0.1N NaOH [10]. For the alkaline-treatment, the sludge was pretreated with adding 4 M NaOH and the pH was kept at 12.0 for 24 h, and was then adjusted back to pH 7.0 with the addition of 0.1N HCl [12]. The temperature was around 25 °C for both acid- and alkaline-treatments.

2.2. Experiments

Hydrogen production experiments were conducted in triplicate in a 5-L fermentor (Baoxin Biotech Ltd., Shanghai) equipped with an impeller and four baffles. Agitation of the fermentation broth was provided by using a six-bladed impeller. One litre of heat-, acid- or alkaline-treated anaerobic sludge with volatile suspended solids (VSS) of 18.0 g/L and 3 mL of nutrients solution were added to the fermentor. The working volume was adjusted to 3.0 L with distilled water. The solution in the fermentor was composed as follows (unit in mg/L): sucrose 25,000; NH₄HCO₃ 2025; K₂HPO₄·3H₂O 800; CaCl₂ 50; MgCl₂·6H₂O 100; FeCl₂ 25; NaCl 10; CoCl₂·6H₂O 5; MnCl₂·4H₂O 5; AlCl₃ 2.5; (NH₄)₆Mo₇O₂₄ 15; H₃BO₄ 5; NiCl₂·6H₂O 5; CuCl₂·5H₂O 5; ZnCl₂ 5. Prior to operation, the fermentor was purged with nitrogen gas for 10 min to ensure anaerobic conditions.

The pH of the mixed liquor was kept at constant by feeding 4 M NaOH or 2 M HCl solutions via respective peristaltic pumps. The agitation rate in the fermentor was kept at 120 rpm. A biogas sampling port was installed between the meter and the reactor to allow direct biogas sampling with a syringe. In each trail, the pH of the mixed liquor was chosen as 5.5, which was reported to be optimum for H₂ production by mixed anaerobic cultures [14,15]. The temperature was kept at 35.0 °C with a temperature controller, and the initial sucrose concentration was 25 g/L.

2.3. Analytical methods

The amount of biogas produced was recorded using water-replace method. The biogas composition was analyzed by using a gas chromatograph (Lunan, Model SP-6800A) equipped with a thermal conductivity detector and a 1.5 m stainless-steel column packed with 5 Å molecular sieve. The temperatures of injector, detector and column were kept at 100 °C, 105 °C and 60 °C, respectively. Argon was used as carrier gas at a flow rate of 30 mL/min. The concentrations of ethanol and volatile fatty acids (VFA), including acetate, propionate, butyrate, *i*-butyrate, valerate, and caporate in the effluent were determined by second gas chromatograph (Agilent, Model 6890NT) equipped with a flame ionization detector and a 30 m × 0.25 mm × 0.25 µm fused-silica capillary column (DB-FFAP). The liquor samples were first centrifuged at 12,000 rpm for 5 min, and were then acidified by formic acid and filtrated through 0.2 µm membrane and finally measured for free acids. The temperatures of the injector and detector

were 250 °C and 300 °C, respectively. The initial temperature of oven was 70 °C for 3 min followed with a ramp of 20 °C/min for 5.5 min and to final temperature of 180 °C for 3 min. Nitrogen was used as carrier gas with a flow rate of 2.6 mL/min. The sucrose concentration was measured using anthrone-sulfuric acid method [16]. VSS concentration was measured according to Standard Methods [17].

A following modified Logistic equation was employed to model the kinetics of biohydrogen production [19]:

$$H = \frac{H_{\text{max}}}{1 + \exp[((R_{\text{max}, \text{H}_2} \times e)/H_{\text{max}})(\lambda - t) + 1]}$$
(1)

where H (mL) is the total amount hydrogen produced at reaction time t (h), H_{max} (mL) the potential maximal amount hydrogen produced, R_{max} (mL/h) the maximum hydrogen production rate and λ (h) is the lag time to exponential hydrogen production.

Once cumulative hydrogen production curves were obtained over the course of an entire batch experiment, a curve was drawn using the modified Logistic equation and the values of H_{max} , R_{max} , λ , H_{max} and R_{max} were also determined. The data were analyzed by using software Origin 6.1 in the study.

3. Results and discussion

3.1. Substrate degradation

The degradation profiles of sucrose with acid-, heat- and alkaline-treated sludge are illustrated in Fig. 1. After a lag time, sucrose was degraded rapidly and became nearly depleted within 45 h in the three cases (Fig. 1).

3.2. H_2 production

The H₂ partial pressure profiles in the reactor headspace were in accord with the conversion of sucrose to H₂ for the three cases and no methane was detected (Fig. 2). This might be due to the fact that, in the enrichment, the H₂-utilizing methanogens were inactivated or inhibited, but that the H₂-producing microorganisms, e.g., clostridia, survived. These endospores are resistant to heat, and cannot be inactivated easily even by harsh chemicals [18]. After the end of lag phase, the H₂ partial pressure increased and peaked of 0.61 atm for the heat-treated sludge at hour 26, 0.59 atm for the acid-treated one at hour 23.3, and 0.51 atm for the alkaline-treated one at hour 27. Then, the H₂ partial pressure declined for the three cases. Such a decrease was



Fig. 1. Sucrose degradation profiles for the heat-, acid- and alkaline-treated sludge.

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