



# Theoretical and comparative investigations of $\text{Yb}^{3+}$ ion in $\text{MWO}_4$ and $\text{M}'\text{MoO}_4$ scheelites crystals ( $\text{M}=\text{Sr, Pb, Ca, Ba}$ ) and ( $\text{M}'=\text{Sr, Pb, Ca, Cd}$ )

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## ABSTRACT

The crystal-field model is applied to a series of scheelites crystals ( $\text{CaWO}_4$ ,  $\text{SrWO}_4$ ,  $\text{PbWO}_4$ ,  $\text{BaWO}_4$ ,  $\text{CdMoO}_4$ ,  $\text{CaMoO}_4$ ,  $\text{SrMoO}_4$  and  $\text{PbMoO}_4$ ) doped with the  $\text{Yb}^{3+}$  ion. The calculated crystal-field parameters present a general trend of variation with  $\text{M}^{2+}$  ionic radius of the host cation. The maximum splitting  $\Delta E$  of the  $^2\text{F}_{7/2}$  manifold of the  $\text{Yb}^{3+}$  ion is then obtained as a function of  $N_V$  crystal-field strength parameters. The agreement between experimental results and theoretical predictions for all investigated systems is very satisfactory. The crystal-field effects are very important for the prediction of emission energies of the  $\text{Yb}^{3+}$  ion in different scheelites.

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## 1. Introduction

Over the last decade there has been increasing interest in the research of  $\text{Yb}^{3+}$  activated laser crystals [1]. The  $\text{Yb}^{3+}$  ion in a laser host crystal is the simplest example of electronic levels in an impurity-doped solid, since the  $\text{Yb}^{3+}$  ion possesses only two electronic states: the  $^2\text{F}_{7/2}$  ground state and the  $^2\text{F}_{5/2}$  excited state. The two energy levels are split under the crystal-field, which means that all Yb lasers are quasi-three-level systems. The Yb-doped crystals display some advantages over Nd-doped crystals: the  $\text{Yb}^{3+}$  doped media generally have longer radiative lifetimes and higher quantum efficiency; the  $\text{Yb}^{3+}$  ions in the host materials possess no up-conversion effect or excited state absorption in principle, which will greatly reduce thermal effects in the laser crystal; the absorption spectrum shows prominent bands in the range 900–980 nm; the emission spectra of Yb doping materials are broader than those of Nd doping materials, so Yb doping materials may be used to produce pulsed laser or mode-lock lasers. Many kinds of Yb-doped crystals have been reported for efficient Yb laser outputs [2–13]. Scheelites doped with the ytterbium ion have been investigated extensively through paramagnetic resonance, spectroscopy and other techniques, resulting in a large body of data on these systems.

Until now, the most attention has been paid to the scintillation properties of molybdates with a scheelite-type structure:  $\text{M}'\text{MoO}_4$  ( $\text{M}'=\text{Ca, Cd, Sr, Pb}$ ); however, all of the crystals studied have a number of drawbacks, i.e., low luminescence intensity, even at low temperatures ( $\text{BaMoO}_4$ ); large  $Z$  ( $\text{PbMoO}_4$ ); or the presence of

radioactive cation isotopes ( $\text{CdMoO}_4$ ,  $\text{PbMoO}_4$ ). Currently,  $\text{CaMoO}_4$  crystals are considered as the most appropriate.  $\text{MWO}_4$  ( $\text{M}=\text{Sr, Pb, Ba and Ca}$ ) with scheelite structure has attracted great interest due to its useful properties, such as electro-optics [14,15], high-pressure phase transformation [16,17] and unique spontaneous Raman behaviors [18–20], which are proven to be important for the stimulated Raman scattering (SRS) technique in Raman laser development [21].

In this work, we are interested in ytterbium doped  $\text{CaWO}_4$ ,  $\text{PbWO}_4$ ,  $\text{SrWO}_4$ ,  $\text{BaWO}_4$ ,  $\text{CaMoO}_4$ ,  $\text{PbMoO}_4$ ,  $\text{SrMoO}_4$  and  $\text{CdMoO}_4$  and we estimate a general trend for ytterbium ions in  $\text{MWO}_4$  and  $\text{M}'\text{MoO}_4$  scheelites.

## 2. Calculation

Scheelite-type  $\text{MWO}_4$  (or  $\text{M}'\text{MoO}_4$ ) has a tetragonal crystal structure of space group  $\text{I}4_1/\text{a}$  ( $\text{C}_{4h}^6$ ). The  $\text{M}^{2+}$  (or  $\text{M}'^{2+}$ ) site has eight nearest-neighbor oxygen ions, which can be grouped into two sets of distorted interpenetrating tetrahedra [18,22–24].

The  $\text{Yb}^{3+}$  ion in  $\text{MWO}_4$  or  $\text{M}'\text{MoO}_4$  crystal occupies the  $\text{M}^{2+}$  site. Strictly speaking, the  $\text{M}^{2+}$  site has  $\text{S}_4$  point symmetry. Since this point symmetry is very nearly  $\text{D}_{2d}$ , some authors [25,26] applied  $\text{D}_{2d}$  symmetry as a good and realistic approximation to study the optical spectra for rare-earth ions in  $\text{MWO}_4$  or  $\text{M}'\text{MoO}_4$  crystals. For simplicity, we apply  $\text{D}_{2d}$  symmetry here. In  $\text{MWO}_4\cdot\text{Yb}^{3+}$  or  $\text{M}'\text{MoO}_4\cdot\text{Yb}^{3+}$ , the energy levels and wave functions of the above-mentioned Kramer's doublets can be calculated by diagonalizing a  $14 \times 14$  energy matrix of the  $4\text{f}^{13}$  ion in tetragonal ( $\text{D}_{2d}$ ) symmetry related to the Hamiltonian

$$H = H_{\text{CF}}(\text{D}_{2d}) + H_{\text{SO}} \quad (1)$$

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where  $H_{SO}$  is the spin–orbit coupling term and the crystal-field term is

$$\hat{H}_{CF}(D_{2d}) = B_0^2 \hat{C}_0^2 + B_0^4 \hat{C}_0^4 + B_4^4 (\hat{C}_{-4}^4 + \hat{C}_4^4) + B_0^6 \hat{C}_0^6 + B_4^6 (\hat{C}_{-4}^6 + \hat{C}_4^6) \quad (2)$$

The empirical crystal-field parameters (CFPs) are obtained by a least-square fitting program that minimizes the root mean square (rms) deviation between calculated and experimental energy levels derived from optical spectra. This rms is given by [27]

$$\sigma = \sqrt{\sum_i \frac{(E_i^{cal} - E_i^{exp})^2}{N-P}} \quad (3)$$

where  $N$  denotes the number of levels and  $P$  is the number of independent free-parameters.

As useful tools to measure the CF interaction strength, mainly in order to establish quantitative comparisons between  $\text{CaWO}_4$ ,  $\text{PbWO}_4$ ,  $\text{SrWO}_4$ ,  $\text{BaWO}_4$ ,  $\text{CaMoO}_4$ ,  $\text{PbMoO}_4$ ,  $\text{SrMoO}_4$  and  $\text{CdMoO}_4$ , the crystal-field strength parameters have also been calculated with the formalism defined by Ref. [28]

$$N_V = \left[ \sum_{k,q} \frac{4\pi}{2k+1} (B_q^k)^2 \right]^{1/2} \quad (4)$$

### 3. Results and discussion

#### 3.1. $\text{Yb}^{3+}$ doped in $\text{MWO}_4$ and $\text{M}'\text{MoO}_4$ crystals

Trivalent ytterbium has the simplest energy level structure with only two manifolds, the  $^2F_{7/2}$  ground state and the  $^2F_{5/2}$  excited state, which are separated by approximately  $10,000 \text{ cm}^{-1}$ .

The starting sets of CFPs were taken from Ref. [29] corresponding to the  $\text{Yb}^{3+}$  ion in  $\text{MWO}_4$  ( $M=\text{Sr, Pb, Ba}$  and  $\text{Ca}$ ) and  $\text{M}'\text{MoO}_4$  ( $M'=\text{Sr, Pb, Cd}$  and  $\text{Ca}$ ) crystals. The fitting procedure was described in our previous work ( $\text{Yb}^{3+}$  doped YAB) [30]. In this case, the number of energy levels for  $\text{Yb}^{3+}$  in different scheelites included in the fits was 7. Table 1 shows the final set of CFPs of ytterbium in  $\text{CaWO}_4$ ,  $\text{PbWO}_4$ ,  $\text{SrWO}_4$  and  $\text{BaWO}_4$  obtained for the best fit, which was (with quite good rms) about 3.5, 7.9, 9.5 and  $17 \text{ cm}^{-1}$ , respectively. All obtained parameters are meaningful then compared to the CFPs of  $\text{Yb}^{3+}$  in  $\text{CaWO}_4$  occupying the  $D_{2d}$  symmetry site [31]. Table 2 shows the final set of CFPs of ytterbium in  $\text{CaMoO}_4$ ,  $\text{PbMoO}_4$ ,  $\text{SrMoO}_4$  and  $\text{CdMoO}_4$  obtained for the best fit, which was (with quite good rms) about 7.5, 10.8, 7.3 and  $3.7 \text{ cm}^{-1}$ , respectively. This resulting root square deviation confirms the consistency of the fit. A good agreement is obtained between experimental and calculated energy levels. Table 3 presents a comparison of experimental and calculated  $^2F_{7/2}$  and  $^2F_{5/2}$  Stark energy levels of the  $\text{Yb}^{3+}$  ion occupying a  $D_{2d}$  symmetry site inside  $\text{MWO}_4$  ( $M=\text{Sr, Pb, Ba}$  and  $\text{Ca}$ ) and  $\text{M}'\text{MoO}_4$  ( $M'=\text{Sr, Pb, Cd}$  and  $\text{Ca}$ ) crystals.

**Table 1**  
Fitted crystal-field parameters values ( $\text{cm}^{-1}$ ) for  $\text{Yb}^{3+}$  doped  $\text{CaWO}_4$ ,  $\text{PbWO}_4$ ,  $\text{SrWO}_4$  and  $\text{BaWO}_4$  crystals.

Crystal	$B_{20}$	$B_{40}$	$B_{60}$	$B_{44}$	$B_{64}$	$\zeta$	$\sigma$
$\text{CaWO}_4$ : $\text{Yb}^{3+}$ (this work)	500.7	−483.8	−386	802	366	2904	3.5
$\text{PbWO}_4$ : $\text{Yb}^{3+}$ (this work)	468.8	−327.9	−364.6	718.4	259.6	2903.4	7.9
$\text{SrWO}_4$ : $\text{Yb}^{3+}$ (this work)	487.3	−328.8	−339.5	759.3	202.9	2903.2	9.5
$\text{BaWO}_4$ : $\text{Yb}^{3+}$ (this work)	463.4	−283	−379	653	145	2902.2	17
$\text{CaWO}_4$ : $\text{Yb}^{3+}$ [31]	484	−614	−16	739	509	2906	15.2

**Table 2**

Fitted crystal-field parameters values ( $\text{cm}^{-1}$ ) for  $\text{Yb}^{3+}$  doped  $\text{CaMoO}_4$ ,  $\text{PbMoO}_4$ ,  $\text{SrMoO}_4$  and  $\text{CdMoO}_4$  crystals.

Crystal (this work)	$B_{20}$	$B_{40}$	$B_{60}$	$B_{44}$	$B_{64}$	$\zeta$	$\sigma$
$\text{CdMoO}_4$ : $\text{Yb}^{3+}$	436.3	−495	−388	876	324.6	2901.3	3.7
$\text{CaMoO}_4$ : $\text{Yb}^{3+}$	470	−435.7	−358	809	284	2901.3	7.5
$\text{PbMoO}_4$ : $\text{Yb}^{3+}$	427	−313	−334	731.5	281	2900.3	10.8
$\text{SrMoO}_4$ : $\text{Yb}^{3+}$	444	−297.4	−336	773	203	2902.2	7.3

**Table 3**

Observed and calculated energy levels ( $\text{cm}^{-1}$ ) of  $\text{Yb}^{3+}$  in  $\text{CaWO}_4$ ,  $\text{PbWO}_4$ ,  $\text{SrWO}_4$ ,  $\text{BaWO}_4$ ,  $\text{CaMoO}_4$ ,  $\text{PbMoO}_4$ ,  $\text{SrMoO}_4$  and  $\text{CdMoO}_4$  crystals.

M label <sup>a</sup>	<b><math>\text{CaWO}_4</math>: <math>\text{Yb}^{3+}</math></b>			<b><math>\text{PbWO}_4</math>: <math>\text{Yb}^{3+}</math></b>		
	Cal. val. <sup>b</sup>	Exp. val. <sup>d</sup>	$\Delta$	Cal. val. <sup>b</sup>	Exp. val. <sup>d</sup>	$\Delta$
$^2F_{5/2}$	10663	10665	−2	10599	10604	−5
	10367	10366	1	10316	10314	2
	10279	10278	1	10263	10260	3
$^2F_{7/2}$	490	489	1	427	423	4
	367	366	1	324	323	1
	213	214	−1	160	162	−2
	0	0	0	0	0	0
M label <sup>a</sup>	<b><math>\text{SrWO}_4</math>: <math>\text{Yb}^{3+}</math></b>			<b><math>\text{BaWO}_4</math>: <math>\text{Yb}^{3+}</math></b>		
	Cal. val. <sup>b</sup>	Exp. val. <sup>d</sup>	$\Delta$	Cal. val. <sup>b</sup>	Exp. val. <sup>d</sup>	$\Delta$
$^2F_{5/2}$	10612	10605	7	10560	10549	11
	10315	10318	−3	10286	10291	−5
	10260	10264	−4	10248	10254	−6
$^2F_{7/2}$	442	446	−4	400	408	−8
	331	332	−1	282	291	−3
	152	150	2	131	127	4
	0	0	0	0	0	0
M label <sup>a</sup>	<b><math>\text{CaMoO}_4</math>: <math>\text{Yb}^{3+}</math></b>			<b><math>\text{PbMoO}_4</math>: <math>\text{Yb}^{3+}</math></b>		
	Cal. val. <sup>c</sup>	Exp. val. <sup>d</sup>	$\Delta$	Cal. val. <sup>c</sup>	Exp. val. <sup>d</sup>	$\Delta$
$^2F_{5/2}$	10633	10628	5	10579	10586	−7
	10347	10349	−2	10312	10309	3
	10260	10263	−3	10250	10246	4
$^2F_{7/2}$	469	473	−4	411	406	5
	356	357	−1	329	327	2
	194	193	−1	163	166	−3
	0	0	0	0	0	0
M label <sup>a</sup>	<b><math>\text{SrMoO}_4</math>: <math>\text{Yb}^{3+}</math></b>			<b><math>\text{CdMoO}_4</math>: <math>\text{Yb}^{3+}</math></b>		
	Cal. val. <sup>c</sup>	Exp. val. <sup>d</sup>	$\Delta$	Cal. val. <sup>b</sup>	Exp. val. <sup>d</sup>	$\Delta$
$^2F_{5/2}$	10596	10601	5	10655	10653	2
	10315	10313	2	10380	10381	−1
	10254	10251	3	10265	10267	−2
$^2F_{7/2}$	424	420	4	487	489	−2
	334	333	1	385	385	0
	152	154	−2	226	226	0
	0	0	0	0	0	0

<sup>a</sup> Multiplet label reflects principal SLJ parentage of the calculated energy level.

<sup>b</sup> Calculated levels based on the Hamiltonian parameters listed in Table 1.

<sup>c</sup> Calculated levels based on the Hamiltonian parameters listed in Table 2.

<sup>d</sup> Experimental energy levels taken directly from Ref. [29].

The confidence in these phenomenological parameters and the physical meaning of the fits are supported not only by the low  $\sigma$  values obtained, but also by their smooth variation with ionic radius of the host cation. Figs. 1 and 2 show the evolution of calculated CFPs with ionic radius of substitutional ions  $\text{M}^{2+}$  in  $\text{MWO}_4$  and  $\text{M}'\text{MoO}_4$ , respectively. We noted a smooth

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